


# The biosynthesis of nitrous oxide in the green alga *Chlamydomonas reinhardtii*

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Received 20 January 2017; revised 27 February 2017; accepted 17 March 2017; published online 23 March 2017.

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## SUMMARY

Over the last decades, several studies have reported emissions of nitrous oxide (N<sub>2</sub>O) from microalgal cultures and aquatic ecosystems characterized by a high level of algal activity (e.g. eutrophic lakes). As N<sub>2</sub>O is a potent greenhouse gas and an ozone-depleting pollutant, these findings suggest that large-scale cultivation of microalgae (and possibly, natural eutrophic ecosystems) could have a significant environmental impact. Using the model unicellular microalga *Chlamydomonas reinhardtii*, this study was conducted to investigate the molecular basis of microalgal N<sub>2</sub>O synthesis. We report that *C. reinhardtii* supplied with nitrite (NO<sub>2</sub><sup>-</sup>) under aerobic conditions can reduce NO<sub>2</sub><sup>-</sup> into nitric oxide (NO) using either a mitochondrial cytochrome *c* oxidase (COX) or a dual enzymatic system of nitrate reductase (NR) and amidoxime-reducing component, and that NO is subsequently reduced into N<sub>2</sub>O by the enzyme NO reductase (NOR). Based on experimental evidence and published literature, we hypothesize that when nitrate (NO<sub>3</sub><sup>-</sup>) is the main Nitrogen source and the intracellular concentration of NO<sub>2</sub><sup>-</sup> is low (i.e. under physiological conditions), microalgal N<sub>2</sub>O synthesis involves the reduction of NO<sub>3</sub><sup>-</sup> to NO<sub>2</sub><sup>-</sup> by NR followed by the reduction of NO<sub>2</sub><sup>-</sup> to NO by the dual system involving NR. This microalgal N<sub>2</sub>O pathway has broad implications for environmental science and algal biology because the pathway of NO<sub>3</sub><sup>-</sup> assimilation is conserved among microalgae, and because its regulation may involve NO.

**Keywords:** climate change, greenhouse gases, microalgae, nitric oxide, nitrite response, nitrous oxide.

## INTRODUCTION

Nitrous oxide (N<sub>2</sub>O) is a potent greenhouse gas and a major ozone-depleting pollutant (Ravishankara *et al.*, 2009) that can be emitted from algal cultures (Weathers, 1984; Weathers and Niedzielski, 1986), including verified axenic cultures (Guieysse *et al.*, 2013) and eutrophic ecosystems characterized by a high level of primary activity (Mengis *et al.*, 1997; Wang *et al.*, 2006). Although little is known about N<sub>2</sub>O synthesis in microalgae, nitrite (NO<sub>2</sub><sup>-</sup>) has long been suspected to be a substrate (Weathers, 1984; Weathers and Niedzielski, 1986). Guieysse *et al.* (2013) later proposed that *Chlorella vulgaris* synthesizes N<sub>2</sub>O via the reduction of NO<sub>2</sub><sup>-</sup> to nitric oxide (NO) or nitroxyl (HNO) by nitrate reductase (NR), followed by the reduction of NO to N<sub>2</sub>O by a NO reductase (NOR) or the spontaneous

dimerization of HNO to N<sub>2</sub>O. However, if NO can indeed be reduced to N<sub>2</sub>O under oxic conditions, several alternative scenarios for NO generation could also lead to N<sub>2</sub>O synthesis. Firstly, a dual system of NR and the amidoxime-reducing component NOFNiR (NO-forming nitrite reductase) was recently shown to mediate NO production (both *in vitro* and *in vivo*) in *Chlamydomonas* when the intracellular concentration of NO<sub>2</sub><sup>-</sup> increases in the presence of NO<sub>3</sub><sup>-</sup> (Chamizo-Ampudia *et al.*, 2016). Secondly, the reduction of NO<sub>2</sub><sup>-</sup> to NO could be catalyzed *in vitro* and under anoxia by molybdoenzymes similar to the xanthine oxidase/dehydrogenase found in animals (Maia and Moura, 2014) and vascular plants (Gupta *et al.*, 2011). Thirdly, NO<sub>2</sub><sup>-</sup> can be reduced to NO by mitochondrial cytochrome *c* oxidase

(COX) and alternative oxidase (AOX), as previously proposed for plants (Planchet *et al.*, 2005; Gupta *et al.*, 2016) and the microalga *Chlorella sorokiniana* (Tischner *et al.*, 2004). Fourthly, L-arginine can be oxidized to NO by nitric oxide synthase (NOS) activity, described in plants, although the protein involved has not yet been identified (an animal-type NOS exist in several algae; Gupta *et al.*, 2011; Jeandroz *et al.*, 2016). Finally,  $\text{NO}_2^-$  can be reduced to NO by hemoglobin under hypoxic/anoxic conditions in vascular plants, algae and many cyanobacteria (Sturms *et al.*, 2011; Tiso *et al.*, 2012; Ciaccio *et al.*, 2015).

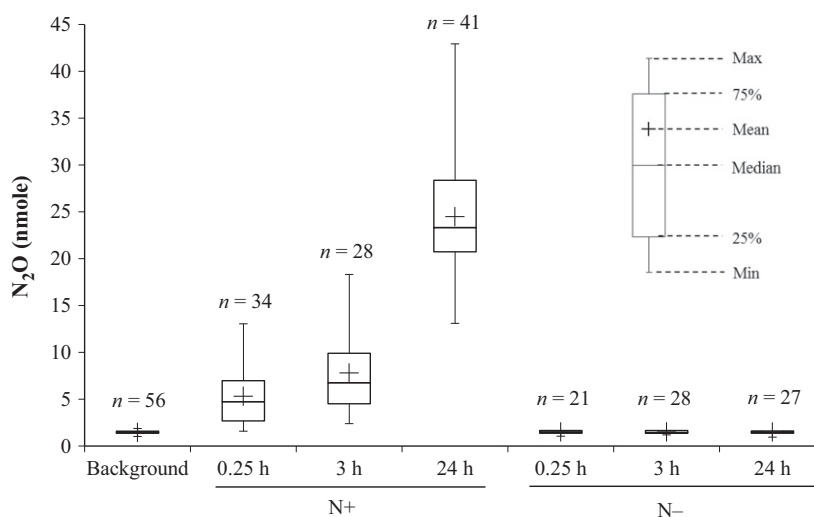
This study was conducted to determine the pathway of  $\text{N}_2\text{O}$  synthesis in microalgae, with the view that this knowledge is critical for the understanding of the nitrogen cycle in aquatic ecosystems (Weathers, 1984; Hayatsu *et al.*, 2010; Kamp *et al.*, 2013). This information will also allow better assessments of the environmental impacts of algal biotechnologies and anthropogenic eutrophication (Guieysse *et al.*, 2013). The model unicellular green microalga *Chlamydomonas reinhardtii* is especially suitable for investigating  $\text{N}_2\text{O}$  synthesis because: (i) nitrogen assimilation and regulation have been extensively studied in this species (Navarro *et al.*, 2000; Fernández and Galván, 2008; Schmollinger *et al.*, 2014; Park *et al.*, 2015; Sanz-Luque *et al.*, 2015a); (ii) the cellular reduction of  $\text{NO}_2^-$  to NO has already been shown (Sakihama *et al.*, 2002; Wei *et al.*, 2014; Chamizo-Ampudia *et al.*, 2016); and (iii) biological and genomic resources are available for this organism (Navarro *et al.*, 2000; Harris, 2001; Gonzalez-Ballester *et al.*, 2005; Pröschold *et al.*, 2005; Merchant *et al.*, 2007; *Chlamydomonas* Resource Centre, <http://www.chlamycollection.org/>). This alga was not known to synthesize  $\text{N}_2\text{O}$  prior to

the current study, so this ability was first demonstrated in axenic cultures. The pathway(s) involved in  $\text{N}_2\text{O}$  synthesis was then determined using mutants and/or specific inhibitors, and these findings were elaborated using RNA sequencing analysis.

## RESULTS

### Nitrous oxide emissions from axenic *C. reinhardtii* cultures

While significant  $\text{N}_2\text{O}$  emissions (average  $1770 \pm 500$  nmol g DW<sup>-1</sup> after 24 h,  $n = 41$ ; DW, dry weight) were recorded from *C. reinhardtii* 6145c cultures supplied with  $\text{NO}_2^-$  in the dark (Figure 1),  $\text{N}_2\text{O}$  production was negligible in autoclaved cultures or sterile medium (Table S1). The  $\text{N}_2\text{O}$  production recorded from *C. reinhardtii* cultures was unlikely to be a result of bacterial or archaeal contamination because sensitive PCR-based detection assays provided no evidence for the presence of these microorganisms (see Appendix S1 in the Supporting Information). These results conclusively demonstrate that *C. reinhardtii* can synthesize  $\text{N}_2\text{O}$  under oxa, which was unknown until now. The production of  $\text{N}_2\text{O}$  was confirmed for all *C. reinhardtii* strains tested, although the rates of emission were strain-dependent (Table S2).  $\text{N}_2\text{O}$  production was linearly correlated with microalgal biomass concentration (Figure S1), providing further evidence that biological processes in the alga were the source of  $\text{N}_2\text{O}$ . In the dark the kinetics of  $\text{N}_2\text{O}$  biosynthesis was characterized by an immediate and short period of  $\text{N}_2\text{O}$  production lasting for 1–4 h, followed by a phase of slow production lasting up to approximately 20 h of incubation, and a final phase of



**Figure 1.** Box plot of  $\text{N}_2\text{O}$  production in cultures of *Chlamydomonas reinhardtii*.

Box plot of  $\text{N}_2\text{O}$  production (nmole) recorded in cultures of *C. reinhardtii* wild type 6145c supplied with 10 mM  $\text{NO}_2^-$  (shown as 'N+') and N-free cultures of 6145c (shown as 'N-') after 0.25, 3 and 24 h of incubation in the dark. The amount of  $\text{N}_2\text{O}$  initially present in flasks supplied with medium (no inoculum) is shown as 'background' and  $n$  represents the number of independent replicates.

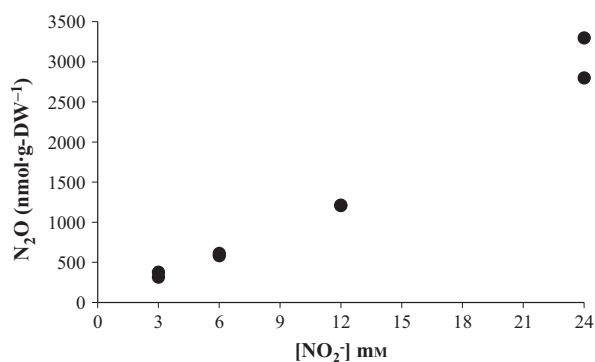
vigorous production over the remaining duration of the experiment (Figure S2a). The initial short phase of N<sub>2</sub>O production was not observed in culture incubated under illumination; here N<sub>2</sub>O biosynthesis started approximately 20 h after the addition of NO<sub>2</sub><sup>-</sup> (Figure S2b). The implications of the kinetics, and the influence of light on N<sub>2</sub>O production, are discussed below in the context of the potential pathways involved.

### NO<sub>2</sub><sup>-</sup> acts as substrate during microalgal N<sub>2</sub>O synthesis

Supplying NO<sub>3</sub><sup>-</sup> or NH<sub>4</sub><sup>+</sup> to the axenic *C. reinhardtii* 6145c cultures did not trigger significant synthesis of N<sub>2</sub>O (Table S1), suggesting that this process depends on the availability of NO<sub>2</sub><sup>-</sup> as a substrate. Supporting this idea, N<sub>2</sub>O production in the dark was found to be linearly correlated to the extracellular concentration of NO<sub>2</sub><sup>-</sup> (Figure 2). This strong association between the reduction of NO<sub>2</sub><sup>-</sup> and the synthesis of N<sub>2</sub>O is in agreement with previous studies (Weathers, 1984; Weathers and Niedzielski, 1986; Guieysse *et al.*, 2013; Kamp *et al.*, 2013; Alcántara *et al.*, 2015). The NO<sub>2</sub><sup>-</sup>-independent synthesis of N<sub>2</sub>O via conversion of L-arginine by NOS was unlikely because N<sub>2</sub>O emissions were not observed in NO<sub>2</sub><sup>-</sup>-free cultures supplied with L-arginine and because N<sub>2</sub>O production in NO<sub>2</sub><sup>-</sup>-laden cultures increased slightly in the presence of the NOS inhibitor N<sup>o</sup>-nitro-L-arginine (L-NNA) (Table S1). These results are consistent with previous findings achieved over shorter durations (4 h) using *C. reinhardtii* (Sakihama *et al.*, 2002) and *C. vulgaris* (Guieysse *et al.*, 2013), and the fact that NOS appears to be absent in *C. reinhardtii* (Jeandroz *et al.*, 2016).

### NO is a key intermediate during N<sub>2</sub>O synthesis

Guieysse *et al.* (2013) suggested that NO is an intermediate during N<sub>2</sub>O synthesis in *C. vulgaris*, and this compound is indeed known to be synthesized by *C. reinhardtii* and other microalgae (Mallick *et al.*, 1999, 2000; Sakihama *et al.*, 2002; Tischner *et al.*, 2004; Estevez and Puntarulo, 2005; Kim *et al.*, 2008; Chang *et al.*, 2013; Wei *et al.*, 2014;



**Figure 2.** Impact of NO<sub>2</sub><sup>-</sup> concentration on N<sub>2</sub>O specific production (nmol g DW<sup>-1</sup>) in duplicate cultures of *C. reinhardtii* 6145c (0.25 g L<sup>-1</sup>) supplied with NO<sub>2</sub><sup>-</sup> at different concentrations and incubated for 24 h in the dark. DW, dry weight.

Sanz-Luque *et al.*, 2015b; Chamizo-Ampudia *et al.*, 2016). In *C. reinhardtii*, NO synthesis has been associated with nitrogen assimilation (Sanz-Luque *et al.*, 2013, 2015a) and both the hypoxic and the nitrogen stress response (Zhang and Mehta, 2008; Hemschemeier *et al.*, 2013; Wei *et al.*, 2014). In our study, the generation of green fluorescence in *C. reinhardtii* 6145c cells pre-incubated with 4-amino-5-methylamino-2',7'-difluorofluorescein diacetate (DAF-FM diacetate) (inside cells, this permeable non-fluorescent chemical is converted into the non-permeable DAF-FM, and the latter reacts with NO to form a fluorescent derivative; see Appendix S2 for details) and subsequently exposed to NO<sub>2</sub><sup>-</sup> suggests the reduction of NO<sub>2</sub><sup>-</sup> caused synthesis of NO, in agreement with the results of Chamizo-Ampudia *et al.* (2016). This fluorescence was not observed when the NO scavenger bovine hemoglobin was added to the microalgal suspension before incubation with DAF-FM diacetate and supply of NO<sub>2</sub><sup>-</sup> (see the full set of results from positive and negative controls in Appendix S2). Taken together these results demonstrate that the conditions triggering N<sub>2</sub>O synthesis (Table 1) also trigger NO production in *C. reinhardtii*.

As HNO has also been proposed as a possible intermediate in algal N<sub>2</sub>O synthesis via the reduction of NO<sub>2</sub><sup>-</sup> (Guieysse *et al.*, 2013), an additional control was performed where *C. reinhardtii* 6145c cells pre-incubated with DAF-FM diacetate were supplied with Angeli's salt as a HNO donor (Appendix S2). The fluorophore DAF-FM is known to react with the NO oxidation products N<sub>2</sub>O<sub>3</sub> and NO<sup>+</sup>; however, the aqueous degradation of Angeli's salt releases NO<sub>2</sub><sup>-</sup> and HNO under physiological conditions (Dutton *et al.*, 2004; Miranda *et al.*, 2005), but can also generate NO at low pH (Miranda *et al.*, 2005). The weak fluorescence reported in these experiments (Appendix S2) may therefore have been caused by 'indirect' NO<sub>2</sub><sup>-</sup> supply causing NO synthesis, or direct NO synthesis in low-pH cellular compartments (Kurkdjian and Guern, 1989). While our results strongly suggest that NO was generated during N<sub>2</sub>O synthesis, further work is needed to fully verify this hypothesis given that knowledge about the biochemistry of HNO production is still very limited (Fukuto *et al.*, 2005).

### Immediate N<sub>2</sub>O synthesis involves NR, but late synthesis involves other enzymes

The catalysis of the reduction of NO<sub>2</sub><sup>-</sup> to NO by NR has long been suspected in plants and *C. reinhardtii* (Yamasaki, 2000; Rockel *et al.*, 2002; Sakihama *et al.*, 2002; Meyer *et al.*, 2005). More recently, this enzyme was shown to provide electrons to a NO-forming NiR, thereby enabling reduction of NO<sub>2</sub><sup>-</sup> to NO by a dual NR-NOFNiR system of two molybdoenzymes (Chamizo-Ampudia *et al.*, 2016). In *C. vulgaris*, the involvement of NR during N<sub>2</sub>O synthesis was evidenced by the repression of N<sub>2</sub>O production in cells pre-cultivated with tungstate, an inhibitor of

**Table 1** Total amount of N<sub>2</sub>O (nmol ± standard error) and specific N<sub>2</sub>O production (nmol g DW<sup>-1</sup> ± standard error) in *Chlamydomonas reinhardtii* 6145c, 2929, M3, M4, 704 and amiCYP55 cultures supplied with NO<sub>2</sub><sup>-</sup> (or NO<sub>3</sub><sup>-</sup> for strains M3 and M4) and incubated in the dark (or illumination for strains M3 and M4)

Strain	Conditions	N source	Effector	N <sub>2</sub> O after 0.25 h			N <sub>2</sub> O after 3 h			N <sub>2</sub> O after 24 h		
				Total (n)	Production (n)	Total (n)	Production (n)	Total (n)	Production (n)			
6145c	Darkness	NO <sub>2</sub> <sup>-</sup>		5.30 ± 3.01 (34) <sup>a</sup>	300 ± 230 (34)	7.80 ± 4.43 (28) <sup>a</sup>	490 ± 345 (28)	24.5 ± 6.59 (41) <sup>a</sup>	1770 ± 500 (41)			
	Darkness	NO <sub>2</sub> <sup>-</sup>	CN <sup>-</sup>	1.59 ± 0.04 (3)	<LOQ	1.89 ± 0.31 (3)	<LOQ	3.78 ± 0.07 (3)	180 ± 20.0 (3)			
	Darkness	NO <sub>2</sub> <sup>-</sup>		1.16 ± 0.25 (8)	<LOQ	2.30 ± 0.76 (8)	<LOQ	29.0 ± 9.20 (8)	2100 ± 700 (8)			
	Darkness	NO <sub>2</sub> <sup>-</sup>	CN <sup>-</sup>	–	–	2.71 ± 0.05 (2) <sup>b</sup>	<LOQ	3.16 ± 0.10 (2)	130 ± 20 (2)			
M3	Darkness	NO <sub>2</sub> <sup>-</sup>		11.4 ± 0.26 (2)	800 ± 50 (2)	55.8 ± 24.1 (2) <sup>b</sup>	4200 ± 2000 (2)	75.6 ± 11.1 (2)	6000 ± 900 (2)			
	Darkness	NO <sub>3</sub> <sup>-</sup>		2.03 ± 0.35 (2)	<LOQ	1.97 ± 0.08 (2) <sup>b</sup>	<LOQ	4.06 ± 1.45 (2)	200 ± 110 (2)			
M4	Illumination	NO <sub>2</sub> <sup>-</sup>		4.64 ± 0.78 (2)	250 ± 60.0 (2)	370 ± 8.20 (2) <sup>b</sup>	(2.8 ± 0.16) × 10 <sup>4</sup> (2)	3480 ± 60.0 (2)	[27.0 ± 1.40] × 10 <sup>4</sup> (2)			
		NO <sub>3</sub> <sup>-</sup>		4.37 ± 0.00 (2)	230 ± 20.0 (2)	29.6 ± 3.42 (2) <sup>b</sup>	2180 ± 290 (2)	720 ± 71.0 (2)	(5.60 ± 0.61) × 10 <sup>4</sup> (2)			
	Darkness	NO <sub>2</sub> <sup>-</sup>		–	–	58.5 ± 0.16 (2) <sup>b</sup>	4400 ± 220 (2)	95.6 ± 6.56 (2)	7300 ± 600 (2)			
	Darkness	NO <sub>3</sub> <sup>-</sup>		2.04 ± 0.01 (2)	<LOQ	2.04 ± 0.01 (2) <sup>b</sup>	<LOQ	3.15 ± 0.70 (2)	130 ± 60 (2)			
704	Illumination	NO <sub>2</sub> <sup>-</sup>		81.9 ± 0.00 (2)	6240 ± 0.00 (2)	177 ± 11.5 (2) <sup>b</sup>	(1.36 ± 0.11) × 10 <sup>4</sup> (2)	3700 ± 300 (2)	(28.6 ± 2.70) × 10 <sup>4</sup> (2)			
		NO <sub>3</sub> <sup>-</sup>		3.21 ± 0.09 (2)	137 ± 20.0 (2)	33.6 ± 0.92 (2) <sup>b</sup>	2500 ± 150 (2)	1830 ± 34.5 (2)	14.2 ± 0.76 × 10 <sup>4</sup> (2)			
amiCYP55	Darkness	NO <sub>2</sub> <sup>-</sup>		2.03 ± 0.06 (3)	<LOQ	6.72 ± 0.99 (3)	410 ± 80.0 (3)	138 ± 9.42 (3)	[1.06 ± 0.9] × 10 <sup>4</sup> (3)			
	Darkness	NO <sub>2</sub> <sup>-</sup>		1.78 ± 0.02 (3)	<LOQ	2.17 ± 0.05 (3)	<LOQ	8.41 ± 2.40 (3)	540 ± 200 (3)			

LOQ, limit of quantification.

<sup>a</sup>Values given in Figure 1 used to calculate N<sub>2</sub>O production.<sup>b</sup>After 6 h.

molybdoenzymes such as NR (Deng *et al.*, 1989). The hypothesis of NR-mediated N<sub>2</sub>O synthesis was therefore tested in *C. reinhardtii*.

Tungstate is known to partially inhibit NR activity in *C. reinhardtii* (Llamas *et al.*, 2000), and pre-cultivation of the 6145c strain with this compound indeed caused partial repression of immediate N<sub>2</sub>O production (Table S3). Interestingly, tungstate inhibits the terminal activity of NR without affecting the diaphorase-associated activity of this enzyme (Barea *et al.*, 1976). Tungstate is therefore likely to have inhibited immediate N<sub>2</sub>O synthesis by inhibiting the NOFNiR activity of the dual molybdoenzyme system catalyzing NO synthesis (Chamizo-Ampudia *et al.*, 2011, 2016). To confirm this, N<sub>2</sub>O emission was recorded during the cultivation of NR-deficient *C. reinhardtii* 2929 (Table 1). Immediate N<sub>2</sub>O production was not recorded in NO<sub>2</sub><sup>-</sup>-replete cultures of the NR-deficient strain, although similar 0–24 h N<sub>2</sub>O emissions were recorded in the 2929 mutant (2100 ± 700 nmol g DW<sup>-1</sup>) and the 6145c control strain (1770 ± 500 nmol g DW<sup>-1</sup>) after 24 h of incubation. Furthermore, fluorescence was not observed in NO<sub>2</sub><sup>-</sup>-replete cultures of NR mutant 2929 pre-incubated with DAF-FM diacetate (this protocol could only be used following short-term exposure to NO<sub>2</sub><sup>-</sup> due to the reactivity of DAF-FM diacetate; Appendix S2). These results indicate that N<sub>2</sub>O synthesis involved the reduction of NO<sub>2</sub><sup>-</sup> into NO by the dual system NR-NOFNiR immediately following the addition of NO<sub>2</sub><sup>-</sup>, and is consistent with a different enzyme system being involved in the late N<sub>2</sub>O response.

#### NR and NiR activities impact N<sub>2</sub>O synthesis under supply of NO<sub>3</sub><sup>-</sup>

NO<sub>3</sub><sup>-</sup> is the main nitrogen source available to microalgae in many 'natural' environments (Raven and Giordano, 2013) and it is commonly used during commercial cultivation of microalgae (Borowitzka, 2005). NR and NiR sequentially catalyze the assimilatory reduction of NO<sub>3</sub><sup>-</sup> to NO<sub>2</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>, thereby contributing to regulation of the intracellular concentration of NO<sub>2</sub><sup>-</sup>. As NO<sub>2</sub><sup>-</sup> 'fuels' N<sub>2</sub>O synthesis via NO, the activities of NR and NiR are likely to affect N<sub>2</sub>O emissions under physiological conditions (when NO<sub>3</sub><sup>-</sup> is the main N source and the intracellular concentration of NO<sub>2</sub><sup>-</sup> is relatively low): this was demonstrated when the NiR-deficient mutants supplied with NO<sub>3</sub><sup>-</sup> produced large amounts of N<sub>2</sub>O under illumination (Table 1). These mutants could carry out the stoichiometric reduction of NO<sub>3</sub><sup>-</sup> to NO<sub>2</sub><sup>-</sup> because they possess a NR activated by light and NO<sub>3</sub><sup>-</sup> (Navarro *et al.*, 2000; Kaiser and Huber, 2001), but could not reduce NO<sub>2</sub><sup>-</sup> to NH<sub>4</sub><sup>+</sup>. Nitrite is therefore likely to have over-accumulated in illuminated NiR-deficient cells (Navarro *et al.*, 2000), which resulted in increased turnover of NO boosting N<sub>2</sub>O emissions. Unlike the case of *C. reinhardtii* 6145c NO<sub>3</sub><sup>-</sup>-replete cultures, where N<sub>2</sub>O production was repressed in the dark

(Table S1) as cells could not sustain reduction of NO<sub>3</sub><sup>-</sup> to NO<sub>2</sub><sup>-</sup>, NiR-deficient mutants supplied with NO<sub>2</sub><sup>-</sup> could release N<sub>2</sub>O in light and dark conditions (Table 1), further highlighting the role of NO<sub>2</sub><sup>-</sup> as a substrate for N<sub>2</sub>O synthesis. In agreement with past studies (Weathers, 1984; Weathers and Niedzielski, 1986; Guieysse *et al.*, 2013; Kamp *et al.*, 2013), illumination repressed the immediate production of N<sub>2</sub>O by *C. reinhardtii* supplied with NO<sub>2</sub><sup>-</sup> (Table 1), probably due to the reduction of NO<sub>2</sub><sup>-</sup> by NiR during photosynthesis (Guieysse *et al.*, 2013).

#### Late N<sub>2</sub>O synthesis involves the reduction of NO<sub>2</sub><sup>-</sup> to NO by mitochondrial COX

The reduction of NO<sub>2</sub><sup>-</sup> via the mitochondrial electron transport chain was first demonstrated in the green microalga *C. sorokiniana* (Tischner *et al.*, 2004), and mitochondria are a known source of NO in higher plants (Planchet *et al.*, 2005; Igamberdiev *et al.*, 2014; Gupta *et al.*, 2016). Mitochondrial COX or AOX can reduce NO<sub>2</sub><sup>-</sup> to NO under hypoxic and anoxic conditions (Tischner *et al.*, 2004; Planchet *et al.*, 2005; Gupta and Igamberdiev, 2011) and, at a low rate, under normoxia (Tischner *et al.*, 2004; Planchet *et al.*, 2005). As experimental evidence linking these mitochondrial enzymes to N<sub>2</sub>O synthesis was previously lacking, the wild-type *C. reinhardtii* 6145c and the NR-deficient mutant *C. reinhardtii* 2929 were supplied with NO<sub>2</sub><sup>-</sup> in the presence of CN<sup>-</sup>, an inhibitor of heme-containing enzymes (Sakihama *et al.*, 2002; Tischner *et al.*, 2004) known to affect mitochondrial COX (Way, 1984) at concentrations lower than 5 mM (Gans and Wollman, 1995). This treatment with 2 mM CN<sup>-</sup> resulted in the immediate inhibition of both N<sub>2</sub>O (Table 1) and NO (Appendix S2), which could be explained by the fact that CN<sup>-</sup> can also inhibit NR (Tischner *et al.*, 2004; Planchet *et al.*, 2005). However, and in contrast to chemical inhibition or repression of NR, N<sub>2</sub>O synthesis was inhibited by 90% and 96% in the wild-type and NR mutant cultures, respectively, after 24 h of exposure to CN<sup>-</sup> and NO<sub>2</sub><sup>-</sup> (Table 1). Together, these results suggest that mitochondrial COX catalyzes the reduction of NO<sub>2</sub><sup>-</sup> to NO during the late period of N<sub>2</sub>O production.

Given that the synthesis of N<sub>2</sub>O by *C. reinhardtii* 6145c was not completely eliminated following the addition of CN<sup>-</sup>, AOX may also be partially responsible for the reduction of NO<sub>2</sub><sup>-</sup> to NO. Tischner *et al.* (2004) indeed demonstrated that the *C. sorokiniana* NiR mutant continuously generated high levels of NO under anoxia and normoxia, and that this production could be catalysed by both COX and AOX. The AOX-mediated reduction of NO<sub>2</sub><sup>-</sup> to NO has also been demonstrated in the model plant *Arabidopsis thaliana* (Gupta *et al.*, 2016). The potential involvement of AOX could not be experimentally dismissed because we found that the chemical reaction of the known AOX inhibitor salicylhydroxamic acid (SHAM) generated N<sub>2</sub>O from NO<sub>2</sub><sup>-</sup> (44 ± 8 nmol N<sub>2</sub>O from abiotic tests incubated for

3 h). It should also be noted that  $\text{NO}_2^-$  is a known antidote to cyanide, and that NO may affect the inhibitory effect of  $\text{CN}^-$  (Leavesley *et al.*, 2008).

### The synthesis of $\text{N}_2\text{O}$ by *C. reinhardtii* involves the reduction of NO to $\text{N}_2\text{O}$ by NOR

Although the concurrent synthesis of NO and  $\text{N}_2\text{O}$  in response to  $\text{NO}_2^-$  supply was repeatedly observed in *C. reinhardtii* cultures, NO is typically oxidized into nitrogen dioxide ( $\text{NO}_2$ ) and other products (e.g.  $\text{ONOO}^-$ ) under oxic conditions (Murphy *et al.*, 1998; Lamattina *et al.*, 2003). *Chlamydomonas reinhardtii*, however, harbors the gene CYP55, which is similar to a fungal gene encoding a NOR (Chang *et al.*, 2011; Guieysse *et al.*, 2013; and further evidence below) that is capable of reducing NO into  $\text{N}_2\text{O}$  under hypoxia (Morozkina and Kurakov, 2007; Shoun *et al.*, 2012). When supplied with  $\text{NO}_2^-$ , a *C. reinhardtii* NOR knockdown mutant (amiCYP55 strain with the gene CYP55 silenced using artificial micro-iRNA; Appendix S3) synthesized 70–90% less  $\text{N}_2\text{O}$  than its respective parent (Table 1), suggesting that wild-type *C. reinhardtii* synthesizes  $\text{N}_2\text{O}$  via NOR-mediated NO reduction under oxic conditions. Interestingly,  $\text{N}_2\text{O}$  synthesis increased when nitrite-laden *C. reinhardtii* 6145c was incubated under anoxic conditions (Table S1). These results may be explained by the absence of competition between  $\text{NO}_2^-$  and  $\text{O}_2$  for the COX reaction sites (Gupta *et al.*, 2005), as well as reduced rates of NO oxidation (Gupta *et al.*, 2016) favoring reduction of NO to  $\text{N}_2\text{O}$  by NOR.  $\text{N}_2\text{O}$  synthesis was not fully inhibited, probably due to the fact NOR expression was not fully repressed in the amiCYP55 strain (e.g. 54% of the wild-type strain 704).

### RNA sequencing

To study the changes in expression of key enzymes potentially involved in  $\text{N}_2\text{O}$  synthesis in *C. reinhardtii* (e.g. NR, NOR, COXs), RNA sequencing (RNA-seq) was performed on cultures exposed to  $\text{NO}_2^-$  in the dark (treatment) and compared with  $\text{NO}_2^-$ -free cultures grown under identical conditions (control). RNA-seq was carried out on triplicate samples collected at 0.25, 3 and 24 h after  $\text{NO}_2^-$  addition, based on the three phases of  $\text{N}_2\text{O}$  kinetics observed in the dark (Figure S2a). As shown in Table 2, genes encoding NR, NiR and NOR were not significantly differentially expressed (the adjusted *P*-value threshold was 0.05) between the treatments and the controls at any of the time points tested. Overall, the transcriptomics data were consistent with the biochemical evidence that reduction of  $\text{NO}_2^-$  to NO leading to 'late'  $\text{N}_2\text{O}$  synthesis is not mediated by NR. NR would only be needed immediately following  $\text{NO}_2^-$  supply. The nitrite response via transcriptional regulation of NOR may not have been triggered/required under the experimental conditions tested, but the data clearly provide evidence for the presence of abundant NOR

transcripts (the average normalized read count across all samples for NOR was 3156,  $n = 18$ ).

In contrast to the genes discussed above, THB1 and THB2 were significantly differentially expressed after 24 h of exposure to  $\text{NO}_2^-$  (Table 2). Plant hemoglobins are known to reduce  $\text{NO}_2^-$  to NO (Hoy and Hargrove, 2008; Gupta *et al.*, 2011) and a microalgal hemoglobin could catalyze this reaction under hypoxic/anoxic conditions (Sturms *et al.*, 2011), a mechanism thought to be related to either  $\text{NO}_2^-$  detoxification (Sturms *et al.*, 2011) or survival under anoxia (Hemschemeier *et al.*, 2013). *Chlamydomonas reinhardtii* possesses 12 truncated hemoglobins (THB1–12), and some of these enzymes can indeed reduce  $\text{NO}_2^-$  (Ciaccio *et al.*, 2015; Huwald *et al.*, 2015). *Chlamydomonas reinhardtii* THB1, which was significantly differentially expressed in the treatment cells after 3 and 24 h of exposure to  $\text{NO}_2^-$  [ $\log_2$  fold change (FC) of 0.7 and 0.5], is also involved in NO signaling,  $\text{NO}_3^-$  assimilation, regulation of NR activity and, of particular relevance, NO scavenging under normoxia via its dioxygenase activity (Sanz-Luque *et al.*, 2015b). Consequently, during  $\text{NO}_3^-$  assimilation, THB1 may help to regulate intracellular accumulation of NO by promoting the transformation of NO into  $\text{NO}_3^-$  in parallel with the reduction of NO to  $\text{N}_2\text{O}$ . The change in transcript abundances for THB1 (positive  $\log_2$  FC) and THB2 (negative  $\log_2$  FC) agrees with the transcriptional NO response described by Sanz-Luque *et al.* (2015b), and therefore provides additional evidence of both the involvement of NO as a  $\text{N}_2\text{O}$  intermediate and the involvement of THBs in the response to NO synthesis.

Transcriptomic data also showed that while numerous transcripts for several COX genes could be identified, only one COX gene was significantly differentially expressed ( $\log_2$  FC of  $-0.5$ ) after 24 h of  $\text{NO}_2^-$  exposure (Table 2). As in the case of NOR, our failure to observe changes in transcript abundance of the COX genes suggests that transcription regulation might not have been required under the experimental conditions tested. In contrast, AOX1 and AOX2 were significantly differentially expressed ( $\log_2$  FC of 0.5 and 1.4, respectively) after 24 h of exposure to  $\text{NO}_2^-$ . AOXs are regulatory enzymes that balance electron transport in mitochondria (Vanlerberghe and McIntosh, 1997). Considering that NO has been shown to induce AOX in *Arabidopsis* (Huang *et al.*, 2002), AOX synthesis could have been activated in response to the generation of NO in *C. reinhardtii*. Further analysis is still needed to evaluate the full metabolic effect of high  $\text{NO}_2^-$  in *C. reinhardtii* as 180, 740 and 3914 genes were significantly differentially expressed between control and treatment groups after 0.25, 3 and 24 h of  $\text{NO}_2^-$  exposure, respectively. The transcriptomic data also showed significantly lower transcript abundances of several eukaryotic initiation factors in  $\text{NO}_2^-$ -replete cultures (Table S4), indicative of cells that have activated stress response pathways (Langland *et al.*,

**Table 2** Log<sub>2</sub> fold expression changes (Log<sub>2</sub>FC) between control and treatment groups of candidate genes potentially involved in microalgal N<sub>2</sub>O synthesis. Numbers in parenthesis represent mean normalized counts

Locus ID	Gene	Annotation	Time (h) after exposure to NO <sub>2</sub> <sup>-</sup>		
			0.25	3	24
Cre09.g410950	<i>NIA1</i>	Nitrate reductase (NR)	0.0 (1)	0.0 (0)	0.0 (0)
Cre09.g410750	<i>NII1</i>	Nitrite reductase (NiR)	-0.2 (2)	-0.4 (4)	-0.7 (1)
Cre01.g007950	<i>CYP55</i>	Nitric oxide reductase (NOR)	0.2 (5150)	0.3 (2974)	-0.1 (1487)
Cre09.g389089	<i>ARC</i>	Amidoxime reducing component	0.0 (10)	-0.2 (7)	-0.8 (15)
Cre14.g615400	<i>THB1</i>	Truncated hemoglobin	<b>0.7 (1142)</b>	0 (1142)	<b>0.5 (1052)</b>
Cre14.g615350	<i>THB2</i>	Truncated hemoglobin	<b>-0.6 (133)</b>	-0.3 (342)	<b>-1.4 (57)</b>
Cre03.g154350	<i>COX2a</i>	Cytochrome <i>c</i> oxidase subunit II	-0.1 (945)	-0.2 (534)	-0.4 (420)
Cre04.g221700	<i>COX3</i>	Cytochrome <i>c</i> oxidase subunit III	-0.1 (1583)	-0.1 (920)	<b>-0.5 (921)</b>
Cre03.g157700	<i>COX5c</i>	Cytochrome <i>c</i> oxidase subunit	-0.1 (1445)	-0.2 (824)	-0.3 (518)
Cre06.g304350	<i>MT-CO1</i>	Mitochondrial cytochrome <i>c</i> oxidase subunit 1	0.0 (2)	-0.3 (2)	0.4 (0)
Cre01.g049500	<i>COX2b</i>	Cytochrome <i>c</i> oxidase subunit	-0.1 (3801)	-0.2 (2409)	0.0 (1482)
Cre01.g055550	<i>COX11</i>	Involved in the insertion of copper into the CuB center of subunit Cox1p	-0.1 (2)	0.0 (1)	0.4 (3)
Cre12.g537450	<i>COX13</i>	Cytochrome <i>c</i> oxidase subunit 10	-0.1 (2219)	-0.1 (1462)	-0.2 (742)
Cre02.g082700	<i>COX15</i>	Cytochrome <i>c</i> oxidase assembly factor	0.2 (2)	0.0 (0)	0.5 (7)
Cre03.g165400	<i>COX16</i>	Cytochrome <i>c</i> oxidase assembly factor	0.0 (2)	0.0 (2)	-0.1 (5)
Cre05.g232850	<i>COX17</i>	Involved in the delivery of copper to mitochondria	0.2 (8)	-0.1 (5)	1.2 (7)
Cre16.g690200	<i>COX18</i>	Cytochrome <i>c</i> oxidase assembly factor	0.2 (4)	0.1 (4)	0.0 (4)
Cre10.g454550	<i>COX19</i>	Cytochrome <i>c</i> oxidase assembly factor	0.2 (21)	0.2 (8)	-0.1 (6)
Cre01.g069107	<i>COX191</i>	Involved in cytochrome <i>c</i> oxidase expression	0.1 (3)	0.0 (9)	0.4 (19)
Cre03.g213425	<i>COX23</i>	Cytochrome <i>c</i> oxidase assembly factor involved in copper homeostasis	-0.2 (49)	-0.4 (37)	0.0 (51)
Cre16.g69185	<i>COX90</i>	Cytochrome <i>c</i> oxidase subunit	-0.1 (910)	-0.2 (465)	0.0 (473)
Cre09.g395950	<i>AOX1</i>	Alternative oxidase	<b>0.5 (1050)</b>	-0.5 (575)	<b>0.5 (736)</b>
Cre03.g169550	<i>AOX2</i>	Alternative oxidase	-0.2 (17)	0.4 (6)	<b>1.4 (14)</b>

Log<sub>2</sub> fold expression change values shown in bold are significant based on a Benjamini–Hochberg adjusted *P*-value < 0.05.

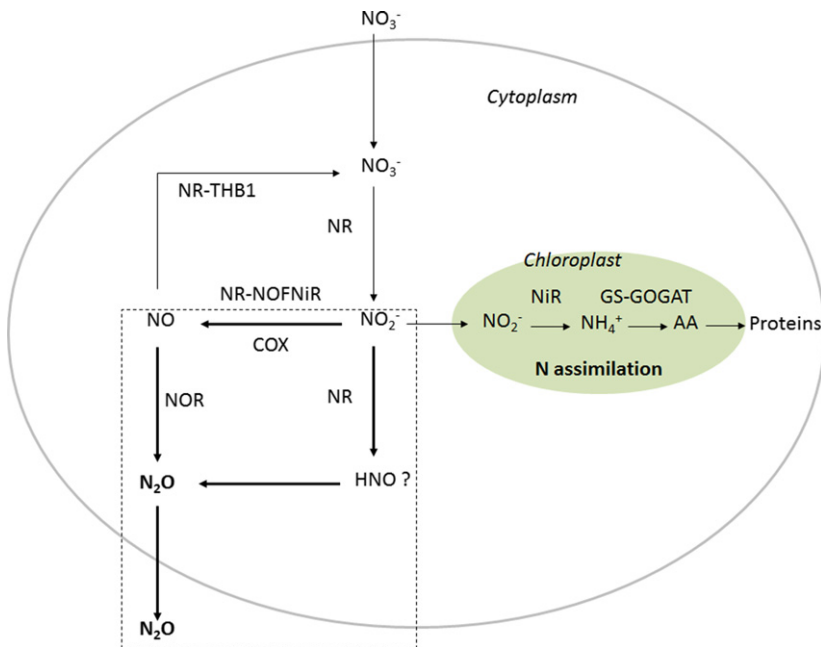
1996; Hinnebusch, 1997; Roy *et al.*, 2010; Pakos-zebrucka *et al.*, 2016). The activation of these pathways possibly explains the large amount of differential expression we observed in our transcriptomic comparisons, and this complicates the identification of other genes potentially involved in this process. A detailed analysis of the RNA-seq study is currently being carried out to fully characterize the molecular responses to NO<sub>2</sub><sup>-</sup> loading.

## DISCUSSION

### Biological implications

This study showed that axenic *C. reinhardtii* supplied with NO<sub>2</sub><sup>-</sup> can synthesize N<sub>2</sub>O under oxia via the reduction of NO<sub>2</sub><sup>-</sup> to NO followed by the reduction of NO to N<sub>2</sub>O. The reduction of NO<sub>2</sub><sup>-</sup> to NO was initially catalyzed by the dual NR:NOFNiR enzyme system, but this activity rapidly ceased and was later substituted by the reduction of NO<sub>2</sub><sup>-</sup> to NO by COX. Under physiological conditions, NO<sub>3</sub><sup>-</sup> is the main source of nitrogen and the intracellular concentration of NO<sub>2</sub><sup>-</sup> is low. The reduction of NO<sub>3</sub><sup>-</sup> to NO<sub>2</sub><sup>-</sup> by NR therefore provides both the nitrogen source for growth and, under specific circumstances, the substrate for NO biosynthesis by the dual NR:NOFNiR system (Chamizo-Ampudia

*et al.*, 2016). Moderate production of NO would be expected to take place at moderate cytosolic NO<sub>2</sub><sup>-</sup> concentrations and, under such conditions, the dioxygenase activity of THB1 could efficiently modulate the intracellular concentrations of NO, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> (Sanz-Luque *et al.*, 2015b). In contrast, significant exposure to high intracellular concentrations of NO<sub>2</sub><sup>-</sup> (as in this work) appears to cause COX to reduce significant amounts of NO<sub>2</sub><sup>-</sup> to NO, as indicated in the model proposed in Figure 3. Both NO<sub>2</sub><sup>-</sup> and NO are known to generate harmful products (Faure *et al.*, 1991; Beckman and Koppenol, 1996) and cells have developed mechanisms to prevent the intracellular accumulation of these molecules, such as the oxidation of NO into NO<sub>3</sub><sup>-</sup> by THBs (Sanz-Luque *et al.*, 2015b) and the excretion of NO<sub>2</sub><sup>-</sup> (Faure *et al.*, 1991; Navarro *et al.*, 2000). The reduction of NO<sub>2</sub><sup>-</sup> to N<sub>2</sub>O via NO could therefore provide algal cells that periodically experience exposure to anoxia or hypoxia (e.g. wastewater stabilization ponds) with a means to detoxify NO<sub>2</sub><sup>-</sup> and NO. This activity maybe the fortuitous consequence of COX activity under oxia, or it may provide additional detoxification capacity in microalgae (i.e. NOR would be acting as a security valve eliminating NO). Interestingly, the yield of algal N<sub>2</sub>O synthesis was lower under oxia (0.001–0.03% g N-N<sub>2</sub>O synthesized per g N-input after 24 h



**Figure 3.** Proposed N<sub>2</sub>O pathway in *Chlamydomonas reinhardtii*.

Abbreviations/symbols: NO<sub>3</sub><sup>-</sup>, nitrate; NO<sub>2</sub><sup>-</sup>, nitrite; NO, nitric oxide; HNO, nitroxyl; N<sub>2</sub>O, nitrous oxide; NR, nitrate reductase; NiR, nitrite reductase; NR-NOFNR, dual enzyme system NO-forming nitrite reductase; COX, cytochrome *c* oxidase; THB1, truncated hemoglobin; NH<sub>4</sub><sup>+</sup>, ammonium; AA, amino acid; GS-GOGAT, glutamine synthetase–glutamine oxoglutarate amino transferase. [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)]

of incubation in the dark for strain 6145c) than under anoxia (0.8% g N-N<sub>2</sub>O synthesized per g N-input after 24 h of incubation in the dark for strain 6145c), suggesting the existence of a competitive NO metabolism under oxia (e.g. NO<sub>x</sub> generation; Lamattina *et al.*, 2003). The low yield of N<sub>2</sub>O synthesis under oxia also probably explains why the reduction of NO to N<sub>2</sub>O has seldom been reported in past studies focusing on nitrogen metabolism in algae. Clearly, our understanding of the function and regulation of the enzymes and substrates involved in N<sub>2</sub>O synthesis in microalgae remains limited. Nevertheless, the evidence presented here (such as the ability of *C. reinhardtii* to reduce NO to N<sub>2</sub>O under oxia) provides new advances in algal biology in light of the importance of nitrate assimilation and NO metabolism in these organisms.

### Environmental implications

As discussed above, the synthesis of N<sub>2</sub>O under physiological conditions is linked to assimilation of NO<sub>3</sub><sup>-</sup> in *C. reinhardtii* in a mechanism similar to the fungal N<sub>2</sub>O denitrification pathway (Shoun *et al.*, 2012), with the significant difference that the microalgae carry out N<sub>2</sub>O synthesis under oxia. This finding has broad implications, because the pathway of NO<sub>3</sub><sup>-</sup> assimilation is conserved among microalgae (Sanz-Luque *et al.*, 2015a) and its regulation involves NO (de Montaigu *et al.*, 2010; Sanz-Luque *et al.*, 2013, 2015a). Moreover, several microalgal species have the ability to synthesize NO (Mallick *et al.*, 1999; Tischner *et al.*, 2004; Kim *et al.*, 2008; Kumar *et al.*, 2015) and/or N<sub>2</sub>O (Weathers, 1984; Weathers and Niedzielski, 1986; Guieysse *et al.*, 2013; Kamp *et al.*, 2013; Alcántara *et al.*, 2015), and *Chlorella variabilis* harbors a close homolog of

the *Chlamydomonas NOR* gene (Guieysse *et al.*, 2013). These observations suggest that numerous species of microalgae could be capable of N<sub>2</sub>O synthesis, which in turn could explain why correlations between primary productivity and N<sub>2</sub>O emission rate have been reported for decades (Pierotti and Rasmussen, 1980; Outdot *et al.*, 1990; Mengis *et al.*, 1997; Wang *et al.*, 2006), even under very low exogenous NO<sub>2</sub><sup>-</sup> concentrations (probably due to the intracellular generation of NO<sub>2</sub><sup>-</sup> during the assimilation of NO<sub>3</sub><sup>-</sup>; Pierotti and Rasmussen, 1980). Because microalgae are ubiquitous in the environment and often associated with anthropogenic pollution (e.g. algal blooms), the potential global significance of biosynthesis of N<sub>2</sub>O by microalgae needs to be assessed. Fortunately, the evidence that mandates a broader recognition of the potential issue also provides guidance for mitigation strategies: distinct differences in species' synthetic abilities (Weathers, 1984; Weathers and Niedzielski, 1986; Guieysse *et al.*, 2013; Kamp *et al.*, 2013) and the kinetics of microalgal N<sub>2</sub>O emissions have been reported as, for example, nitrite-laden *C. vulgaris* was previously shown to constantly produce N<sub>2</sub>O over 48 h of incubation in the dark (Guieysse *et al.*, 2013). These divergences indicate differences in pathways and regulatory mechanisms that, in turn, could be exploited to reduce N<sub>2</sub>O emissions during algal cultivation.

### EXPERIMENTAL PROCEDURES

#### Strains selection and maintenance

The wild type and mutant *C. reinhardtii* strains 6145c, 704, 409, 217, 112, 530, 124– and 125+ (Table S5) were used in this research. The amiCYP55 strain was created from strain 704 via



transformation with the vector pChlamiRNA3-CYP55 by following the protocol described by Molnar *et al.* (2009). Specific primers for artificial miRNA were amiForCYP55 and amiRevCYP55 (Appendix S3).

All strains were maintained on sterilized solid medium containing 15 g L<sup>-1</sup> of agar agar in buffered TRIS-acetate-phosphate (TAP) medium (Harris, 2009), i.e. using NO<sub>3</sub><sup>-</sup> as the N source (0.75 M), and cultivated in TAP medium. The NR- and NiR-deficient mutants were cultivated in similar media where NH<sub>4</sub><sup>+</sup> was substituted for NO<sub>3</sub><sup>-</sup> as the N source because these strains were unable to grow on NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup>. Each week, 250-ml Erlenmeyer-flasks (E-flasks) were filled with 125 ml of autoclaved medium and inoculated with a single colony of *C. reinhardtii*. The microalgae were cultivated in a Minitron incubator (Infors HT, <http://www.infors-ht.com/>) at 25 ± 1°C under continuous agitation (180 rpm) and illumination [21 W of photosynthetically active radiation (PAR) m<sup>-2</sup> at the culture surface, using five 15-W Sylvania GRO-LUX coolwhite tubes; <https://www.sylvania.com/>], and under an atmosphere of 2% (vol.) CO<sub>2</sub> in air to prevent CO<sub>2</sub> limitation. The E-flasks were closed with cotton tops and autoclaved with the medium prior to inoculation.

### Nitrous oxide synthesis bioassays

N<sub>2</sub>O production was quantified using gas chromatography as described in Guieysse *et al.* (2013) and the results presented herein show the total amount of N<sub>2</sub>O produced in the flasks assuming equilibrium between aqueous and gaseous N<sub>2</sub>O concentrations (considering N<sub>2</sub>O losses and pressure changes; Appendix S4). Biomass concentration was quantified as cell dry weight (DW) using the protocol described by Béchet *et al.* (2015), while aqueous NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> concentrations were quantified using ionic chromatography, as described by Alcántara *et al.* (2015). Purity of the cultures was assessed by extracting and amplifying DNA using the primers and methods described by Guieysse *et al.* (2013). To provide algae cultures free of extracellular N, 25–50 ml of 5.5–10-day-old axenic *C. reinhardtii* cultures (described above) were centrifuged at 2900 g for 3.5 min, and the cell pellets were re-suspended in N-free and acetate-free TAP medium to reach a concentration of 0.25 g algae L<sup>-1</sup> to serve as inocula in subsequent assays. Glass flasks were filled with N-free algal suspension, sealed with rubber septa and incubated for 24–48 h at 25 ± 1°C under continuous agitation (180 rpm) and either under continuous illumination (21 W PAR m<sup>-2</sup>) or in the dark. The following experiments were thus conducted (see Tables S1 and S2 and corresponding results for further details):

- 1 *Chlamydomonas reinhardtii* 6145c cultures were supplied with different N sources (10 mM NO<sub>2</sub><sup>-</sup>, NO<sub>3</sub><sup>-</sup> or NH<sub>4</sub><sup>+</sup>) or NO<sub>2</sub><sup>-</sup> at various concentrations (2–24 mM).
- 2 Cultures of various *C. reinhardtii* strains were supplied with 10 mM NO<sub>2</sub><sup>-</sup> in the dark.
- 3 Cultures of *C. reinhardtii* 6145c were pre-cultivated in modified TAP where molybdate (MoO<sub>4</sub><sup>4-</sup>) was replaced by orthotungstate (WO<sub>4</sub><sup>4-</sup>) to inhibit NR.
- 4 *Chlamydomonas reinhardtii* 6145c cultures were supplied with 10 mM of the NOS substrate L-arginine (≥ 98%; Sigma-Aldrich, <http://www.sigmaaldrich.com/>) and/or the NOS-inhibitor L-NNA (≥ 98%; Sigma-Aldrich).
- 5 Cultures of *C. reinhardtii* 6145c and *C. reinhardtii* NR mutant 2929 were supplied with cyanide (KCN ≥ 97%; Sigma-Aldrich) at 2 mM as a COX inhibitor.
- 6 Cultures of *C. reinhardtii* NiR mutant M3 and M4 were supplied with 10 mM NO<sub>2</sub><sup>-</sup> or NO<sub>3</sub><sup>-</sup> in the dark or under illumination.

7 Cultures of *C. reinhardtii* 704 and the NOR knockdown mutant amiCYP55 were supplied with 10 mM NO<sub>2</sub><sup>-</sup> in the dark.

Over the duration of the study, a positive control (cultures of wild type 6145c supplied with 10 mM nitrite in the dark) was repeated 40 times (Figure 1). The 'background' N<sub>2</sub>O level initially found in the flask was experimentally estimated to 1.45 ± 0.18 nmol (average ± SE; *n* = 56). Based on this value and the standard error of 57% recorded at *t* = 0.25 h (Figure 1), the limit of quantification was estimated to 2.9 nmol in the flasks (i.e. 200% of the background value). When N<sub>2</sub>O production was higher than this threshold, specific production values (nmol g DW<sup>-1</sup>) were calculated as the amount of N<sub>2</sub>O produced a time *t* (nmole) minus the background level (1.45 nmole), divided by the initial cell concentration (g DW L<sup>-1</sup>).

### RNA sequencing assays

Aliquots of *C. reinhardtii* 6145c microalgal suspensions were withdrawn and poured into serum flasks incubated in the dark (centrifugation was avoided to prevent mechanical stress). Six flasks were thus incubated at constant temperature (25 ± 1°C) and agitation (180 rpm) for 1 h before NO<sub>2</sub><sup>-</sup> (10 mM) was injected into three of the flasks, henceforth referred to as the 'treatments', and the remaining flasks were used as controls. Based on the N<sub>2</sub>O kinetics recorded in the dark (see Results), gas and liquid samples were withdrawn from each flask after 1.25 (t1, 15 min after NO<sub>2</sub><sup>-</sup> supply), 4 (t2, 3 h after NO<sub>2</sub><sup>-</sup> supply) and 25 h (t3, 24 h after NO<sub>2</sub><sup>-</sup> supply) of incubation to quantify N<sub>2</sub>O and extract RNA (Appendix S5). This protocol was repeated three times to generate biological replicates for the RNA-seq experiment. The amount of N<sub>2</sub>O recorded for each time point between triplicates over the three repeated experiments never exceeded 14% variability (for each time point *n* = 9, *P* = 0.05).

RNA was extracted using a plant NucleoSpin RNA kit (Macherey-Nagel, <http://www.mn-net.com/>). The presence of high-quality RNA was confirmed in each sample using an Agilent 2100 Bioanalyzer System (Agilent Technologies, <http://www.agilent.com/>). Illumina TruSeq stranded mRNA libraries were synthesized from poly-A purified RNA and the resulting libraries were sequenced using an Illumina HiSeq2500 machine (Otago Genome and Bioinformatics Facility, <http://www.otago.ac.nz/genomics/>). Adaptor sequences were removed from the 125-bp paired end of resulting reads using the software FASTQC (Aronesty, 2013) before quality trimming (with *P* = 0.01) using SolexaQA++ v.3.3 (Cox *et al.*, 2010). The reads were aligned to the *C. reinhardtii* v.5.5 genome model (Merchant *et al.*, 2007) using TopHat v.2.1.0 (Kim *et al.*, 2013). Read counts were generated from the alignments using htseq-count (Anders *et al.*, 2015) based on the *C. reinhardtii* v.5.5 genome model. Finally, differential expression analysis was performed on the dataset using the R-package DESeq2 (Love *et al.*, 2014) as described in the package vignette. The RNA-seq data have been deposited in the SRA archive under the number GSE90609.

### Nitric oxide detection

The fluorescent dye sensitive to NO oxidation, DAF-FM diacetate (10 μM), was added to microalgal suspensions that were then incubated for 0.5–1 h in serum flasks (25°C, constant agitation at 180 rpm). Aliquots (5 ml) of these suspensions were withdrawn, centrifuged, re-suspended in the same volume of N-free medium and supplied with 10 mM NO<sub>2</sub><sup>-</sup> in the darkness. The cell suspensions (200 μl) were then observed using a fluorescence microscope (Olympus BX51 microscope, <http://www.olympus-ims.com/>) equipped with a U-MWIBA2 filter block (excitation 460–490 nm,

emission 510–550 nm) (Chang *et al.*, 2013). The specificity of DAF-FM for NO or HNO was tested using diethylamine NONOate (DEA NONOate  $\geq 98\%$ , Sigma-Aldrich) and Angeli's salt (HNO donor  $\geq 99\%$ ; Sigma-Aldrich), respectively.

## ACKNOWLEDGEMENTS

This research was supported by Massey University and the Marsden Fund Council from New Zealand government funding, administrated by the Royal Society of New Zealand (grant MAU1102). Co-funding from MINECO (Ministerio de Economía y Competitividad, Spain, grant no. BFU2015-70649-P) with support of the European FEDER program, Junta de Andalucía (BIO-502) and the University of Cordoba (Plan Propio) is also gratefully acknowledged.

## CONFLICT OF INTEREST

The authors declare no conflict of interest.

## SUPPORTING INFORMATION

Additional Supporting Information may be found in the online version of this article.

**Figure S1.** N<sub>2</sub>O synthesis is correlated to biomass concentration in *Chlamydomonas reinhardtii*.

**Figure S2.** Kinetics of N<sub>2</sub>O synthesis by *Chlamydomonas reinhardtii* 6145c supplied with nitrite.

**Table S1.** N<sub>2</sub>O synthesis by *Chlamydomonas reinhardtii* 6145c under various conditions.

**Table S2.** N<sub>2</sub>O synthesis from various *Chlamydomonas reinhardtii* strains supplied with NO<sub>2</sub><sup>-</sup> and incubated in the dark.

**Table S3.** N<sub>2</sub>O synthesis of non- or tungstate-treated cells of *Chlamydomonas reinhardtii* 6145c cultures supplied with NO<sub>2</sub><sup>-</sup> and incubated in the dark.

**Table S4.** Log<sub>2</sub> fold expression changes (Log<sub>2</sub>FC) between T3 control and T3 treatment for the 19 annotated eukaryotic initiation factor (EIF) genes. Numbers in parenthesis represent mean normalized counts.

**Table S5.** Wild-type and mutant strains used during this study.

**Appendix S1.** Polymerase chain reaction analysis.

**Appendix S2.** NO generation during N<sub>2</sub>O synthesis in *Chlamydomonas reinhardtii*.

**Appendix S3.** NOR amistrains construction.

**Appendix S4.** N<sub>2</sub>O quantification.

**Appendix S5.** RNA sequencing optimization and results summary.

## REFERENCES

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