Contents lists available at ScienceDirect



International Journal of Biological Macromolecules

journal homepage: www.elsevier.com/locate/ijbiomac



Three-dimensional solution structure, dynamics and binding of thioredoxin m from *Pisum sativum*^{*}

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ARTICLE INFO

Keywords: Thioredoxin Protein-protein interactions NMR Fluorescence Fructose-1,6-bisphosphatase Peptide

ABSTRACT

Thioredoxins (TRXs) are ubiquitous small, globular proteins involved in cell redox processes. In this work, we report the solution structure of TRX m from *Pisum sativum (pea)*, which has been determined on the basis of 1444 nuclear Overhauser effect- (NOE-) derived distance constraints. The average pairwise root-mean-square deviation (RMSD) for the 20 best structures for the backbone residues (Val7-Glu102) was 1.42 ± 0.15 Å, and 1.97 ± 0.15 Å when all heavy atoms were considered. The structure corresponds to the typical fold of TRXs, with a central five-stranded β -sheet flanked by four α -helices. Some residues had an important exchange dynamic contribution: those around the active site; at the C terminus of β -strand 3; and in the loop preceding α -helix 4. Smaller NOE values were observed at the N and C-terminal residues forming the elements of the secondary structure or, alternatively, in the residues belonging to the loops between those elements. A peptide derived from pea fructose-1,6-biphosphatase (FBPase), comprising the preceding region to the regulatory sequence of FBPase (residues of Int2), was bound to TRX m with an affinity in the low micromolar range, as measured by fluorescence and NMR titration experiments. Upon peptide addition, the intensities of the cross-peaks of all the residues of TRX m was larger than that of the intact FBPase, indicating that there are additional factors in other regions of the polypeptide chain of the latter protein affecting the binding to thioredoxin.

1. Introduction

Thioredoxins (TRXs) are a family of small, globular proteins (\sim 10–13 kDa) present in all living systems from eukaryotic to prokaryotic cells [1–4], but plants exhibit the most complex TRX multigenic family, as shown by the analyses performed in Arabidopsis and rice genomes [5]: for instance, the *Arabidopsis thaliana* genome encodes >25 different TRXs and other TRX-like proteins with several TRX domains [6]. TRXs are heat-stable proteins with an acidic isoelectric point, working as disulfide oxido-reductase enzymes, with a well-known scaffolding: a five-stranded β -sheet, sandwiched by four α -helices: a $\beta\alpha\beta\alpha\beta\alpha\beta\beta\alpha$. This fold has been identified in numerous proteins, including glutaredoxins [7], glutathione-S-transferase [8], protein disulfide isomerase [9] and Dsb proteins from *E. coli* periplasm [10]. Their catalytic mechanism involves, in most of the examples described so far, a conserved pentapeptide sequence: WC(G/P)PC motif [11,12]; this motif remains unaltered in all the TRX protein sequences reported to date, regardless of the residue variation in the rest of the sequences. The active site cysteines of TRXs are oxidized, reducing other proteins, through a

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https://doi.org/10.1016/j.ijbiomac.2024.129781

Received 14 December 2023; Received in revised form 23 January 2024; Accepted 24 January 2024 Available online 1 February 2024

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^{*} This paper is dedicated to the memory of Manuel Rico (a mentor, colleague and friend) and Julio López-Gorge (a mentor, friend, colleague and husband).

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Abbreviations		
CD	circular dichroism	
FBPase	fructose-1,6-bisphosphatase	
FBPpep	fragment derived from FBPase, with the sequence E ¹⁵² SLPDYGDDSDDNTLGTEEQRSIVNVSQ ¹⁷⁹	
Fd	ferredoxin	
HSQC	heteronuclear single quantum coherence	
IPTG	isopropyl-β-D-1-thiogalactopyranoside	
NMR	nuclear magnetic resonance	
NOE	nuclear Overhauser enhancement	
NADP	nicotinamide adenine dinucleotide phosphate	
NADPH	reduced nicotinamide-adenine dinucleotide-phosphate	
NOESY	2D nuclear Overhauser enhancement spectroscopy	
RMSD	root-mean-square deviation	
TOCSY	2D total correlation spectroscopy	
TPPI	time proportional phase increment	
TRX	thioredoxin	
TSP	3-(trimethylsilyl) propionic acid-2,2,3,3- ² H ₄ -sodium	
	salt	
UV	ultraviolet	

reversible disulfide-dithiol reaction of the two SH groups [1]; in fact, the reduced TRXs are the major cellular protein disulfide reductases in living organisms, and they also work as an electron donor for particular enzymes of primary metabolism. The reaction mechanism of TRXs involves a nucleophilic attack on one of the cysteines in a target protein by the catalytic cysteine in the TRX redox-active motif, forming an intermolecular disulfide between TRX and the target protein. This intermolecular disulfide bond is reduced, resulting in the reduced target protein and oxidized TRX. The oxidized TRX is reactivated through reduction by a specific enzyme called thioredoxin reductase [13,14] (in fact, both proteins form the so-called thioredoxin system). Therefore, TRXs are capable of interacting with other targets, having a crucial role in many cellular redox metabolism processes, activation of transcription factors, regulation of photosynthetic events in plants, and in regulation pathways in other species such as cell proliferation, insulin degradation,

blood clotting, seed germination and repair of oxidative damage [11,15–19] (Fig. 1). As a consequence, in animals TRXs are associated with cancer, cardiac disease or viral infections [20,21]; furthermore, recently, TRXs have been used as potential therapeutic regulators of cell growth, apoptosis and inflammation [16,22] (Fig. 1).

Based on their gene sequences, there are two families of TRXs: family I includes proteins with a single TRX domain, whereas family II is composed of multiple TRX-like domains. At least twenty TRX-encoding genes belonging to family I have been reported in Arabidopsis thaliana, while in mammals (humans included) there are only two, as well as in E. coli [23-25]: TRX1 (TXN1 in humans) and TRX2 (TXN2). Whereas the protein codified by TRX2 is mitochondrial, that from TRX1 appears to be cytosolic (but it is also present in the nucleus and at another cell locations). In plant chloroplasts, two TRX systems are present with different sources of reducing power. In the first system, ferredoxin (Fd)-thioredoxin reductase (FTR) is involved in the Fd-TRX system, and it activates the chloroplasts TRXs by reducing equivalents from photosynthetically reduced (activated) Fd [26–28]. The second system is the reduced nicotinamide-adenine dinucleotide-phosphate (NADPH)dependent chloroplast thioredoxin reductase that is reduced by NADPH, present in chloroplasts. The chloroplasts contain multiple types of TRXs: two TRX f; four TRX m; and two y isoforms, as well as x- and z-type of TRX [29-31]. TRXs also intervene in plastid transcription, ATP synthesis, chloroplast biogenesis, carbon, nitrogen and sulfur metabolism, chloroplast biogenesis and oxidative stress responses [27,32] (Fig. 1). Due to that plethora of functions, some of the plastid TRXs are involved in regulation of several enzymes [11,28,33]: for instance, TRX m activates the nicotinamide-adenine dinucleotide-phosphate- (NADP) dependent malate dehydrogenase [33] and TRX f activates the fructose-1,6-bisphosphatase (FBPase), among other Calvin cycle enzymes [11,26,33,34]. However, it has been shown that TRX m also activates chloroplast FBPase [35-38] [39]. FBPpase catalyzes the breakdown of fructose-1,6-bisphosphate to fructose-6-phosphate and phosphorous; this reaction has a large free energy change and it is a regulatory step for pathways where FBPase intervenes. In general, FBPase activity is regulated by electron availability through a modulating mechanism involving the Fd-thioredoxin system [26,40-42]. Our groups have shown that the preceding region of the FBPase regulatory sequence, comprising residues Leu154 to Glu170 is the thioredoxin docking site [38,40,42,43]. Therefore, we wondered whether such isolated region



REDOX ACTIVITY OF TRXs

- 1. Cell redox environment control
- 2. Transcription factors regulation
- 3. Metabolic enzymes regulation

PLANTS

Self-incompatibility in flowers Calvin cycle enzymes regulation by light Seed germination and development Anther development Intercellular messenger Oxidative damage repair MAMMALS

Immune response Insulin degradation Involvement in diseases (cancer, Alzheimer..) Sperm and embryo development Intercellular messenger Apoptosis control

Fig. 1. Functions of thioredoxins: Scheme showing some well-established roles for plant (left-hand-side figure) and mammals (right-hand-side figure) thioredoxins (TRXs) based on their molecular redox activity (centered figure).

could be capable of binding to the intact TRX m, and if so whether the measured affinity was similar to that of the intact FBPase. If the TRX-peptide affinity was similar to that of TRX-FBPase, we could modulate the binding between the two proteins with such peptide.

In this work, we report the ¹⁵N, ¹³C and ¹H assignments of TRX m (108 residues) from Pisum sativum (pea), its three-dimensional solution structure, its dynamics in the ns-ps time regime, and its binding to a peptide derived from FBPase, comprising the TRX-binding site of the intact protein (residues Glu152 to Gln179) [38,39,43]. The obtained structure was similar to that of TRX m from spinach [44], adopting the TRX fold, with minor differences in the length of some of the strands comprising the β -sheet and some of the helices: namely, a five-stranded β -sheet, sandwiched by four α -helices, located at the external side of the β -sheet, with a general $\beta\alpha\beta\alpha\beta\alpha\beta\beta\alpha$ scaffolding. The active-site center (preceding the second α -helix); the C terminus of β -strand 3; and the Cterminal segment (comprising the last fourth α -helix) appeared to be highly mobile. Binding of the peptide derived from the pea FBPase, containing the thioredoxin docking site, occurred with a dissociation constant, K_d of ~2 μ M, as determined by fluorescence, and K_d of ~50 µM, as determined from NMR titrations (followed by 2D ¹⁵N, ¹H-HSOC spectra). However, this value differed from that of the intact FBPase (0.019 µM), obtained by a Koshland-Nemethy-Filmer model [39], indicating the TRX cooperative binding to the FBPase. In our case, probably, the isolated nature of the pea FBPase peptide showed only the energetics of the interactions (and, probably, folding) of the peptide in its binding site on the TRXs, without any cooperativity effect. Therefore, our results show that other factors in the polypeptide chain of TRX might intervene in the binding to FBPase.

2. Materials and methods

2.1. Materials

Imidazole, Trizma base, DNase, SIGMAFAST protease tablets, deuterium oxide (99 % atom in ${}^{2}\text{H}_{2}\text{O}$), NaCl, Ni ${}^{2+}$ -resin, 3-(trime-thylsilyl) propionic acid-2,2,3,3- ${}^{2}\text{H}_{4}$ -sodium salt (TSP), DTT (dithio-threitol) and Amicon centrifugal devices with a molecular weight cut-off of 3 kDa were from Sigma (Madrid, Spain). The β -mercaptoethanol was from BioRad (Madrid, Spain). Kanamycin and isopropyl- β -D-1-thio-galactopyranoside (IPTG) were obtained from Apollo Scientific (Stock-port, UK). Triton X-100, Tris(2-carboxyethyl)phosphine (TCEP), dialysis tubing with a molecular weight cut-off of 3500 Da, and the SDS protein marker (PAGEmark Tricolor) were from VWR (Barcelona, Spain). The rest of the materials were of analytical grade. Water was deionized and purified on a Millipore system.

2.2. TRX m cloning, expression and purification

The cDNA from TRX m from Pisum sativum [37] was subcloned using BamHI and HindIII restriction sites into pETM11 plasmid (Novagen), and modified to yield a protease cleavable N-terminal hexahistidine (His₆)-tag. The thrombin cleavage sequence is formed by -GLVPRGSH-, where the protease cleaves between the R and G residues. The cloned protein had a K at position 87 instead of an R, as the originally deposited pea TRX m in UniProt. Plasmid encoding the (His₆)-tagged protein was introduced into competent E. coli BL21 (Merck, Madrid, Spain) cells. Transformed cells were grown at 37 °C of LB media (containing 30 mg/ mL kanamycin) until an OD₆₀₀ of 0.6–0.9 was reached. The expression in M9 minimal media was supplemented with 1 g/L of ¹⁵NH₄Cl for ¹⁵Nlabelled samples, or, alternatively for double labelled samples used in TRX m assignment, with 1 g/L of ¹⁵NH₄Cl and 2 g/L of ¹³C-glucose. The addition of a final concentration of 1 mM IPTG (1 mL of a stock solution 1 M IPTG to 1 L of media) when the above indicated OD₆₀₀ was reached, induced protein expression either in LB or in M9 minimal media. After adding IPTG, the cells were incubated overnight at 37 °C and harvested by centrifugation.

Protein (either in LB or minimal media) was purified by Ni²⁺-affinity chromatography in a prepacked HiTrap affinity column (GE Healthcare, Barcelona, Spain) according to the manufacturer's indications. Briefly, cell pellets were resuspended in 50 mL containing 50 mM Tris (pH 8.0), 1 % Triton X-100, 0.5 M NaCl, 1 mM β -mercaptoethanol and 10 mM imidazole (buffer A) with 1 % Triton X-100 and a single SIGMAFAST tablet. The suspension of the cells was lysed by sonication on ice with 10 bursts of 45 s at maximum power interleaved with 15 s keeping in ice (the sonicator was a Branson model 102C). Insoluble cell debris was removed from the cell lysate by centrifugation. The supernatant was loaded onto the HiTrap affinity column, equilibrated with buffer A. The protein was eluted with a linear gradient of buffer B (buffer $A\,+\,1$ M imidazole). Fractions containing the target protein were pooled together, concentrated by using Amicon centrifugal devices, and loaded onto a Superdex 75 16/60 size-exclusion column (GE Healthcare) equilibrated with 25 mM phosphate buffer (pH 7.0) with 150 mM NaCl, on an AKTA FPLC (GE Healthcare) by following the absorbance at 280 nm. The (His₆)-tag was cleaved off by using commercial thrombin (GE Healthcare) as described previously [45,46]. After cleavage, the protein had the additional residues at its N terminus: GSHM-. The concentration of pure protein was determined by measuring the absorbance at 280 nm with the extinction coefficients as determined by the sequence [47].

2.3. Peptide design

Peptide was synthesized and purified by GenScript (Amsterdam, Netherlands) with a purity larger than 98 %; purity was checked by mass spectrometry, and by SDS-PAGE in our own laboratory. The sequence of the peptide (FBPpep), coming from the pea fructose-1,6-bisphosphatase, was E¹⁵²SLPDYGDDSDDNTLGTEEQRSIVNVSQ¹⁷⁹, with the C terminus amidated. The following residues of the original FBPase sequence: (a) Cys153, Cys173 and Cys178 were mutated to serine to avoid the formation of intra- or inter-disulfide bridges; and (b) Phe157 was mutated to tyrosine to allow for reliable absorbance detection of the peptide at 280 nm [47]. In the crystal structure of oxidized pea FBP, Cys153 and Cys173 form a disulfide bond [48], but in the structure of the C153S mutant the secondary structure is conserved independently of the disulfide bridge formation. Although this mutation affects the redoxdependent regulation of the enzyme [49], since we are going to measure binding, and not the redox properties of the peptide, we decided to work with the peptide where the three cysteines where mutated to serine to avoid peptide aggregation by intermolecular cysteine oxidation.

2.4. Fluorescence binding experiments

Fluorescence spectra were collected on a Cary Varian spectrofluorometer (Agilent, Santa Clara, CA, USA), interfaced with a Peltier unit. A 1-cm-pathlength quartz cell (Hellma, Kruibeke, Belgium) was used. Excitation and emission slit widths were set to 5 nm. Voltage of the photomultiplier was set to 650 V. Data were collected every 1 nm. For the titration between FBPpep and TRX m, increasing amounts of FBPpep, in the concentration range 0-10 µM (final concentration), were added to a solution with a fixed concentration of TRX m (1.9 μ M) in the presence of DTT (1 mM). The samples were prepared the day before and left overnight at 5 °C; before the measurements, they were incubated for 1 h at 30 °C. Experiments were carried out in phosphate buffer (50 mM), pH 5.8 at 30 °C. The samples were excited at 280 and 295 nm. In all cases, the appropriate blank-corrections were made by subtracting the signal obtained with the corresponding amount of FBPpep with KaleidaGraph (Synergy software, Reading, PA, USA). Spectra were corrected for inner-filter effects during fluorescence excitation [50]. The titration was repeated three times, using new samples; variations in the results among the three experiments were lower than 10 %.

The dissociation constant of the complex, K_d , was calculated by fitting the binding isotherm constructed by plotting the observed fluorescence change as a function of FBPpep concentration to the general

binding model, explicitly considering protein depletion due to binding [51,52]:

$$F = F_0 + \frac{\Delta F_{max}}{2[\text{TRX m}]_T} \left([\text{FBPpep}]_T + [\text{TRX m}]_T + K_d \right) - \sqrt{\left(\left([\text{FBPpep}]_T + [\text{TRX m}]_T + K_d \right)^2 - 4[\text{FBPpep}]_T [\text{TRX m}]_T \right)}, \quad (1)$$

where *F* is the measured fluorescence at any particular concentration of FBPpep after subtraction of the spectrum of the sample containing only the same concentration of such polypeptide (i.e., *F* is the differential, or difference fluorescence); ΔF_{max} is the largest change in the fluorescence of FBPpep when all polypeptide molecules were forming the complex, compared to the fluorescence of each isolated protein (at the same corresponding concentration); *F*₀ is the fluorescence intensity when no FBPpep was added; [TRX m]_T is the constant, total concentration of TRX m (1.9 µM); and [FBPpep]_T is that of FBPpep, which was varied during the titration. Fitting to Eq. (1) was carried out by using KaleidaGraph.

2.5. Circular dichroism (CD)

Far-UV CD spectra of FBBpep were collected on a Jasco J810 spectropolarimeter (Jasco, Tokyo, Japan) with a thermostated cell holder and interfaced with a Peltier unit at 5 °C. The instrument was periodically calibrated with (+)-10-camphorsulfonic acid. A cell of path length of 0.1 cm was used (Hellma, Kruibeke, Belgium). All spectra were corrected by subtracting the corresponding baseline obtained from the spectrum of a solution containing only the buffer with KaleidaGraph. The buffer was the same used in the fluorescence titration experiments. Isothermal wavelength spectrum of FBPpep was acquired as an average of 6 scans, at a scan speed of 50 nm/min, with a response time of 2 s and a band-width of 1 nm.

2.6. Nuclear magnetic resonance (NMR)

NMR samples for homonuclear experiments were prepared by concentrating and exchanging the solution containing the TRX m obtained from the gel filtration purification step in Amicon 3 K centrifugal devices, in a final 9:1 $\rm H_2O{:}^2H_2O$ solution, with a 0.02 % $\rm NaN_3$ in phosphate buffer (50 mM), pH 5.8. The solution was centrifuged briefly to remove insoluble protein and then transferred to a 5 mm NMR tube. Spectra were recorded on a Bruker Avance 500 MHz spectrometer (Bruker GmbH, Karlsruhe, Germany), with z-pulse field gradients and a TXI probe, equipped with Topspin 2.1 software (Bruker, Karlsruhe, Germany). Experiments were run at 30 °C; the temperature of the probe was calibrated with methanol [53]. The pH of the solution was measured at the beginning and end of the complete series of experiments by using a Russell glass electrode, without finding any difference between both measurements. Values of the pH reported here represent apparent values of pH, without correction for isotope effects. TSP was used as the internal chemical shift reference and corrected for the pH value in all spectra described below [53].

The NMR experiments with ¹³C/¹⁵N-labelled TRX m samples were performed in the same magnet with the same buffer (and 1 mM DTT), and at the same temperature as the homonuclear experiments by using Amicon 3 K centrifugal devices to concentrate the sample, in a final 9:1 H_2O :² H_2O solution, with a 0.02 % NaN₃ in phosphate buffer (50 mM), pH 5.8. Thus, we were determining the structure of the reduced protein, as we ran the experiments in the presence of a reducing agent; this redox agent should avoid the problems of having some population of the oxidized state, due to the oxygen dissolved in the solution. Protein concentration was 2.5 mM for either double labelled or unlabelled samples.

(a) Assignment of TRX m: The NMR experiments used for sequence specific assignment of HN, N, C_{α} and C_{β} resonances of a $^{13}C/^{15}N$ -labelled TRX m were HNCACB, HNCA, HNCO, HCCCONH [53]. Standard 2D

NOESY (mixing time 90 ms) and 2D TOCSY (mixing time, 80 ms) homonuclear experiments spectra were also run to allow for identification of additional protons in the side-chain of residues and for finding NOE restrictions [54,55]. The details of the acquisition parameters of all the experiments used are shown in Supplementary Table 1 (Table ST1). The chemical shifts have been deposited in the BMRB (Biomagnetic Resonance Bank) with number 52150. The 2D ¹⁵N-HSQC spectra were also acquired in the TPPI (time proportional phase increment) mode at the beginning, in-between and at the end of the triple-resonance suite of experiments to ensure sample integrity during the long acquisition times. Processing and analysis of data was carried out by using the Bruker software (Topspin 2.1).

(b) Relaxation measurements of TRX m: We ran ${}^{15}N-R_1, {}^{15}N-R_2$ and ¹H-¹⁵N NOE experiments. All the relaxation measurements were determined in an interleaved manner to ensure that the experimental conditions were the same for the different relaxation delays. The temperature was the same used in the assignment (30 °C). All experiments were acquired by using gradient pulse sequences developed by Kay and co-workers [56]. The T_1 (=1/ R_1) relaxation time was measured with typically 10 inversion-recovery delays, varying from 5 to 650 ms; T_2 $(=1/R_2)$ was determined by collecting 8 time points ranging from 50 to 400 ms. Two relaxation delays were randomly repeated in each series of rates to ensure reproducibility. For the T_1 and T_2 pulse sequences, the delay between scans was 5 s. Cross-peaks intensities at each time were measured by integration of signals in Topspin 2.1. Intensity variation in the T_1 and T_2 experiments of each cross-peak was fitted to simple exponential functions, with a pre-exponential factor by using Kaleida-Graph. Errors shown in the T_1 and T_2 were from fitting errors.

The ¹H—¹⁵N NOE experiment was measured by recording interleaved spectra with or without proton saturation. The ¹H—¹⁵N NOE spectrum recorded in the presence of proton saturation was acquired with a saturation time of 10 s. The ¹H—¹⁵N NOE spectrum recorded without proton saturation incorporated a relaxation delay of 10 s. The ¹H—¹⁵N NOE experiment (with and without saturation) was repeated twice. The NOE ratio was defined as NOE = I_{sat}/I_{ref} , where I_{sat} is the intensity of the corresponding cross-peak in the saturated spectrum and I_{ref} is that in the reference (non-saturated) spectrum. The intensities of the cross-peaks in both spectra were measured by using Topspin 2.1. Errors in the NOEs were estimated to be 10 % as judged from the integrated regions where no peaks were observed.

Spectra were recorded with 2048 \times 180 complex matrices in the F_2 and F_1 dimensions, respectively, with typically 32 scans (NOE experiment) and 16 scans (R_1 and R_2 experiments) *per* F_1 experiment. Spectral widths of 1935 and 7000 Hz were used in F_1 and F_2 dimensions, respectively; the ^{15}N carrier was set at 120 ppm and that of 1H was set on the water signal in all the experiments.

(c) Assignment of FBPpep: Two-dimensional homonuclear spectra of the peptide were acquired in each dimension in phase-sensitive mode by using the TPPI technique [57] and a spectral width of 5500 Hz in both dimensions. Peptide concentration was 2.0 mM, and the temperature was 5 °C; the pH was 5.8, phosphate buffer (50 mM). Standard 2D TOCSY [58] and NOESY [59] (with mixing times of 80 and 250 ms, respectively) were performed by acquiring a data matrix size of 2048 imes256 points. The relaxation time in both experiments was 1 s. The DIPSI (decoupling in the presence of scalar interactions) spin-lock sequence [60] was used in the TOCSY experiments. A number of 88 scans were acquired per increment in the first dimension, and the residual water signal was removed by using the WATERGATE sequence [61]. NOESY spectra were collected with 128 scans per increment in the first dimension, using again the WATERGATE sequence [61]. In both spectra, data were zero-filled, resolution-enhanced with a square sine-bell window function optimized in each spectrum, baseline-corrected and processed with Topspin 2.1. The ¹H resonances were assigned by standard sequential assignment processes [53]. The chemical shift values of H_{α} protons in random-coil regions were obtained from tabulated data, corrected by neighboring residue effects [53,62-64]. Spectra were

referenced with internal TSP considering the pH-dependence of the methyl signal [53].

(d) Temperature coefficients of FBPpep: The chemical shifts of the amide protons usually exhibit a linear dependence with temperature, whose slope is referred to as "temperature coefficient". The temperature coefficients ($\Delta\delta/\Delta T$) for solvent-accessible amide protons are in the range – 6 to –10 ppb K⁻¹. Temperature coefficients small in absolute values (i.e., $|\Delta\delta/\Delta T| \leq 4.0$ ppb K⁻¹) are indicators of solvent inaccessibility, suggesting that the corresponding amide protons are probably involved in hydrogen-bonds [65–67]. We measured the temperature coefficients of the amide protons of FBPpep by acquiring TOCSY experiments at 5, 10, 15 and 17 °C at the same pH described above. The TOCSY experiments were referenced by using TSP, considering the pH-dependence of the methyl signal [53].

(e) Titration of FBPpep over TRX m followed by NMR: The cross-peaks in the 2D ¹H, ¹⁵N HSQC NMR spectra [68] of TRX m, at different FBPpep concentrations, were identified by using the assignment provided in this work. Spectra were acquired in the TPPI mode at 25 °C (the temperature was chosen to allow for comparison with the titration fluorescence experiments), in phosphate buffer (50 mM), pH 5.8 in the presence of 1 mM DTT. The concentration of ¹⁵N-labelled TRX m was 70 µM either in isolation or in the presence of increasing concentrations of FBPpep (from 0 to 943 µM, final concentration). The spectra were typically acquired with 2048 complex points in the ¹H dimension, 200 complex points in the ¹⁵N dimension, with 192 scans. Typical spectral widths for the 2D 1 H 15 N HSQC spectra were 6000 (1 H) and 1500 (15 N) Hz. Water was removed by using the WATERGATE sequence [61]. After the acquisition of each experiment, at a particular peptide concentration, the sample was lyophilized and the corresponding amount of a FBPpep stock solution (2 mM) was added to the dried powder to yield the next peptide concentration. It could be thought that this procedure of lyophilization after the acquisition of each titration point could result in artifacts and they would be responsible for the behavior observed in the intensities of the cross-peaks of the protein (Section 3.3.2), as the peptide concentration was raised. Control experiments carried out with samples of ¹⁵N-TRX m concentrated by using lyophilization or Amicon-concentration devices yielded similar 2D ¹H-¹⁵N HSQC spectra, with no variations in the signal intensity (volume). Besides, it is quite unlikely that any artifact associated with lyophilization would result in a decrease of intensities in a sigmoidal-like way.

The resulting matrix of each experiment was zero-filled to double the number of original points in all dimensions, and shifted squared sinebell apodization functions were applied before Fourier transformation. NMR data were processed and analyzed using TopSpin 2.1. Cross-peaks intensities (volumes) were corrected by the corresponding value of the receiver gain for each experiment. All the 2D 1 H, 15 N HSQC NMR spectra were referenced with the methyl protons of external TSP for 1 H and for the indirect 15 N dimension [53].

The variations in the intensities of the cross-peaks of 2D 1 H, 15 N HSQC NMR spectra of TRX m, corrected by the corresponding receiver gain, were fitted to:

$$F = F_0 + \frac{\left(\Delta F_{max} \left[\text{FBPpep}\right]_T\right)}{\left(\left[\text{FBPpep}\right]_T + K_d\right)} (2),$$

which is a simplified expression of Eq. (1). The meanings of all symbols are the same as those previously indicated in Eq. (1).

2.7. Structure calculations

All spectra were analyzed manually by using the Sparky software [69]. After integration, the peak volumes and chemical shifts list were output to a CYANA 2.1.-compatible format [70–72]. The automatic calibration method implemented in CYANA, CALIBA [72], was run to transform the peak volumes into distances. The upper limits obtained were used without modification for the initial structure calculations.

Structures of TRX m were calculated by using the NOE distance restraints with the program CYANA. This was an iterative process because many chemical shifts of TRX m could be assigned to more than a single nucleus, and therefore, many NOE cross-peaks were ambiguous, meaning that a cross-peak in the NOESY spectrum many times could be assigned to a set of possible pairs of protons [73,74]. At subsequent steps during the calculation of the structures, those distance restraints were combined in an iterative fashion with: (i) backbone ϕ and ψ angle restraints as predicted by TALOS+ [75] based on the chemical shifts of C_{α} , C_{β} , H_{α} , H_{β} , NH, amide N and CO atoms (only when angular predictions were classified as "good" were used in the calculations); and (ii) hydrogen-bonds obtained from the remaining amide protons observed in hydrogen-exchange experiments carried out by dissolving the protein in 99.99 % D₂O, at 0.2 mM final concentration, and acquiring a short 2D ¹H—¹⁵N HSQC NMR spectrum. The introduced hydrogen-bond restraints were in all cases well-suited with the observed, experimental distance restraints from NOESY spectra, and they were only introduced as restraints, when the first runs of CYANA, predicted the presence of such hydrogen bonds in at least 15 of the best 20 conformers (with the lowest target functions). Each hydrogen-bond was specified by two distance restraints: HN-O distance of 1.7-2.3 Å and N-O distance of 2.4-3.3 Å [73,74,76].

The resulting initial structures, obtained with CYANA (without the addition of hydrogen-bonds as restrictions), and which had the lowest target functions in the ensemble of conformers, were used to assign additional, ambiguous peaks (because either they were mis-assigned or there was chemical shift overlapping) in the NOESY spectrum. It is important to stress out that we did not use all the NOEs from the NOESY spectrum at the first calculations, but only those which had been unambiguously assigned. The quality of the introduced constraints was checked at every step by analyzing the restraint violations of the calculated conformers. The NOE cross-peaks corresponding to restraints that were consistently violated in a significant number of structures were checked for possible overlapping in the NOESY spectra and the corresponding restraints were modified. The cycle was repeated until no consistent violation was detected. The cycle of calculations and assignments was repeated until no further assignments were possible.

A total of 300 random conformers were annealed in 12,000 steps. The 20 conformers with the lowest target function constituted the final family of structures of TRX m. The quality of the structures was evaluated in terms of deviations from ideal bond lengths, angles, and through Ramachandran plots by using the PSVS server [77].

3. Results

3.1. Description of the structure of TRX m from pea

Several repeated rounds of ambiguous long-range and nonsequential NOEs were assigned, with the help of the first structures, obtained from the calculations, with the lowest target function, and converted to distance constraints to be used as input data in CYANA, together with the previously unambiguously assigned NOEs. We also removed restraints that were repeatedly violated in the structure calculations. Table 1 summarizes the number of structurally relevant intraresidual, sequential, medium- and long-range constraints used in the calculations. The RMSD values for all backbone atoms and heavy atoms between residues Val7 and Glu102 were 1.42 \pm 0.15 Å and 1.97 \pm 0.15 Å, respectively. The target function of CYANA had an average value of 0.51 ± 0.12 (in the range 0.30 to 0.70). These values of the RMSD among the conformers were similar to those of other solution structures of TRXs from other species (see for instance [78]). There were no distances smaller than 2.2 Å for heavy atoms, nor distances smaller than 1.6 Å for hydrogens. The RMSD for covalent bonds relative to the standard value was 0.001 Å, and that for covalent angles was 0.2 degrees, as indicated by the PSVS web-server [77].

The overall fold of TRX m from pea indicates a core formed by a five-

Table 1

NMR restraints and structural statistics for the best 20 structures.

Restraints			
Total number of restraints			
NOE restraints ^a	1444		
Intra-residue (i-j = 0)	446		
Sequential ($ i-j = 1$)	400		
Medium-range ($ i-j \le 4$)	238		
Long-range ($ i-j \ge 4$)	360		
Dihedral angle restraints	114 (for ϕ and ψ angles)		
Hydrogen-bond restraints	29 x 2 ^b		
RMSD to the mean structure residues Val7-Glu102 (in Å)			
Average backbone	1.42 ± 0.15		
Heavy atoms	1.97 ± 0.15		
Ramachandran summary for all residues in the 20 conformers from Procheck (in %) ^c			
Most favored	52.2		
Additionally allowed regions	33.3		
Generously allowed regions	10.5		
Disallowed regions	4.1		

^a The i and j represent two residues of the sequence of TRX m.

 $^{\rm b}\,$ For each hydrogen-bond two restrictions were introduced: the HN-O and the N—O distances.

^c Obtained from PSVS server [77].

stranded mixed β -sheet, surrounded by four α -helices (Fig. 2). The α 1 and α 3 helices pack on one side of the β -sheet. Conversely, the α 2 and α 4 helices pack on the other side of the β -sheet, as judged by the proximity of several of the side-chains of their residues, as concluded from the NOEs between the side-chains of Leu46 and Lys103, or between Tyr50 and Glu105, among others. However, these interactions do not prevent the fraying of α 4, as it has been observed in the structure of TRX m from spinach [44]. The scaffolding of β -sheets and α -helices is similar to those observed in other TRXs (see [44,79] and references therein), but some of the latter were shorter and more irregular in the TRX m from pea (see below). In the TRX m from spinach, the Pro77 is in a cis conformation; however, we could not detect any NOE between the H $_{\alpha}$ proton of Ile76 (at 4.65 ppm) and the H $_{\alpha}$ proton of Pro77 (at 5.17 ppm) because of proximity to residual water signal; then, during the calculations Pro77 adopted a trans conformation in TRX m from pea.

The use of the chemical shifts of ${}^{13}C_{\alpha}$, ${}^{13}C_{\beta}$, ${}^{13}CO$, H_{α} , H_{β} , NH and

amide ^{15}N atoms in TALOS+ [75] predicted the following elements of secondary structure: a five-stranded pleated β -sheet, which is twisted, and composed of parallel (p) and antiparallel (a) strands following the arrangement: Gln6-Val8 (β-strand 1 p); Ile54-Leu59 (β-strand 3 p); Val24-Trp29 (β-strand 2 p); Pro77-Lys83 (β-strand 4 a); and Ser90-Ile92 $(\beta$ -strand 5 a). The presence of these strands was confirmed by: (i) longdistance NOEs involving backbone amides and H_{α} protons of residues located in the different β -strands (Fig. S1), together with long-distance NOEs of backbone amides with side-chains of residues located in the different β-strands; and, (ii) consistency between the observed slowestexchanging protons and the hydrogen-bond scaffolding characteristic of a β -sheet conformation (Fig. 3). This five-stranded β -sheet forms the core of the molecule. The β -strand 1 did not show any solvent-exchange protected amide proton (Fig. 3), but: (i) the observed NOE between the NH of Lys58 and the H_{α} proton of Gln6; and, (ii) the presence of a hydrogen-bond between the NH of Val7 and the CO of Lys58 in the first obtained structures during the CYANA calculations allowed the identification of the region Gln6-Val8 as a β -strand.

The four helices were also predicted by TALOS+ software. The presence of such helices (Fig. 2) was confirmed by: (i) strong or medium NN(i,i + 1), medium α N(i,i + 2) and medium $\alpha\beta(i,i + 3)$ NOEs (Fig. 3); and, (ii) the amide protons protected against solvent-exchange (Fig. 3). The helices were roughly spanning residues: Trp13-Leu16 (a1), Ile42-Glu49 (α 2), Asn66-Lys70 (α 3) and Ala98-Glu105 (α 4). The four α -helices were displayed on the external surface of the globular molecule. The helices did not seem to be regular and there were deviations for the main chain torsion angles, ϕ and ψ , when compared to the standard helical values (Fig. 2). For instance, in α1, Glu15 did have both angles different to those observed in regular α -helices; in $\alpha 2$, Glu49 appeared to show also deviations from the ideal ϕ and ψ values, and $\alpha 2$, located after the active site loop, appeared to be bent; α 3, where turns of a 3₁₀-helix structure were observed (hydrogen-bond scaffolding of CO(i)-NH(i + 3)), probably because the formation of a regular α -helix is hindered by Pro65; and finally, α 4, with the last C-terminal residues of the helix having the largest deviations from the regular ϕ and ψ values.

The active site, involving Trp32-Cys33-Gly34-Pro35-Cys36 residues, was poorly defined; in fact, analyses of the 20 structures showed a large variation in the ϕ and ψ values, and Cys36 was among the outlier



Fig. 2. Comparison of the solution structure of TRXs m from two plants. Solution structures of TRX m from pea (this work) (A) (in blue) and that of spinach (1GL8) (B) (in yellow) in the same orientation. The ribbon representation of each protein is shown together with the 20 best conformers. NMR model structures were superimposed over residues Val7-Glu102 and Val14-Leu117 for the pea and spinach enzymes, respectively. The figure was produced with PyMOL (Schrödinger).



Fig. 3. Summary of NMR data. NOEs are classified into strong, medium or weak according to the height of the bar underneath the sequence, as concluded from peak integration. Dotted bars indicate NOEs which could not be unambiguously assigned due to overlapping or proximity to the residual water signal. The corresponding sequential H_{α} NOE with the following H_{δ} of the proline residue are indicated by an open bar in the row corresponding to the α N NOEs. The white squares indicate those residues that were detected in a 2D ¹H, ¹⁵N HSQC spectrum after dissolving the sample in ²H₂O at pH 5.8 and 30 °C; that is, they are the most solvent-exchange protected residues.

residues (together with Gln6, Ala51, Glu63, Ile76 and Ala30) in the arrangement of such angles for most of the 20 structures (the same feature happened in such active site cysteine for TRX m from spinach [44]). The active site was located at the N terminus of $\alpha 2$. The sidechains of the residues involved in the active site protruded from the protein surface and they did not show interactions with other protein regions, although the preceding (β -strand2) and the following (α 2) polypeptide patches contact with other residues distant in the sequence. Interestingly enough, the signal of Gly34 in the 2D ¹H, ¹⁵N-HSQC spectrum did not appear (it was possible to identify the NH in the 2D ¹H⁻¹H NOESY spectrum); the absence of the cross-peak in the ¹H, ¹⁵N HSQC spectrum for such glycine was also reported in TRX m from spinach [44], where the residues after the last cysteine did not show any further NOE with other regions [44]. Furthermore, cross-peak broadening of all those residues close to or belonging to the active site has been also observed in the reduced species of the TRX of Ehrlichia chaffeensis [78].

3.2. Backbone dynamics of TRX m

Backbone ¹⁵N relaxation parameters were obtained for 78 residues (Fig. 4) out of the 100 residues (all amino acids, except the N-terminal one and the 7 prolines). The R_1 values were relatively uniform through all the sequence with a mean of 3.37 s⁻¹ and a standard deviation of 0.27 s⁻¹ (Fig. 4 A), excluding the first and last four residues in the sequence.

Conversely, the R₂ values showed a larger variation, with a mean of

7.18 s⁻¹ and a standard deviation of 1.78 s⁻¹, excluding the first and last four residues in the sequence. The residues with the largest R_2 values ($\geq 8 \text{ s}^{-1}$) were: Cys33, Ile39 and Ala 40 (all of them involved or close to the active site); Ile54, Lys58, Leu59, Asn60 and Thr61 (all of them at the C terminus of β -strand 3); and Ala94, Val95 and Leu100 (all of them located in the loop following β -strand 5 and at the N terminus of α 4). The large values of the R_2 in those residues indicate a high conformational exchange (i.e., in the microsecond-to-millisecond time regime) contribution [53].

The NOE values for residues involved in the elements of the secondary structure were not far from the theoretical one of 0.82, expected for an amide proton in a rigid environment at this field strength [53]. Smaller NOE ratios were observed at the N and C-terminal residues forming the elements of secondary structure as well as in many of the residues belonging to the regions connecting those elements of the secondary structure. For instance, the region preceding α 4 had smaller NOE values (and it also showed the largest R_2 values, see above). Furthermore, residues belonging to the active site also had smaller NOE ratios than the rest of amino acids. Apart from the N terminus of the protein, the residues with the lowest NOE values (< 0.3) were: Ile18 (at the C terminus of α 1) and Leu59 (at the end of β -strand 3).

An estimation of the correlation time of 2.1 ± 0.5 ns was obtained from a trimmed, averaged R_2/R_1 (excluding the last first and four terminal residues and those described above with large R_2 values). This value is smaller than that expected theoretically, for a 108-residue long protein, which should be ~4.3 ns [53]. Another examples of proteins with highly dynamic regions leading to anomalous estimations of the



Fig. 4. Relaxation parameters of TRX m. The R_1 (A), R_2 (B) and NOE (C) values for residues of TRX m acquired at 500 MHz (11.14 T). Error bars in R_1 and R_2 are from fitting to exponential curves. Errors in NOE are estimated to be 10 % of the reported value.

correlation times, when compared to the theoretical ones, have been described in the literature [80,81]. In fact, it has been argued that the presence of nonterminal residues with a small R_2/R_1 ratio suggests the presence of two internal motions on different time scales [82–84].

3.3. Conformational preferences and binding to TRX m of the FBPpep

Before studying the binding of FBPpep to TRX m of pea, we studied its conformational preferences in isolation, by using two different spectroscopic techniques.

3.3.1. The isolated FBPpep was mainly disordered

We used far-UV CD to elucidate whether there was some secondary structure in the isolated peptide. The spectrum of the peptide showed an intense minimum at \sim 202 nm (Fig. S2), indicating that it acquired mostly a random-coil conformation. The spectrum also showed a small shoulder at 222 nm, which could be attributed to the presence of the sole aromatic residue in the sequence, Tyr157 [85–87]. In fact, fluorescence spectra of isolated FBPpep showed a maximum wavelength at 308 nm, which is characteristic of polypeptides containing a fluorescent tyrosine residue (data not shown).

We assigned FBPpep by using 2D ¹H NMR sequential assignment methods (Table ST2). Several lines of evidence from the analyses of 2D ¹H NMR spectra further confirmed the disordered nature of FBPpep in isolation in aqueous solution. First, the conformational shifts ($\Delta\delta$) of H_a protons [62-64] for all the assigned residues were within the commonly accepted range for random-coil peptides ($\Delta \delta < 0.1$ ppm) (Table ST2). Second, no long- or medium-range NOEs were detected, but only sequential ones (i.e. αN (i, i + 1) and βN (i, i + 1)) (Fig. 5). However, at the second half of FBPpep (residues Thr165 to Gln179) there was evidence of weak, sequential contacts between the amide protons of some residues, namely Asn164-Thr165, Gly167-Thr168, Thr168-Glu169, Val175-Asn176, Val177-Ser178 (for which all their H_{α} protons had $\Delta \delta \leq 0.1$ ppm, Fig. 5). And third, the temperature coefficients of the amide protons, which measure their solvent-accessibility, were in all residues in the range of -6.0 to -9.0 ppb K⁻¹ (Table ST2), which are typical of solvent-exposed amide protons [65-67]; thus, all the amide protons of FBPpep were not hydrogen-bonded. A similar conclusion was reached when FBPpep was dissolved in D₂O and acquisition of a 1D ¹H NMR spectrum was carried out: no amide proton was observed after 5 min of dissolving the sample (data not shown).

3.3.2. Binding of FBPpep to TRX m

We measured the affinity of FBPpep in vitro for TRX m, by following a two-part experimental approach. First, we carried out a fluorescence titration to follow changes in the fluorescence of the protein upon peptide addition. And second, we carried out a 2D 1 H, 15 N HSQC NMR titration to follow the changes in the intensity or in chemical shifts of the spectrum of TRX m as the peptide concentration was increased.

In the fluorescence experiments, there was a decrease in the fluorescence as the peptide concentration was raised. The value of the K_d for the binding reaction from these experiments was $1.7\pm0.5\,\mu M$ (Fig. 6 A); the fact that binding can be followed by excitation at 295 nm indicates that at least one of the tryptophan residues of TRX m (namely, Trp13, Trp29 or Trp32) was affected by peptide binding. On the other hand, in the NMR experiments, the intensity of all the signals decreased in a



Fig. 5. NOE diagram of the FBPpep. NOEs are classified into strong, medium or weak, as represented by the height of the bar underneath the sequence; signal intensity was judged from the NOESY experiments by visual inspection. The dotted lines indicate NOE contacts that could not be unambiguously assigned due to overlapping with the chemical shifts of other nuclei. A white square indicates an $\alpha\delta(i, i + 1)$ NOEs involving the Pro in the row corresponding to the αN NOEs.



(caption on next column)

Fig. 6. Binding of FBPpep to TRX m as mapped by spectroscopic probes. (A) Titration curve obtained by monitoring the changes in the fluorescence at 315 nm when FBPpep was added to TRX m. The fluorescence intensity on the y-axis is the relative signal after removal of the corresponding blank. The line through the data is the fitting to Eq. (1). Experiments were carried out at 30 °C in phosphate buffer (50 mM), pH 5.8. The changes in the intensity of the backbone amide proton of Gly19 (B) and the indole chain of Trp13 (C) in the 2D ¹H, ¹⁵N HSQC NMR spectra upon addition of increasing concentrations of FBPpep. Experiments were carried out at 30 °C in phosphate buffer (50 mM), pH 5.8 in the presence of 1 mM DTT. The lines through the data of the intensities of the two selected cross-peaks are the corresponding fittings to Eq. (2).

sigmoidal way, as the FBPpep concentration was raised; furthermore, we did not observe any change in chemical shifts as the peptide concentration was increased. Furthermore, it seems there were no preferred regions in TRX m for peptide binding, or, alternatively, there was allosteric changes in all the molecule upon binding of the peptide. This general decrease in the intensity of all the signals precluded an unambiguous definition of the region of TRX m involved directly in peptide binding. Most of the sigmoidal variations of the cross-peaks intensities of TRX m could be fitted to Eq. (2), yielding similar K_d values, within the error. For instance, the K_d obtained from the intensity decrease in the cross-peak of the backbone amide proton of Gly19 was 104 \pm 50 μM (Fig. 6 B), and that obtained from the variation of the cross-peak intensity of the indole side-chain of Trp13 was 52 \pm 20 μM (Fig. 6 C). Although both values are similar within the error, the large fitting errors in the values from each cross-peak are probably due to: (i) the error in the intensities measured from the HSQC spectra (even after correcting for the values of the receiver gain); and, (ii) the absence of enough points where the binding curve is close to reach the saturation.

To sum up, both spectroscopic techniques unambiguously indicated that there was binding between TRX m and FBPpep, with a dissociation constant in the low micromolar range.

4. Discussion

4.1. Structure and dynamics of TRX m from pea

The solution structure of pea TRX m was similar to that of TRX m from spinach (which was also similar to the crystal structure of the same protein) [44]. However, one important difference with the structure of TRX m from spinach is that in this one, no NOE contacts were observed between β -strand 1 (residues 13 to 17) and β -strand 3 (residues 63 to 69) [44]; this absence of NOE contacts of β -strand 1 with the rest of the strands has also been observed in TRX h of *C. reinhardtii* [79]. Nevertheless, in the structure of TRX m from pea, NOE contacts were observed between the hydrogens involved in those strands (Fig. S1); for instance, we observed contacts between the NH of Lys58 and: (i) the side-chain of Val5; (ii) the amide proton of Val7 and the side-chain; and (iii) the H_{α} proton of Gln6, thus, pinpointing to the presence of this β -strand (Fig. S1).

Comparison with the solution structure of TRX m from spinach [44] showed two regions where the differences were more important. First, the active-site region and the following α 2 appeared more separated from the rest of the molecule in the TRX m from pea, and the first turns of α 2 did not appear to be well-formed (Fig. 2); in general terms, α 2 and α 4 appeared to be the less well-formed helices of TRX m from pea when compared to the similar helices of TRX m from spinach (Fig. 2). Furthermore, residues Cys33, Ile39 and Ala 40, at the N terminus of this helix, appeared to have a large flexibility as concluded from the large R_2 values (Fig. 4). And second, α 4 in TRX m from pea appeared also more separated from the rest of the molecule at its N terminus and in a less parallel-like arrangement with α 2 (Fig. 2) than in TRX m from spinach. In TRX m, we observed NOE contacts of the aromatic side-chain of Tyr50 (in α 2) with the side-chains of Val104, Glu105 and Lys106 (all of them in α 4), and we also detected NOEs of Ile42 with the side chain of Lys97;

that is, we observed NOEs with residues at the two termini of $\alpha 2$ with residues of $\alpha 4$. Interestingly enough, amino acids Ala94, Val95 and Leu100, at the N terminus of $\alpha 4$ had large R_2 values, indicating a high flexibility in the microsecond-to-millisecond time regime (Fig. 4 B). Taken together, these results suggest that the N-terminal of $\alpha 4$ in TRX m had a large flexibility and could move away from the rest of the protein. Movements of helical regions or of polypeptide patches nearby to them have been observed in: (a) a mutant of TRX of *Staphylococcus aureus*, involving a domain-swapping of the amino-terminal helix [88]; and (b) the wild-type PipX protein, where a helical region can adopt a flexed conformation in the free-target state of the protein involving residues with a high flexibility in the microsecond-to-millisecond time regime [89].

Another important difference with the TRX m from spinach is that the helical regions of TRX m from pea appeared more disordered and less regular. This could further support the hypothesis that TRX m from pea was more flexible. In fact, we observed that the picosecond-tonanosecond dynamics of the pea protein showed for the amino acids involved in the elements of secondary structure NOE values close to the number expected for a residue in a rigid environment (0.82), but also the polypeptide patches in-between those regions showed NOE values very small, indicating a high flexibility (Fig. 4 C). Furthermore, residues around the active site, at both termini of β -strand 3, and at the N-terminal region of a4 showed a high mobility in the microsecond-tomillisecond time regime (i.e., high R2 values). In addition, the fact that the trimmed, averaged R_2/R_1 yielded a lower correlation time value than that expected theoretically further supports the presence of several type of motions at different time scales for the amide protons [83,84]. Altogether, these findings suggest that the flexibility of pea TRX m was much larger than that of other members of the family. Although at this moment, we do not know the reasons behind such higher flexibility, it could be related to an improved ability to interact with a larger number of molecules, as it has been suggested in other small proteins such as PipX [89].

4.2. Binding of TRX m with the FBPpep

FBPase catalyzes the breakdown of fructose-1,6-bisphosphate to fructose-6-phosphate and inorganic phosphorous. FBPase activity is modulated by electron availability through the ferredoxin-thioredoxin system [26]: in fact, chloroplast FBPase can be activated by TRX f and TRX m [35,39,40,42]. It has been proposed that a long loop around residue 170 in FBPase is the region responsible for binding to either TRX f and TRX m [40]; this region contains a large amount of negatively charged residues (Asp and Glu), and appears partially disordered in the crystal structures of the FBPase from pea. Binding of the intact TRX m to FBPase, both from pea, resulted in a dissociation constant of 0.0188 µM (following a Koshland-Nemethy-Filmer model) [39]. In the case of spinach TRX f and FBPase, a K_d value of 800 nM has been reported for the non-covalent complex between the two proteins [90]. In this work, we wondered whether an isolated peptide derived from the interacting loop of FBPase could also bind to TRX m. Our results indicate that binding occurred with a dissociation constant in the range 1.3 (fluorescence)-50 µM (NMR); thus, there was binding between both molecules. At this stage, it is important to note that although NMR and fluorescence led to different values for the K_d , both of them are within the same range (low micromolar), and, those different values reflect the diverse range of protein concentrations employed in the experiments. The fact that the K_d was larger for FBPpep than for the intact FBPase indicates that, apart from the highly negatively charged loop region, other regions of FBP, and probably, its oligomeric state, are important in binding to TRX m (perhaps involving other hydrophobic interactions, as it has been suggested by protein engineering studies with the intact protein [40]). Furthermore, the lower affinity of the peptide for TRX m could indicate that since the peptide is disordered in isolation, as we have shown (Section 3.3.1), binding to TRX m also requires that the

peptide acquires a native-like conformation as that in the intact FBPase. Then, our measurements (Fig. 6) would be reporting a folding-uponbinding event.

Residues Arg37, Lys70, Arg74 and Lys97 of TRX m have been shown to be important in binding to intact FBPase [37]. This region is also involved in binding to other monomer of the TRX h, where dimers of TRXs have been observed in crystal structures [91]. However, from the NMR titration, all the residues of pea TRX m were involved in binding (Figs. 6 B and C), since all of them had a sigmoidal-like behavior with dissociation constants within the same order of magnitude upon FBPpep addition. Thus, binding of FBPpep seemed to induce allosteric conformational changes through the whole molecule.

5. Conclusion

To sum up, we have solved the solution structure of thioredoxin m of *Pisum sativum*. The folding and arrangement of secondary structure elements are similar to those found in other thioredoxins m of other plants, but there seems to be a higher flexibility at the α -helix 4. Binding to a fragment of short of FBPase, containing the thioredoxin binding site, occurs with an affinity lower than that between the two intact proteins, suggesting that there are other factors, apart from the residues involved in the TRX-binding site, modulating the binding between the two intact TRX and FBPase proteins.

Funding and acknowledgements

This research was funded by Consellería de Innovación, Universidades, Ciencia y Sociedad Digital (Generalitat Valenciana) [CIAICO 2021/0135 to ACA and JLN]. The funders had no role in the study design, data collection and analysis, decision to publish, or preparation of the manuscript.

The authors thank people in the PDBe (Sudakshina Ganguly) for helpful suggestions and corrections on the structure deposition. We thank María Angeles Jiménez from Instituto de Química Física Blas Cabrera (CSIC) in Madrid for helpful discussions.

The authors thank the two anonymous reviewers for helpful suggestions and discussions. The authors thank Prof. Aichun Dong for handling the manuscript.

Deposition

The chemical shifts of TRX m from pea (¹³C, ¹⁵N and ¹H) have been deposited in the BMRB under the accession number 52150. The coordinates of the 20 conformers with the lowest target function have been deposited in the PDB, together with the list of restrictions (distances and angles) used, under the accession number 8QPD.

CRediT authorship contribution statement

José L. Neira: Writing – review & editing, Writing – original draft, Resources, Methodology, Investigation, Funding acquisition, Formal analysis, Conceptualization. Martina Palomino-Schätzlein: Writing – review & editing, Methodology, Investigation, Formal analysis. Virginia Rejas: Writing – review & editing, Investigation, Formal analysis. José A. Traverso: Writing – review & editing, Resources, Methodology, Investigation, Formal analysis. Manual Rico: Conceptualization. Julio López-Gorgé: Resources. Ana Chueca: Resources. Ana Cámara-Artigas: Writing – review & editing, Writing – original draft, Resources, Methodology, Investigation, Funding acquisition, Formal analysis, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

The datasets generated during and/or analyzed during the current study are available from the corresponding authors on reasonable request.

Appendix A. Supplementary data

There are one figure describing the alignment of the β -strands in TRX m of pea and the observed NOEs (Fig. S1); the far-UV CD spectrum of isolated FBPpep (Fig. S2). Also, there are two tables containing the acquisition parameters of the full suite of acquired NMR spectra used to assign TRX m (Table ST1) and the ¹H assignment of FBPpep (Table ST2). Supplementary data to this article can be found online at https://doi.org/10.1016/j.ijbiomac.2024.129781.

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