

ELECTRONIC SUPPLEMENTARY MATERIALS

Deep insight into the minor fraction of virgin olive oil by using LC-MS and GC-MS multi-class methodologies

Lucía Olmo-García¹, Juan J. Polari², Xueqi Li³, Aadil Bajoub¹, Alberto Fernández-Gutiérrez¹, Selina C. Wang^{2,3}, Alegría Carrasco-Pancorbo^{1*}

¹*Department of Analytical Chemistry, Faculty of Sciences, University of Granada,
Ave. Fuentenueva s/n, E-18071 Granada, Spain*

²*Department of Food Science and Technology, University of California Davis, One Shields Avenue, Davis, CA
95616, USA*

³*Olive Center, University of California Davis, One Shields Avenue, Davis, CA 95616, USA.*

Index and brief description (in order of appearance)

- **Table 1a SM.** General overview of the available methods for the determination of the main families of VOO minor compounds (Single-class methodologies).
- **Table 1b SM.** Examples of multi-class approaches.
- **Figure 1 SM.** Efficiency (for each chemical class) of the different solvents (or mixtures of solvents) used to carry out the extraction.
- **Table 2 SM.** Evaluation of the performance of the extraction protocol.
- **Fig. 2 SM.** Base Peak Chromatogram of a VOO extract of Cayon in LC-MS and GC-MS.

Table 1a SM. Main characteristics of diverse methodologies, selected as example, for the specific determination of different minor VOO chemical classes: Single-class methodologies. (The official method, when available, is described in the shaded row).

Sample treatment	Separation conditions	Detection conditions	Number of Analytes	Ref.
Phenolic compounds				
UAE (15 min) 2 g sample + 5 mL MeOH/H ₂ O (80:20, v/v)	RP-LC C18 (250 × 4.6 mm, 5 μm) Gradient: H ₂ O (0.2% H ₃ PO ₄)/MeOH/ACN (82 min)	UV 280 nm	27 (total referred to TY)	(International Olive Council, 2009)
LLE 2g sample + 6 mL MeOH/H ₂ O (60:40, v/v) → 1 mL MeOH	RP-LC C18 (150 × 4.6 mm, 1.8 μm) Gradient: H ₂ O (0.5% AcH)/ACN (25 min)	MS (ESI-IT/TOF) Negative polarity	20	(Bajoub et al., 2015)
LLE 1 g sample + 2 mL MeOH → 850 μL MeOH/H ₂ O (80:20, v/v)	RP-LC C18 (100 × 2.1 mm, 1.8 μm) Gradient: H ₂ O (0.1% CO ₂ H ₂)/ACN (0.1% CO ₂ H ₂) (30 min)	MS (ESI-QTOF) Negative polarity	34 (no quantification)	(Capriotti et al., 2014)
SPE (diol-bonded phase cartridge) 2.5 g sample (10 mL MeOH) → 100 μL derivatization reagent	GC ZB-5MS (30 m × 0.25 mm, 0.25 μm) T: 150-295 °C (71 min)	MS (EI-IT) Positive polarity	21 (no quantification)	(Ríos, Gil, & Gutiérrez-Rosales, 2005)
SPE (diol-bonded phase cartridge) 60 g sample (40 mL MeOH) → 2 mL MeOH/H ₂ O (50:50, v/v)	CE Fused silica capillary (400 × 0.375 mm, 50 μm) Running buffer: (45 mM sodium tetraborate, pH 9.3) (7 min)	DAD 200, 240, 280, 340 nm	16	(Carrasco-Pancorbo et al., 2006)
Triterpenic compounds				
Saponification, TLC 5 g sample → 50 μL silylation reagent/mg analyte	GC SE-52 or SE-54 (20-30 m × 0.25-0.32 mm, 0.10-0.30 μm) T: 260 °C (30-60 min)	FID	2 (plus 15 sterols)	(European Commission, 2016)
SPE (bonded aminopropyl phase cartridge) 0.2 g sample (6 mL diethyl ether/acetic acid (98:2, v/v)) → 200 μL silylation reagent	GC SGL-5 (25 m × 0.25 mm, 0.25 μm) T: 260-320 °C (34 min)	FID	3	(Pérez-Camino & Cert, 1999)
SPE (bonded aminopropyl phase cartridge) 0.2 g sample (6 mL diethyl ether/acetic acid (98:2, v/v)) → 200 μL silylation reagent	GC Rtx-65TG (30 m × 0.25 mm, 0.10 μm) T: 260-320 °C	MS (EI-IT MS) Positive polarity	1	(Guinda, Albi, Pérez-Camino, & Lanzón, 2004)
LLE 0.8 g sample (9.6 mL MeOH/EtOH (50:50, v/v)) → 2.4 mL MeOH	RP-LC C18 (250 × 4.6 mm, 5 μm) Isocratic: 8% H ₂ O (H ₃ PO ₄)/92% MeOH (20 min)	UV 210 nm	2 (plus 2 fatty acids)	(García, Brenes, Dobarganes, Romero, & Ruiz-Méndez, 2008)
UAE 0.2 g sample (10 mL MeOH) → 1 mL MeOH	RP-LC C18 (100 × 4.6 mm, 1.8 μm) Isocratic: 10% ammonium buffer (pH 9.6)/54% ACN/36% MeOH (13 min)	MS (ESI-IT) Negative/positive polarities	6	(Olmo-García, Bajoub, Monasterio, Fernández-Gutiérrez, & Carrasco-Pancorbo, 2018)

Tocopherols

Dilution 2 g sample + 25 mL hexane	NP-LC Si60 (250 × 4.0 mm, 5 μm) Isocratic: 0.5% 2-propanol/99.5% hexane (30 min)	FLD λ _{ex} 290 nm, λ _{em} 330 nm UV 290 nm	4	(IUPAC, 1991)
Dilution 2.5 g sample + 5 mL hexane	NP-LC Si60 (250 × 4.6 mm, 5 μm) Isocratic: 1% 2-propanol/98.5% hexane/0.5% EtOH	FLD λ _{ex} 290 nm, λ _{em} 330 nm	4	(Franco, Galeano-Díaz, Sánchez, De Miguel, & Martín-Vertedor, 2014)
LLE 4 g sample + 20 mL MeOH + 10 mL MeOH/2-propanol (80:20 (v/v)) □ 1 mL MeOH	Nano-RP-LC Monolithic C18 (250 × 0.1 mm) Isocratic: 75% ACN/8% MeOH/17% H ₂ O (0.2% AcH) (20 min)	UV 295 nm	4	(Cerretani, Lerma-García, Herrero-Martínez, Gallina-Toschi, & Simó-Alfonso, 2010)
Dilution (1:10, v/v) with isopropanol	RP-LC C18 (150 × 2.1 mm, 5 μm) Isocratic: 50% MeOH/50% ACN (12 min)	FLD λ _{ex} 290 nm, λ _{em} 330 nm	4	(Bakre, Gadmale, Toche, & Gaikwad, 2015)
Saponification 0.025 g sample → 1 mL MeOH	RP-LC FPF (200 × 4.6 mm, 5 μm) Isocratic: 95% MeOH/5% H ₂ O (15 min)	MS (APCI/ESI-Q) Negative polarity	4	(Lanina, Toledo, Sampels, Kamal-Eldin, & Jastrebova, 2007)

Sterols

Saponification, TLC 5 g sample → 50 μL silylation reagent/mg analyte	GC SE-52 or SE-54 (20-30 m × 0.25-0.32 mm, 0.10-0.30 μm) T: 260 °C (30-60 min)	FID	15 (plus 2 triterpenic alcohols)	(European Commission, 2016)
Saponification, solid supported liquid extraction, SPE (base-activated silica cartridge) 0.2 g sample (60 mL diethyl ether), (10 mL hexane/diethyl ether (60:40, v/v)) → 250 μL silylation reagent	GC HP-5MS (30 m × 0.25 mm, 0.25 μm) Isothermal T: 260 °C (55 min)	FID	15 (plus 2 triterpenic alcohols)	(Mathison & Holstege, 2013)
Saponification, TLC 5 g sample → 50 μL silylation reagent/mg analyte	RP-LC C18 (150 × 2.1 mm, 5 μm) Gradient: H ₂ O (0.01% AcH)/ACN (20 min)	MS (APCI-Q) Positive polarity	5 (plus 2 triterpenic alcohols)	(Cañabate-Díaz et al., 2007)
Saponification, SPE (silica-based octadecyl bonded phase) 1 g sample (15 mL 5% MeOH in chloroform (v/v)) → 1 mL MeOH (dilution 1:10, v/v)	Nano-RP-LC C18 (150 × 0.1 mm, sub-2 μm) Isocratic: MeOH (22 min)	DAD 195 nm	5	(Rocco & Fanali, 2009)
Saponification, TLC 5 g sample → 50 μL silylation reagent/mg analyte	CEC Methacrylate ester-based monolithic column (8.5 cm) Mobile phase: 85% ACN/10% 2-propanol/5% aqueous Tris buffer (5 mM, pH 8.0) (8 min)	DAD 210 nm	5	(Lerma-García, Simó-Alfonso, Ramis-Ramos, & Herrero-Martínez, 2008)

Fatty acids

<i>Trans</i> -esterification 0.1 g → 2 mL heptane	GC Polar polysiloxane column (60 m × 0.20-0.32 mm, 0.10-0.20 μm) T: 165-210 °C (37 min)	FID	23	(European Commission, 2016)
Saponification, <i>trans</i> -esterification 0.1 g → H ₂ SO ₄ 0.5 M in MeOH	GC Supelcowax 10 (30 m × 0.25 mm, 0.25 μm) T: 180- 240 °C (15 min)	FID	6	(Bodoira et al., 2015)
<i>Trans</i> -esterification 0.1 g → 2 mL hexane (dilution 1:50, v/v)	GC SPTM-2380 (60 m × 0.25 mm, 0.20 μm) T: 120-250 °C (36 min)	FID	10	(Sánchez de Medina, Calderón-Santiago, El Riachy, Priego-Capote, & Luque de Castro, 2015)
<i>Trans</i> -esterification 0.1 g → 5 mL hexane	GC DB-1 (30 m × 0.25 mm × 0.10 μm) T: 90-250 °C (9 min)	MS (EI-Q) Positive polarity	18	(Alves et al., 2016)
Dilution (1:20, v/v) with isopropanol	RP-LC C18 (50 × 2.1 mm, 1.9 μm) Gradient: ACN/H ₂ O (8 min)	MS (ESI-QqQ) Negative polarity	12	(Wabaidur et al., 2016)

Table 1b SM. Main characteristics of different methodologies for simultaneous determination of two (or more) minor VOO chemical classes: Multi-class approaches.

Sample treatment	Separation conditions	Detection conditions	Number of Analytes	Ref
LLE 10 g sample + 50 mL MeOH + 25 mL MeOH/2-propanol (80:20 (v/v)) → 5 mL MeOH/2-propanol/hexane (1:3:1, v/v/v)	RP-LC C18 (250 × 4 mm, 5 μm) Gradient: H ₂ O (2% AcH)/MeOH/ACN/2-propanol (70 min)	DAD 280 nm	10 phenolic compounds, 2 tocopherols	(Tasioula-Margari & Okogeri, 2001)
LLE 1 g sample + 15 mL MeOH → 1 mL MeOH	GC HP-5 MS (30 m × 0.25 mm, 0.25 μm) T: 70-300 °C (60.5 min)	MS (EI-Q) Positive polarity	13 phenolic compounds, 3 triterpenic compounds	(Kalogeropoulos, Kaliora, Artemiou, & Giogios, 2014)
Saponification 5 g sample → 0.5 mL ACN	RP-LC C18 (250 × 3 mm, 5 μm) Gradient: H ₂ O (0.01% AcH)/ACN (65 min)	MS (APCI-IT) Positive polarity	9 sterols, 3 tocopherols, 2 triterpenic dialcohols	(Zarrouk, Carrasco-Pancorbo, Zarrouk, Segura-Carretero, & Fernández-Gutiérrez, 2009)
Saponification 1 g sample → 130 μL silylation reagent/mg analyte	GC SPB-5 (30 m × 0.53 mm, 0.5 μm) T: 180-270 °C (76 min)	FID	4 sterols, α-tocopherol, squalene, 7 aliphatic alcohols	(Giacometti, 2001)
Dilution 0.5 g sample → 10 mL 2-propanol/hexane (1.5:98.5, v/v)	NP-LC Si60 (250 × 4.6 mm, 5 μm) Isocratic: 1.5% 2-propanol/98.5% hexane (30 min)	DAD 409, 430, 433, 452 nm FLD λ _{ex} 295 nm, λ _{em} 330 nm	3 tocopherols, 3 tocotrienols, 2 chlorophylls, 2 pheophytins, β-carotene	(Seppanen, Rahmani, & Csallany, 2003)

The symbol “→” means evaporation and reconstitution.

For solid supported liquid extraction and SPE, only the extraction or elution solvent (not considering cleaning steps) is indicated between brackets.

Abbreviations (used in Tables 1a and b SM)

methanol (MeOH), ethanol (EtOH), liquid-liquid extraction (LLE), solid phase extraction (SPE), ultrasounds assisted extraction (UAE), liquid/gas chromatography (LC/GC), capillary electrophoresis (CE), diode array detector (DAD), fluorescence detector (FLD), flame ionization detector (FID), mass spectrometry (MS), electron impact (EI), electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), ion trap (IT), time of light (TOF), quadrupole (Q), triple quadrupole (QqQ).

References (cited in Tables 1a and b SM)

- Alves, E., Melo, T., Rey, F., Moreira, A. S. P., Domingues, P., & Domingues, M. R. (2016). Polar lipid profiling of olive oils as a useful tool in helping to decipher their unique fingerprint. *LWT - Food Science and Technology*, *74*, 371–377. <https://doi.org/10.1016/j.lwt.2016.07.071>
- Bajoub, A., Hurtado-Fernández, E., Ajal, E. A., Ouazzani, N., Fernández-Gutiérrez, A., & Carrasco-Pancorbo, A. (2015). Comprehensive 3-year study of the phenolic profile of Moroccan monovarietal virgin olive oils from the Meknès region. *Journal of Agricultural and Food Chemistry*, *63*(17), 4376–4385. <https://doi.org/10.1021/jf506097u>
- Bakre, S. M., Gadmale, D. K., Toche, R. B., & Gaikwad, V. B. (2015). Rapid determination of alpha tocopherol in olive oil adulterated with sunflower oil by reversed phase high-performance liquid chromatography. *Journal of Food Science and Technology*, *52*(5), 3093–3098. <https://doi.org/10.1007/s13197-014-1309-7>
- Bodoira, R., Torres, M., Pierantozzi, P., Taticchi, A., Servili, M., & Maestri, D. (2015). Oil biogenesis and antioxidant compounds from “Arauco” olive (*Olea europaea* L.) cultivar during fruit development and ripening. *European Journal of Lipid Science and Technology*, *117*, 377–388. <https://doi.org/10.1002/ejlt.201400234>
- Cañabate-Díaz, B., Segura Carretero, A., Fernández-Gutiérrez, A., Belmonte Vega, A., Garrido French, A., Martínez Vidal, J. L., & Duran Martos, J. (2007). Separation and determination of sterols in olive oil by HPLC-MS. *Food Chemistry*, *102*(3), 593–598. <https://doi.org/10.1016/j.foodchem.2006.05.038>
- Capriotti, A. L., Cavaliere, C., Crescenzi, C., Foglia, P., Nescatelli, R., Samperi, R., & Laganà, A. (2014). Comparison of extraction methods for the identification and quantification of polyphenols in virgin olive oil by ultra-HPLC-QToF mass spectrometry. *Food Chemistry*, *158*, 392–400. <https://doi.org/10.1016/j.foodchem.2014.02.130>
- Carrasco-Pancorbo, A., Gómez-Caravaca, A. M., Cerretani, L., Bendini, A., Segura-Carretero, A., & Fernández-Gutiérrez, A. (2006). Rapid quantification of the phenolic fraction of Spanish virgin olive oils by capillary electrophoresis with UV detection. *Journal of Agricultural and Food Chemistry*, *54*(21), 7984–7991. <https://doi.org/10.1021/jf0617925>
- Cerretani, L., Lerma-García, M. J., Herrero-Martínez, J. M., Gallina-Toschi, T., & Simó-Alfonso, E. F. (2010). Determination of tocopherols and tocotrienols in vegetable oils by nanoliquid chromatography with ultraviolet-visible detection using a silica monolithic column. *Journal of Agricultural and Food Chemistry*, *58*(2), 757–761. <https://doi.org/10.1021/jf9031537>
- European Commission. (2016). Commission implementing regulation (EU) 2016/1095 of 6 July 2016 amending Regulation (EEC) No 2568/91 on the characteristics of olive oil and olive-residue oil and on the relevant methods of analysis. *Official Journal of the European Union*.
- Franco, M. N., Galeano-Díaz, T., Sánchez, J., De Miguel, C., & Martín-Vertedor, D. (2014). Total phenolic compounds and tocopherols profiles of seven olive oil varieties grown in the south-west of Spain. *Journal of Oleo Science*, *63*(2), 115–125. <https://doi.org/10.5650/jos.ess13098>
- García, A., Brenes, M., Dobarganes, M. C., Romero, C., & Ruíz-Méndez, M. V. (2008). Enrichment of pomace olive oil in triterpenic acids during storage of “Alpeorujo” olive paste. *European Journal of Lipid Science and Technology*, *110*(12), 1136–1141. <https://doi.org/10.1002/ejlt.200800070>
- Giacometti, J. (2001). Determination of aliphatic alcohols, squalene, alpha-tocopherol and sterols in olive oils: direct method involving gas chromatography of the unsaponifiable fraction following silylation. *The Analyst*, *126*(4), 472–475. <https://doi.org/10.1039/b007090o>

- Guinda, A., Albi, T., Pérez-Camino, M. C., & Lanzón, A. (2004). Supplementation of oils with oleanolic acid from the olive leaf (*olea europaea*). *European Journal of Lipid Science and Technology*, *106*, 22–26. <https://doi.org/10.1002/ejlt.200300769>
- International Olive Council. (2009). Determination of biophenols in olive oils by HPLC. *COI/T.20/Doc No 29*, 1–8.
- IUPAC. (1991). Method 2432. Determination of tocopherols and tocotrienols in vegetable oils and fats by high performance liquid chromatography. In *Standard Methods for the Analysis of Oils, Fats and Derivatives* (7th ed., pp. 1–151). Oxford: Blackwell. <https://doi.org/10.1351/pac198254061257>
- Kalogeropoulos, N., Kaliora, A. C., Artemiou, A., & Giogios, I. (2014). Composition, volatile profiles and functional properties of virgin olive oils produced by two-phase vs three-phase centrifugal decanters. *LWT - Food Science and Technology*, *58*(1), 272–279. <https://doi.org/10.1016/j.lwt.2014.02.052>
- Lanina, S. A., Toledo, P., Sampels, S., Kamal-Eldin, A., & Jastrebova, J. A. (2007). Comparison of reversed-phase liquid chromatography-mass spectrometry with electrospray and atmospheric pressure chemical ionization for analysis of dietary tocopherols. *Journal of Chromatography A*, *1157*, 159–170. <https://doi.org/10.1016/j.chroma.2007.04.058>
- Lerma-García, M. J., Simó-Alfonso, E. F., Ramis-Ramos, G., & Herrero-Martínez, J. M. (2008). Rapid determination of sterols in vegetable oils by CEC using methacrylate ester-based monolithic columns. *Electrophoresis*, *29*(22), 4603–4611. <https://doi.org/10.1002/elps.200800247>
- Mathison, B., & Holstege, D. (2013). A rapid method to determine sterol, erythrodiol, and uvaol concentrations in olive oil. *Journal of Agricultural and Food Chemistry*, *61*(19), 4506–4513. <https://doi.org/10.1021/jf400254k>
- Olmo-García, L., Bajoub, A., Monasterio, R. P., Fernández-Gutiérrez, A., & Carrasco-Pancorbo, A. (2018). Development and validation of LC-MS-based alternative methodologies to GC – MS for the simultaneous determination of triterpenic acids and dialcohols in virgin olive oil. *Food Chemistry*, *239*, 631–639. <https://doi.org/10.1016/j.foodchem.2017.06.142>
- Pérez-Camino, M. C., & Cert, A. (1999). Quantitative determination of hydroxy pentacyclic triterpene acids in vegetable oils. *Journal of Agricultural and Food Chemistry*, *47*(4), 1558–1562.
- Ríos, J. J., Gil, M. J., & Gutiérrez-Rosales, F. (2005). Solid-phase extraction gas chromatography-ion trap-mass spectrometry qualitative method for evaluation of phenolic compounds in virgin olive oil and structural confirmation of oleuropein and ligstroside aglycons and their oxidation products. *Journal of Chromatography A*, *1093*(1–2), 167–176. <https://doi.org/10.1016/j.chroma.2005.07.033>
- Rocco, A., & Fanali, S. (2009). Analysis of phytosterols in extra-virgin olive oil by nano-liquid chromatography. *Journal of Chromatography A*, *1216*(43), 7173–7178. <https://doi.org/10.1016/j.chroma.2009.03.081>
- Sánchez de Medina, V., Calderón-Santiago, M., El Riachy, M., Priego-Capote, F., & Luque de Castro, M. D. (2015). Influence of genotype on the fatty acids composition of virgin olive oils from advanced selections obtained by crosses between Arbequina, Picual, and Frantoio cultivars along the ripening process. *European Journal of Lipid Science and Technology*, *117*(8), 1261–1270. <https://doi.org/10.1002/ejlt.201400488>
- Seppanen, C. M., Rahmani, M., & Csallany, A. S. (2003). Simultaneous determination of chlorophylls, pheophytins, beta-carotene, tocopherols, and tocotrienols in olive and soybean oils by high-performance liquid chromatography. *Journal of Food Science*, *68*(5), 1644–1647. <https://doi.org/10.1111/j.1365-2621.2003.tb12306.x>
- Tasioula-Margari, M., & Okogeri, O. (2001). Simultaneous determination of phenolic compounds and tocopherols in virgin olive oil using HPLC and UV detection. *Food Chemistry*, *74*(3), 377–383. [https://doi.org/https://doi.org/10.1016/S0308-8146\(01\)00176-5](https://doi.org/https://doi.org/10.1016/S0308-8146(01)00176-5)
- Wabaidur, S. M., AlAmmari, A., Aqel, A., AL-Tamrah, S. A., Alothman, Z. A., & Ahmed, A. Y. B. H. (2016). Determination of free fatty acids in olive oils by UPHLC–MS. *Journal of Chromatography B*, *1031*, 109–115. <https://doi.org/10.1016/j.jchromb.2016.07.040>
- Zarrouk, W., Carrasco-Pancorbo, A., Zarrouk, M., Segura-Carretero, A., & Fernández-Gutiérrez, A. (2009). Multi-component analysis (sterols, tocopherols and triterpenic dialcohols) of the unsaponifiable fraction of vegetable oils by liquid chromatography-atmospheric pressure chemical ionization-ion trap mass spectrometry. *Talanta*, *80*(2), 924–934. <https://doi.org/10.1016/j.talanta.2009.08.022>

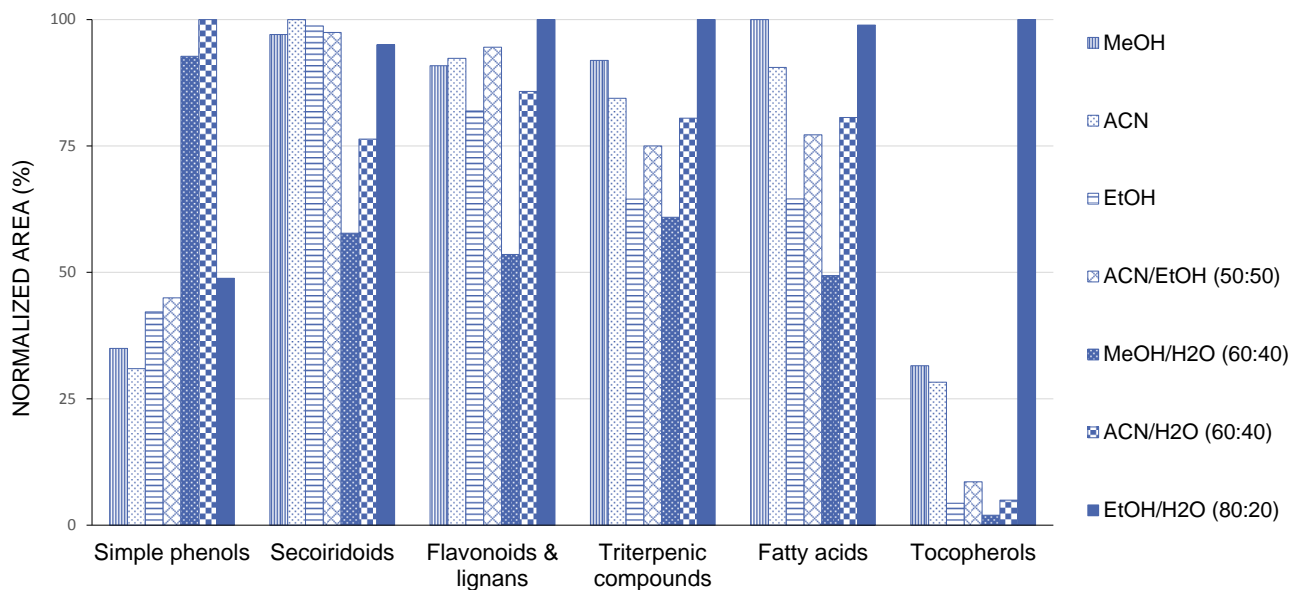


Fig. 1 SM. Total (normalized) peak areas of the different VOO minor chemical classes (simple phenols, secoiridoids, flavonoids and lignans, triterpenic compounds, fatty acids, and tocopherols) found in the extracts obtained with the seven extractant agents selected in the first stage of the sample treatment optimization (MeOH, ACN, EtOH, ACN/EtOH (50:50, v/v), MeOH/H₂O (60:40, v/v), ACN/H₂O (60:40, v/v) and EtOH/H₂O (80:20, v/v)). Analyses were made in the LC-IT MS system. Normalization was made considering as 100% the total area value of the most effective extractant agent (for each chemical class) and referring the rest to that value. The compounds considered as members of each chemical class are those included in Table 2 SM.

Table 2 SM. Evaluation of the performance (efficiency) of the extraction protocol considering the percentage of different target compounds obtained in each successive extraction stage as well as the extraction repeatability.

Chemical class	Compound	1st%	2nd%	3rd%	Repeatability (RSD)	
Simple phenols	quinic acid	99.5	0.5	-	9.7	LC-MS
	hydroxytyrosol	75.6	24.1	0.3	3.6	
	tyrosol	93.6	6.4	-	6.4	
	<i>p</i> -coumaric acid	100.0	-	-	1.1	
	vanillin	100.0	-	-	0.7	
	ferulic acid	100.0	-	-	8.8	
Secoiridoids	desoxy elenolic acid	99.9	0.1	-	0.6	
	elenolic acid	93.7	5.2	1.1	3.3	
	decarboxymethyl oleuropein aglycone	76.8	14.8	8.4	7.8	
	oleuropein aglycone	84.7	11.9	3.4	7.9	
	decarboxymethyl ligstroside aglycone	77.9	11.2	10.9	3.0	
	ligstroside aglycone	89.7	5.5	4.7	9.8	
Flavonoids	luteolin	99.8	0.2	-	0.9	
	apigenin	99.2	0.8	-	6.5	
Lignans	pinoresinol	98.7	1.3	-	5.0	
	acetoxypinoresinol	97.9	2.1	-	5.4	
Triterpenic compounds	maslinic acid	96.4	2.8	0.7	2.7	
	betulinic acid	99.8	0.2	-	4.3	
	oleanolic acid	95.7	2.4	2.0	7.4	
	erythrodiol	75.5	24.5	-	8.0	
Fatty Acids	linolenic acid	97.5	2.3	0.2	2.8	
	linoleic acid	98.4	1.5	0.2	3.5	
	oleic acid	94.5	5.0	0.5	5.9	
	palmitic acid	95.2	3.7	1.1	1.5	
	palmitoleic acid	98.9	1.1	0.0	1.7	
	stearic acid	88.3	7.5	4.1	1.5	
Tocopherols	α -tocopherol	78.1	15.9	5.9	1.3	GC-MS
	β -tocopherol	86.6	10.3	3.0	2.5	
	γ -tocopherol	82.9	13.1	4.0	0.1	
	δ -tocopherol	90.5	7.3	2.2	4.0	
Sterols	β -sitosterol	70.6	22.4	7.0	4.4	
	campesterol	78.6	17.2	4.2	5.4	
	stigmasterol	80.7	15.4	3.9	0.7	
	Δ^5 -avenasterol	76.1	18.6	5.2	7.2	
	cycloartenol	75.3	19.2	5.6	0.5	
	methylcycloartanol	70.6	22.7	6.6	5.0	

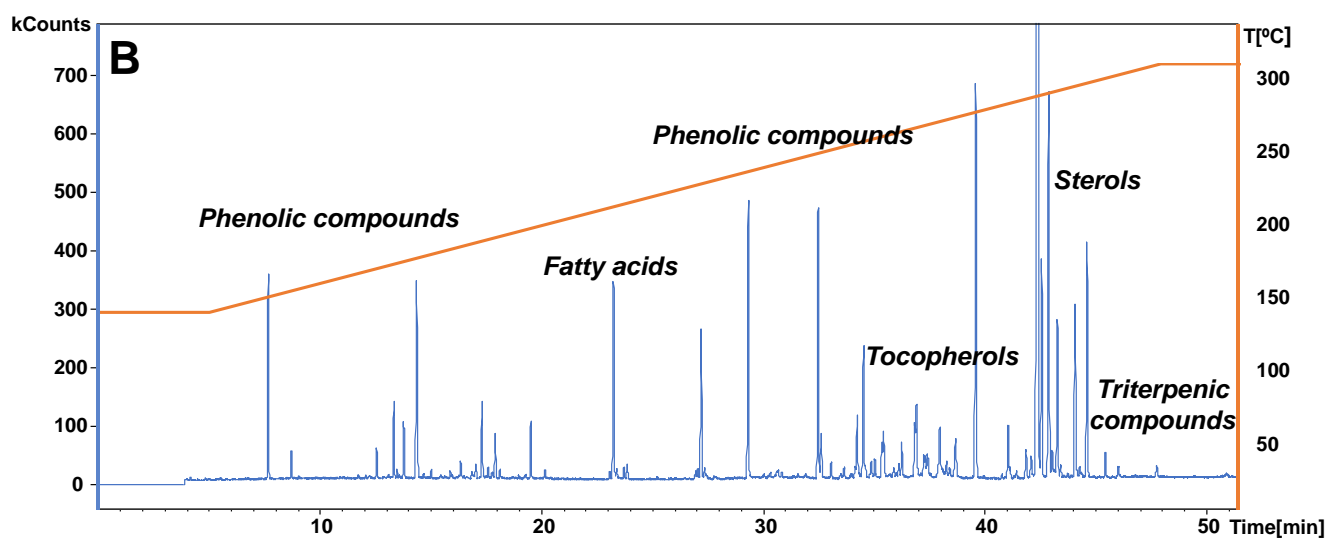
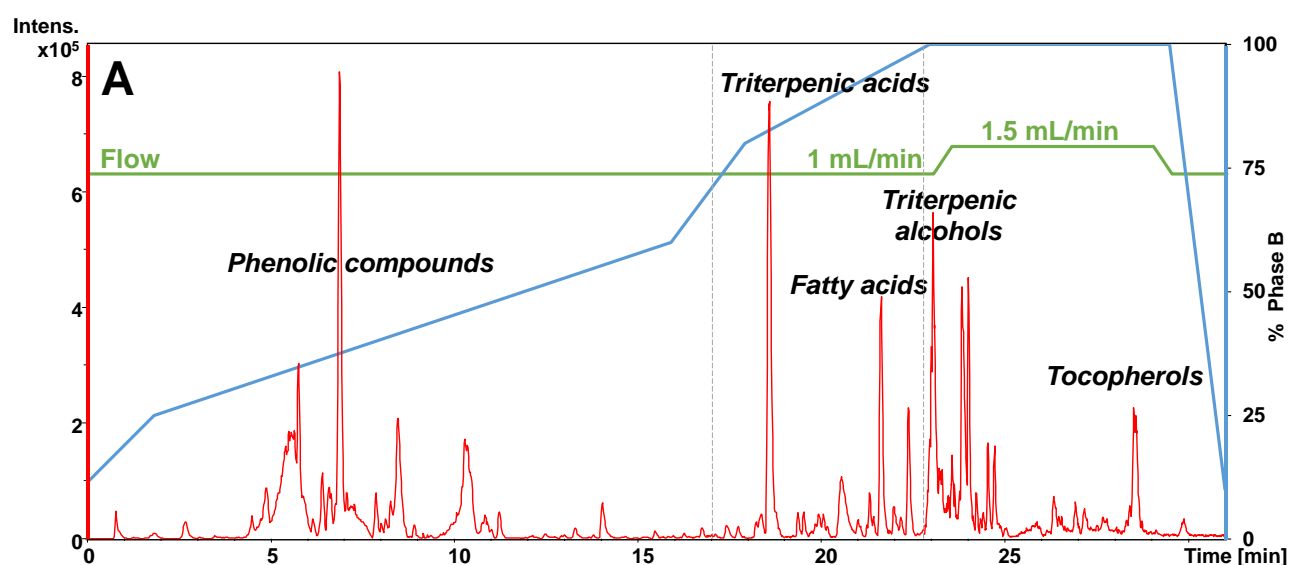


Fig. 2 SM. A) LC-MS Base Peak Chromatogram (BPC) of a VOO extract of Cayon variety obtained under the optimum conditions. The optimized gradient composition of mobile phases and flow are also shown in the figure, as well as the elution areas of each one of the determined families. B) GC-MS BPC of the same VOO extract together with the optimized temperature ramp gradient and elution areas of each chemical class.