



Unravelling the main mechanism responsible for nocturnal CO₂ uptake by dryland soils

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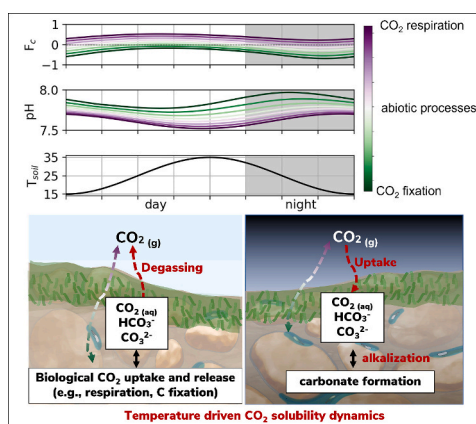
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HIGHLIGHTS

- Nocturnal CO₂ uptake by dryland soils is explained with a mechanistic model.
- Diel temperature cycles were the main driver by increasing CO₂ solubility at night.
- Alkalinization of thin soil water films at night can induce carbonate precipitation.
- Biological soil crusts have the potential to modulate the abiotic CO₂ uptake.
- These mechanisms can contribute to the global residual terrestrial carbon sink.

GRAPHICAL ABSTRACT



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ABSTRACT

Soil respiration, or CO₂ efflux from soil, is a crucial component of the terrestrial carbon cycle in climate models. Contrastingly, many dryland soils absorb atmospheric CO₂ at night, but the exact mechanisms driving this uptake are actively debated. Here we used a mechanistic model with heuristic approaches to unravel the underlying processes of the observed patterns of soil-atmosphere CO₂ fluxes. We show that the temperature drop during nighttime is the main driver of CO₂ uptake by increasing CO₂ solubility and local water pH of a thin water film on soil particle surfaces, providing favourable conditions for carbonate precipitation. Our data demonstrate that the nocturnal inorganic carbon absorption is a common soil process, but often offset by biological CO₂ production. The uptake rates can be impacted by different successional stages of biocrusts that consume or produce CO₂ and

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modify the pH of the soil water film, which can be maintained by non-rainfall water inputs, such as pore space condensation. Annual estimates of nocturnal carbon uptake, based on in situ continuous measurements at the soil level in drylands are still very scarce, but fluxes of up to several tens of $\text{g C m}^{-2} \text{y}^{-1}$ have been reported, potentially accounting for a considerable fraction of the global residual terrestrial carbon sink.

1. Introduction

Soils are a major reservoir of carbon (C), storing organic C (1530 Pg), solid inorganic C as mostly carbonates (940 Pg), and dissolved inorganic C in soil solution (1404 Pg) (Monger et al., 2015). Each pool amounts to more C than is contained in the vegetation (450–650 Pg) or the atmosphere (828 Pg) globally (IPCC, 2013). Soil C is continuously exchanged with the atmosphere through CO_2 fluxes (F_c) resulting from CO_2 production, consumption, and transport processes in soil at an estimated rate of $87 \pm 9 \text{ Pg y}^{-1}$ (Jian et al., 2022). Locally, F_c can be either positive (CO_2 efflux) or negative (CO_2 influx), depending on the balance between CO_2 production and consumption. In most environments F_c has been found to be positive, and research has mainly focused on understanding CO_2 production through respiration of roots, microorganisms and decomposition of organic matter (Goffin et al., 2015; Jochheim et al., 2022; Lu et al., 2021; Ryan and Law, 2005; Waring and Powers, 2016).

However, dryland soils around the globe have been increasingly reported to absorb atmospheric CO_2 at night suggesting a possible role of soil CO_2 consumption processes on the C balance of water-limited ecosystems (Fa et al., 2015; Hamerlynck et al., 2013; Sagi et al., 2021; Sánchez-Cañete et al., 2018; Yates et al., 2013). Due to the large extension of drylands representing around 45 % of Earth's land surface (Lal, 2019), the observed nocturnal CO_2 uptake could contribute to explain the residual/missing terrestrial C sink ($3.1 \pm 0.9 \text{ Pg y}^{-1}$) (Houghton et al., 2018), i.e., the imbalance in the global C budget that remained unresolved for decades. On the other hand, some authors have shown skepticism regarding the relevance and significance of these fluxes at the global scale (Schlesinger, 2017; Schlesinger et al., 2009), essentially arguing that the reported rates of CO_2 uptake were incompatible with plausible mechanisms of CO_2 consumption and therefore due to inaccurate instrumentation.

Drylands offer unique conditions to study CO_2 consumption processes in soil. Although these processes are likely ubiquitous across ecosystems, soil CO_2 consumption is often masked by high respiration rates in mesic or humid environments where it can remain undetected by flux measurements. In contrast, drylands maintain low CO_2 production rates in soil: the scarcity of plants strongly reduces root respiration, and the decomposition of organic matter is limited by the low water availability as well as low soil organic carbon (SOC) content and SOC bounding to carbonates. With 80 to 97 % of the soil solid inorganic carbon (SIC) (mainly as CaCO_3) located in water-limited ecosystems, SIC stocks often largely exceed SOC stocks (Plaza et al., 2018). However, despite such favourable conditions to detect CO_2 consumption in drylands, the mechanisms involved in nocturnal CO_2 uptake remain actively debated (Sagi et al., 2021).

It is important to unravel the mechanisms of nocturnal CO_2 uptake by dryland soils to predict the impact of climate change on the C storage of drylands, and to explore novel strategies to mitigate climate change. Some studies have found contradictory results on this topic, for example, the observation of CO_2 uptake only during dry conditions (Hamerlynck et al., 2013) or only during wet conditions (Fa et al., 2015, 2016b; Yates et al., 2013). This highlights the complexity of disentangling the mechanisms involved in the CO_2 uptake. Mostly, the following processes have been suggested to explain this uptake: (1) pH- and temperature-driven CO_2 dissolution, and subsequent leaching to aquifers (Ma et al., 2013, 2014; Wang et al., 2020); (2) calcium carbonate dissolution (Fa et al., 2016a, 2016b); (3) thermal convection, diel expansion/contraction of soil air promoted by temperature changes (Gao et al., 2021; Yang et al., 2020); (4) CO_2 adsorption on soil particles (Davidson et al., 2013;

Lopez-Canfin et al., 2022a); (5) precipitation of secondary carbonates favoured by the weathering of Ca-bearing minerals other than CaCO_3 (Lopez-Canfin et al., 2022b). In particular, the latter mechanism deserves special attention since it represents a long-term C sink due to the exogenic origin of Ca (Monger et al., 2015; Sanderman, 2012).

The dynamics of carbonate formation are usually considered to be slower than the nocturnal CO_2 uptake reported for drylands. For example, in semiarid areas, C sequestration rates through carbonate precipitation is estimated to range between 0.3 and $12.4 \text{ g C m}^{-2} \text{yr}^{-1}$ (Lal, 2001) while the nocturnal CO_2 uptake can be slightly faster, with estimates from continuous in situ measurements at the soil level of up to several tens of $\text{g C m}^{-2} \text{yr}^{-1}$ (Lopez-Canfin et al., 2022a). However, those estimates are not incompatible and remain scarce, thus uncertain at the global scale. In addition, soil microbiota could potentially accelerate this geochemical process through biomineralization (Liu et al., 2020). In drylands, biological soil crusts (hereafter biocrusts) are communities of cyanobacteria, algae, bacteria, lichens and mosses that inhabit the top few centimetres of the soil (Garcia-Pichel, 2023; Weber et al., 2022). These are hotspots of microbial activity that drive biogeochemical processes sustaining important ecological functions in drylands. Such functions include the alteration of F_c (Darrrouzet-Nardi et al., 2018), the modification of soil physico-chemical properties such as pH (Büdel et al., 2004; Wu et al., 2013), porosity (Miralles-Mellado et al., 2011), soil water and SOC content (Chamizo et al., 2016), as well as enhanced mineral weathering and biomineralization (Benzerara et al., 2014; Büdel et al., 2004; Chen et al., 2014; Souza-Egipsy et al., 2004). However, considering that the lack of water hinders biological activity and most soil CO_2 consumption processes, it is important to comprehend how these components interact during dry spells. So far, little attention has been given to abiotic CO_2 uptake in soils covered by biocrusts with modified physicochemical properties and predictive mechanistic or process-based models are needed to explain these field observations at the sub-diurnal resolution.

In this study, we aim at unravelling the dominant mechanism responsible for nocturnal soil CO_2 uptake using a mechanistic model, the simplified field application of Desert Biocrust Model (sfDBM), to simulate processes governing F_c . We disentangle different CO_2 uptake processes: (1) the dissolution of CO_2 driven by an increase in solubility at night; (2) the role of inorganic C partitioning that depends on soil water pH affecting inorganic C dynamics; (3) the contributions of these processes by biocrusts at different successional stages (Fig. S1). We validate our model predictions using continuous measurements of soil-atmosphere CO_2 fluxes and micrometeorological variables from a semiarid desert where nocturnal CO_2 uptake occurs on a regular basis (Lopez-Canfin et al., 2022b, 2024).

2. Results

2.1. Hourly measurements of soil-atmosphere CO_2 exchange and other environmental variables

The soil-atmosphere CO_2 exchange (F_c) displayed strong diel variability (coefficient of dispersion (CV) of all crusts = 135 %), which was comparable to the seasonal variability (CV = 126 %; Fig. 1a). The distribution of F_c was skewed (Fisher-Pearson coefficient of skewness = 0.72) due to sporadic rainfall events that triggered enhanced F_c for all crusts (Fig. S2, Fig. S3). At nighttime, atmospheric CO_2 uptake by soil was very frequent in early successional stages and even more frequent than CO_2 release (F_c median = $-0.10 \mu\text{mol m}^{-2} \text{s}^{-1}$) (Fig. S2). By

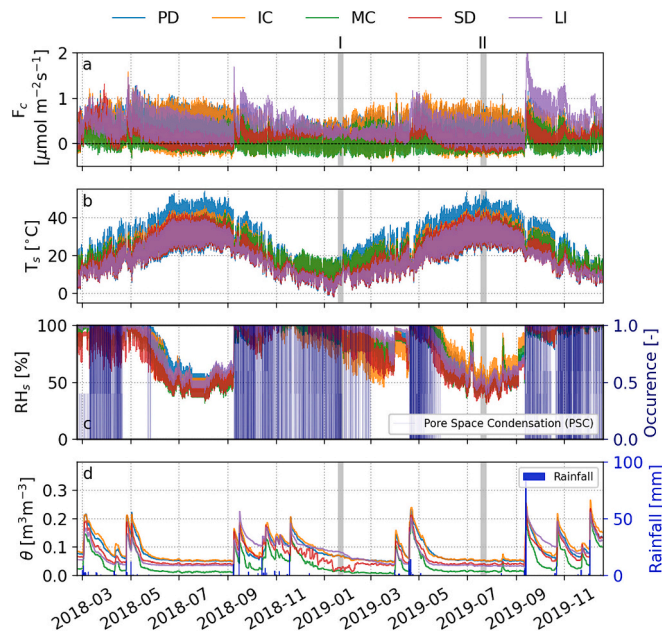


Fig. 1. Hourly time series of environmental measurements averaged by crust type. (a) soil-atmosphere CO₂ exchange (F_c); (b) soil temperature (T_s); (c) soil relative humidity (RH_s) and possible pore-space condensation in soil (navy bars); and (d) daily time series of soil water content (θ) and daily rainfall shown with blue bars. The time series of θ are displayed at daily resolution, instead of hourly resolution, because sensors were not accurate enough to capture the diel variability in this variable. Line colours of each panel indicate biocrust type; PD, physical depositional crust ($n = 3$ replicates) in blue; IC, incipient cyanobacteria ($n = 3$) in orange; MC, mature cyanobacteria ($n = 5$) in green; SD, lichen community dominated by *Squamarina lentigera* and *Diploschistes diacapsis* ($n = 3$) in red; LI, lichen community characterized by *Lepraria isidiata* ($n = 3$) in purple. Grey shaded areas indicate the dry periods with contrasting temperature conditions selected to test the predictions of the mechanistic model: test periods are numbered as (I) dry and cool; (II) dry and hot.

contrast, soil CO₂ release was more frequent than CO₂ uptake in late successional stages (median = $0.19 \mu\text{mol m}^{-2} \text{s}^{-1}$) (Fig. S2). The soil temperature (T_s) showed strong diel and seasonal variability, with a similar pattern across crust types but remarkably greater diel fluctuations in the physical depositional (PD) crust (Fig. 1b). The soil relative humidity (RH_s) and possible pore-space condensation (PSC, defined as the condition when soil temperature is equal or lower than the dew point at 5 cm depth) were substantially lower during summer than the rest of the year (Fig. 1c). The occurrence of PSC was more likely during nighttime and under wet or cool conditions (Fig. S4). The soil water content (θ) dynamics exhibited a pattern of pulses governed by rainfall events and subsequent drying (Fig. 1d).

2.2. Diel and seasonal patterns of CO₂ uptake by soil

The daily averaged time series of CO₂ release, uptake, and resultant net F_c as well as the average diel F_c cycle during contrasting periods in terms of both moisture and temperature are shown in Fig. 2. In addition, hourly time series and daily cumulative values of F_c during two dry periods under contrasting temperature conditions (dry and cool - winter vs. dry and hot - summer, marked as grey shaded areas I and II in Fig. 1) are plotted in Fig. 3. Overall, CO₂ uptake by soil was shaped by wetness, temperature and biocrust type. Uptake was greater in early successional stages (from physical depositional to cyanobacterial crusts; PD, IC, and MC) and persisted throughout the whole study period (Fig. 2a-e). It occurred during nighttime whereas CO₂ release occurred during daytime (Fig. 2f-i). Overall, the CO₂ uptake was greater during dry periods compared to wet periods (Fig. 2f-i). Among dry periods, the daily CO₂

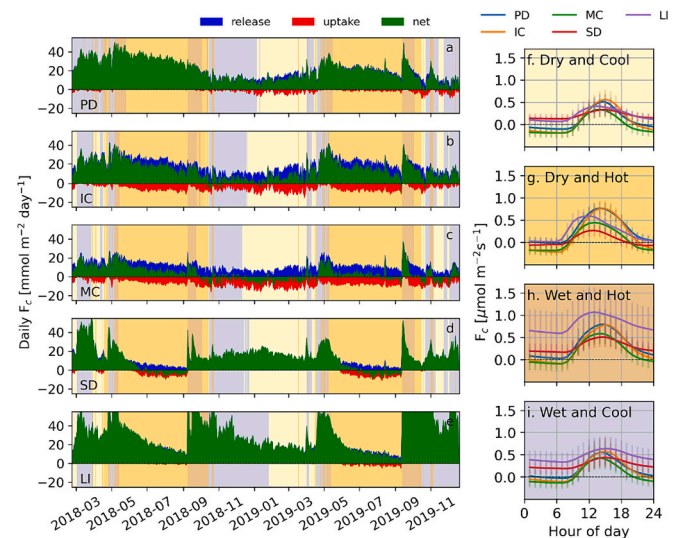


Fig. 2. Daily averaged time series of soil-atmosphere CO₂ flux (F_c) and diel F_c patterns under contrasting climatic periods. (a-e) Daily time series of F_c (averaged by biocrust type) subdivided into release (sum of positive hourly flux), uptake (sum of negative hourly flux), and net (daily sum of flux) values; (f-i) Average diel F_c patterns. Values are averaged by crust type: PD, physical depositional; (b) IC, incipient cyanobacteria; (c) MC, mature cyanobacteria; (d) SD, lichen community dominated by *Squamarina lentigera* and *Diploschistes diacapsis* and (e) LI, lichen community characterized by *Lepraria isidiata*. Note that the high values of LI in (e) are truncated for comparison across crust types. Classified dry/wet, and cool/hot periods are colour-shaded as follows: dry and cool (bright yellow), dry and hot (orange), wet and hot (brown), and wet and cool (purple) (f-i) The error bars represent the hourly standard deviation.

uptake increased under hot conditions and occurred even in late succession stages (Lichen crusts: SD and LI) (Fig. 3). Daily net CO₂ uptake occurred mainly under dry conditions in the MC crusts during winter and in the SD crusts during summer.

2.3. Effect of temperature-dependent CO₂ solubility on carbon uptake dynamics

The effect of temperature dynamics on F_c was examined with the sfDBM using a heuristic approach, representing a method of learning that uses simplified boundary conditions compared to field data. Here, to simulate the diel cycle of temperature, a sinusoidal function with a 24-h-period was used (Eq. (1)). The soil domain was set to be at a dry condition ($\theta \approx 0.05 \text{ m}^3 \text{ m}^{-3}$, effective film thickness on soil particles $\approx 1 \text{ m}^{-7}$). The model prediction showed a non-linear response of F_c dynamics to temperature changes (Fig. 4). At lower average temperature, the amplitude of the diel CO₂ cycle was higher, as influenced by greater temporal changes in temperature-driven CO₂ solubility (Henry's law). Essentially, the F_c dynamics followed the dynamics of time derivatives of temperature (dT/dt in Fig. 4b), reflecting CO₂ dissolution and degassing caused by temperature-dependent saturation of CO₂ in the thin soil water films. The soil CO₂ uptake occurred when the solubility of CO₂ was enhanced by the decrease in temperature ($dT/dt < 0$). The symmetric cycle of temperature yielded a diel hysteresis of F_c (Fig. 4d) and the dependency between F_c and dT/dt showed a pinched-hysteresis loop (Fig. 4e).

2.4. Effect of the water film Ca²⁺ concentration and pH

Under a modelled diel cycle of temperature ($25 \pm 10 \text{ }^\circ\text{C}$ at a dry state of $\theta \approx 0.05 \text{ m}^3 \text{ m}^{-3}$), we examined the effect of pH variation in the soil water film on F_c (Fig. 5). Here, Ca²⁺ was used to control the chemical status of the film under the assumption of local charge balance. Higher

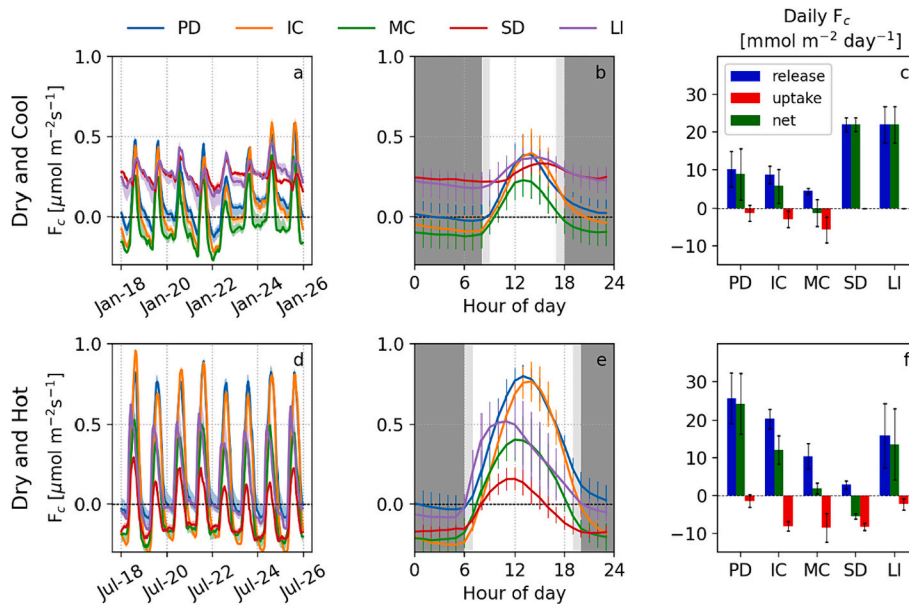


Fig. 3. Diel patterns and daily balances of soil-atmosphere CO₂ exchange (F_c) averaged by crust type during two contrasting dry periods. (a-c) dry and cool (d-f) dry and hot conditions, marked as I and II in Fig. 1. (a and d) Average F_c over a 8-day timespan by crust type: PD, physical depositional crust; (b) IC, incipient cyanobacteria; (c) MC, mature cyanobacteria; (d) SD, lichen community dominated by *Squamarina lentigera* and *Diploschistes diacapsis* and (e) LI, lichen community characterized by *Lepraria isidiata*. Lines are medians and shaded areas delimit the 1st and 3rd quartiles. (b and e) Mean diel cycles with hourly standard deviation over each period. The grey shading indicates nocturnal hours with low photosynthetically active radiation ($PAR < 5 \mu\text{mol m}^{-2} \text{s}^{-1}$). (c and f) Daily sum of soil CO₂ release, uptake, and resultant net F_c averaged by crust type. Error bars represent one standard deviation.

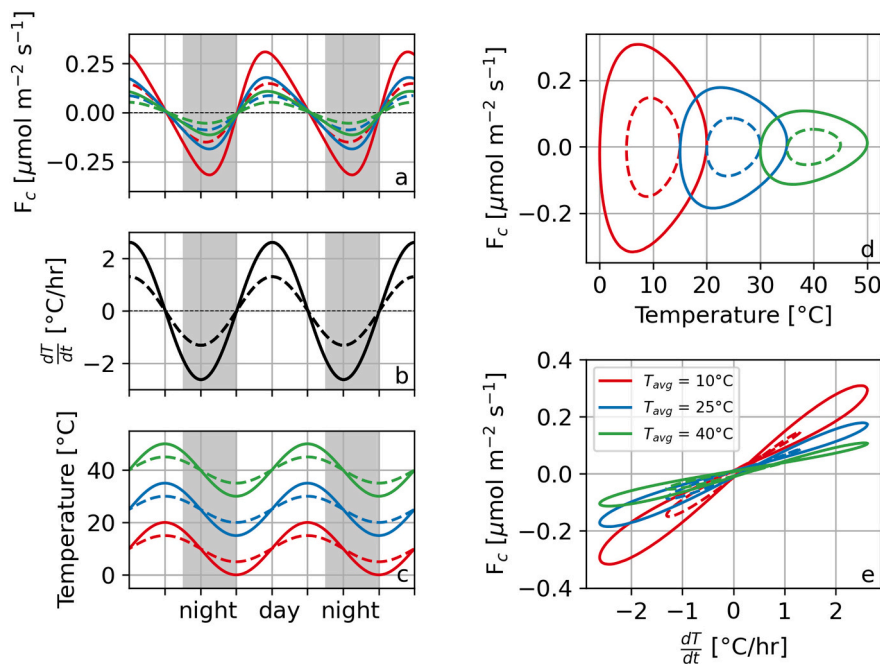


Fig. 4. Modelled effect of simulated diel temperature cycle on soil-atmosphere CO₂ exchange (F_c). (a) Prediction of F_c by the model under a simple temperature cycle, (b) the time derivatives, and (c) the assumed sinusoidal functions (Eq. (1)). We used three mean temperatures (T_{avg}): 10 °C (red), 25 °C (blue), and 40 °C (green) with two amplitudes (ΔT): 5 °C (dashed lines) and 10 °C (solid lines). The grey shading indicates fictitious nocturnal hours with decreasing temperature. F_c values are plotted against (d) temperature and (e) time derivatives.

values of Ca²⁺ concentration result in higher pH (lower values of H⁺ concentration for the charge compensation). The kinetics of inorganic C coupled with diel cycles of temperature had an impact on F_c and local pH in several ways: (1) a more basic pH increased the magnitude of the F_c diel cycle (Fig. 5a); (2) the CO₂ solubility dynamics caused the diel cycle of pH with alkalisation during night time (Fig. 5b); (3) depending on chemical conditions, the shape of the hysteresis between F_c and

temperature changed (Fig. 5d); (4) higher pH also increased the asynchrony from the temperature-driven solubilisation dynamics shown in Fig. 4. This is summarized in Fig. 5e where the daily F_c minimum (F_c^{min} i. e., the maximum CO₂ uptake) and its lag compared to the model driven by temperature alone (t_{lag}) are plotted against pH. When the film pH was close to 8 or lower, F_c dynamics could be explained solely by Henry's law (i.e., black dashed lines in Fig. 5a). As the local pH value increased up to

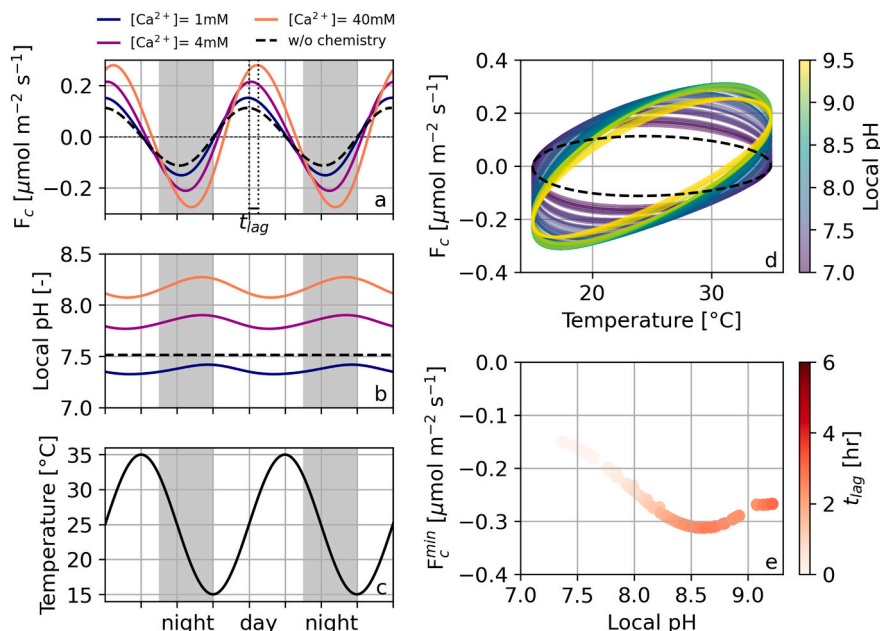


Fig. 5. Modelled effect of calcium concentration and pH on soil-atmosphere F_c during diel temperature cycles. Different Ca^{2+} concentrations were simulated to modulate pH levels at chemical equilibria. Model predictions of (a) the diel dynamics of F_c and (b) local pH dynamics when (c) a temperature forcing ($25 \pm 10^\circ\text{C}$) is prescribed. The grey shading indicates fictitious nocturnal hours with decreasing temperature. (d) Diel hysteresis between F_c and temperature for varying conditions of local pH. The dashed lines in (a), (b), and (d) correspond to the model prediction without kinetics of DIC species (i.e. when only Henry's law is applied). (e) The F_c minimum (F_c^{\min}) and its lag compared to the model driven only by Henry's law (t_{lag}) as a function of local pH.

pH ~ 8.5 , both the magnitude of F_c^{\min} and t_{lag} increased, indicating that CO_2 dissolution was limited with increased pH. The CO_2 uptake reached a maximum at pH 8.5, which corresponds to a transition point between DIC species (i.e., when HCO_3^- starts to decrease and CO_3^{2-} starts to increase). At pH > 8.5 , the magnitude of F_c^{\min} decreased slightly and stabilized around pH > 9 .

2.5. Effect of CO_2 consumption and production

Effects of additional undifferentiated (i.e., biotic or abiotic) CO_2 production and consumption processes other than mineral dissolution in the soil water film were explored by modulating the reaction term $R_{\text{CO}_2}(t)$ in (Eq. (3)). Resulting F_c and film pH are displayed in Fig. 6a and b under the prescribed diel temperature cycle (Fig. 6c). The addition of CO_2 consumption and production at a constant rate changed the magnitude of F_c , inducing a switch from positive to negative values and vice versa (almost constant shifts), and modified t_{lag} in a non-linear manner. The CO_2 consumption in the soil water film increased the pH, thus, triggering uptake of atmospheric CO_2 for the mass balance at the gas-liquid interfacial surface (green lines in Fig. 6a, b). On the other hand, CO_2 production in the soil water film decreased the pH and resulted in CO_2 release from the soil (purple lines in Fig. 6a, b). The diel cycle of local pH showed repositioning of the daily maximum and minimum depending on CO_2 reaction rates caused by shifts in chemical status (Fig. 6b).

2.6. Model prediction of F_c dynamics in physical depositional crusts

By using environmental measurements (Fig. 1b-d), the sfDBM predicted F_c under field conditions. As an example, predicted F_c of physical depositional crusts (where biological activity is assumed to be the lowest) during the selected dry and cool period (marked as period I in Fig. 1) are presented together with field observations (Fig. 7a). Here we present two predictions; (1) the model without chemistry (i.e., in which only Henry's law is considered) and (2) the model including chemical kinetics of DIC species, which captured the F_c diel dynamics more

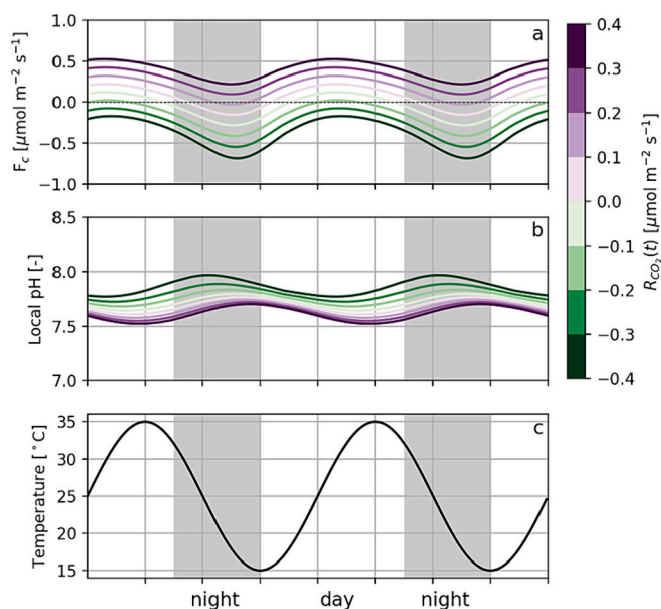


Fig. 6. Effect of CO_2 consumption and production rate ($R_{\text{CO}_2}(t)$) on the soil-atmosphere CO_2 exchange (F_c). (a) F_c and (b) local pH dynamics of the water film under (c) the sinusoidal temperature cycle ($25 \pm 10^\circ\text{C}$) prescribed. The grey shading indicates fictitious nocturnal hours with decreasing temperature. The purple-green colour scale indicates a constant rate of CO_2 consumption (negative $R_{\text{CO}_2}(t)$) or CO_2 production (positive $R_{\text{CO}_2}(t)$) in soil water.

accurately. The patterns and amplitudes of the CO_2 uptake and release were well-captured by the latter model without the necessity to incorporate additional processes of CO_2 consumption or production. The model also predicted the diel dynamics of soil water pH on soil surface and at 5 cm depth based on the temperature input (Fig. 7b). Alkalinization of the medium during nocturnal hours was observed in synchrony with negative F_c .

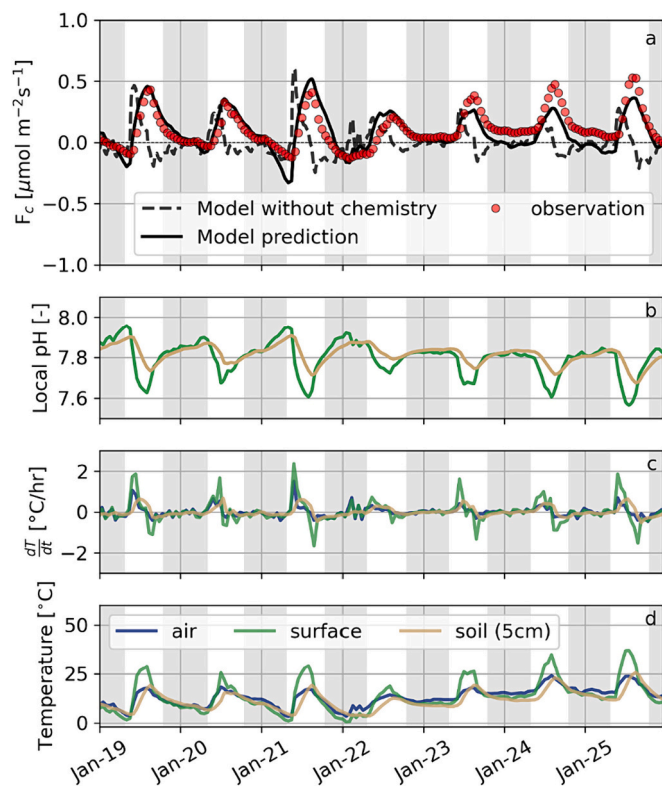


Fig. 7. Model prediction of the soil-atmosphere CO₂ exchange (F_c) for a physical depositional crust during a dry/cool period (period I in Fig. 1). (a) Field observations of F_c (red circles) are compared with predictions by the model without chemistry (dashed black line) and the model including chemistry (solid black line). The grey shading indicates nocturnal hours with low photosynthetically active radiation ($\text{PAR} < 5 \mu\text{mol m}^{-2} \text{s}^{-1}$). (b) Model prediction of the local pH of the soil water film of surface (green), and soil (brown). (c) Measured temporal derivative of temperature of air (blue), surface (green), and soil (brown). (d) Temperature measurements of air (blue), surface (green), and soil (brown), used as input variables in the model.

2.7. Model prediction of abiotically driven F_c along biocrusts succession

By assuming that biological activities are limited during dry periods, we made predictions of abiotically driven F_c during dry/cool – winter and dry/hot – summer periods for different biocrusts along succession ($R_{\text{CO}_2}(t) = 0$). Results are shown in comparison with the measured F_c for each crust type (Fig. 8). Overall, the model predictions of abiotically driven F_c are similar to F_c measurements. The timing of maximum CO₂ release/uptake matches the model predictions, especially for early stage biocrusts during the dry and cool periods. However, the model underestimated daytime CO₂ release during the dry and hot period in early successional stages (PD and IC) and showed the biggest discrepancy under dry and cool conditions for the SD lichen crusts, when CO₂ uptake was predicted but not observed.

2.8. Potential respiration by biocrusts under dry and cool conditions

According to the model prediction of abiotically driven F_c from different biocrusts, we detected a large deviation from the field observations for the SD crusts under dry and cool conditions (Fig. 8). In Fig. 9, we show the effect of possible additional CO₂ consumption and production by SD crusts on the resulting F_c dynamics. Using the model outcome with varying R_{CO_2} values (Fig. 9a), we further estimated the rates of additional CO₂ reaction by minimizing the error between the observations and predictions (Fig. 9b) under the given temperature and relative humidity (Fig. 9c). The estimated respiration rates were high

when the conditions of PSC ($T_s \leq T_{\text{dew point}}$) met. By assuming that estimated $R_{\text{CO}_2}(t)$ mainly represented the biological activities, we plotted the diel patterns of decomposed biotic (estimated $R_{\text{CO}_2}(t)$) and abiotic (observed $F_c(t) - R_{\text{CO}_2}(t)$) contributions on F_c compared to the model prediction of abiotically driven F_c when $R_{\text{CO}_2} = 0$ (Fig. 9d). The diel patterns show that biological CO₂ production offset the abiotic CO₂ uptake as it mainly occurred during nighttime. The model prediction of abiotic processes well-aligns with the estimated values, although the modelled uptake rate was slightly lower than the estimated rate during nighttime owing to the chemical effects. In addition, estimated $R_{\text{CO}_2}(t)$, the potential respiration, is plotted as a function of T_s and RH_s (Fig. 9e) indicating that respiration was enhanced by the combination of high RH_s and low T_s conditions that were favourable to PSC during this dry period.

3. Discussion

For dryland soils covered with biocrusts, one expects daytime CO₂ uptake and nighttime CO₂ emission to occur due to biocrust photosynthesis and respiration as long as the biocrusts are moist (Tamm et al., 2018; Weber et al., 2018). However, using continuous measurements of the soil-atmosphere CO₂ flux (F_c) along an ecological succession of biocrusts in calcareous soils, we found that biocrusts at early stages (cyanobacteria) exhibited a substantial nocturnal CO₂ uptake under dry conditions (Fig. 2, Fig. 3, Fig. S2, Fig. S3). Only two crust types showed a net daily CO₂ uptake during contrasting periods: mature cyanobacterial crusts (MC) showed a net uptake under dry and cool conditions, whereas lichen crusts dominated by *Squamarina lentigera* and *Diploschistes diacapsis* (SD) showed it under dry and hot conditions (Fig. 3 c and f).

Although many dryland soils have been reported to nocturnally absorb CO₂ (Sagi et al., 2021), to our knowledge only one study from our site has reported annual estimates of CO₂ uptake based on in situ continuous measurements at the soil level (Lopez-Canfin et al., 2022b, 2024). This data is particularly valuable, since reports of larger annual rates obtained by the eddy-covariance measurements have received sharp criticism which questioned the role of abiotic CO₂ uptake in the global C cycle (Schlesinger, 2017; Schlesinger et al., 2009). According to the two-year dataset used in our previous study, annual CO₂ uptake was $-17 \pm 15 \text{ g C m}^{-2} \text{ y}^{-1}$ (mean \pm spatial standard deviation associated to biocrust succession stages) with a maximum uptake of $55 \text{ g C m}^{-2} \text{ y}^{-1}$. Interestingly, our mean estimate is very close to the uptake flux of $16 \text{ g C m}^{-2} \text{ y}^{-1}$ reported for silicate weathering in a temperate forest (Andrews and Schlesinger, 2001) and close to the upper estimate of calcite precipitation ($12.4 \text{ g C m}^{-2} \text{ yr}^{-1}$) reported for drylands (Lal, 2001). Upscaling our average and maximum values to the total surface of drylands (i.e. $66.7 \times 10^6 \text{ km}^2$) results in an estimated average and maximum C uptake of ~ 1.1 and 3.7 Pg y^{-1} , respectively, by dryland soils. The magnitude of these values is comparable with the most recent estimate of the missing/residual terrestrial C sink ($3.1 \pm 0.9 \text{ Pg y}^{-1}$) (Houghton et al., 2018). Therefore, even after considering that dryland surfaces absorb C at heterogeneous rates and that biocrust coverages are uncertain, it remains reasonable to assume that the nocturnal CO₂ uptake by dryland soils can significantly contribute to balance the global C budget. The potentially overlooked role of non-rainfall water inputs to sustain abiotic CO₂ consumption processes in drylands will be discussed later.

Using a mechanistic model with a heuristic approach (i.e., simplified boundary conditions), we highlighted the pivotal role of temperature cycles in determining the magnitude and temporal patterns of CO₂ release/uptake under dry conditions (Fig. 4). Due to the temperature-dependency of CO₂ solubility (Henry's law) in the water film on soil particles, atmospheric CO₂ uptake was predicted when $dT_{\text{soil}}/dt < 0$ subject to temporal increase in CO₂ solubility. Several studies have already pointed out dT_{soil}/dt as an important driver of the CO₂ uptake based on statistical approaches (Sagi et al., 2021). The model in this

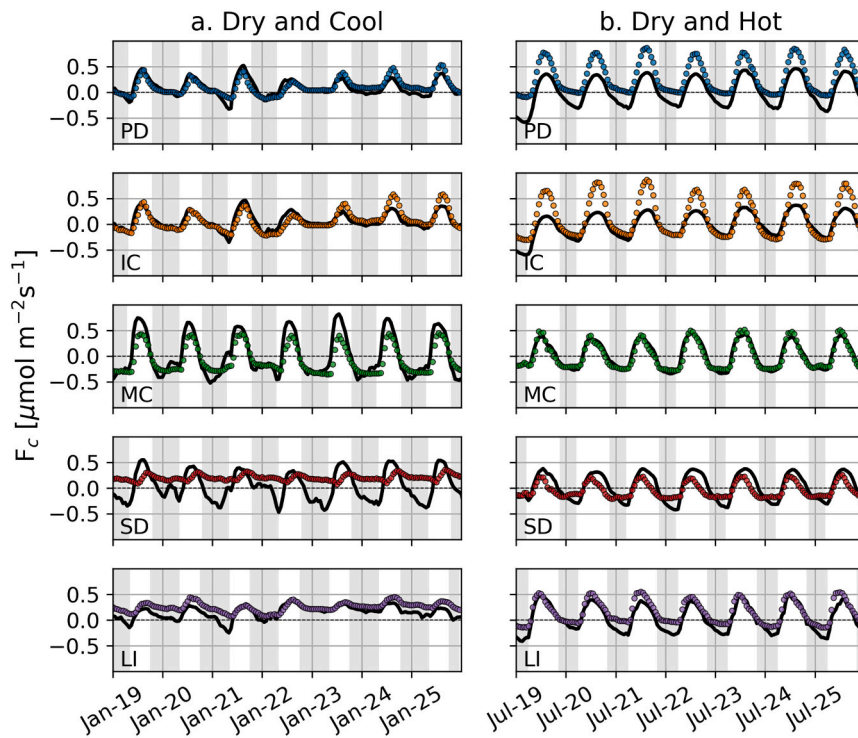


Fig. 8. Model predictions of abiotically driven soil-atmosphere CO_2 exchange (F_c) of different biocrusts during two dry periods with contrasting temperature. Black solid lines indicate the model predictions when $R_{\text{CO}_2} = 0$ and circles indicate measurements. The grey shading indicates nocturnal hours with low photosynthetically active radiation ($\text{PAR} < 5 \mu\text{mol m}^{-2} \text{s}^{-1}$). (a) F_c under dry and cool conditions (period I in Fig. 1) (b). F_c under dry and hot conditions (period II in Fig. 1). The predictions are compared to field measurements averaged over replicates for each biocrust type. From top to bottom, results are presented along successional stages; PD, physical depositional crust; IC, incipient cyanobacteria; MC, mature cyanobacteria; SD, lichen community dominated by *Squamarina lentigera* and *Diploschistes diacapsis*; and LI, lichen community characterized by *Lepraria isidiata*.

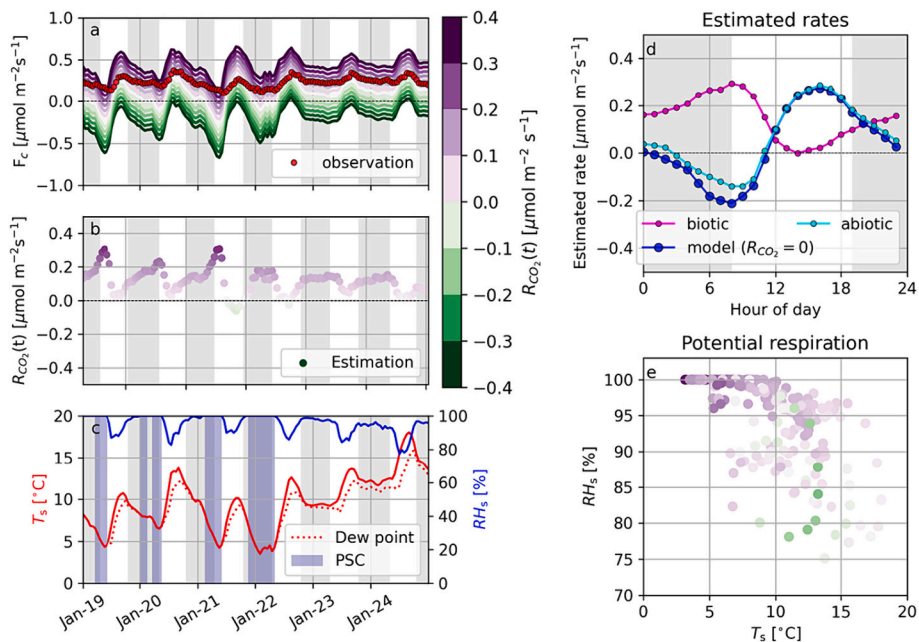


Fig. 9. Effects of CO_2 consumption and production in a lichen crust community dominated by *Squamarina lentigera* and *Diploschistes diacapsis* (SD) under dry/cool conditions. (a) Model prediction of the soil-atmosphere CO_2 exchange (F_c) with varying biological contribution $R_{\text{CO}_2}(t)$. The purple-green colour scale indicates the constant rate of CO_2 consumption (negative $R_{\text{CO}_2}(t)$) or CO_2 production (positive $R_{\text{CO}_2}(t)$) in soil water. (b) Estimated $R_{\text{CO}_2}(t)$ values by minimizing the error between the observations and predictions. (c) Diel cycle of soil temperature (T_s) and relative humidity in soil (RH_s) used as input in the model. The dotted red line indicates the estimated dew point at 5 cm depth and the navy shading indicates the time windows of possible pore-space condensation (PSC). (d) Diel patterns of estimated biotic (magenta) and abiotic (cyan) effects on CO_2 dynamics. The blue line indicates the model prediction when $R_{\text{CO}_2}(t) = 0$. (e) Estimation of potential respiration by the SD crust as a function of T_s and RH_s . The purple-green circles indicate the estimated $R_{\text{CO}_2}(t)$ following the same scale as the figure (a) and (b). The grey shading indicates nocturnal hours with low photosynthetically active radiation ($\text{PAR} < 5 \mu\text{mol m}^{-2} \text{s}^{-1}$).

study supports the previous finding with further implications. Here, we show that the diel cycle of temperature (T_{avg} and ΔT) determines the magnitude of F_c exhibiting a diel hysteresis loop, and a pinched hysteresis loop with dT/dt (Fig. 4). The hysteresis with F_c lagging after temperature has been explained to be caused by both biological (e.g., transport of photosynthates to the soil) and physical processes (e.g., thermal convection and CO_2 diffusion) (Dusza et al., 2020; Fa et al., 2016b; Hamerlynck et al., 2013; Phillips et al., 2011; Riveros-Iregui et al., 2007; Song et al., 2015; Zhang et al., 2015). Here, we showed that under dry conditions the temperature cycle alone can cause a lag of the CO_2 uptake and release due to its influence on the CO_2 solubility. In addition, the pinched hysteresis between F_c and dT/dt reveals that CO_2 dissolution in the soil water film occurs at a slower rate than CO_2 degassing in the model.

The magnitude of the CO_2 uptake as well as t_{lag} increased under near-basic to high pH values (Fig. 5). At $pH > 8.5$, most of the dissolved CO_2 converts to HCO_3^- or CO_3^{2-} species in the water film, allowing more CO_2 to dissolve under Henry's law. Furthermore, the diel cycle of pH in the soil water film was predicted in the model under the assumption of charge balance (Fig. 5b) that are usually not considered in most soil models, where pH values are assumed to be constant (Pumpanen et al., 2003; Šimůnek and Suarez, 1993; Taylor et al., 2017). Our approach of tracking the temporal variation of pH may provide insight into time windows of certain chemical/biological reactions. For instance, alkalinization at decreasing temperatures during nighttime (Fig. 5b) may support the formation of HCO_3^- and CO_3^{2-} , inducing calcite precipitation and triggering positive feedback for CO_2 uptake. This is in agreement with several recent studies which report a rapid translocation of C from atmospheric CO_2 to pedogenic carbonates in drylands based on isotopic measurements (Carmi et al., 2019; Liu et al., 2015, 2018, 2020; Wang et al., 2016). However, there is a common conception in the literature according to which $CaCO_3$ dissolution constitutes a CO_2 sink while $CaCO_3$ precipitation represents a CO_2 source to the atmosphere (Gallagher and Breecker, 2020; Ortiz et al., 2022; Soper et al., 2017) which can seem contradictory to these results. Though this perspective is true in terms of instantaneous gas exchange of the mineral reactions alone, it is less mentioned that $CaCO_3$ dissolution/precipitation are associated with an immediate loss/storage of stable mineral C, respectively. In terms of short-term net C balance, $CaCO_3$ precipitation captures one mole of C from dissolved CO_2 that may have an atmospheric origin while $CaCO_3$ dissolution releases on mole of C as dissolved CO_2 . To correctly assess the net C balance of these reactions over longer periods, transient storage of dissolved CO_2 in groundwater, and the presence of other readily available sources of Ca^{2+} need to be considered. Briefly, whether calcite precipitation constitutes a net C sink ultimately depends on the origin of Ca^{2+} , with only Ca^{2+} inherited from an external source (other than $CaCO_3$, i.e. Ca-bearing minerals like gypsum and silicates, irrigation water or dust) leading to net C sequestration (Monger et al., 2015; Sanderman, 2012). At our study site where soil contains gypsum, CO_2 uptake could be driven by coupled gypsum dissolution- calcite precipitation (Yu et al., 2019), although more empirical evidence is necessary for validation.

When additional CO_2 reactions (biotic or abiotic) were considered, the model showed non-linear shifts of F_c dynamics as related to magnitude and temporal asynchrony with temperature (Fig. 6). Overall, CO_2 consumption in the soil water film by a third agent, such as dark microbial CO_2 fixation (Braun et al., 2021; Spohn et al., 2020) induces alkalinization, thus leading to an enhancement of the CO_2 uptake. In contrast, CO_2 production (e.g., biological respiration) leads to acidification and near-constant shifts in F_c dynamics towards CO_2 release (Fig. 6b). Most F_c research has focused on soil CO_2 production and assumed soil CO_2 consumption processes as being negligible. Under close to neutral/acidic conditions, considering only the effect of abiotic and biotic CO_2 production may be reasonable because the effects of DIC fractionation and mineralization are minimized. However, for basic conditions like those of dryland alkaline soils, contributions of abiotic

and biotic CO_2 reactions to F_c dynamics do not obey a simple additive relation. This is why detailed reaction dynamics in space and time need to be analysed and an adequate upscaling strategy of soil processes with highly heterogeneous water availability, biological activity and mineral distributions needs to be developed.

Investigating F_c dynamics over the ecological succession of biocrusts yielded the unexpected outcome that biological activities and their masking effects on the abiotic CO_2 uptake under dry conditions are not linearly related with biocrust successional stages with increasing SOC content (Figs. 7, 8, 9, and Fig. S5). Other factors such as soil pH (or amount of cations) and/or other physical properties affect the magnitude and diel patterns of F_c dynamics under various conditions as shown in the model. We indeed found similar patterns with soil pH, and also the relative fraction of cyanobacteria (Miralles et al., 2020) in the microbial communities (Fig. S5). Under a diel cycle of temperature, the optimum soil pH for CO_2 uptake is around 8.5 (Fig. 5e) which is associated with enhanced carbonate precipitation and coincides with the pH of mature cyanobacterial crusts (MC) (Fig. S5) that exhibited the highest nocturnal CO_2 uptake over the whole succession (Fig. 3 and Fig. S3). In fact, Cyanobacteria might have enhanced this uptake by inducing this pH alteration. They are known to have a CO_2 concentrating mechanism (CCM) that induces alkalization by releasing OH^- into the surrounding medium during photosynthesis (Büdel et al., 2004; Garcia-Pichel and Belnap, 1996). In addition, they have also been suggested to promote gypsum dissolution to extract its water of crystallization (Huang et al., 2020), potentially fostering $CaCO_3$ precipitation through Ca^{2+} release by weathering of gypsum or Ca-silicates (Jansson and Northen, 2010). It is therefore reasonable to expect cyanobacterial contributions to $CaCO_3$ accumulation in biocrust soils enhancing CO_2 uptake.

The net CO_2 uptake by biocrusts was greatly reduced in lichenized late successional stages (SD and LI) compared to early stages (IC and MC) (Fig. 2). As larger organisms like lichens or mosses develop in late stages of the succession, the abundance of cyanobacteria decreases while SOC increases (Miralles et al., 2020). The activity of green algal lichens is known to lower the pH of surroundings by secreting organic acids, and to increase SOC content (Ghiloufi et al., 2023), hence inhibiting the abiotic processes of CO_2 dissolution and precipitation of carbonates. Similarly, the CO_2 uptake and release by soil were also lower and higher, respectively, in the physical depositional crusts compared to cyanobacterial crusts (IC and MC) (Fig. 2). While the reduced CO_2 uptake can be attributed to the lowest pH under the influence of the lowest calcium carbonate contents, the substantial CO_2 release (Fig. 1b) is likely due to greater T_s fluctuations enhancing non-diffusive CO_2 transport by thermal convection (Ganot et al., 2014; Phillips et al., 2011; Roland et al., 2013; Spohn and Holzheu, 2021).

Some studies using chamber measurements have also attributed the CO_2 uptake by dryland soils to thermal convection and expansion/contraction of soil air (Fa et al., 2016b; Gao et al., 2021; Yang et al., 2020). However, although we acknowledge that such processes can play a significant role in CO_2 transport within the soil-atmosphere continuum, here we postulate that they alone cannot explain the observed CO_2 uptake as it arose from the CO_2 molar fraction (χ_c) of soil dropping below atmospheric χ_c values. Therefore, this necessarily involves CO_2 consumption, not only CO_2 transport which would at best equilibrate soil χ_c with atmosphere χ_c . Although our model does not include thermal convective venting, most of the F_c variation was explained by the temperature- and pH-driven dissolution dynamics of CO_2 in the top soil.

This study focuses on the mechanisms of nocturnal CO_2 uptake under dry conditions. With limited water availability for biological activity, F_c dynamics could be explained mostly by abiotic processes like in physical depositional crusts (Figs. 7 and 8). This agrees with the previous finding that biocrust respiration is close to zero during summer in response to the extreme drought and thermal stress (Miralles et al., 2018). Therefore, differences in F_c dynamics among crust types during drought were related to edaphic properties and environmental conditions. However, the model revealed that SD crusts during wintertime (dry/cool

conditions) were able to maintain biological respiration under high RH_s through possible pore-space condensation (PSC) (Fig. 9). Although the water content was low during this period ($0 \sim 0.06 \text{ m}^3 \text{ m}^{-3}$), biological activity could be temporally enabled by non-rainfall water input such as PSC, since soil air was regularly close to saturation with respect to water vapor ($RH_s \sim 100 \%$) with temperature meeting the condition for PSC, in contrast to the dry and hot summer periods when RH_s values were frequently below 50 % and temperature was too high to induce PSC (Fig. 1b and c). This supports the finding that lichens, especially of *D. diacapsis* studied here, can regain metabolic activity even at low water contents in presence of high relative humidity (Baldauf et al., 2021; Richardson, 2002).

In the model, we use the effective water film thickness as the hydrological basis that enables chemical reactions and biological activity even under relatively dry conditions. The effective water film thickness is estimated from measured water contents at daily scale. Furthermore, in this study, we assume the homogeneity of physical structures for simplified upscaling (fixed soil physical parameter, Table S5). However, biocrusts create a discontinuous boundary at the interface between soil and atmosphere, as well as micro-scale heterogeneity in terms of soil porosity, texture, and distribution of biological and chemical substances (Kim and Or, 2017, 2019; Kratz et al., 2022). This structural effect becomes crucial under wet conditions, especially owing to the swelling-shrinking dynamics of extracellular polymeric substances in the biocrusts influencing the properties of liquid-gas transport (Kidron et al., 2020; Sun et al., 2022, 2023). However, under dry conditions near the surface, gas transport may not be limited by structural obstacles, and can even be enhanced due to crust-cracking, which possibly caused the increased F_c during summertime in PD and IC crusts (Fig. 8b). Therefore, we argue that, under dry conditions, our approach using the effective water film thickness was reasonable to estimate F_c and yielded good predictions.

The notion of water film thickness can be further extended by including adsorption and condensation processes under the mass conservation rule of water vapor within soils (Tuller et al., 1999). The adsorption of atmospheric water vapor by soil has been shown to potentially drive the nocturnal CO_2 uptake in summer at this site (Lopez-Canfin et al., 2022a). Here, we additionally show that outside of summer periods, whereas surface conditions of temperature and relative humidity at the soil-atmosphere interface are often unfavourable to dew formation (Agam and Berliner, 2006; Lopez-Canfin et al., 2022a), these conditions in the soil below crusts were frequently favourable to PSC at night (Fig. S4). Especially after major rain events, we observe that below-crust soils sustained high RH_s allowing condensation to occur within soil pores and the conditions could last days or even weeks depending on precipitation and crust types (Fig. 1c). This implies that biocrusts may contribute to the prolonged storage of rain and its possible usage to sustain biological activity in drylands even at the small scale, which would mirror the processes at landscape scale (Baldauf et al., 2023; Eldridge et al., 2020). We argue that due to the immense surface area of inner soils (especially in fine-textured soils) compared to the soil surface, the water storage effect through PSC can be substantial in drylands. By contrast, it has been stated that the lack of water in drylands would prevent CO_2 uptake associated with the dynamics of Ca-bearing minerals (Schlesinger et al., 2009). However, after considering the contribution of overlooked non-rainfall water inputs (PSC and water vapor adsorption), our annual estimates of CO_2 uptake may be compatible with rates of CO_2 -consuming mineral reactions. Hence, further research is necessary to provide quantitative estimates of water vapor adsorption and PSC and their coupling with many ecosystem functions in drylands (Wang et al., 2017). This is of major importance as climate change is expected to affect the duration and frequency of wet conditions as well to increase the extension of dry conditions in time and space (Grünzweig et al., 2022).

4. Conclusions

The objective of this research was to unravel the main mechanism responsible for the nocturnal CO_2 uptake by soil and biocrusts in drylands using a mechanistic model. We focused on dry periods, during which biological activity is minimized but a thin water film remains on the surface of soil particles possibly sustained by water vapor adsorption and pore-space condensation. Under these conditions, abiotic processes alone can predict most of the nocturnal CO_2 uptake; in particular, dissolution/degassing of CO_2 driven by temperature dynamics in combination with diel pH variations in the soil water film. The alkalization of the water film at night, with pH conditions >8.5 , is particularly favourable for CO_2 uptake and calcite precipitation in agreement with the maximum CO_2 uptake measured in mature cyanobacterial crusts. By consuming or producing CO_2 and modifying the pH of the soil water film, biocrusts further modulate carbon dynamics. The coupling between water and inorganic C cycling in drylands has been largely understudied compared to mesic ecosystems, and therefore further research is required to better constrain their rates and confirm that the mechanism of CO_2 uptake unravelled in this study is widespread. Expecting changes in precipitation, temperature, and disturbance regimes globally, it is becoming increasingly important to comprehend and forecast the soil-atmosphere CO_2 exchange capacity of drylands, especially regarding biocrusts as the primary carbon sequester. By elucidating temperature-water-carbon processes within biocrusts, this study provides a missing puzzle piece of nocturnal CO_2 uptake and emphasizes the interplay between abiotic processes and soil biological agents mitigating CO_2 emissions in the face of changing climates.

5. Material and methods

5.1. Experimental site

The measurements were conducted in the experimental site of El Cautivo (N $37^\circ 00' 37''$, W $2^\circ 26' 30''$, $\sim 250 \text{ m a.s.l.}$), an area of badlands located in the Tabernas Desert. The climate is classified as semi-arid thermo-Mediterranean with a mean annual temperature of 18°C and mean annual precipitation of 230 mm. The soil is classified as a silty-loam with $>60 \%$ gypsum-calcareous and siliceous particles of silt-size, 20–30 % fine sand, 5–10 % clay. The calcium carbonate equivalent content ranges from 11 % to 28 % and gypsum content is highly heterogeneous between 0.1 % to 35 %. A detailed pedological characterization of the soil horizons can be found in Cantón et al., 2003 and soil properties specific to the 0–5 cm horizon are detailed in Lopez-Canfin et al., 2022b.

A third of the territory is bare and eroded soil and another third is covered by short vascular vegetation with biocrusts in the interspaces. The rest is covered mainly by biocrusts. Five different biocrust types can be distinguished and assigned to stages of ecological succession (Lázaro et al., 2008): (1) physical depositional crust (PD); (2) incipient cyanobacteria-dominated crust (IC); (3) mature cyanobacteria-dominated crust (MC); (4) lichen-dominated crust with *Squamaria lentigera* (Web.) Poelt and *Diploschistes diacapsis* (Ach.) Lumbsch as key species (SD); and (5) lichen-dominated crust with *Lepraria isidiata* (Llimona) Llimona and Crespo as key species (LI). Lichen-dominated stages (SD and LI) were considered as late successional stages while previous stages were considered as early successional stages (PD, IC and MC) (Fig. S1).

5.2. Environmental measurements

Continuous measurements of soil and above-surface variables were conducted from February 2018 to December 2019. At 5 cm depth, the CO_2 molar fraction (χ_c) was measured by GMP252 and GMM222 probes (total of 14 and 3 probes, respectively) (Vaisala, Vantaa, Finland; hereafter Vaisala). All GMP252 sensors were new and GMM222 sensors

were calibrated for CO₂ just before installation by using the GMK220 carbon dioxide calibrator (Vaisala) and a two-points calibration curve: using (variable but known) ≈ 350 ppm and ≈ 1500 ppm for spans. The soil water content (θ) was measured at 5 cm depth, below crusts, by EC-5 and 5TM sensors (Meter Group, Pullman, WA). The θ sensors were calibrated using the manufacturer's general equation for mineral soils. All mineral soils up to an electrical conductivity (EC) of 8 dS m⁻¹ in saturation extract should be covered by this equation, according to the manufacturer's instructions. In 1:5 soil-to-water ratios, preliminary testing revealed that most soils at our study site had an EC < 1 dS m⁻¹. Based on those values, it was reasonable to expect that EC in saturation extracts would not exceed the recommended threshold (Kargas et al., 2020). The 5TM sensors measured the soil temperature (T_s) that was also measured with thermistors (108, Campbell Scientific, Logan, UT, USA; hereafter CSI). All these measurements were performed in the different biocrust types in triplicates. Since mature cyanobacterial patches were also present within the SD site, for comparison purpose, measurements were also performed for this MC crusts within the SD site in duplicate, thus results in five MC replicates. Within each biocrust type, the relative humidity at 5 cm depth and in the atmosphere directly in contact with soil (RH_s and RH_a , respectively) were measured with iButton® DS1923 loggers (Maxim Integrated, San Jose, CA, USA); air temperature (T_a) and relative humidity at 30 cm aboveground were measured by a S-THB-M006 Smart Sensor (Onset Computer Corporation, Bourne, MA, USA; hereafter Onset); surface temperature (T_{surf}) was measured by a S-TMB-M0xx Smart Sensor (Onset) and the photosynthetically active radiation (PAR) by a S-LIA-M003 Smart Sensor (Onset); precipitation was measured by a Rain-O-Matic-Pro tipping-bucket rain gauge of 0.25 mm resolution (Pronamic, Ringkoebing, Denmark) at 1.5 m above the ground surface. At 2 cm aboveground, atmosphere χ_c was monitored in the PD and SD sites by GMP252 probes (Vaisala), and atmospheric temperature and pressure was monitored in the LI site by a SPIA air pressure transmitter (PM Ecology, Gdynia, Poland). The rain gauge was connected to an on-off Hobo Event data logger (Onset) and all other variables were measured every 30 s and stored as 20-min averages by data-loggers CR1000 (CSI) and H21 (Onset). More details about this experimental design can be found in Lopez-Canfin et al., 2022a.

5.3. CO₂ flux calculation and gap-filling

Measurements in different time resolutions were harmonized to hourly averages. Then, the whole dataset was split into subsets of variables of the same type (e.g. temperatures) and each subset was imputed separately to avoid generating spurious correlations. A non-parametric method based on random forests was chosen to calculate missing values (Stekhoven and Bühlmann, 2012), as the method can deal with mixed-type data and is particularly effective in handling complex interactions and non-linear data structures.

The soil-atmosphere CO₂ fluxes (F_c) were estimated from χ_c using the gradient method based on Fick's first law of molecular diffusion: $F_c = -\rho_a k_s \frac{d\chi}{dz}$ where F_c is the soil-atmosphere flux ($\mu\text{mol m}^{-2} \text{s}^{-1}$), ρ_a is the average molar density of air (mol m^{-3}), $d\chi$ is the difference in CO₂ molar fraction ($\mu\text{mol mol}^{-1}$) between atmosphere and soil, and dz is the vertical difference between atmosphere and soil (m), k_s is the CO₂ diffusion coefficient or empirical soil transfer coefficient ($\text{m}^2 \text{s}^{-1}$), calibrated empirically from flux chamber measurements as previously recommended (Sánchez-Cañete et al., 2017). Negative F_c values correspond to CO₂ uptake from the atmosphere to the soil (i.e., influx) whereas positive fluxes correspond to CO₂ release from the soil to the atmosphere (i.e., efflux). Further details about the flux estimation are available in Lopez-Canfin et al. (2022b, 2024).

5.4. Classification of wet/dry and cold/hot periods

We split the dataset in wet/dry and cool/hot periods for subsequent

data analysis. Based on daily averaged values of θ and T_a during the whole measurement period (from February 2018 to December 2019), each day was classified as wet or dry and cold or hot. To perform these classifications, θ and T_a thresholds were selected based on the principle of minimizing cross-entropy. For the automated threshold selection, we used an algorithm that does not make any a priori assumption on the underlying distribution of values (Li and Lee, 1993). The daily classification was performed for each site to consider the effect of microclimate on biotic and abiotic processes. The result of the classification process defines four distinctive periods: (a) dry and cool, (b) wet and cool, (c) dry and hot, and (d) wet and hot. Based on this output, we selected two dry sub-periods with contrasting temperature to compare the field observations and the model predictions of the F_c time series; (I) dry and cool (from 2019-01-18 to 2019-01-25, $\theta = 0.05 \pm 0.03 \text{ m}^3 \text{ m}^{-3}$; $T_s = 11.08 \pm 3.97 \text{ }^\circ\text{C}$; $RH_s = 95.61 \pm 5.82 \%$); (II) dry and hot (from 2019-07-18 to 2019-07-25, $\theta = 0.04 \pm 0.02 \text{ m}^3 \text{ m}^{-3}$, $T_s = 33.50 \pm 6.92 \text{ }^\circ\text{C}$; $RH_s = 49.36 \pm 7.93 \%$). These two periods are marked as grey shaded area in Fig. 1.

5.5. Mechanistic modelling of CO₂ fluxes from biocrusts

The influence of abiotic and biotic processes on F_c was explored mechanistically by using the desert biocrust model (DBM) (Kim and Or, 2017, 2019). The DBM simulates biophysical and chemical processes in biocrusts (including biotic and abiotic C and N-cycling) under dynamic conditions of hydration, light, and temperature. In this study, the individual-based description of microbial activity was replaced by a population-based description to elucidate the chemical processes that are influenced by biological processes. Furthermore, we included a module of gypsum dissolution and calcite precipitation dynamics.

In this study, we investigated the roles of temperature dynamics on F_c under field conditions. Unlike the previous release of the model that simulated pre-assigned laboratory conditions of temperature and water dynamics (Kim and Or, 2017, 2019), we used environmental measurements as input variables for boundary conditions; soil water content (θ), air-surface-soil temperature (T_a , T_{surf} , T_s) for the temperature profile, and atmospheric CO₂ molar fraction (χ_c). The porosity and soil pH values were also used to set initial conditions of the physical/chemical soil domain at the interface with atmospheric conditions. The detailed model description is provided in the supplementary information. The modified DBM in this study is named as the simplified field DBM (sfDBM) and the script written in Python 3 is archived on Zenodo (Kim, 2024).

Although a realistic empirical input is required to ultimately test the model prediction of F_c , the interactive effect of many variables complexifies interpretation under field conditions. Therefore, in order to understand the mechanisms and to disentangle tightly coupled processes involved in the CO₂ uptake, we first used a heuristic approach, which is a simplified learning method to test the effect of a reduced set of variables controlled by model simulations. Following this approach, simplified boundary conditions of temperature were used to evaluate the effect of temperature. To this end, a diel temperature cycle was simulated with a periodic function at a fixed hydration condition (a dry condition equivalent to $\theta \sim 0.05 \text{ m}^3 \text{ m}^{-3}$, which is about 0.03 g g^{-1}). Here, we used a sinusoidal function to simulate a diel cycle of air temperature as follows:

$$T(t) = T_{avg} + \Delta T \sin \left[\frac{2\pi}{24} (t - 6) \right], \quad (1)$$

where T_{avg} is the mean and ΔT is the amplitude of a diel temperature cycle [$^\circ\text{C}$], t is the time in the unit of hour [hr]. With this function, the temperature minimum $T_{avg} - \Delta T$ occurs at $t = 0$ and the maximum $T_{avg} + \Delta T$ at $t = 12$ [hr] (Fig. 4c). Regarding the chemical effect of inorganic C partitioning, the Ca²⁺ content in the water film was used as a representative control parameter of pH. Other input

parameters used in this work are listed in the supplementary information.

The essential process in this model is the solubility of CO₂ which is a function of the temperature dependent solubility (Sander, 2015);

$$H_{cc}(T(t)) = H_{cc}^{\Theta} e^{-\frac{\Delta_{\text{soln}}H}{R} \left(\frac{1}{T(t)+273.15} - \frac{1}{T^{\Theta}} \right)}, \quad (2)$$

where $H_{cc}^{\Theta} \equiv k_H RT^{\Theta}$ is the gas-specific (here, e.g., CO₂) dimensionless Henry's constant. The superscript Θ refers to the standard temperature condition at $T^{\Theta} = 298.15\text{K}$, $\Delta_{\text{soln}}H$ is the enthalpy, and R is the gas constant. According to this Arrhenius formular, the solubility increases with decreasing temperature.

$$C^{l,*}(\vec{r}, t) = H_{cc}(T(t))C^g(\vec{r}, t), \quad (3)$$

where $C^{l,*}(\vec{r}, t)$ is the concentration of dissolved CO₂ in soil water at equilibrium and $C^g(\vec{r}, t)$ is the concentration of gaseous CO₂ at a position \vec{r} at time t based on the temperature dependent air-density $\rho^g(T(t))$ and the partial pressure of CO₂, $p_{\text{CO}_2}(t)$, resulting $C^g(\vec{r}, t) = \rho^g(T(\vec{r}, t))p_{\text{CO}_2}(t)$. Here, the temperature-dependent air density includes the effect of thermal expansion and contraction. We use observed partial pressure of CO₂ in the air and in soil at 5 cm depth as model inputs and the thermal convection from deeper soils are not considered in this work. The concentration of dissolved CO₂, $C^{l,*}(\vec{r}, t)$, can be calculated based on the diffusion-convection-reaction equation with the mass transfer term based on the Henry's law:

$$\frac{\partial C^l(\vec{r}, t)}{\partial t} = \frac{dC^l(\vec{r}, t)_{\text{degassing}}}{dt} + \frac{dC^l(\vec{r}, t)_{\text{diffusion}}}{dt} + \frac{dC^l(\vec{r}, t)_{\text{convection}}}{dt} + \frac{dC^l(\vec{r}, t)_{\text{reaction}}}{dt}$$

$$= -k_L a(C^l(\vec{r}, t) - C^{l,*}(\vec{r}, t)) + \nabla \cdot (D(\vec{r}, t) \nabla C^l(\vec{r}, t)) - \nabla \cdot (\vec{v} C^l(\vec{r}, t)) + R(t)$$

$$\approx -k_L a(C^l(\vec{r}, t) - H_{cc}(T(t))C^g(\vec{r}, t)) + R_{\text{chem}}(t) + R_{\text{CO}_2}(t), \quad (4)$$

where k_L is the mass-transfer coefficient and a is the gas-liquid interfacial surface area and $D(\vec{r}, t)$ is a diffusion coefficient which is a function of position and time. However, we assume that on the top soil $k_L a \gg D(\vec{r}, t)$ and the convection on the liquid phase is negligible under dry conditions, thus Eq. (4) is used for calculating $C^l(\vec{r}, t)$. Here, the reaction terms are described with two factors; $R(t) \approx R_{\text{chem}}(t) + R_{\text{CO}_2}(t)$ where $R_{\text{chem}}(t)$ indicates the chemical reactions under the explicit kinetics with the charge conservation constraint and $R_{\text{CO}_2}(t)$ is an additional source/sink of CO₂. Since the major chemical contributions are already included in $R_{\text{chem}}(t)$, $R_{\text{CO}_2}(t)$ can be interpreted as biological contributions such as respiration (positive) or fixation (negative). $R_{\text{CO}_2}(t)$ can vary in time, however, in our heuristic approach, we assumed the reaction rates are constant to investigate the roles of additional production or consumption of CO₂.

By assuming that the gas-liquid partitioning of gases under Henry's law, concentration of CO₂, in liquid and gas phases, $C^l(\vec{r}, t)$ and $C^g(\vec{r}, t)$ respectively, can be calculated at each time step. Between the interval Δt , the amount of degassed CO₂ (loss from the liquid phase calculated based on Eqs. (3) and (4) under the principle of mass conservation) is defined as the gaseous efflux of CO₂, $F_c(\vec{r}, t)$, which is calculated as follows:

$$F_c(\vec{r}, t) = \frac{1}{\Delta t} \int_t^{t+\Delta t} \frac{dC^g(\vec{r}, \tau)_{\text{degassing}}}{d\tau} + \frac{dC^g(\vec{r}, \tau)_{\text{diffusion}}}{d\tau} + \frac{dC^g(\vec{r}, \tau)_{\text{convection}}}{d\tau} d\tau$$

$$\approx \frac{1}{\Delta t} \int_t^{t+\Delta t} \frac{dC^g(\vec{r}, \tau)_{\text{degassing}}}{d\tau} d\tau$$

$$F_c(t) = \iint k_L a (C^l(\vec{r}, t) - H_{cc}(T(t)) \rho^g(T(\vec{r}, t)) p_{\text{CO}_2}(t)) dt d\vec{r} \quad (5)$$

In this work, in the calculation of the gaseous concentration of CO₂, convective flow and diffusion is ignored by assuming the dominant process as the degassing by temperature-dependent solubility dynamics. However, by including the temperature dependent air density, thermal expansion and related dilution effect is included indirectly in the model. Details of kinetics included in the model are described in the previous publications and supplementary information (Kim and Or, 2017, 2019). The script of sfDBM in Python 3 is available in Github (https://github.com/Graz-Institute-of-Biology/sfDBM_STOTEN) and archived on Zenodo (Kim, 2024). Used parameters and input variables for the soil models are provided in the supplementary information and supplementary data.

CRedit authorship contribution statement

Minsu Kim: Writing – review & editing, Writing – original draft, Visualization, Validation, Software, Methodology, Formal analysis, Conceptualization. **Clément Lopez-Canfin:** Writing – review & editing, Writing – original draft, Validation, Methodology, Investigation, Data curation, Conceptualization. **Roberto Lázaro:** Writing – review & editing, Resources, Investigation, Funding acquisition. **Enrique P. Sánchez-Cañete:** Writing – review & editing, Resources, Investigation, Funding acquisition. **Bettina Weber:** Writing – review & editing, Supervision, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

The model scripts is available in a GitHub repository (https://github.com/Graz-Institute-of-Biology/sfDBM_STOTEN) and archived on Zenodo (<https://doi.org/10.5281/zenodo.10843357>)

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.scitotenv.2024.171751>.

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