En este anexo se presentan:
A. Los textos completos con las oraciones numeradas.
B. Matriz de cada texto analizado con las unidades léxicas que han establecido repetición, indicando el tipo de repetición.
C. Matriz de cada texto con la contabilización de unidades léxicas.
D. Tabla de cada texto analizado con el número de conexiones que las oraciones presentan.
E. Los textos resultantes después de eliminar las oraciones marginales.
F. Listado de las unidades léxicas que han establecido repetición en los dos tipos de texto analizados.
G. Los textos originales del corpus analizado.
H. Listado de abreviaciones utilizadas.

## 1. TEXTOS PERTENECIENTES A LAS INTRODUCCIONES DEL ARTÍCULO DE INVESTIGACIÓN.

## 1. 1. Texto 1: Bioconversion of solid food wastes to ethanol.

1. Energy and environmental issues take turns driving the development and use of alternative fuels for motor vehicles. 2. As the availability of petroleum-derived fuels and industrial feedstocks decreases owing to depletion and also economic and political developments, renewable sources of organic compounds are tested for their suitability as alternatives to petroleum-based substances. 3. Recent environmental concerns such as ozone non-attainment, solid waste management and control of toxic air pollutants have been other reasons for finding clean-burning alternative fuels.
2. Ethanol production from agricultural products has been in practice for the past 80 years. 5. Ethanol can be produced from many kinds of raw material that contains starch, sugar or cellulose. 6. Wastes from food processing industries represent a severe pollution problem and need better waste management techniques. 7. Utilization of food processing wastes to produce fuel alcohol with an increased efficiency has been under investigation in our laboratory for the past few years. 8. We were able to develop a novel and highly efficient cofermentation system for food wastes containing starch and lactose.
3. Fermentation is an anaerobic, energy-releasing transformation of carbohydrates by living organisms. 10. Yeast can ferment a wide variety of sugars and oligosaccharides other than glucose. 11. The D-hexoses and oligosaccharides fermented most often by yeast are glucose, mannose, fructose, galactose, maltose, lactose, melibiose, trehalose and raffinose. 12. The yeast in most widespread use for alcoholic fermentation is Saccharomyces cerevisiae. 13. Several studies on ethanol production via fermentation and the effects of different factors on the fermentation have been published in the past decade. 14. Utilization of cheese whey as the liquid portion of a fermenting corn mash has been investigated by Whalen et al. 15. Their work involved the fermentation of lactose/corn mash by the use of a dual yeast inoculum (Kluyveromyces marxianus and distillerís yeast). 16. This lactose/glucose cofermentation process took $60-72 \mathrm{~h}$ for completion. 17. We investigated the use of whey with bakery products and other starchy waste products by the application of lactose hydrolysis in conjunction with a single yeast inoculum to reduce the fermentation time and an increase in alcohol yield.
4. The objectives of this work were to study the effect of low- and hightemperature enzymes on hydrolysis of food wastes, to compare the fermentation of bakery products with mixed waste products and to study the cofermentation of cheese whey and starchy food wastes.

## 1. 1. 1. Matriz de repetición de unidades léxicas.

| 2 | rs. alternative alternatives rs. fuels - fuels | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. <br> environmental environmental rs. alternative alternative <br> rs. fuels - fuels | rs. fuels - fuels rs. alternatives alternative | 3 |  |  |
| 4 |  |  |  | 4 |  |
| 5 |  |  |  | rs. ethanol - ethanol <br> rc. production - <br> produced <br> psm. products - material | 5 |
| 6 |  |  | rs. waste - waste <br> rs. management management rc. pollutants pollution |  |  |
| 7 | rs. fuels - fuel | rs. fuels - fuel | rs. fuels - fuel | hip. ethanol - alcohol <br> rc. production - produce <br> rs. past - past <br> rs. years - years | hip. ethanol alcohol rs. produced produce |
| 8 |  |  | rs. waste - wastes |  | rs. containing containing <br> rs. starch - starch |
| 9 |  |  |  |  |  |
| 10 |  |  |  |  | rs. sugar - sugars |
| 11 |  |  |  |  |  |
| 12 |  |  |  |  |  |
| 13 |  |  |  | rs. ethanol - ethanol <br> rs. production production <br> rs. past - past | rs. ethanol - ethanol rc. produced production |
| 14 |  |  |  |  |  |
| 15 |  |  |  |  |  |
| 16 |  |  |  |  |  |


| 17 |  |  | rs. waste - waste | hip. ethanol - alcohol <br> psp. production - yield <br> rs. products - products | hip. ethanol - alcohol <br> rc. produced - products |
| :---: | :--- | :--- | :--- | :--- | :--- |
| 18 |  |  | rs. waste - <br> wastes | rs. products - products | rc. produced - products <br> rc. starch - starchy |


| 7 | rs. wastes wastes <br> rs. food - food rs. processing processing | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | rs. wastes wastes rs. food - food | rs. food - food <br> rs. wastes wastes <br> rc. efficiency efficient rs. our - we+ | 8 |  |  |
| 9 |  |  | rc. cofermentation fermentation | 9 |  |
| 10 |  |  | rc. cofermentation ferment | rc. fermentation ferment | 10 |
| 11 |  |  | rc. cofermentation fermented <br> rs. lactose - lactose | rc. fermentation fermented | rs. yeast - yeast <br> rs. ferment - fermented <br> rs. oligosaccharides oligosaccharides <br> rs. glucose - glucose |
| 12 |  | psm. utilization use rc. alcohol alcoholic | rc. cofermentation fermentation | rs. fermentation fermentation | rs. yeast - yeast rc. ferment fermentation |
| 13 |  | tr. alcohol ethanol rc. produce production psm. investigation studies rs. past - past | rc. cofermentation fermentation | rs. fermentation fermentation | rc. ferment fermentation |
| 14 |  | rs. utilization utilization rc. investigation - investigated | rc. cofermentation fermenting | rc. fermentation fermenting | rs. ferment - fermenting |
| 15 |  |  | rc. cofermentation fermentation rs. lactose - lactose | rs. fermentation fermentation | rs. yeast - yeast rc. ferment fermentation |
| 16 |  |  | rs. cofermentation cofermentation rs. lactose - lactose | rc. fermentation cofermentation | rc. ferment cofermentation rs. glucose - glucose |

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| 17 |  | psm. utilization use <br> rs. wastes waste rc. investigation - investigated rs. our - we+ | rs. we - we + <br> rc. cofermentation - <br> fermentation <br> hip. food - products <br> rs. wastes - waste <br> rc. starch - starchy <br> rs. lactose - lactose | rs. fermentation fermentation | rs. yeast - yeast rc. ferment fermentation |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 18 | rs. wastes - wastes rs. food - food | rs. food - food <br> rs. wastes wastes pc. investigation - study (study) | rc. cofermentation fermentation rs. food - food rs. wastes - wastes rc. starch - starchy | rs. fermentation fermentation | rc. ferment fermentation |



| 16 |  |  |
| :--- | :--- | :--- |
| 17 | rs. lactose - lactose <br> rc. cofermentation - fermentation | rc. cofermentation - fermentation |
| 18 | psm. investigated - study <br> rs. whey - whey <br> rs. bakery - bakery <br> rs. products - products <br> rs. starchy - starchy <br> r. waste - wastes <br> rs. fermentation - fermentation |  |

## 1. 1. 2 . Matriz con el número de unidades léxicas.




## 1. 1. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 1)[1]$ | 2. $(0,0)[0]$ | 3. $(1,1)[2]$ |
| :--- | :--- | :--- |
| 4. $(0,4)[4]$ | 5. $(1,0)[1]$ | 6. $(1,1)[2]$ |
| 7. $(2,4)[6]$ | 8. $(1,2)[3]$ | 9. $(0,0)[0]$ |
| 10. $(0,1)[1]$ | 11. $(1,3)[4]$ | 12. $(0,1)[1]$ |
| 13. $(2,2)[4]$ | 14. $(0,3)[3]$ | 15. $(2,1)[3]$ |
| 16. $(1,0)[1]$ | 17. $(8,1)[9]$ | 18. $(5,-)[5]$ |

## 1. 1. 4. Texto resultante tras eliminar las oraciones marginales.

1. Energy and environmental issues take turns driving the development and use of alternative fuels for motor vehicles. 3. Recent environmental concerns such as ozone non-attainment, solid waste management and control of toxic air pollutants have been other reasons for finding clean-burning alternative fuels.
2. Ethanol production from agricultural products has been in practice for the past 80 years. 5. Ethanol can be produced from many kinds of raw material that contains starch, sugar or cellulose. 6. Wastes from food processing industries represent a severe pollution problem and need better waste management techniques. 7. Utilization of food processing wastes to produce fuel alcohol with an increased efficiency has been under investigation in our laboratory for the past few years. 8. We were able to develop a novel and highly efficient cofermentation system for food wastes containing starch and lactose.
3. Yeast can ferment a wide variety of sugars and oligosaccharides other than glucose. 11. The D-hexoses and oligosaccharides fermented most often by yeast are glucose, mannose, fructose, galactose, maltose, lactose, melibiose, trehalose and raffinose. 12. The yeast in most widespread use for alcoholic fermentation is Saccharomyces cerevisiae. 13. Several studies on ethanol production via fermentation and the effects of different factors on the fermentation have been published in the past decade. 14. Utilization of cheese whey as the liquid portion of a fermenting corn mash has been investigated by Whalen et al. 15. Their work involved the fermentation of lactose/corn mash by the use of a dual yeast inoculum (Kluyveromyces marxianus and distillerís yeast). 16. This lactose/glucose cofermentation process took $60-72 \mathrm{~h}$ for completion. 17. We investigated the use of whey with bakery products and other starchy waste products by the application of lactose hydrolysis in conjunction with a single yeast inoculum to reduce the fermentation time and an increase in alcohol yield.
4. The objectives of this work were to study the effect of low- and hightemperature enzymes on hydrolysis of food wastes, to compare the fermentation of bakery products with mixed waste products and to study the cofermentation of cheese whey and starchy food wastes.

## 1. 2. Texto 2: Speciation as an analytical aid in trace element research in infant nutrition.

1. During the prenatal period, the fetus is supplied with minerals and trace elements via maternal circulation and controlled placental transfer. 2. After separation from the mother, the newborn has to develop its own functions and regulatory systems, including respiration, digestion and immune defenses. 3. Trace elements are involved in the form of metalloproteins and enzymes at all stages in the development of these processes. 4. Infancy is further characterized by an extremely high rate of synthesis of tissue cells, which leads to the infant's doubling its birth mass in a period of only 4 months. 5. The infant's trace element requirement is supplied not only by amounts transferred via the mother's milk in specific binding forms or by formula, but also from prenatal stores. 6. Special attention must be paid to very low birth mass, premature infants because they are born with lower stores of essential micronutrients. 7. Trace elements must be added to pre-term infants' formulas to satisfy their higher dietary requirements.
2. In early infancy, breast milk or cow's-milk-based and soy- based formulas are the only dietary source of essential trace elements. 9. The mother's milk provides an adequate supply of all micronutrients for the full-term infant. 10. The concentrations and the fairly well defined binding pattern of the essential trace elements in human milk are therefore used as a reference. 11. On the other hand, the trace elements chromium, copper, zinc, iron, manganese, molybdenum, iodine and, recently, selenium have been added to the formulas as compounds and at concentration levels that are different from those found in breast milk. 12. With the sole exception of selenium, the trace element intake of infants via formula is significantly higher than via breast milk. 13. The iron supply was found to be up to 20 times higher despite the fact that the high hemoglobin of newborns forms a reservoir. 14. In the case of manganese, the supply of the formulafed infant can be as much as 1000 times higher than that of the breast-fed infant. 15. During the first months of life such high values are critical with respect to known Fe $\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ interactions. 16. As negative effects of high iron supplementation ( $>4 \mathrm{mg} 1-^{1}$ ) significantly lower levels of glutathione peroxidase in serum and superoxide dismutase in erythrocytes have been observed in formula-fed infants. 17. In addition given the prooxidant effects of excessive amounts of iron in the iron(II) form,
the balance between the formation and inactivation of free radicals generated by the rapid growth rate of premature newborns during the first months of life might be disturbed.
3. In spite of the significantly lower trace element intake of breast-fed infants, their serum concentrations of the essential elements $\mathrm{Cu}, \mathrm{Fe}$ and Zn are comparable to those of formula-fed infants. 19. Further, mass and length gains, as developmental parameters, were comparable for the two groups over a period of 4 months. 20 Because no signs of deficiency were observed in breast-fed infants, the bio availibility of copper, iron and zinc of the special binding proteins in human milk must be considerably higher than that in cow's milk or soy-based formula.
4. In the light of these facts, we considered it of importance to investigate the concentration, chemical form and nutritive value of trace elements in both human milk and infant formulas, with our ultimate goal being to obtain as much information as possible about adequate infant nutrition. 22. We therefore carried out speciation studies to determine the binding form of trace elements in these nutritive fluids, combining methods for protein separation with methods for trace element determination in the eluted fractions.

## 1. 2. 1. Matriz de repetición de unidades léxicas.

| 2 | pc. maternal mother (motherly) | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. trace - trace rs. elements elements | rc. develop development <br> d. own functions .... defenses - these | 3 |  |  |
| 4 |  | psm. newborn infant |  | 4 |  |
| 5 | rs. prenatal prenatal <br> rs. supplied supplied rs. trace - trace rs. elements element rs. via - via pc. maternal mother (motherly) rc. transfer transferred | rs. mother mother <br> psm. newborn infant | rs. trace trace rs. elements element | rs. infant infant | 5 |
| 6 | psm. trace elements micronutrients | psm. newborn infants | psm. trace elements micronutrients | rs. infant infants rs. birth birth rs. mass mass | rs. infant - infants psm. trace element micronutrients rs. stores - stores |
| 7 | rs. trace - trace rs. elements elements | psm. newborn infants | rs. trace trace rs. elements elements | rs. infant infants | rs. infant - infants <br> rs. trace - trace <br> rs. element - elements <br> rs. requirement - requirements <br> rs. formula - formulas |
| 8 | rs. trace - trace <br> rs. elements elements tr. maternal breast | tr. mother - breast pc. newborn infancy (infant) | rs. trace trace rs. elements elements | rs. infancy <br> - infancy | rc. infant - infancy <br> rs. trace - trace <br> rs. element - elements <br> psm. mother - breast <br> rs. milk - milk <br> rs. formula - formulas |
| 9 | rc. supplied supply psm. trace elements micronutrients pc. maternal mother (motherly) | rs. mother mother psm. newborn infant | psm. trace elements micronutrients | rs. infant infant | rs. infant - infant psm. trace elements micronutrients rc. supplied - supply rs. mother - mother rs. milk - milk |
| 10 | rs. trace - trace <br> rs. elements elements <br> tr. maternal human | tr. mother - human | rs. trace trace rs. elements elements |  | rs. trace - trace <br> rs. element - elements <br> hip. mother - human rs. milk - milk psm. specific - defined rs. binding - binding psm. forms - pattern |

$\qquad$

| 11 | rs. trace - trace <br> rs. elements elements tr. maternal breast | tr. mother breast | rs. trace - trace rs. elements elements |  | rs. trace - trace <br> rs. element - elements <br> psm. mother - breast <br> rs. milk - milk <br> rs. formula - formulas |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 12 | rs. trace - trace rs. elements element rs. via - via tr. maternal breast | tr. mother breast psm. newborn - infants | rs. trace - trace rs. elements element | rs. infant - infants | rs. infant - infants <br> rs. trace - trace <br> rs. element - element <br> rs. via - via <br> psm. mother - breast <br> rs. milk - milk <br> rs. formula - formula |
| 13 | rc. supplied supply tr. trace elements - iron | rs. newborn newborns | tr. trace elements iron | psm. infant newborns | psm. infant - newborns tr. trace element - iron rc. supplied - supply psm. stores - reservoir |
| 14 | rc. supplied supply tr. trace elements manganese tr. maternal breast | tr. mother breast psm. newborn - infant | tr. trace elements manganese | rs. infant - infant | rs. infant - infant <br> tr. trace element manganese <br> rc. supplied - supply psm. mother - breast rs. formula - formula |
| 15 | tr. trace elements - Fe $\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ |  | tr. trace elements -$\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ |  | tr. trace elements $-\mathrm{Fe}-$ $\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ |
| 16 | tr. trace elements - iron | psm. newborn - infants | tr. trace elements iron | rs. infant - infants | rs. infant - infants tr. trace element - iron rs. formula - formula |
| 17 | tr. trace elements - iron | rs. newborn newborns | tr. trace elements iron | psm. infant newborns rs. months months | psm. infant - newborns tr. trace element - iron rs. amounts - amounts |
| 18 | rs. trace - trace <br> rs. elements element tr. maternal breast | tr. mother breast psm. newborn - infants | rs. trace - trace rs. elements element | rs. infant - infants | rs. infant - infants <br> rs. trace - trace <br> rs. element - element <br> psm. mother - breast <br> rs. formula - formula |
| 19 |  |  |  | rs. mass - mass <br> rs. period - period <br> rs. 4-4 <br> rs. months months |  |
| 20 | tr. trace elements copper, iron and zinc tr. maternal breast | tr. mother breast psm. newborn - infants | tr. trace elements copper, iron and zinc <br> rc. metalloproteins - proteins | rs. infant - infants | rs. infant - infants <br> tr. trace element copper, iron and zinc hip. mother - human rs. milk - milk psm. specific - special* rs. binding - binding* rs. formula - formula |

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| 21 | rs. trace - trace <br> rs. elements elements <br> tr. maternal human | tr. mother human psm. newborn - infant | rs. trace - trace rs. elements elements | rs. infant - infant | rs. infant - infant <br> rs. trace - trace <br> rs. element - elements <br> hip. mother - human <br> rs. milk - milk <br> rs. forms - form* <br> rs. formula - formulas |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 22 | rs. trace - trace rs. elements elements |  | rs. trace - trace <br> rs. elements elements rc. metalloproteins - protein |  | rs. trace - trace <br> rs. element - elements <br> hip. milk - fluids <br> rs. binding - binding <br> rs. forms - form |

## 6

| 7 | psm. premature - pre-term rs. infants infants psm. micronutrients trace elements | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | rc. infants infancy rs. essential essential psm. micronutrients trace elements | rs. trace trace rs. elements elements rc. infants infancy rs. formulas formulas rs. dietary dietary | 8 |  |  |
| 9 | a. premature -full-term rs. infants infant rs. micronutrients micronutrients | psm. trace elements micronutrients <br> a. pre-term -full-term rs. infants infant | rc. infancy infant <br> psm. breast mother rs. milk - milk psm. trace elements micronutrients | 9 |  |
| 10 | rs. essential essential psm. <br> micronutrients trace elements | rs. trace trace rs. elements elements | tr. breast - human <br> rs. milk - milk <br> rs. essential essential <br> rs. trace - trace <br> rs. elements elements | hip. mother - human rs. milk - milk psm. micronutrients trace elements | 10 |
| 11 | psm. <br> micronutrients - <br> trace elements | rs. trace trace rs. elements elements rs. added added rs. formulas formulas | rs. breast - breast <br> rs. milk - milk <br> rs. formulas formulas <br> rs. trace - trace <br> rs. elements elements | psm. mother - breast rs. milk - milk psm. micronutrients trace elements | rs. concentrations concentration rs. trace - trace rs. elements elements tr. human - breast rs. milk - milk |
| 12 | rs. infants infants <br> psm. <br> micronutrients - <br> trace element | rs. trace trace rs. elements element rs. infants infants rs. formulas formula | rc. infancy infants <br> rs. breast - breast <br> rs. milk - milk <br> rs. formulas formula <br> rs. trace - trace rs. elements element | psm. mother - breast rs. milk - milk psm. micronutrients trace elements rs. infant - infants | rs. trace - trace <br> rs. elements - element <br> tr. human - breast <br> rs. milk - milk |
| 13 | psm. infants newborns tr. micronutrients iron | tr. trace elements iron psm. infants newborns | pc. infancy newborns (infant) tr. trace elements - iron | rs. supply - supply tr. micronutrients iron psm. infant newborns | tr. trace elements iron |

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|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. infants - infant tr. micronutrients manganese | tr. trace elements manganese rs. infants infant rs. formulas formula | rc. infancy - infant <br> rs. breast - breast <br> rs. formulas formula <br> tr. trace elements manganese | psm. mother breast rs. supply - supply tr. micronutrients manganese rs. infant - infant | tr. trace elements - manganese <br> tr. human - breast |
| 15 | tr. micronutrients - $\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ | tr. trace elements - Fe $\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ | tr. trace elements -$\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ | tr. micronutrients -$\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ | tr. trace elements - $\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ |
| 16 | rs. infants - infants <br> tr. micronutrients <br> - iron | tr. trace elements - iron rs. infants infants rs. formulas formula | rc. infancy infants <br> rs. formulas formula <br> tr. trace elements iron | tr. micronutrients iron <br> rs. infant - infants | tr. trace elements - iron |
| 17 | rs. premature premature psm. infants newborns tr. micronutrients - iron | tr. trace elements - iron psm. pre-term premature psm. infants newborns | pc. infancy newborns (infant) tr. trace elements iron | tr. micronutrients iron psm. infant newborns | tr. trace elements - iron |
| 18 | rs. infants - infants <br> rs. essential - <br> essential <br> psm. <br> micronutrients - <br> trace element | rs. trace - trace <br> rs. elements element rs. infants infants rs. formulas formula | rc. infancy infants <br> rs. breast - breast <br> rs. formulas formula <br> rs. essential essential <br> rs. trace - trace rs. elements element | psm. mother - <br> breast <br> psm. <br> micronutrients - <br> trace elements <br> rs. infant - infants | rs. concentrations <br> - concentrations <br> rs. essential essential <br> rs. trace - trace <br> rs. elements element <br> tr. human - breast |
| 19 | rs. mass - mass* |  |  |  |  |
| 20 | rs. infants - infants tr. micronutrients copper, iron and zinc | tr. trace elements copper, iron and zinc rs. infants infants rs. formulas formula | rc. infancy infants rs. breast - breast rs. milk - milk <br> rs. cow - cow <br> rs. soy - soy <br> rs. based - based <br> rs. formulas - <br> formula <br> tr. trace elements copper, iron and zinc | psm. mother breast <br> rs. milk - milk <br> tr. micronutrients copper, iron and zinc rs. infant - infants | psm defined special* <br> rs. binding binding* <br> tr. trace elements - copper, iron and zinc <br> rs. human human <br> rs. milk - milk |


| 6 |  | 7 | 9 | 10 |
| :--- | :--- | :--- | :--- | :--- | :--- |

## Anexo

| 12 | rs. trace - trace <br> rs. elements element <br> rs. selenium selenium <br> rs. formulas formula <br> rs. breast - breast <br> rs. milk - milk | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | rs. iron iron <br> rs. found - found | tr. trace element <br> - iron <br> psm. infants - <br> newborns <br> rs. higher higher <br> e. than via breast milk - 0 | 13 |  |  |
| 14 | rs. manganese manganese rs. formulas formula rs. breast - breast | tr. trace element <br> - manganese <br> rs. infants infant <br> rs. formula formula <br> rs. higher higher <br> rs. breast breast | tr. iron manganese rs. supply supply tr. 20-1000 rs. times - times rs. higher higher psm. newborns infant | 14 |  |
| 15 | psm. copper -Cu <br> psm. zinc - Zn <br> psm. iron -Fe <br> psm. manganese Mn | tr. trace element <br> $-\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ rs. higher - high | psm. iron -Fe hip. 20 - values rs. higher - high | psm. manganese - <br> Mn <br> hip. 1000 - values <br> rs. higher - high | 15 |
| 16 | rs. iron - iron <br> rs. formulas formula <br> rs. levels - levels* | tr. trace element <br> - iron <br> rs. infants infants <br> rs. formula formula rs. higher - high | rs. iron - iron rs. higher - high psm. newborns infants | tr. manganese iron <br> rs. formula formula <br> rs. fed - fed <br> rs. infant - infants | rs. high - high psm. Fe - iron |
| 17 | rs. iron - iron | tr. trace element - iron psm. infants newborns | rs. iron - iron <br> rs. newborns newborns | tr. manganese iron psm. infant newborns | rs. first - first <br> rs. months - months <br> rs. life - life <br> psm. Fe - iron |


| 11 |  | 12 | 12 |
| :--- | :--- | :--- | :--- | :--- | :--- |

## Anexo



## 1. 2. 2. Matriz con el número de unidades.




1. 2. 3. Tabla representativa del número de conexiones entre oraciones.
1. $(-, 2)[2]$
2. $(0,0)$ [0]
3. $(0,0)[0]$
4. $(0,1)[1]$
5. $(1,12)[13]$
6. $(0,0)[0]$
7. $(1,5)[6]$
8. $(2,8)[10]$
9. $(2,4)[6]$
10. $(4,5)[9]$
11. $(7,6)[13]$
12. $(2,5)(2,6)[7][8]$
13. $(5,4)[9]$
14. $(1,2)[3]$
15. $(2,1)[3]$
16. $(1,0)[1]$
17. $(9,2)$ [11]
18. $(1,0)[1]$
19. $(8,2)(9,2)[10][11]$
20. $(10,1)[11]$
21. (4,-) [4]

## 1. 2. 4. Texto resultante tras eliminar las oraciones marginales.

1. During the prenatal period, the fetus is supplied with minerals and trace elements via maternal circulation and controlled placental transfer. 4. Infancy is further characterized by an extremely high rate of synthesis of tissue cells, which leads to the infant's doubling its birth mass in a period of only 4 months. 5. The infant's trace element requirement is supplied not only by amounts transferred via the mother's milk in specific binding forms or by formula, but also from prenatal stores. 7. Trace elements must be added to pre-term infants' formulas to satisfy their higher dietary requirements.
2. In early infancy, breast milk or cow's-milk-based and soy- based formulas are the only dietary source of essential trace elements. 9. The mother's milk provides an adequate supply of all micronutrients for the full-term infant. 10. The concentrations and the fairly well defined binding pattern of the essential trace elements in human milk are therefore used as a reference. 11. On the other hand, the trace elements chromium, copper, zinc, iron, manganese, molybdenum, iodine and, recently, selenium have been added to the formulas as compounds and at concentration levels that are
different from those found in breast milk. 12. With the sole exception of selenium, the trace element intake of infants via formula is significantly higher than via breast milk. 13. The iron supply was found to be up to 20 times higher despite the fact that the high hemoglobin of newborns forms a reservoir. 14. In the case of manganese, the supply of the formula- fed infant can be as much as 1000 times higher than that of the breast-fed infant. 15. During the first months of life such high values are critical with respect to known $\mathrm{Fe}-\mathrm{Zn}, \mathrm{Fe}-\mathrm{Cu}$ and $\mathrm{Mn}-\mathrm{Fe}$ interactions. 16. As negative effects of high iron supplementation ( $>4 \mathrm{mg} \mathrm{T}^{-1}$ ) significantly lower levels of glutathione peroxidase in serum and superoxide dismutase in erythrocytes have been observed in formula-fed infants. 17. In addition given the prooxidant effects of excessive amounts of iron in the iron(II) form, the balance between the formation and inactivation of free radicals generated by the rapid growth rate of premature newborns during the first months of life might be disturbed.
3. In spite of the significantly lower trace element intake of breast-fed infants, their serum concentrations of the essential elements $\mathrm{Cu}, \mathrm{Fe}$ and Zn are comparable to those of formula-fed infants. 19. Further, mass and length gains, as developmental parameters, were comparable for the two groups over a period of 4 months. 20. Because no signs of deficiency were observed in breast-fed infants, the bio availibility of copper, iron and zinc of the special binding proteins in human milk must be considerably higher than that in cow's milk or soy-based formula.
4. In the light of these facts, we considered it of importance to investigate the concentration, chemical form and nutritive value of trace elements in both human milk and infant formulas, with our ultimate goal being to obtain as much information as possible about adequate infant nutrition. 22. We therefore carried out speciation studies to determine the binding form of trace elements in these nutritive fluids, combining methods for protein separation with methods for trace element determination in the eluted fractions.

## 1. 3. Texto 3: Analysis of carbonaceous aerosols: interlaboratory comparison.

1. Many workplace and environmental settings contain aerosols composed primarily of carbon. 2. Cabonaceous aerosols encountered in these settings include asphalt fumes, oil mists, cigarette and wood smokes, carbon black, and diesel exhaust. 3. Some of these aerosols are known or suspect human carcinogens (e,g., cigarette smoke and diesel exhaust, respectively) and have been linked to other adverse health effects (e.g., asthma, heart disease) 4. Exposure to diesel exhaust is of particular concern because it has been classified a probable human carcinogen and diesel equipment use is widespread in (e,g., trucking, transit, mining, railroads, agriculture). 5. An estimated 1.35 million workers are routinely exposed to diesel engine exhaust and exposures in some industries are relatively high (e.g., $>0.5 \mathrm{mg} \mathrm{m}^{-3}$ ). 6. Unfortunately, health-based exposure criteria for diesel particulate have not yet been established. 7. A Threshold Limit Value (TLV) of $150 \mu / \mathrm{m}^{-3}$ has been proposed but has not yet been adopted.
2. Particulate diesel exhaust, like fine particulate air pollution in general, also is of concern with respect to noncancer health effects. 9. The US Environmental Protection Agency (EAP) has proposed an inhalation Reference Concentration (RfC) of $5 \mu / \mathrm{m}^{-3}$ for the noncancer health effects of diesel exhaust and the State of California Office of Environmental Health Hazard Assessment (OEHHA) has proposed adoption of this value for the chronic inhalation reference exposure level in California. 10. The RfC for a substance is an estimate of a daily exposure of humans, including sensitive subgroups, that is 'likely to be without appreciable risk of deleterious effects during a lifetime of exposure'. 11. Comprehensive reviews of the potential health effects of exposure to diesel exhaust exposure have been recently published.
3. Because diesel exhaust is a chemically complex mixture containing thousands of compounds, some measure of exposure must be selected. 13. Given the high carbon content of diesel particulate, a carbon-based method was investigated. 14. The method, recently published as National Institute for safety and Health (NIOSH) Method 5040, is based on an evolved gas analysis technique called the 'thermal-optical method'. 15. With this technique, speciation of organic and elemental carbon (OC and EC, respectively) is accomplished through temperature and atmosphere control and by an optical feature that corrects for pyrolytically generated carbon, or 'char', formed during the analysis of some materials. 16. Although both organic and elemental carbon are determined in the analysis, EC is the superior marker of diesel particulate because it constitutes a large fraction of the particulate mass, it can be quantified at background (i.e., environmental) levels, and its only significant source in most workplaces is the diesel engine. 17. Different approaches can be applied for OC-EC analysis, but a thermal-optical method was selected because the instrumentation has desirable design features not present in other carbon analyzers. 18. An in-depth discussion on Method 5040, including both technical and exposure-related issues, has been published elsewhere.
4. In a previous study, different methods gave widely varying results in the speciation of organic and elemental carbon. 20. For this reason, OC-EC methods are considered operational in the sense that the method itself defines the analyte. 21. Given its operational nature, it is important to examine interlaboratory variability of the method; however, when the thermal-optical method was initially evaluated, only one
instrument was available, so interlaboratory variability could not be examined. 22. More recently, additional instruments were constructed by a commercial laboratory and an interlaboratory comparison was conducted. 23. Seven laboratories that perform thermaloptical analysis participated in the comparison. 24. Six of them used NIOSH Method 5040 (i.e., they used identical instrumentation and thermal program), while the seventh used a variation on the method. 25. Another thermal technique based on coulometric detection of $\mathrm{CO}_{2}$ is being used in Europe for occupational monitoring of diesel particulate. 26. Four laboratories employing the coulometric method also participated in the interlaboratory comparison, giving a total of eleven laboratories (seven thermaloptical and four coulometric). 27. Discussion of the methods and a summary of the results of the intercomparison are reported in this paper.

## 1. 3. 1. Matriz de repetición de unidades léxicas.



| 1 |  | 3 |  | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 10 |  |  | rs. human - humans psm. adverse deleterious rs. effects - effects | rs. exposure exposure rs. human humans | hip. workers humans rs. exposures exposure |
| 11 |  | rs. diesel - diesel rs. exhaust exhaust | rs. diesel - diesel <br> rs. exhaust exhaust <br> rs. health - health <br> rs. effects - effects | rs. exposure exposure rs. diesel - diesel rs. exhaust exhaust | rs. diesel - diesel <br> rs. exhaust - exhaust <br> rs. exposures exposure |
| 12 | rs. contain containing* | rs. diesel - diesel rs. exhaust exhaust | rs. diesel - diesel rs. exhaust exhaust | rs. exposure exposure <br> rs. diesel diesel <br> rs. exhaust exhaust | rs. diesel - diesel <br> rs. exhaust - exhaust <br> rs. exposures exposure |
| 13 | rc. contain content rs. carbon carbon | rc. carbonaceous <br> - carbon <br> rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel |
| 14 |  |  |  |  |  |
| 15 | rs. carbon carbon | rc. carbonaceous <br> - carbon |  |  |  |
| 16 | rs. workplace workplaces <br> rs. environmental - environmental hip. aerosols mass <br> rs. carbon carbon | rc. carbonaceous <br> - carbon <br> hip. aerosols - <br> mass <br> rs. diesel - diesel | hip. aerosols - mass rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel <br> rs. engine - engine |
| 17 | rs. carbon carbon | rc. carbonaceous <br> - carbon |  |  |  |
| 18 |  |  |  | rs. exposure exposure | rs. exposures exposure |
| 19 | rs. carbon carbon | rc. carbonaceous <br> - carbon |  |  |  |
| 20 | psm. carbon - C | pc. carbonaceous <br> - C (carbon) |  |  |  |
| 21 |  |  |  |  |  |
| 22 |  |  |  |  |  |
| 23 |  |  |  |  |  |
| 24 |  |  |  |  |  |
| 25 |  | rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel |
| 26 |  |  |  |  |  |
| 27 |  |  |  |  |  |

## Anexo

| 6 |  | 7 | 8 |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 7 |  |  |  |  |  |
| 8 | rs. health health rs. diesel - diesel rs. particulate particulate |  |  |  |  |
| 9 | rs. health health rs. exposure exposure rs. diesel - diesel | rs. value value* <br> rs. proposed <br> - proposed rc. adopted adoption | rs. diesel - diesel <br> rs. exhaust - exhaust <br> rs. noncancer noncancer <br> rs. health - health <br> rs. effects - effects | 9 |  |
| 10 | rs. exposure exposure |  | rs. effects - effects | rs. $\mathrm{RfC}-\mathrm{RfC}$ <br> rs. effects - effects <br> rs. exposure exposure | 10 |
| 11 | rs. health health rs. exposure exposure r. diesel - diesel |  | rs. diesel - diesel <br> rs. exhaust - exhaust <br> rs. health - health <br> rs. effects - effects | rs. health - health <br> rs. effects - effects <br> rs. diesel - diesel <br> rs. exhaust - <br> exhaust <br> rs. exposure exposure | rs. effects - effects <br> rs. exposure exposure |
| 12 | rs. exposure exposure psm. criteria measure rs. diesel - diesel |  | rs. diesel - diesel <br> rs. exhaust - exhaust | rs. diesel - diesel <br> rs. exhaust exhaust <br> rs. exposure exposure | psm. substance compounds <br> rs. exposure exposure |
| 13 | rs. diesel - diesel rs. particulate particulate |  | rs. diesel - diesel rs. particulate particulate | rs. diesel - diesel |  |
| 14 | rs. health health* |  | rs. health - health* | rs. health - health* |  |
| 15 |  |  |  |  |  |
| 16 | rs. diesel - diesel rs. particulate particulate |  | rs. diesel - diesel rs. particulate particulate | rs. diesel - diesel rs. level-levels |  |
| 17 |  |  |  |  |  |
| 18 | rs. exposure exposure |  |  | rs. exposure exposure | rs. exposure exposure |
| 19 |  |  |  |  |  |
| 20 |  |  |  |  |  |
| 21 |  |  |  |  |  |
| 22 |  |  |  |  |  |


| 6 |  | 7 |  | 8 |  |
| :--- | :--- | :--- | :--- | :--- | :--- |

## Anexo



| 11 |  | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 21 |  |  | rs. method method psm. investigated examined | rs. thermal thermal rs. optical - optical rs. method method | rs. optical optical |
| 22 |  |  |  |  |  |
| 23 |  |  |  | rs. analysis analysis <br> rs. thermal thermal rs. optical - optical | psp. is accomplished perform rs. optical optical rs. analysis analysis |
| 24 |  |  | rs. method method | rs. NIOSH NIOSH rs. method method rs. $5040-5040$ rs. thermal thermal |  |
| 25 | rs. diesel - diesel | rs. diesel - diesel | rs. diesel - diesel rs. particulate particulate rs. based based* | rs. based - based <br> rs. technique technique <br> rs. thermal thermal | rs. technique technique |
| 26 |  |  | rs. method method* | tr. technique method rs. thermal thermal rs. optical - optical | rs. opticaloptical |
| 27 | psm. published reported* |  | rs. method methods | rs. method method psm. published reported |  |

## Anexo

| 17 | psp. organic carbon - OC <br> rs. $\mathrm{EC}-\mathrm{EC}$ <br> rs. analysis analysis | 17 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 18 |  | rs. method - method | 18 |  |  |
| 19 | rs. organic organic rs. elemental elemental rs. carbon - carbon | psp. OC - organic <br> carbon <br> psp. EC - elemental <br> carbon <br> rs. method - methods | psm. discussion - study rs. method methods | 19 |  |
| 20 | psp. organic carbon $-\mathrm{OC}$ <br> rs. $\mathrm{EC}-\mathrm{EC}$ | rs. $\mathrm{OC}-\mathrm{OC}$ <br> rs. $\mathrm{EC}-\mathrm{EC}$ <br> rs. method - methods | rs. method methods | d. oración 19 - this rs. methods - methods psp. organic carbon OC <br> psp. elemental carbon - EC | 20 |
| 21 |  | rs. thermal - thermal <br> rs. optical - optical <br> rs. method - method <br> rc. instrumentation - <br> instrument | rs. method method | rs. methods - method | rs. operational operational rs. method method |
| 22 |  | rc. instrumentation instruments |  |  |  |
| 23 | rs. analysis analysis | rs. analysis - analysis rs. thermal - thermal rs. optical - optical |  |  |  |
| 24 |  | rs. thermal - thermal rs. method - method rs. instrumentation instrumentation | rs. method method <br> rs. $5040-5040$ | rs. methods - method | rs. method method |
| 25 | rs. diesel - diesel <br> rs. particulate particulate | rs. thermal - thermal |  |  |  |
| 26 |  | rs. thermal - thermal <br> rs. optical - optical <br> rs. method - method | rs. method method* | rs. methods - method | rs. method method |
| 27 |  | rs. method - methods | rs. discussion discussion rs. method methods psm. published reported | psm. study - <br> discussion <br> rs. methods - methods <br> rs. results - results | rs. method methods |


| 22 | rs. <br> interlaboratory interlaboratory rs. instrument instruments | 22 |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 23 | rc. interlaboratory laboratories rs. thermal thermal rs. optical optical | rs. laboratory laboratories rs. comparison comparison psm. conducted - perform* | 23 |  |  |  |
| 24 | rc. variability variation rs. method method rs. thermal thermal rc. instrument instrumentation | rc. instruments instrumentation | s. laboratories them rs. thermal thermal | 24 |  |  |
| 25 | rs. thermal thermal |  | rs. thermal thermal | rs. used used rs. thermal thermal | 25 |  |
| 26 | rs. <br> interlaboratory interlaboratory rs. method method rs. thermal thermal rs. optical optical | rs. laboratory laboratories rs. interlaboratory interlaboratory rs. comparison comparison | rs. seven - seven <br> rs. laboratories - <br> laboratories <br> rs. thermal thermal <br> rs. optical optical <br> rs. participated participated rs. comparison comparison | psm. used employing rs. method method rs. thermal thermal | rs. thermal thermal psp. technique method <br> rs. coulometric coulometric psm. used employing | 26 |
| 27 | rs. method methods | rc. comparison intercomparison | rc. comparison intercomparison | rs. method methods |  | rs. method methods rc. comparison inetercomparison |

## 1. 3. 2. Matriz con el número de unidades léxicas.




## 1. 3. 3. Tabla representativa del número de conexiones entre oraciones

1. $(-, 2)[2]$
2. $(1,2)[3]$
3. $(1,6)[7]$
4. $(1,5)[6]$
5. $(2,3)[5]$
6. $(0,4)[4]$
7. $(0,0)(0,1)[0][1]$
8. $(3,2)$ [5]
9. $(5,3)(6,3)[8][9]$
10. $(2,0)[2]$
11. $(6,1)(6,2)[7][8]$
12. $(5,0)[5]$
13. $(0,2)(0,3)[2][3]$
14. $(0,8)(1,8)[8][9]$
15. $(1,5)[6]$
16. $(4,2)[6]$
17. $(3,6)[9]$
18. $(1,1)$ [2]
19. $(4,2)[6]$
20. $(3,0)$ [3]
21. $(2,3)[5]$
22. $(0,1)(0,2)[1][2]$
23. $(4,1)(5,1)[5][6]$
24. $(3,1)[4]$
25. $(1,1)(2,1)[2][3]$
26. $(7,0)[7]$
27. (2,-) [2]

## 1. 3. 4. Texto resultante tras eliminar las oraciones marginales.

1. Many workplace and environmental settings contain aerosols composed primarily of carbon. 2. Cabonaceous aerosols encountered in these settings include asphalt fumes, oil mists, cigarette and wood smokes, carbon black, and diesel exhaust. 3. Some of these aerosols are known or suspect human carcinogens (e,g., cigarette smoke and diesel exhaust, respectively) and have been linked to other adverse health effects (e.g., asthma, heart disease) 4. Exposure to diesel exhaust is of particular concern because it has been classified a probable human carcinogen and diesel equipment use is widespread in (e,g., trucking, transit, mining, railroads, agriculture). 5. An estimated 1.35 million workers are routinely exposed to diesel engine exhaust and exposures in some industries are relatively high (e.g., $>0.5 \mathrm{mg} \mathrm{m}^{-3}$ ). 6. Unfortunately, health-based exposure criteria for diesel particulate have not yet been established. ${ }^{1} 8$. Particulate diesel exhaust, like fine particulate air pollution in general, also is of concern with respect to noncancer health effects. 9. The US Environmental Protection Agency (EAP) has proposed an inhalation Reference Concentration (RfC) of $5 \mu / \mathrm{m}^{-3}$ for the noncancer health effects of diesel exhaust and the State of California Office of Environmental Health Hazard Assessment (OEHHA) has proposed adoption of this value for the chronic inhalation reference exposure level in California. 10. The RfC for a substance is an estimate of a daily exposure of humans, including sensitive subgroups, that is 'likely to be without appreciable risk of deleterious effects during a lifetime of exposure'. 11. Comprehensive reviews of the potential health effects of exposure to diesel exhaust exposure have been recently published.
2. Because diesel exhaust is a chemically complex mixture containing thousands of compounds, some measure of exposure must be selected. 13. Given the high carbon content of diesel particulate, a carbon-based method was investigated. 14. The method, recently published as National Institute for safety and Health (NIOSH) Method 5040, is based on an evolved gas analysis technique called the 'thermal-optical method'. 15. With this technique, speciation of organic and elemental carbon (OC and EC, respectively) is accomplished through temperature and atmosphere control and by an optical feature that corrects for pyrolytically generated carbon, or 'char', formed during the analysis of some materials. 16. Although both organic and elemental carbon are determined in the analysis, EC is the superior marker of diesel particulate because it constitutes a large fraction of the particulate mass, it can be quantified at background (i.e., environmental) levels, and its only significant source in most workplaces is the diesel engine. 17. Different approaches can be applied for OC-EC analysis, but a thermal-optical method was selected because the instrumentation has desirable design features not present in other carbon analyzers. 18. An in-depth discussion on Method 5040, including both technical and exposure-related issues, has been published elsewhere.
3. In a previous study, different methods gave widely varying results in the speciation of organic and elemental carbon. 20. For this reason, OC-EC methods are considered operational in the sense that the method itself defines the analyte. 21. Given its operational nature, it is important to examine interlaboratory variability of the method; however, when the thermal-optical method was initially evaluated, only one instrument was available, so interlaboratory variability could not be examined. 22. More recently, additional instruments were constructed by a commercial laboratory and an interlaboratory comparison was conducted. 23. Seven laboratories that perform thermal-optical analysis participated in the comparison. 24. Six of them used NIOSH
[^0]Method 5040 (i.e., they used identical instrumentation and thermal program), while the seventh used a variation on the method. 25. Another thermal technique based on coulometric detection of $\mathrm{CO}_{2}$ is being used in Europe for occupational monitoring of diesel particulate. 26. Four laboratories employing the coulometric method also participated in the interlaboratory comparison, giving a total of eleven laboratories (seven thermal-optical and four coulometric). 27. Discussion of the methods and a summary of the results of the intercomparison are reported in this paper.

## 1. 4. Texto 4: High-precision conductometric detector for the measurement of atmospheric carbon dioxide.

1. The recent increase in atmospheric $\mathrm{C}_{2}$ mixing ratio is one of the most significant changes in the trace gas composition of the atmosphere. 2. The observed $30 \%$ rise, from 280 to 360 ppmv since the beginning of the industrial revolution, accounts for only $\sim 50 \%$ of the $\mathrm{C}_{2}$ released into the atmosphere from anthropogenic sources. 3 . The remainder of the $\mathrm{C}_{2}$ released from fossil fuel burning and deforestation is assumed to have been absorbed by the oceans and terrestrial biosphere. 4. Direct measurements of $\mathrm{C}_{2}$ fluxes are needed in order to determine the strengths of these sinks and to close regional and global carbon budgets. 5. In addition, flux measurements are necessary to improve the global circulation models that predict future $\mathrm{C}_{2}$ concentrations and climate change.
2. Currently, $\mathrm{CO}_{2}$ concentrations are determined either by collecting air in flasks for analysis offsite or by continuous monitoring in the field. 7. Offsite analysis is usually performed by GC/TCD, GC/FID with a methanizer, or nondispersive infrared absorption (NDIR). 8. The disadvantages of batch analysis include sample storage and transport problems, limitation of the number of measurements by the number of available flasks, and a significant time lag between flask sample collection and analysis. 9. For example, in a recent field campaign aimed at measuring the fluxes of greenhouse gases in the Amazon rain forest of Peru, we were limited to six flask samples to characterize each vertical profile through the convective boundary layer. 10. Continuous monitoring is almost exclusively performed by NDIR 11. The limitations and errors associated with open- and closed-path NDIR analyzers have been extensively discussed by Leuning and Judd. 12. Disadvantages of in situ analysis by NDIR include instrument
expense (and therefore limited sampling sites) and the inability to use NDIR from kite or small balloon platforms because of excessive weight and power requirements.
3. The new technique described here for measurement of $\mathrm{C}_{2}$ mixing ratios is based on the increase in conductivity that occurs when deionized water makes contact with air by use of microporous hollow fiber membranes. 14. The detector is sufficiently small and lightweight to be operated from kite and balloon platforms for continuous vertical profiling of the atmosphere and has adequate precision and accuracy to determine landscape-scale fluxes of $\mathrm{C}_{2}$ from vertical profile measurements.
4. There are previous reports of conductometric techniques for measuring gasphase $\mathrm{C}_{2}$. 16. Initial designs were cumbersome and slow because they incorporated large amounts of air and water and required time for degassing. 17. Van Kempen and Kreuzer and Himpler et al. used microsensors and semipermeable membranes but did not study atmospheric levels of $\mathrm{C}_{2}$. 18. Symanski et al. designed microsensors for atmospheric $\mathrm{C}_{2}$ and were successful at measuring concentrations that would be found in highly polluted air. 19. The instruments measured $\mathrm{C}_{2}$ mixing ratios in the range 0 $3 \%$ and were not tested extensively at concentrations characteristic of "clean" air ( $\sim 350-$ 370 ppmv). 20. Furthermore, the continuous microsensor developed by Symanski et al exhibited a RSD of $\sim 2 \%$. 21. This precision is adequate for polluted air measurements but does not meet the precision required $(\sim 0.1 \%)$ for monitoring the small concentration variations that are found in relatively unpolluted air, e.g., in the atmosphere above a forest canopy.

## 1. 4. 1. Matriz de repetición de unidades léxicas.

| 2 | psm. increase - rise <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. atmosphere <br> - atmosphere |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 2 |  |  |  |
| 3 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. released released <br> e. into the atmosphere - 0 hip. antr. sources fossil ...deforestat. | 3 |  |  |
| 4 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ hip. fuel carbon hip. oceans ... biosphere sinks | 4 |  |
| 5 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ psm. mixing ratio concentrations rs. changes change* | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. measurements measurements <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. fluxes - flux <br> psm. needed - <br> necessary | 5 |
| 6 | psm. mixing <br> ratio - <br> concentrations <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. determine determined | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. concentrations concentrations |
| 7 |  |  |  |  |  |
| 8 |  |  |  | rs. measurements measurements | rs. measurements measurements |
| 9 | hip. $\mathrm{CO}_{2}-$ grennhouse gases | hip. $\mathrm{CO}_{2}-$ greenhouse gases | hip. $\mathrm{CO}_{2}-$ greenhouse gases | rc. measurements measuring <br> hip. $\mathrm{CO}_{2}-$ <br> greenhouse gases <br> rs. fluxes - fluxes | rs. flux - fluxes <br> rc. measurements measuring <br> hip. $\mathrm{CO}_{2}-$ greenhouse gases |
| 10 |  |  |  |  |  |
| 11 |  |  |  |  |  |
| 12 |  |  |  |  |  |
| 13 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. mixing mixing <br> rs. ratio ratios | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. measurements measurement rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | psm. concentrations mixing ratios rs. measurements measurement rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |

## Anexo

| 1 |  | 2 | 3 |  | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. atmosphere atmosphere | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. atmosphere <br> - atmosphere | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. measurements measurements <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. fluxes - fluxes <br> rs. determine determine | rs. flux - fluxes rs. measurements measurements rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |
| 15 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |
| 16 |  |  |  |  |  |
| 17 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ psm. mixing ratio - levels rc. atmosphere atmospheric | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ rc. atmosphere - atmospheric | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> psm. concentrations - levels |
| 18 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ psm. mixing ratio concentrations rc. atmosphere atmospheric | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rc. atmosphere - atmospheric | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. concentrations concentrations |
| 19 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. mixing mixing <br> rs. ratio - ratios | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measured rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rc. measurements measured <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. concentrations concentrations |
| 20 |  |  |  |  |  |
| 21 | rs. psm. mixing ratio concentrations psm. changes variations rs. atmosphere atmosphere | rs. atmosphere <br> - atmosphere |  | rs. measurements measurements | rs. measurements measurements <br> rs. concentrations concentration psm. change - variations* |


| 7 | rs. analysis - <br> analysis <br> rs. offsite - <br> offsite |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |

## Anexo

|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. determined determine rs. continuous continuous |  | rs. measurements measurements | rc. measuring measurements rs. fluxes - fluxes tr. greenhouse gases $\mathrm{CO}_{2}$ <br> rs. vertical - vertical <br> rs. profile - profile |  |
| 15 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |  | rc. measurements measuring | rs. measuring measuring <br> tr. greenhouse gases $\mathrm{CO}_{2}$ |  |
| 16 |  |  |  |  |  |
| 17 | psm. concentrations <br> - levels <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |  |  | tr. greenhouse gases $\mathrm{CO}_{2}$ |  |
| 18 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. concentrations concentrations rs. air - air |  | rc. measurements measuring | rs. measuring - <br> measuring <br> tr. greenhouse gases $\mathrm{CO}_{2}$ |  |
| 19 | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. concentrations concentrations rs. air - air |  | rc. measurements measured | rs. measuring - <br> measured <br> tr. greenhouse gases $\mathrm{CO}_{2}$ |  |
| 20 | rs. continuous continuous |  |  |  | rs. continuous continuous |
| 21 | rs. concentrations concentration rs. air - air rs. monitoring monitoring |  | rs. measurements measurements | rc. measuring measurements | rs. monitoring monitoring |


| 12 | rc. limitations <br> - limited <br> rs. NDIR - <br> NDIR <br> rc. analyzers analysis | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 |  | 13 |  |  |  |
| 14 |  | psm. use operated rs. kite - kite rs. balloon balloon rs. platforms platforms <br> a. excessive weight lightweight | rs. measurement - measurements <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | 14 |  |
| 15 |  |  | rs. technique - techniques <br> pc. described - reports <br> (description) <br> rc. measurement - measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rc. conductivity - conductometric | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rc. measurements measuring | 15 |
| 16 |  |  | rs. water - water <br> rs. air - air |  |  |
| 17 |  |  | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> psm. mixing ratios - levels <br> rc. use - used <br> rs. membranes - membranes | rc. atmosphere atmospheric rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |
| 18 |  |  | rc. measurement - measuring <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> psm. mixing ratios concentrations rs. air - air | rc. atmosphere atmospheric <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rc. measurements measuring | rs. measuring - measuring rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |
| 19 |  |  | rc. measurement - measured <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. mixing - mixing <br> rs. ratios - ratios <br> rs. air - air | rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rc. measurements measured | rs. measuring - measured rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |
| 20 |  |  |  |  |  |
| 21 |  |  | rs. measurement - measurements psm. mixing ratios - <br> concentration rs. air - air | rs. atmosphere atmosphere rs. precision precision rs. measurements measurements | rc. measuringmeasurements |

## Anexo

| 17 |  | 17 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | 18 |  |  |
| 18 | rc. designs designed | rs. <br> microsensors microsensors rs. atmospheric atmospheric psm. levels concentrations rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ |  |  |  |
|  |  |  |  | 19 |  |
| 19 |  | hip. <br> microsensors instruments psm. levels mixing ratios rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ | hip. microsensors - instruments <br> rs. $\mathrm{CO}_{2}-\mathrm{CO}_{2}$ <br> rs. measuring - measured <br> rs. concentrations concentrations <br> a. polluted - clean <br> rs. air - air |  |  |
| 20 | pc. designs developed (designed) | rs. microsensors microsensor | rs. Symanski - Symanski psm. designed - developed rs. microsensors - microsensor | tr. instruments microsensor | 20 |
| 21 |  | psm. levels concentrations | rc. atmospheric - atmosphere <br> rc. measuring - measurements <br> rs. concentrations - <br> concentration <br> rs. found - found <br> rs. polluted - polluted <br> rs. air - air | rc. measured measurements rs. concentrations concentration psm. clean unpolluted rs. air - air | hip. RSD - <br> precision |

## 1. 4. 2. Matriz con el número de unidades léxicas.




## 1. 4. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 6)[6](-, 7)[7]$ | 2. $(1,1)[2]$ | 3. $(1,1)[2]$ |
| :--- | :--- | :--- |
| 4. $(1,3)[4]$ | 5. $(1,5)[6](2,6)[8]$ | 6. $(0,8)[8]$ |
| 7. $(0,2)[2]$ | 8. $(1,2)[3]$ | 9. $(4,2)[6]$ |
| 10. $(2,0)[2]$ | 11. $(0,1)[1]$ | $12 \cdot(4,1)[5]$ |
| 13. $(3,5)[8]$ | $14 \cdot(5,2)[7]$ | $15 \cdot(1,0)[1]$ |
| 16. $(0,0)[0]$ | $17 \cdot(2,2)[4]$ | $18 \cdot(6,3)[9]$ |
| 19. $(6,1)[7]$ | 20. $(1,0)[1]$ | 21. $(6,-)[6](7,-)[7]$ |

## 1. 4. 4. Texto resultante tras eliminar las oraciones marginales.

1. The recent increase in atmospheric $\mathrm{CO}_{2}$ mixing ratio is one of the most significant changes in the trace gas composition of the atmosphere. 2. The observed $30 \%$ rise, from 280 to 360 ppmv since the beginning of the industrial revolution, accounts for only $\sim 50 \%$ of the $\mathrm{CO}_{2}$ released into the atmosphere from anthropogenic sources. 3. The remainder of the $\mathrm{CO}_{2}$ released from fossil fuel burning and deforestation is assumed to have been absorbed by the oceans and terrestrial biosphere. 4. Direct measurements of $\mathrm{CO}_{2}$ fluxes are needed in order to determine the strengths of these sinks and to close regional and global carbon budgets. 5. In addition, flux measurements are necessary to improve the global circulation models that predict future $\mathrm{CO}_{2}$ concentrations and climate change.
2. Currently, $\mathrm{CO}_{2}$ concentrations are determined either by collecting air in flasks for analysis offsite or by continuous monitoring in the field. 7. Offsite analysis is usually performed by GC/TCD, GC/FID with a methanizer, or nondispersive infrared absorption (NDIR). 8. The disadvantages of batch analysis include sample storage and transport problems, limitation of the number of measurements by the number of available flasks, and a significant time lag between flask sample collection and analysis. 9. For example, in a recent field campaign aimed at measuring the fluxes of greenhouse gases in the Amazon rain forest of Peru, we were limited to six flask samples to characterize each vertical profile through the convective boundary layer. 10. Continuous monitoring is almost exclusively performed by NDIR 11. The limitations and errors associated with open- and closed-path NDIR analyzers have been extensively discussed by Leuning and Judd. 12. Disadvantages of in situ analysis by NDIR include instrument expense (and therefore limited sampling sites) and the inability to use NDIR from kite or small balloon platforms because of excessive weight and power requirements.
3. The new technique described here for measurement of $\mathrm{CO}_{2}$ mixing ratios is based on the increase in conductivity that occurs when deionized water makes contact with air by use of microporous hollow fiber membranes. 14. The detector is sufficiently small and lightweight to be operated from kite and balloon platforms for continuous vertical profiling of the atmosphere and has adequate precision and accuracy to determine landscape-scale fluxes of $\mathrm{CO}_{2}$ from vertical profile measurements.
4. There are previous reports of conductometric techniques for measuring gasphase $\mathrm{CO}_{2}$. 17. Van Kempen and Kreuzer and Himpler et al. used microsensors and semipermeable membranes but did not study atmospheric levels of $\mathrm{CO}_{2}$. 18. Symanski et al. designed microsensors for atmospheric $\mathrm{CO}_{2}$ and were successful at measuring concentrations that would be found in highly polluted air. 19. The instruments measured $\mathrm{CO}_{2}$ mixing ratios in the range $0-3 \%$ and were not tested extensively at concentrations characteristic of "clean" air ( $\sim 350-370 \mathrm{ppmv}$ ). 20. Furthermore, the
continuous microsensor developed by Symanski et al exhibited a RSD of $\sim 2 \%$. 21. This precision is adequate for polluted air measurements but does not meet the precision required ( $\sim 0.1 \%$ ) for monitoring the small concentration variations that are found in relatively unpolluted air, e.g., in the atmosphere above a forest canopy.

## 1. 5. Texto 5: Refinement of the borohydride reduction method for trace analysis of dissolved and particulate dimethyl sulfoxide in marine water samples.

1. Recent interest in dimethyl sulfoxide (DMSO) in the marine environment stems from its widespread occurrence in nature and its potential role in the biogeochemical cycle of dimethyl sulfide (DMS), a key species in the global sulfur cycle and the precursor of climatically active sulfur aerosols in the atmosphere. 2. However, relatively few measurements of DMSO levels in natural waters have been made to date, essentially because of the scarcity of sufficiently sensitive and selective analytical procedures. 3. During the past few years, five methods for trace analysis of aqueous DMSO have been reported. 4. All involve gas chromatography, either via direct injection of the water aliquot or via reduction and subsequent determination of the evolved DMS. 5. Simó et al. developed a borohydride reduction method which is relatively simple and performs well at nanomolar concentration levels. 6. When used as part of a sequential protocol, this technique allows analysis of a suite of methylated sulfur compounds, eg., DMS, methanethiol, dimethylsufonipropionate, DMSP), and DMSO, in the same water sample. 7. The method has been applied successfully in a number of field studies (rfs 10 and 11 and Simó, unpublished work).
2. In this paper, we report on refinements to the borohydride reduction method for DMSO analysis which resulted from adapting the technique for a different sample preparation and GC analytical system to that described by Simó et al. 9. New insight into the method has been gained, including the need to adjust the proportion of reductant specificity, blank troubleshooting, sample storage, and the first-ever application of the method to analysis of particular DMSO $\left(\mathrm{DMSO}_{\mathrm{p}}\right)$. 10. This information should be useful for those intending to analyze aqueous DMSO by reduction methods.

## 1. 5. 1. Matriz de repetición de unidades léxicas.



| 7 | psm.. technique method |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  | 7 | 8 |  |
| 8 | rs. technique technique <br> rs. analysis - analysis <br> rs. DMSO - DMSO <br> a. same - different <br> rs. sample - sample | rs. method - method rs. Simó - Simó psp work - paper |  |  |
| 9 | psm.. technique - <br> method <br> rs. analysis - analysis <br> rs. DMSO - DMSO <br> rs. sample - sample | rs. method - method <br> rc. applied - application | rc. reduction - reductant <br> rs. method - method <br> rs. DMSO - DMSO <br> rs. analysis - analysis <br> rs. sample - sample | 9 |
| 10 | rc. used - useful psm.. technique methods rc. analysis - analyze <br> rs. DMSO - DMSO | rs. method - methods | rs. reduction - reduction <br> rs. method - methods <br> rs. DMSO - DMSO <br> rc. analysis - analyze | rs. method - methods <br> rc. reductant - reduction <br> rc. analysis - analyze <br> rs. DMSO - DMSO |

## 1. 5. 2. Matriz con el número de unidades léxicas.



## 1. 5. 3. Tabla representativa del número de conexiones entre oraciones.

1. $(-, 0)[0]$
2. $(0,0)[0]$
3. $(0,3)[3]$
4. $(0,0)[0]$
5. $(0,1)[1]$
6. $(1,3)[4]$
7. $(0,0)[0]$
8. $(3,2)[5]$
9. $(2,1)[3]$
10. (4-) [4]

## 1. 5. 4. Texto resultante tras eliminar las oraciones marginales.

3. During the past few years, five methods for trace analysis of aqueous DMSO have been reported. 5. Simó et al. developed a borohydride reduction method which is relatively simple and performs well at nanomolar concentration levels. 6. When used as part of a sequential protocol, this technique allows analysis of a suite of methylated sulfur compounds, eg., DMS, methanethiol, dimethylsufonipropionate, DMSP), and DMSO, in the same water sample.
4. In this paper, we report on refinements to the borohydride reduction method for DMSO analysis which resulted from adapting the technique for a different sample preparation and GC analytical system to that described by Simó et al. 9. New insight into the method has been gained, including the need to adjust the proportion of reductant specificity, blank troubleshooting, sample storage, and the first-ever application of the method to analysis of particular DMSO ( $\mathrm{DMSO}_{p}$ ). 10. This information should be useful for those intending to analyze aqueous DMSO by reduction methods.

## 1. 6. Texto 6: Determination of cyanide in whole blood by capillary gas chromatography with cryogenic oven trapping.

1. Cyanide is known as one of the most rapidly acting and powerful poisons; it inhibits cytochrome oxidase of the mitochondrial respiratory chain. 2. Suicidal, accidental, or homicidal death by cyanide salts is frequently experienced in forensic toxicological practice. 3. Several researchers reported that cyanide occasionally played a significant role in the cause of death of fire cases.
2. For analysis of cyanide, the most classical is a colorimetric method with microdiffusion, fluorometric methods were also reported. 5. Methods using gas chromatography (GC) with electron capture detection (ECD) and with nitrogenphosphorus detection (NPD) and mass spectrometry (MS), after suitable derivatizations, were reported. 6. GC measurements of cyanide with NPD without derivatization were usually made using the headspace (HS) method. 7. In most of these reports, conventional packed columns, which give relatively low sensitivity and poor separation, were used. 8. With wide-bore capillary columns, only a $0.5-\mathrm{mL}$ volume of the HS vapor can be injected; with medium-bore capillary columns, split injection giving less than $5 \%$ of efficiency has to be used. 9. Solid-phase microextraction has been applied to analysis of cyanide in human whole blood.
3. Recently, a microcomputer-controlled device for cooling oven temperatures below $0^{\circ} \mathrm{C}$ has become available for new types of GC instruments. 11. It was originally
designed for rapid cooling of the oven to reduce time for analysis. 12. This new device has been applied for determining chloroform and methylene chloride in blood.
4. In this paper, we have established a new GC technique using cryogenic oven for measuring cyanide in whole blood without any complicated pre-treatment; as much as 5 mL of the HS vapor for cyanide can be introduced without any loss into a mediumbore capillary column by use of a low oven temperature. 14. This means that 10-100 times higher sensitivity can be obtained by this method as compared with that of the previous methods.

## 1. 6. 1 . Matriz de repetición de unidades léxicas.

| 2 | rs. cyanide cyanide | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. cyanide cyanide | rs. death - death rs. cyanide cyanide | 3 |  |  |
| 4 | rs. cyanide cyanide | rs. cyanide cyanide | rs. reported reported rs. cyanide cyanide | 4 |  |
| 5 |  |  | rs. reported reported | rs. reported - reported <br> rs. method - methods | 5 |
| 6 | rs. cyanide cyanide | rs. cyanide cyanide |  | psm. analysis - <br> measurements <br> rs. cyanide - cyanide <br> rs. method - method | rs. methods - method <br> rs. using - using <br> rs. GC - GC <br> rs. NPD - NPD <br> rs. derivatizations derivatization |
| 7 |  |  | rc. reported reports | rc. reported - reports psm. classical conventional | rs. using - used rc reported - reports |
| 8 |  |  |  |  | rs. using - used |
| 9 | rs. cyanide cyanide | rs. cyanide cyanide | rs. cyanide cyanide | rs. analysis - analysis <br> rs. cyanide - cyanide |  |
| 10 |  |  |  |  | rs. $\mathrm{GC}-\mathrm{GC}$ |
| 11 |  |  |  | rs. analysis - analysis |  |
| 12 |  |  |  | pc. analysis determining (analize) |  |
| 13 | rs. cyanide cyanide | rs. cyanide cyanide | pc. reported paper (report) rs. cyanide cyanide | pc. analysis measuring (measurement) rs. cyanide - cyanide psm. method technique pc. reported - paper (report) | psm. methods technique rs. using - using <br> rs. GC - GC <br> pc. reported - paper (report) |
| 14 |  |  |  | rs. method - methods | rs. methods - methods |


| 7 | rs. using - used | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | rs. using - used rs. HS - HS | rs. columns columns <br> rs. give giving psm. sensitivity efficiency rs. used used | 8 |  |  |
| 9 | psm. measurements - analysis <br> rs. cyanide cyanide |  |  | 9 |  |
| 10 | rs. GC-GC |  |  |  | 10 |
| 11 | psm. measurements <br> - analysis |  |  | rs analysis analysis | s. microcomputer controlled device - it rs. cooling - cooling rs. oven - oven |
| 12 | pc. measurements determining (measuring) |  |  | rs. applied - applied <br> pc. analysis - <br> determining <br> (analize) <br> rs. blood - blood | rs. device - device |
| 13 | rs. $\mathrm{GC}-\mathrm{GC}$ <br> rc. measurements measuring <br> rs. cyanide cyanide rs. using - using rs. HS - HS | psp. reports paper rs. columns column rs. used using | rs. $\mathrm{mL}-\mathrm{mL}$ <br> rs. HS - HS <br> rs. vapor - vapor <br> psm. injected - <br> introduced <br> rs. medium - medium <br> rs. bore - bore <br> rs. capillary - capillary <br> rs. columns - column <br> rs. used - using | pc. analysis measuring (analize) <br> rs. cyanide cyanide rs. whole - whole <br> rs. blood - blood | rs. oven - oven <br> rs. temperatures temperature <br> rs. $\mathrm{GC}-\mathrm{GC}$ |
| 14 | rs. method methods | psm. give obtained rs. sensitivity - sensitivity pc. low higher (high) | psm. efficiency sensitivity |  |  |


| 12 | pc. analysis - determining <br> (analize) | 112 |  |
| :---: | :--- | :--- | :--- |
| 13 | rs. oven - oven <br> pc. analysis - measuring <br> (analize) | psm. determining - measuring <br> rs. blood - blood |  |
| 14 |  |  | d. oración 13 - this |

## 1. 6. 2. Matriz con el número de unidades léxicas.



1. 6. 3. Tabla representativa del número de conexiones entre oraciones.
1. $(-, 0)[0]$
2. $(0,0)[0]$
3. $(0,0)[0]$
4. $(0,2)[2]$
5. $(0,2)[2]$
6. $(2,1)[3]$
7. $(0,3)[3]$
8. $(1,1)$ [2]
9. $(0,2)[2]$
10. $(0,2)[2]$
11. $(1,0)[1]$
12. $(1,0)[1]$
13. $(7,0)[7]$
14. $(1,-)[1]$

## 1. 6. 4. Texto resultante tras eliminar las oraciones marginales.

4. For analysis of cyanide, the most classical is a colorimetric method with microdiffusion, fluorometric methods were also reported. 5. Methods using gas chromatography (GC) with electron capture detection (ECD) and with nitrogenphosphorus detection (NPD) and mass spectrometry (MS), after suitable derivatizations, were reported. 6. GC measurements of cyanide with NPD without derivatization were usually made using the headspace (HS) method. 7. In most of these reports, conventional packed columns, which give relatively low sensitivity and poor separation, were used. 8. With wide-bore capillary columns, only a $0.5-\mathrm{mL}$ volume of the HS vapor can be injected; with medium-bore capillary columns, split injection giving less than $5 \%$ of efficiency has to be used. 9. Solid-phase microextraction has been applied to analysis of cyanide in human whole blood.
5. Recently, a microcomputer-controlled device for cooling oven temperatures below $0^{\circ} \mathrm{C}$ has become available for new types of GC instruments. 11. It was originally designed for rapid cooling of the oven to reduce time for analysis. 12. This new device has been applied for determining chloroform and methylene chloride in blood.
6. In this paper, we have established a new GC technique using cryogenic oven for measuring cyanide in whole blood without any complicated pre-treatment; as much as 5 mL of the HS vapor for cyanide can be introduced without any loss into a medium-bore capillary column by use of a low oven temperature. 14. This means that 10-100 times higher sensitivity can be obtained by this method as compared with that of the previous methods.

## 1. 7. Texto 7: RP-HPLC binding domains of proteins.

1. Reversed-phase high-performance liquid chromatography (RP- HPLC) is now a central technique for the analysis and purification of biological molecules as a result of the high level of reproducibility, selectivity, and sensitivity that can be achieved. 2. Due to its ability to monitor subtle changes in molecular conformation, RP-HPLC is also now emerging as a powerful technique for studying the role of lipidlike surfaces in several biorecognition phenomena, such as the action of antimicrobial peptides and the role of hydrophobicity in protein folding. 3. However, further significant progress in the development of RP-HPLC is impeded by the lack of theoretical models which accurately describe the molecular details of peptide and protein interactions in RP-HPLC. 4. The slow development of detailed physicochemical models is largely due to the complex structural equilibria that peptides, and particularly proteins, can undergo in RP-HPLC systems.
2. A full understanding of the chromatographic process requires detailed knowledge of the chemical and physical nature of both the mobile phase and the stationary phase and also information on the types of interactions which occur between
the solute and the ligand or the solvent. 6. While little is known about the detailed molecular structure of proteins at the chromatographic surface, experimental data with species variants of proteins, as well recombinant mutants, indicate that proteins interact with the chromatographic surface in an orientation-specific manner. 7. The retention behavior of proteins, which can be described in terms of the affinity and kinetics of the interaction, is therefore determined by the molecular composition of a specific contact region. 8. Although the contact region for small peptides may involve contributions from the total or a large proportion of the molecular surface of the solute, for larger polypeptides or proteins, retention data suggest that the contact region represents a relatively small portion of the total solute surface. 9. The retention properties of larger polypeptides and proteins are therefore determined by the specific contact amino acid residues rather than by the entire amino acid sequence. 10. However, the location and identity of these chromatographic contact regions of proteins cannot be readily established. 11. Without this information, it is not possible to predict the molecular basis of the retention behavior of a protein, and this limitation constrains the further development of RP-HPLC as a technique to study protein- surface interactions.
3. To address this problem, procedures have been developed in this study to identify the chromatographic contact regions of proteins when adsorbed to reversedphase sorbents. 13. In particular, proteolytic techniques have been used to probe the surface region of horse heart cytochrome c (Cyt c) and bovine growth hormone (bGH) while adsorbed to an n-butyl (C-4) and n-octadecylsilica (C-18) reversed-phase sorbent. 14. Following proteolytic digestion and characterization of the derived fragments, the results were correlated with the known three-dimensional structure of these two proteins and provide insight into the location of the possible contact regions as well as the orientation of these two proteins at the surface of reversed-phase sorbents.

## 1. 7. 1. Matriz de repetición de unidades léxicas.

| 2 | rs. RP-HPLC -RP-HPLC <br> rs. technique technique rc. molecules molecular | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. RP-HPLC -RP-HPLC rc. molecules molecular | rs. molecular molecular <br> rs. RP-HPLC <br> - RP-HPLC <br> rs. peptides peptide rs. protein protein | 3 |  |  |
| 4 | rs. RP-HPLC -RP-HPLC | pc. <br> conformation structural (structure) rs. RP-HPLC - RP-HPLC rs. peptides peptides rs. protein proteins | rs. development development <br> rs. RP-HPLC - RPHPLC <br> rs. models - models <br> rc. details - detailed <br> rs. peptide - <br> peptides <br> rs. protein proteins | 4 |  |
| 5 | rc. chromatography chromatographic | hip. peptides/ protein solute | rc. details - detailed hip. peptide/protein - solute rs. interactions interactions | rs. detailed - detailed psm. physicochemical - chemical and physical <br> hip. peptides/proteins - solute | 5 |
| 6 | rc. <br> chromatography chromatographic rc. molecules molecular | rs. molecular molecular <br> psm. <br> conformation structure <br> rs. surfaces surface rs. protein proteins | rs. molecular molecular rc. details - detailed rs. protein proteins rc. interactions interact | rs. detailed - detailed <br> rc. structural - <br> structure <br> rs. proteins - proteins | rs. <br> chromatographicchromatographic rs. detailed detailed <br> rc. knowledge known rc. interactions interact <br> tr. solute - proteins |
| 7 | rc. molecules molecular | rs. molecular molecular psm. conformation composition rs. protein proteins | rs. describe described rs. molecular molecular rs. protein proteins rs. interactions interaction | pc. structural composition (structure) rs. proteins - proteins | rs. interactions interaction tr. solute - proteins |
| 8 | rc. molecules molecular | rs. molecular molecular rs. surfaces surface rs. peptides peptides rs. protein proteins | rs. molecular molecular rs. peptide peptides rs. protein proteins | rs. peptides - peptides <br> rs. proteins - proteins | rs. solute - solute |

## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 9 |  | rc. peptides polypeptides rs. protein - proteins | rc. peptide polypeptides rs. protein proteins | rc. peptides polypeptides rs. proteins proteins | tr. solute proteins |
| 10 | rc. chromatography chromatographic | rs. protein - proteins | rs. protein proteins | rs. proteins proteins | rs. <br> chromatographicchromatographic tr. solute proteins |
| 11 | rs. RP-HPLC -RP-HPLC rs. technique technique rc. molecules molecular | rs. molecular molecular <br> rs. RP-HPLC - RPHPLC <br> rs. technique technique rs. studying - study rs. surfaces - surface rs. protein - protein | rs. development <br> - development <br> rs. RP-HPLC -RP-HPLC <br> psm. impeded constrains rs. molecular molecular <br> rs. protein protein rs. interactions interactions | rs. development development rs. proteins protein rs. RP-HPLC -RP-HPLC | rs. interactions interactions tr. solute - protein |
| 12 | rs. reversed reversed rs. phase - phase rc. chromatography chromatographic | rs. protein - proteins | rc. development <br> - developed <br> rs. protein proteins | rc. development developed rs. proteinsproteins | rs. <br> chromatographicchromatographic tr. solute proteins |
| 13 | rs. reversed reversed rs. phase - phase | rs. surfaces - surface <br> tr. protein - Cyt c <br> /bGH | tr. protein - Cyt c/bGH | $\begin{aligned} & \text { tr. proteins - Cyt c } \\ & \text { /bGH } \end{aligned}$ |  |
| 14 | rs. reversed reversed rs. phase - phase | psm. conformation structure rs. surfaces - surface rs. protein - proteins | rs. protein proteins | rc. structural structure rs. proteinsproteins | tr. solute proteins |



## Anexo




## 1. 7. 2. Matriz con el número de unidades léxicas.



## 1. 7. 3. Tabla representativa del número de conexiones entre oraciones

| 1. $(-, 0)[0]$ | 2. $(0,5)[5]$ | 3. $(1,4)[5]$ |
| :--- | :--- | :--- |
| 4. $(2,0)[2]$ | 5. $(0,1)[1]$ | 6. $(3,4)[7]$ |
| 7. $(2,6)[8]$ | 8. $(3,3)[6]$ | 9. $(2,0)[2]$ |
| 10. $(1,2)[3]$ | 11. $(5,1)[6]$ | 12. $(3,2)[5]$ |
| 13. $(1,1)[2]$ | 14. $(6,-)[6]$ |  |

## 1. 7. 4. Texto resultante tras eliminar las oraciones marginales.

2. Due to its ability to monitor subtle changes in molecular conformation, RPHPLC is also now emerging as a powerful technique for studying the role of lipid-like surfaces in several biorecognition phenomena, such as the action of antimicrobial peptides and the role of hydrophobicity in protein folding. 3. However, further significant progress in the development of RP-HPLC is impeded by the lack of theoretical models which accurately describe the molecular details of peptide and protein interactions in RP-HPLC. 4. The slow development of detailed physicochemical models is largely due to the complex structural equilibria that peptides, and particularly proteins, can undergo in RP-HPLC systems.
3. A full understanding of the chromatographic process requires detailed knowledge of the chemical and physical nature of both the mobile phase and the
stationary phase and also information on the types of interactions which occur between the solute and the ligand or the solvent. 6. While little is known about the detailed molecular structure of proteins at the chromatographic surface, experimental data with species variants of proteins, as well recombinant mutants, indicate that proteins interact with the chromatographic surface in an orientation-specific manner. 7. The retention behavior of proteins, which can be described in terms of the affinity and kinetics of the interaction, is therefore determined by the molecular composition of a specific contact region. 8. Although the contact region for small peptides may involve contributions from the total or a large proportion of the molecular surface of the solute, for larger polypeptides or proteins, retention data suggest that the contact region represents a relatively small portion of the total solute surface. 9. The retention properties of larger polypeptides and proteins are therefore determined by the specific contact amino acid residues rather than by the entire amino acid sequence. 10. However, the location and identity of these chromatographic contact regions of proteins cannot be readily established. 11. Without this information, it is not possible to predict the molecular basis of the retention behavior of a protein, and this limitation constrains the further development of RP-HPLC as a technique to study proteinsurface interactions.
4. To address this problem, procedures have been developed in this study to identify the chromatographic contact regions of proteins when adsorbed to reversedphase sorbents. 13. In particular, proteolytic techniques have been used to probe the surface region of horse heart cytochrome c (Cyt c) and bovine growth hormone (bGH) while adsorbed to an n-butyl (C-4) and n-octadecylsilica (C-18) reversed-phase sorbent. 14. Following proteolytic digestion and characterization of the derived fragments, the results were correlated with the known three-dimensional structure of these two proteins and provide insight into the location of the possible contact regions as well as the orientation of these two proteins at the surface of reversed-phase sorbents.

## 1. 8. Texto 8: Nanoliter chemistry combined with mass.

1. At the early development stage of a disease such as cancer, only a small population of normal cells undergoes transformation and a change of the proteome is expected to occur in these tumor cells. 2. In cell research, a number of cell lines derived from tumors in in vitro cell culture systems have been used as sources of large numbers of cells of a uniform type and they play an essential role in the process of investigating cell functions. 3. However, because of the difference in the environment of cell growth in the intact organism and the culture, great care must be taken in extrapolating the results of in vitro experiments to the reality in vivo. 4. This is particularly true for proteins, whose identity and abundance can vary greatly at different stages of cell development or expressing conditions. 5. Thus, analyzing the primary cells isolated from a tissue, instead of a cultured cell line, is the only way to provide a direct correlation of the change in protein contents and identities with a
biological event, such as the progression of a disease, without running into a risk of potential artifacts of cell culture. 6. This requires very sensitive analytical methods, because the number of tumor or other disease cells available for investigation from a tissue is often limited.
2. At present, several tracer techniques involving radiolabeling, immunoassay, and fluorescence tagging have been used to provide information on the distribution of usually known proteins in a small number of cells or a single cell. 8. Miniaturized detection schemes based on electrochemical, laser-induced fluorescence detection and, more recently, mass spectrometry have shown great promise in analyzing cellular components including peptides and proteins in single cells. 9. However, unequivocal identification and characterization of trace amounts of unknown or modified proteins in very small volumes associated with tissues, single cells, subcellular compartments, and exocytosis still remain a formidable task. 10. In this report, we describe an analytical approach that combines three rapidly developing techniques, namely, nanoliter or subnanoliter chemistry, matrix-assisted laser desorption/ ionization time-of-flight mass spectrometry (MALDI-TOF MS), and protein database searching, to characterize attomole quantities of proteins from small-volume samples including single cells.

## 1. 8. 1. Matriz de repetición de unidades léxicas.

| 2 | rs. tumor - tumors <br> rs. cells - cell | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | psm. development - growth* rs. cells - cell | rs. cell - cell <br> rs. in vitro - in vitro <br> rs. culture culture | 3 |  |  |
| 4 | rs. development development* rs. stage - stages* pc. change - vary (change) rc. proteome proteins rs. cells - cell | rs. cell - cell | d. great care... in vivo - this rs. cell cell psm. growth development | 4 |  |
| 5 | psm. development <br> - progression <br> rs. disease disease <br> rs. change change rc. proteome proteins rs. cells - cells | rs. lines - line <br> rs. cell - cell <br> rs. culture culture | rs. cell cells rs. culture culture | rs. proteins protein <br> rs. identity identities <br> pc. vary change (change) rs. cell - cells psm. development progression* | 5 |
| 6 | rs. disease disease rs. tumor - tumor rs. cells - cells | rs. number number <br> rs. tumors tumor <br> rs. cells - cells <br> rc. investigating <br> - investigation | rs. cell cells | rs. cell - cells | d. oración 5 - this rc. analyzing analytical rs. cells - cells rs. tissue - tissue rs. disease - disease |
| 7 | rc. proteome proteins rs. cells - cells | rs. used - used <br> a. large - small <br> rs. numbers - <br> number <br> rs. cells - cells | rs. cell cell | rs. proteins proteins rs. cell - cells | rs. cells - cells <br> rs. provide - provide <br> rs. protein - proteins |
| 8 | rc. proteome proteins rs. cells - cells | rs. cells - cells | rs. cell cells | rs. proteins proteins rs. cell - cells | rs. analyzing - analyzing <br> rs. cells - cells <br> rs. protein - proteins |
| 9 | rs. cells - cells pc. change modified (modification) rc. proteome proteins | rs. cells - cells | rs. cell cells | rs. proteins proteins <br> psm. vary modified rs. cell - cells | rs. cells - cells <br> rs. tissue - tissues <br> pc. change - modified <br> (modification) <br> rs. protein - proteins |
| 10 | rs. cells - cells rc. proteome proteins | rs. cells - cells | rs. cell cells | rs. proteins protein <br> rs. cell - cells | rc. analyzing analytical rs. cells - cells rs. protein - proteins |



1. 8. 2. Matriz con el número de unidades léxicas.


## 1. 8. 3. Tabla representativa del número de conexiones entre oraciones

1. $(-, 1)(-, 2)[1][2]$
2. $(0,2)[2]$
3. $(0,0)(1,0)[0][1]$
4. $(0,1)(1,1)[1][2]$
5. $(2,2)[4]$
6. $(2,0)[2]$
7. $(1,3)$ [4]
8. $(1,1)$ [2]
9. $(2,1)[3]$
10. (3,-) [3]

## 1. 8. 4. Texto resultante tras eliminar las oraciones marginales.

1. At the early development stage of a disease such as cancer, only a small population of normal cells undergoes transformation and a change of the proteome is expected to occur in these tumor cells. 2. In cell research, a number of cell lines derived from tumors in in vitro cell culture systems have been used as sources of large numbers of cells of a uniform type and they play an essential role in the process of investigating cell functions. ${ }^{2} 4$. This [that great care must be taken in extrapolating the results of in vitro experiments to the reality in vivo, because of the difference in the environment of cell growth in the intact organism and the culture] is particularly true for proteins, whose identity and abundance can vary greatly at different stages of cell development or expressing conditions. 5. Thus, analyzing the primary cells isolated from a tissue, instead of a cultured cell line, is the only way to provide a direct correlation of the change in protein contents and identities with a biological event, such as the progression of a disease, without running into a risk of potential artifacts of cell culture. 6 . This requires very sensitive analytical methods, because the number of tumor or other disease cells available for investigation from a tissue is often limited.
2. At present, several tracer techniques involving radiolabeling, immunoassay, and fluorescence tagging have been used to provide information on the distribution of usually known proteins in a small number of cells or a single cell. 8. Miniaturized detection schemes based on electrochemical, laser-induced fluorescence detection and, more recently, mass spectrometry have shown great promise in analyzing cellular components including peptides and proteins in single cells. 9. However, unequivocal identification and characterization of trace amounts of unknown or modified proteins in very small volumes associated with tissues, single cells, subcellular compartments, and exocytosis still remain a formidable task. 10. In this report, we describe an analytical approach that combines three rapidly developing techniques, namely, nanoliter or subnanoliter chemistry, matrix-assisted laser desorption/ ionization time-of-flight mass spectrometry (MALDI-TOF MS), and protein database searching, to characterize attomole quantities of proteins from small-volume samples including single cells.

## 1. 9. Texto 9: The determination of food colours by HPLC with on-line dialysis for

## sample preparation.

1. Synthetic colours, mainly azo dyes, have been used in a wide range of food products for many years. 2. The sensory perception of colour is an important quality

[^1]attribute and many processed products have been coloured either to replace natural colours destroyed during processing or to provide colour in goods which would otherwise be colourless, as, for example, soft drinks. 3. The current trend is, however, away from the use of such synthetic dyes despite the extensive toxicological screening which they have undergone. 4. The lists of permitted synthetic dyes are progressively being reduced and a number of food processors are relying on the use of natural colours to impart the desired colour to their products. 5. Unfortunately, many of the natural colours (e.g. anthocyanins, carotenoids and betalaines) do not have the same stability under processing conditions as their synthetic counterparts. 6. There will always, therefore, be a tendency (or at least a temptation) for some food processors to include synthetic dyes in their products without the correct label designation.
7. There is, therefore, a well-defined need for precise and accurate methods for the determination of synthetic dyes in foods, particularly for the following reasons:
(i) to determine whether there are synthetic dyes present in foods and if so, whether they are correctly permitted;
(ii) to determine the levels of such dyes;
(iii) to confirm the absence of added dyes in foods where they are not declared;
(iv) to check on the stability of dyes during processing and storage (Damant et al.,, 1989).
8. There are well-documented methods for the chromatographic separation of synthetic dyes (Saag, 1988). 9. These are either based on ion-exchange methods or now more commonly on ion-pair chromatography under reversed phase conditions. 10. A detailed study of the factors affecting retention under these conditions has recently been published (Damant, 1990). 11. The simplest mobile phase conditions are those based on ammonium acetate buffers. 12. The problem in methods for the quantitative determination of synthetic dyes in foods does not, therefore, lie in their separation, but rather in the means for their quantitative isolation from the food matrix. 13. Traditional methods, such as adsorption on to wool or polyamide powder (Lehmann, 1970) tend not to be quantitative and can lead to dye degradation. 14. A milder means of extraction, either from the food itself (e.g. soft drinks) or from an aqueous extract of the food, would offer considerable advantages and this is the situation encountered with dialysis. 15. This technique has been used as a means of sample preparation for vitamin analysis
by HPLC (Nicholson et al.. 1984). 16. However, only recently has a fully automated system been made commercially available, which allows considerable flexibility in terms of dialysis conditions, coupled with automated injection of the sample into the HPLC column (Green et al., 1989). 17. The power of the technique is further extended by allowing enrichment of the determinand in the dialysate on small trace enrichment cartridge prior to elution to the analytical HPLC column. 18. The combination of dialysis and trace enrichment then leads to a complete sample preparation systems for microconstituents of foods, which is marketed under the acronym ASTED (automated sample treatment through enrichment of dialysates).

## 1. 9. 1. Matriz de repetición de unidades léxicas.

| 2 | a. synthetic natural rs. colours colours <br> e. food - 0 rs. products products | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. synthetic synthetic rs. dyes - dyes rc. used - use | a. natural - synthetic psm. colours - dyes | 3 |  |  |
| 4 | rs. synthetic synthetic rs. colours colours <br> rs. dyes - dyes rc. used - use rs. food - food rs. products products | rc. processed processors rs. natural - natural rs colours - colours psm. provide impart rs. products products | rs. use - use <br> rs. synthetic synthetic rs. dyes dyes | 4 |  |
| 5 | rs. synthetic synthetic rs. colours colours | rs. natural - natural rs colours - colours rs. processing processing | rs. synthetic synthetic psm. dyes colours | rs. synthetic synthetic rc. processors processing rs. natural - natural rs colours - colours | 5 |
| 6 | rs. synthetic synthetic rs. dyes - dyes rs. food - food rs. products products | rc. processed processors a. natural - synthetic psm. colours - dyes rs. products products | psm. trend tendency rs. synthetic synthetic rs. dyes dyes | rs. synthetic synthetic rs. dyes - dyes rs. food - food rs. processors processors rs. products - products | psm. colours - dyes rc. processing processors rs. synthetic synthetic |
| 7 | rs. synthetic synthetic rs. dyes - dyes rs. food - foods | a. natural - synthetic psm. colours - dyes rs. processing processing hip. products - foods | rs. synthetic synthetic rs. dyes dyes | rs. permitted permitted rs. synthetic synthetic rs. dyes - dyes rs. food - foods rc. processors processing | psm. colours - dyes rs. stability stability <br> rs. processing processing rs. synthetic synthetic |
| 8 | rs. synthetic synthetic rs. dyes - dyes | a. natural - synthetic psm. colours - dyes | rs. synthetic synthetic rs. dyes dyes | rs. synthetic synthetic rs. dyes - dyes | psm. colours - dyes rs. synthetic synthetic |
| 9 |  |  |  |  |  |

## Anexo

1
2
3
4
5

| 10 |  |  |  |  |  |
| :---: | :--- | :--- | :--- | :--- | :--- |
| 11 |  |  |  |  |  |
| 12 | rs. synthetic - <br> synthetic <br> rs. dyes - dyes <br> rs. food - foods | hip. products - <br> foods <br> a. natural - <br> synthetic <br> psm. colours - dyes | rs. synthetic - <br> synthetic <br> rs. dyes - dyes | rs. synthetic - <br> synthetic <br> rs. dyes - dyes <br> rs. food - foods | psm. colours - dyes <br> rs. synthetic - <br> synthetic |
| 13 | rs. dyes - dye | psm. colours - dye | rs. dyes - dye | rs. dyes - dye | psm. colours - dye |
| 14 | rs. food - food | hip. products - food <br> rs. soft - soft <br> rs. drinks - drinks |  | rs. food - food |  |
| 15 |  |  |  |  |  |
| 16 |  |  |  |  |  |
| 17 | hip. synthetic dyes <br> - determinand |  | hip. synthetic dyes <br> - determinand | hip. synthetic dyes <br> - determinand | hip. synthetic - <br> determinand |
| 18 | hip. synthetic dyes - <br> microconstituents <br> rs. food - foods | hip. products - <br> foods | hip. synthetic dyes - <br> microconstituents | hip. synthetic dyes <br> rs. mood - foods <br> roconstituents | hip. synthetic - <br> microconstituents |


| 7 | rs. food - foods rc. processors processing rs. synthetic synthetic rs. dyes - dyes | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | rs. synthetic synthetic rs. dyes - dyes | rs. methods methods <br> rs. synthetic synthetic rs. dyes - dyes | 8 |  |  |
| 9 |  | rs. methods methods | rs. methods - methods rc. chromatographic chromatography | 9 |  |
| 10 |  |  |  | rs. conditions conditions | 10 |
| 11 |  |  |  | rs. based - based <br> rs. phase - phase <br> rs. conditions conditions | rs. conditions conditions |
| 12 | rs. food - foods <br> rs. synthetic synthetic rs. dyes - dyes | rs. methods methods <br> rs. determination determination rs. synthetic synthetic rs. dyes - dyes rs. foods - foods | rs. methods - methods <br> rs. separation - <br> separation <br> rs. synthetic - <br> synthetic <br> rs. dyes - dyes | rs. methods methods |  |
| 13 | rs. dyes - dye | rs. methods methods <br> rs. dyes - dye | rs. methods - methods rs. dyes - dye | rs. methods methods |  |
| 14 | rs. food - food | rs. food - food |  |  |  |
| 15 |  |  |  |  |  |
| 16 |  |  |  |  |  |
| 17 | hip. synthetic dyes <br> - determinand | hip. synthetic dyes <br> - determinand | hip. synthetic dyes determinand |  |  |
| 18 | hip. synthetic dyes - microconstituents rs. food - foods | psp. methods treatment hip. synthetic dyes - microconstituents rs. foods - foods | hip. synthetic dyes microconstituents |  |  |

## Anexo



## 1. 9. 2. Matriz con el número de unidades léxicas.



## 1. 9. 3. Tabla representativa del número de conexiones entre oraciones.

1. $(-, 6)[6]$
2. $(1,6)[7]$
3. $(1,2)[3]$
4. $(3,4)[7]$
5. $(2,2)[4]$
6. $(5,2)[7]$
7. $(5,3)[8]$
8. $(1,1)$ [2]
9. $(0,1)[1]$
10. $(0,0)[0]$
11. $(1,0)[1]$
12. $(6,3)[9]$
13. $(1,0)[1]$
14. $(2,2)[4]$
15. $(0,2)$ [2]
16. $(2,2)[4]$
17. $(1,1)[2]$
18. (6-) [6]

## 1. 9. 4. Texto resultante tras eliminar las oraciones marginales.

1. Synthetic colours, mainly azo dyes, have been used in a wide range of food products for many years. 2. The sensory perception of colour is an important quality attribute and many processed products have been coloured either to replace natural colours destroyed during processing or to provide colour in goods which would otherwise be colourless, as, for example, soft drinks. 3. The current trend is, however, away from the use of such synthetic dyes despite the extensive toxicological screening which they have undergone. 4. The lists of permitted synthetic dyes are progressively being reduced and a number of food processors are relying on the use of natural colours to impart the desired colour to their products. 5. Unfortunately, many of the natural colours (e.g. anthocyanins, carotenoids and betalaines) do not have the same stability under processing conditions as their synthetic counterparts. 6. There will always, therefore, be a tendency (or at least a temptation) for some food processors to include synthetic dyes in their products without the correct label designation.
2. There is, therefore, a well-defined need for precise and accurate methods for the determination of synthetic dyes in foods, particularly for the following reasons:
(i) to determine whether there are synthetic dyes present in foods and if so, whether they are correctly permitted;
(ii) to determine the levels of such dyes;
(iii) to confirm the absence of added dyes in foods where they are not declared;
(iv) to check on the stability of dyes during processing and storage (Damant et al.., 1989).
3. There are well-documented methods for the chromatographic separation of synthetic dyes (Saag, 1988). 9. These are either based on ion-exchange methods or now more commonly on ion-pair chromatography under reversed phase conditions. 11. The simplest mobile phase conditions are those based on ammonium acetate buffers. 12. The problem in methods for the quantitative determination of synthetic dyes in foods does not, therefore, lie in their separation, but rather in the means for their quantitative isolation from the food matrix. 13. Traditional methods, such as adsorption on to wool or polyamide powder (Lehmann, 1970) tend not to be quantitative and can lead to dye degradation. 14. A milder means of extraction, either from the food itself (e.g. soft drinks) or from an aqueous extract of the food, would offer considerable advantages and this is the situation encountered with dialysis. 15. This technique has been used as a means of sample preparation for vitamin analysis by HPLC (Nicholson et al.. 1984). 16. However, only recently has a fully automated system been made commercially available, which allows considerable flexibility in terms of dialysis conditions, coupled with automated injection of the sample into the HPLC column (Green et al., 1989). 17. The power of the technique is further extended by allowing
enrichment of the determinand in the dialysate on small trace enrichment cartridge prior to elution to the analytical HPLC column. 18. The combination of dialysis and trace enrichment then leads to a complete sample preparation systems for microconstituents of foods, which is marketed under the acronym ASTED (automated sample treatment through enrichment of dialysates).

## 1. 10. Texto 10: Analysis of serotonin in whole-blood samples - A novel fully automated method.

1. For many years, serotonin (5-hydroxytryptamine) has been known as a pharmacological substance. 2. As early as 1948, Rapport, (1) described the structure of the compound. 3. Today, serotonin is known generally as a neurotransmitter and neuroregulating compound. 4. Serotonin participates in the regulation of important functions, including, circadian rhythm, temperature regulation, aggression control, and sexual function. 5. Researchers have observed changes in serotonin metabolism accompanying psychiatric diseases, including forms of depression. 6. In cases of migraine attacks, the concentration of serotonin in plasma with high platelet concentrations can increase as much as three times. 7. Furthermore, a correlation exists between the severity of the attack and the serotonin level.
2. The analysis of serotonin in whole blood is interesting because the compound is deposited in thrombocytes, which resemble some nerve cells. 9. Disturbances in the central nervous system, where serotonin acts, can in some cases be measured indirectly by monitoring the serotonin metabolism in blood. 10. Using thrombocytes as a model system, we can examine the influence of psychotropic agents. 11. The normal level of serotonin in blood varies from 70 to $160 \mathrm{ng} / \mathrm{mL}$ (10).
3. The current method for measuring serotonin in whole blood or in plateletenriched plasma requires three steps: adding perchloric acid to the sample, centrifuging it, and injecting some of the supernatant into a high performance liquid chromatography (HPLC) system. 13. If we were able to perform an equally reliable, but less tedious and time- consuming, solid-phase extraction (SPE) method, it would be a step forward. 14. Common off-line SPE does not seem to be the proper choice for analysing serotonin in whole blood. 15. In the past, analysts have reported that SPE cartridges become clogged with whole blood samples, which caused disturbed flow patterns and provided irreproducible results.
4. On-line, high-pressure SPE is better suited to viscous and complex matrices such as whole blood. 17. In this article, we will describe a method that uses on-line, high pressure SPE for the automated analysis of serotonin in whole-blood samples.

## 1. 10. 1. Matriz de repetición de unidades léxicas.

| 2 | hip. serotonin - compound | 2 |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. serotonin serotonin rs. known known | rs. compound compound | 3 |  |  |  |
| 4 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin rc. neuroregulating regulation | 4 |  |  |
| 5 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | 5 |  |
| 6 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin | 6 |
| 7 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. attacks attack psm. concentration - level rs. serotonin serotonin |
| 8 | rs. serotonin serotonin | rs. compound compound | rs. serotonin serotonin rs. compound compound | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin |
| 9 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin rs. metabolism metabolism | rs. serotonin serotonin |
| 10 |  |  |  |  |  |  |
| 11 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | pc. changes varies* (variation) rs. serotonin serotonin | psm. <br> concentration <br> - level <br> rs. serotonin serotonin |
| 12 | rs. serotonin serotonin | hip. compound serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin | rs. serotonin serotonin rs. plasma plasma rs. platelet platelet |
| 13 |  |  |  |  |  |  |

## Anexo

| 1 | 2 | 3 | 4 | 6 |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 14 | rs. serotonin - <br> serotonin | hip. compound - <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin - <br> serotonin |
| 15 |  |  |  |  |  |  |
| 16 |  |  |  |  |  |  |
| 17 | rs. serotonin - <br> serotonin | hip. compound <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin - <br> serotonin | rs. serotonin <br> serotonin |



## Anexo

| 13 | rs. method method | 13 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. serotonin serotonin <br> rs. whole whole <br> rs. blood - blood | rs. SPE - SPE <br> e. method -0 | 14 |  |  |
| 15 | rs. whole - <br> whole <br> rs. blood - blood <br> rs. sample samples | rs. SPE - SPE | rs. SPE - SPE <br> rc. analysing analysts <br> rs. whole - whole <br> rs. blood - blood | 15 |  |
| 16 | rs. whole whole rs. blood - blood | rs. SPE - SPE | a. off-line - on-line <br> rs. SPE - SPE <br> rs. whole - whole <br> rs. blood - blood | rs. SPE - SPE <br> rs. whole - whole <br> rs. blood - blood | 16 |
| 17 | rs. method method <br> rs. serotonin serotonin <br> rs. whole whole rs. blood - blood rs. sample samples | rs. we - we+ <br> rs. SPE - SPE <br> rs. method method | a. off-line - on-line <br> rs. SPE - SPE <br> rc. analysing analysis <br> rs. serotonin serotonin <br> rs. whole - whole <br> rs. blood - blood | rc. analysts analysis psm. reported describe rs. SPE - SPE <br> rs. whole - whole <br> rs. blood - blood <br> rs. samples - <br> samples | rs. on-line - on-line <br> rs. high - high <br> rs. pressure - pressure <br> rs. SPE-SPE <br> rs. whole - whole <br> rs. blood - blood |

1. 10. 2. Matriz con el número de unidades léxicas.


## 1. 10. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 0)[0]$ | 2. $(0,0)[0]$ | 3. $(0,0)[0]$ |
| :--- | :--- | :--- |
| 4. $(0,0)[0]$ | 5. $(0,0)[0]$ | 6. $(0,2)[2]$ |
| 7. $(1,0)[1]$ | 8. $(0,5)[5]$ | 9. $(1,1)[2]$ |
| 10. $(0,0)[0]$ | 11. $(0,0)[0]$ | 12. $(3,3)[6]$ |
| 13. $(0,0)(0,1)[0][1]$ | 14. $(2,3)[5]$ | 15. $(3,2)[5]$ |
| 16. $(2,1)[3]$ | 17. $(5,-)(6,-)[5][6]$ |  |

## 1. 10. 4. Texto resultante tras eliminar las oraciones marginales.

6. In cases of migraine attacks, the concentration of serotonin in plasma with high platelet concentrations can increase as much as three times. 7. Furthermore, a correlation exists between the severity of the attack and the serotonin level.
7. The analysis of serotonin in whole blood is interesting because the compound is deposited in thrombocytes, which resemble some nerve cells. 9. Disturbances in the central nervous system, where serotonin acts, can in some cases be measured indirectly by monitoring the serotonin metabolism in blood.
8. The current method for measuring serotonin in whole blood or in plateletenriched plasma requires three steps: adding perchloric acid to the sample, centrifuging it, and injecting some of the supernatant into a high performance liquid chromatography (HPLC) system. ${ }^{3}$ 14. Common off-line SPE does not seem to be the proper choice for analysing serotonin in whole blood. 15. In the past, analysts have reported that SPE cartridges become clogged with whole blood samples, which caused disturbed flow patterns and provided irreproducible results.
9. On-line, high-pressure SPE is better suited to viscous and complex matrices such as whole blood. 17. In this article, we will describe a method that uses on-line, high pressure SPE for the automated analysis of serotonin in whole-blood samples.
[^2]
## 2. TEXTOS PERTENECIENTES A LOS ARTÍCULOS ‘ACADÉMICOS INFORMALES.

## 2. 1. Texto 1: Is it Real Gold?

1. On March 12, 1997, Ann Landers advised a writer to believe her boyfriend, who claimed the necklace he had given her for Christmas was "real gold", despite the fact that it kept turning her neck green. 2. She went on to say that, "Some people have an element in their system that does this." 3. What should a chemist make of this exchange?
2. First of all, what is "real" gold? 5. To a chemist, "real" gold might imply "pure" gold. 6. The gift necklace was surely not "pure" in a chemical sense, because $100 \%$, or 24 carat gold (also spelled "karat", and always marked as "K") is too soft to be practical for use in jewelry. 7. Jewelry is usually made of 18 or 14 carat gold, whose weight fraction of gold is $18 / 24$ or $14 / 24$, respectively. 8. The "carat" system was invented by the British in about the year 1300 to facilitate the use of gold in commerce. 9. In the United States, the lowest allowed carat designation for gold is 10 , but a $1 / 2-$ carat error is allowed, so that " 10 K " can be marketed that is only 9.5 K , or $39.6 \%$ by weight gold. 10. In Britain, items that are only 9 K can be sold, but there is no margin for error on the low side; France's lowest carat designation is 18 K . 11. The rest of the material in the alloy can be a variety of other metals; those most often used are copper, nickel, or silver. 12. The composition of the alloy is not disclosed in the "carat" marking, and different alloying metals are used to make different colors. 13. For example, notice the three colors of gold in the 19th-century English verge pocket watch illustrated in Figure 1.
3. The metals used to make different colors are usually:
```
Yellow: \(\mathrm{Au}, \mathrm{Cu}, \mathrm{Ag}, \mathrm{Zn}\)
White: Au, Cu, Ni, Zn
Red: Au, Cu
Green: Au, Ag
```

15. The alloy called "green gold" (which is only slightly greenish) is rarely used, so the boyfriend of Ann Landers' correspondent was most likely claiming that the gift
necklace was one of the recognized alloys whose minimum gold content has been designated in Britain by Hallmarks and there and elsewhere by the carat system.
16. The common phrase "acid text" comes from the practice of testing gold alloys with nitric acid. 17. An alloy of less than about 9 or 10 carat is quickly turned green. 18. Compositions up to 18 carat gold alloy can be tested with aqua regia (a mixture of nitric and hydrochloric acid, in roughly equal proportions); the small spot subjected to the acid will immediately become pale yellow, as the base metals that provide some of the color are dissolved. 19. Instead of risking damage to the piece of yewelry, tests were often done using a "touchstone", a hard, black, slightly abrasive stone on which the object was rubbed fairly firmly, wiping a small amount of metal onto the stone surface. 20. The tests were done on the stone and the jewelry could easily be repolished to its original condition. 21. It is interesting that so many of the words involved in this testing process have survived to the present time: "Hallmark", "acid test", and "touchstone".
17. A perceptive chemist will recognize that the carat marking specifies the minimum weight percentage of gold (only), but neither the identity nor the concentration of the other parts of the alloy. 23. This means that an 18 carat gold item could have from zero to 25 weight percent copper, which corresponds to zero to 51 mole percent copper. 24. Mixtures involving nickel and zinc result in about the same mole fraction of the base metals because of the similarity of their average atomic masses to that of copper.
18. The question of whether it is possible to oxidize a metal, and therefore to produce the possibility of a colored salt, is largely reflected in the standard potential. 26. For the principal elements of the gold alloys, the pertinent numbers are:

$$
\begin{array}{ll}
\mathrm{Au}^{3+}+3 \mathrm{e} \rightarrow \mathrm{Au} & \mathrm{E}^{0}=1.42 \mathrm{~V} \\
\mathrm{Ag}^{+}+\mathrm{e} \rightarrow \mathrm{Ag} & \mathrm{E}^{0}=0.80 \mathrm{~V} \\
\mathrm{Cu}^{2+}+2 \mathrm{e} \rightarrow \mathrm{Cu} & \mathrm{E}^{0}=0.34 \mathrm{~V} \\
\mathrm{Ni}^{2+}+2 \mathrm{e} \rightarrow \mathrm{Ni} & \mathrm{E}^{0}=0.23 \mathrm{~V} \\
\mathrm{Zn}^{2+} 2 \mathrm{e} \rightarrow \mathrm{Zn} & \mathrm{E}^{0}=0.76 \mathrm{~V}
\end{array}
$$

27. These data suggest why gold is a "noble" metal: the potential required to oxidize it is near the maximum available in aqueous solutions. 28. Consider, for example, combining the half-cells

$$
\mathrm{Au}(\mathrm{~s})=\mathrm{Au}^{3+}+3 \mathrm{e} \quad \mathrm{E}^{0}=-1.42 \mathrm{~V}
$$

or

$$
\mathrm{Cu}(\mathrm{~s})=\mathrm{Cu}^{2+}+2 \mathrm{e} \quad \mathrm{E}^{0}=-0.34 \mathrm{~V}
$$

with the half-cell for a good oxidizer, such as:

$$
\mathrm{NO}_{3}^{-}+4 \mathrm{H}_{3} \mathrm{O}^{+}+3 \text { é }=\mathrm{NO}+6 \mathrm{H}_{2} \mathrm{O} \quad \mathrm{E}^{0}=0.96 \mathrm{~V}
$$

29. It is obvious that nitric acid will not oxidize gold but will easily oxidize copper. $\mathbf{3 0}$. However, the prediction of the conditions under which a metal might be oxidized depends upon more than just the potential for producing the "bare" (or hydrated) metal ion. 31. One must also consider that the metal ion may be stabilized in solution by formation of a complex ion, which is the reason why both the nitric acid oxidant and the hydrochloric acid complexing agent are required when aqua regia (literally, royal water - a phrase coined by alchemists to designate a solvent for "noble" metals) dissolves gold. 32. When gold is dissolved in aqua regia, the reaction is:

$$
\mathrm{Au}(\mathrm{~s})+4 \mathrm{Cl}^{-}+\mathrm{NO}_{3}^{-}+4 \mathrm{H}_{3} \mathrm{O}^{+}=\mathrm{AuCl}_{4}^{-}+\mathrm{NO}(\mathrm{~g})+6 \mathrm{H}_{2} \mathrm{O}
$$

33. Since the potential for

$$
\mathrm{AuCl}_{4}^{-}+3 \mathrm{e} \rightarrow \mathrm{Au}+4 \mathrm{Cl}^{-}
$$

is 1.00 vol , the dissolution of gold in aqua regia becomes thermodynamically favorable.
34. Oxidation by ordinary air (or air contaminated by sulfides) can tarnish silver, copper, and nickel, but pure gold is impervious to attack, even by concentrated nitric or hydrochloric acid acting independently. 35. The chloride ion in a person's perspiration can facilitate the oxidation of the base metals in a gold jewelry alloy. 36. But another factor impacting on whether these metals are leached out of necklaces, earrings, or dental work is the fact that mixtures of gold, silver, and copper with other metals are less reactive than one would predict if their alloys were ideal solutions. 37. Greenwood and Earnshaw say that these materials "can be thought of as nonstoichiometric intermetallic compounds of definite structural types...."
38. When people experience an allergic reaction to "real gold" jewelry, it is almost always one of the base metals that is the culprit, and nickel is by far the most notorious in this respect. 39. It seems that some people develop an amazingly acute sensitivity to this metal, and this most often occurs after ears are pierced and gold-plated
earrings are inserted. 40. Since the gold plating is usually quite thin and it is often applied on top of a layer of nickel plating, it is not too surprising that the wearer is often exposed to significant amounts of nickel as the gold wears, cracks, and is scratched. 41. What is surprising is that the body "learns" to react to these ions only after it has been sensitized by previous exposure. 42. The precise mechanism of this sensitization is not well understood.
43. Consider the original question, "was the necklace gold, or not". 44. If it were "real" 14 K or 18 K , it is unlikely that a person who does not sweat aqua regia would develop a green neck. 45. It is much more likely that the boyfriend had passed off a gold-plated necklace as more expensive jewelry. 46. If some misrepresentation occurred in this case, Georgius Agricola reminds us that it was not the fault of the element: "if by means of gold and silver and gems men can overcome the chastity of women, corrupt the honour of many people, bribe the course of justice and commit innumerable wickednesses, it is not the metals which are to be blamed, but the evil passions of men which become inflamed and ignited"

## 2. 1. 1. Matriz de repetición de unidades léxicas.

| 2 | s. Ana Landers - she <br> d. turning her neck green - this | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | d. turning her neck green - this | rs. this this |  |  |  |
|  |  |  | 3 |  |  |
| 4 | rs. real - real <br> rs. gold - gold |  |  | 4 |  |
| 5 | rs. real - real <br> rs. gold - gold |  | rs. chemist - chemist | rs. real - real <br> rs. gold - gold |  |
|  |  |  |  |  | 5 |
| 6 | rs. necklace - necklace <br> rs. gold - gold |  | rc. chemist - chemical | rs. gold - gold | rc. chemist - chemical <br> rs. pure - pure <br> rs. gold - gold |
| 7 | hip. necklace - jewelry <br> rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 8 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 9 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 10 | hip. necklace - items |  |  | rs. gold - gold |  |
| 11 | hip. necklace - alloy <br> tr. gold - metals |  |  | tr. gold - metals | tr. gold - metals |
| 12 | hip. necklace - alloy tr. gold - metals hip. green - colors |  |  | tr. gold - metals | tr. gold - metals |
| 13 | rs. gold - gold <br> hip. green - colors |  |  | rs. gold - gold | rs. gold - gold |
| 14 | rs. $\mathrm{Au}-\mathrm{AU}$ <br> rs. green - green |  |  | psm. gold - Au | psm. gold - Au |
| 15 | rs. Ann Landers - Ann Landers <br> rs. boyfriend boyfriend <br> rs. claimed - claiming <br> rs. necklace - necklace <br> rs. gold - gold <br> rs. green - green |  |  | rs. gold - gold | rs. gold - gold |

## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 16 | hip. necklace - alloy rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 17 | hip. necklace - alloy rs. turning - turned rs. green - green |  |  |  |  |
| 18 | hip. necklace - alloy rs. gold - gold hip. green - color |  |  | rs. gold - gold | rs. gold - gold |
| 19 | hip. necklace - piece of jewelry <br> tr. gold - metal |  |  | tr. gold - metal | rs. gold - gold |
| 20 | hip. necklace - jewelry |  |  |  |  |
| 21 |  |  |  |  |  |
| 22 | hip. necklace - alloy rs. gold - gold |  | rs. chemist - chemist | rs. gold - gold | rs. chemist - chemist <br> rs. gold - gold |
| 23 | hip. necklace - item rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 24 | hip. necklace mixtures <br> tr. gold - metals |  |  | tr. gold - metals | tr. gold - metals |
| 25 | hip. gold - metal tr. green - colored |  |  | hip. gold - metal | hip. gold - metal |
| 26 | hip. necklace - alloys rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 27 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 28 | psm. gold - Au |  |  | psm. gold - Au | psm. gold - Au |
| 29 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 30 | hip. gold - metal |  |  | hip. gold - metal | hip. gold - metal |
| 31 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 32 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |


| 1 |  | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 33 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 34 | rs. gold - gold |  |  | rs. gold - gold | rs. pure - pure <br> rs. gold - gold |
| 35 | hip. Ann Landers - person hip. necklace - jewelry rs. gold - gold | psp. people - person |  | rs. gold - gold | rs. gold - gold |
| 36 | rs. necklace - necklaces <br> rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 37 |  |  |  |  |  |
| 38 | hip. Ann Landers - people hip. necklace - jewelry rs. real - real rs. gold - gold | rs. people - people |  | rs. real - real rs. gold - gold | rs. real - real rs. gold - gold |
| 39 | hip. Ann Landers - people rs. gold - gold | rs. people - people |  | rs. gold - gold | rs. gold - gold |
| 40 | hip. Ann Landers - wearer rs. gold - gold | tr. people - wearer |  | rs. gold - gold | rs. gold - gold |
| 41 |  | psp. system - body |  |  |  |
| 42 |  |  |  |  |  |
| 43 | rs. necklace - necklace <br> rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 44 | hip. Ann Landers - person <br> s. necklace - it <br> rs. real - real <br> e. gold -0 <br> rs. green - green <br> rs. neck - neck | psp. people - person |  | rs. real - real <br> e. gold -0 | rs. real - real <br> e. gold -0 |
| 45 | rs. boyfriend - boyfriend <br> rs. necklace - necklace <br> rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |
| 46 | rs. gold - gold |  |  | rs. gold - gold | rs. gold - gold |

## Anexo

| 7 | rs. carat - carat <br> rs. gold - gold <br> rs. jewelry jewelry | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | rs. carat - carat rs. use - use rs. gold - gold | rs. carat - carat <br> rs. gold - gold | 8 |  |  |
| 9 | rs. carat - carat <br> rs. gold - gold <br> rs. $K-K$ | rs. carat - carat <br> rs. weight - weight <br> rs. gold - gold | rs. carat - carat <br> rs. gold - gold | 9 |  |
| 10 | hip. necklace items rs. carat - carat rs. $\mathrm{K}-\mathrm{K}$ | tr. jewelry - items <br> rs. $18-18$ <br> rs. carat - carat | rs. carat - carat rc. British Britain | rs. lowest - lowest <br> rs. carat - carat <br> rs. designation designation <br> rs. error - error <br> rs. $\mathrm{K}-\mathrm{K}$ <br> psm. marketed sold | 10 |
| 11 | hip. necklace alloy <br> tr. gold - metals <br> rc. use - used | tr. jewelry - items <br> tr. gold - metals | rc. use - used <br> tr. gold - metals | tr. gold - metals | tr. items - alloy |
| 12 | hip. necklace alloy <br> rs. carat - carat <br> tr. gold - metals <br> rs. marked marking <br> rc. use - used | tr. jewelry - alloy <br> rs. made - make <br> rs. carat - carat <br> tr. gold - metals | rs. carat - carat rc. use - used tr. gold - metals | rs. carat - carat <br> tr. gold - metals | tr. items - alloy <br> rs. carat - carat |
| 13 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 14 | psm. gold - Au rc. use - used | rs. made - make psm. gold - Au | rc. use - used psm. gold - Au | psm. gold - Au |  |
| 15 | rs. gift - gift <br> rs. necklace necklace <br> rs. carat - carat <br> rs. gold - gold <br> rc. use - used | tr. jewelry necklace <br> rs. carat - carat <br> rs. gold - gold | rs. carat - carat <br> rs. system system rc. British Britain rc. use - used rs. gold - gold | psm. lowest minimum <br> rs. carat - carat <br> rc. designation - <br> designated <br> rs. gold - gold | rs. Britain - Britain <br> tr. items - necklace <br> psm. lowest - <br> minimum <br> rs. carat - carat <br> rc. designation - <br> designated |
| 16 | hip. necklace alloys <br> rs. gold - gold | tr. jewelry - alloys <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items - alloys |


| 6 |  | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 17 | hip. necklace - alloy <br> rs. carat - carat | tr. jewelry - alloy <br> rs. carat - carat | rs. carat carat | rs. carat - carat | tr. items - alloy rs. 9-9 <br> rs. carat - carat |
| 18 | hip. necklace compositiuons rs. carat - carat rs. gold - gold <br> tr. jewelry - alloy | tr. jewelry - alloy <br> rs. 18-18 <br> rs. carat - carat <br> rs. gold - gold | rs. carat carat rs. gold gold | rs. carat - carat <br> rs. gold - gold | tr. items - alloy <br> rs. carat - carat <br> rs. $18-18$ |
| 19 | hip. necklace - object tr. gold - metal rs. jewelry - jewelry | rs. jewelry jewelry tr. gold - metal | tr. gold metal | tr. gold - metal | tr. items - piece of jewelry |
| 20 | rs. jewelry - jewelry | rs. jewelry jewelry |  |  | tr. items - jewelry |
| 21 |  |  |  |  |  |
| 22 | hip. necklace - alloy <br> rc. chemical - chemist <br> rs. carat - carat <br> rs. gold - gold <br> rc. marked - marking | tr. jewelry - alloy rs. carat - carat rs. weight - weight psm. fraction percentage rs. gold - gold | rs. carat carat rs. gold gold | psm. lowest minimum rs. carat - carat pc. designation specifies (specification) rs. weight - weight rs. gold - gold | tr. items - alloy psm. lowest minimum rs. carat - carat pc. designation specifies (specification) |
| 23 | hip. necklace - item rs. carat - carat rs. gold - gold | tr. jewelry - item <br> rs. 18-18 <br> rs. carat - carat <br> rs. weight - weight <br> pc. fraction percent <br> (percentage) <br> rs. gold - gold | rs. carat carat rs. gold gold | rs. carat - carat <br> rs. weight - weight <br> rs. gold - gold | rs. items - item rs. carat - carat rs. $18-18$ |
| 24 | hip. necklace mixtures <br> tr. gold - metals | tr. jewelry mixtures <br> rs. fraction fraction tr. gold - metals | tr. gold metals | tr. gold - metals | tr. items - mixtures |
| 25 | hip. gold - metal | hip. gold - metal | hip. gold metal | hip. gold - metal |  |
| 26 | hip. necklace - alloys rs. gold - gold | tr. jewelry - alloys <br> rs. gold - gold | rs. gold gold | rs. gold - gold | tr. items - alloys |
| 27 | rs. gold - gold | rs. gold - gold | rs. gold gold | rs. gold - gold |  |
| 28 | psm. gold - Au | psm. gold - Au | $\begin{aligned} & \text { psm. gold - } \\ & \mathrm{Au} \end{aligned}$ | psm. gold - Au |  |

## Anexo

|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 29 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 30 | hip. gold - metal | hip. gold - metal | hip. gold - metal | hip. gold - metal |  |
| 31 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 32 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 33 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 34 | rs. pure - pure <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 35 | hip. necklace - alloy rs. gold - gold rs. jewelry - jewelry | rs. jewelry - jewelry <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items alloy |
| 36 | rs. necklace - necklaces <br> rs. gold - gold <br> tr. jewelry - alloys | tr. jewelry - necklaces <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items necklaces |
| 37 |  |  |  |  |  |
| 38 | rs. gold - gold <br> rs. jewelry - jewelry | rs. jewelry - jewelry <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items jewelry |
| 39 | tr. jewelry - earrings <br> rs. gold - gold | tr. jewelry - earrings <br> rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 40 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |
| 41 |  |  |  |  |  |
| 42 |  |  |  |  |  |
| 43 | rs. necklace - necklace <br> rs. gold - gold | tr. jewelry - neclace <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items necklace |
| 44 | s. necklace - it <br> rs. K - K | rs. 18-18 <br> rs. $14-14$ <br> psm. carat - K |  | rs. $\mathrm{K}-\mathrm{K}$ | rs. $18-18$ <br> rs. $K-K$ |
| 45 | rs. necklace - necklace <br> rs. gold - gold <br> rs. jewelry - jewelry | rs. jewelry - jewelry <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | tr. items necklace |
| 46 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold |  |



## Anexo

|  | 11 | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 23 | tr. alloy - item tr. metals - gold rs. copper - copper | tr. alloy - item rs. carat - carat tr. metals - gold | rs. gold - gold | psm. Au - gold psm. Cu - copper | hip. necklace - item rs. gold - gold <br> rs. carat - carat |
| 24 | tr. alloy - mixtures rs. metals - metals rs. copper - copper rs. nickel - nickel | tr. alloy mixtures rs. metals metals | tr. gold - metals | rs. metals - metals psm. Cu - copper psm. Zn - zinc psm. Ni - nickel | hip. necklace mixtures tr. gold - metals |
| 25 | rs. metals - metal | rs. metals metal rc. colors colored | rc. colors colored <br> hip. gold - metal | rs. metals - metal <br> rc. colors - colored | tr. green - colored hip. gold - metal |
| 26 | rs. alloy - alloys tr. metals - gold psm. copper -Cu psm. nickel - Ni psm. silver - Ag | rs. alloy - alloys <br> tr. metals - gold | rs. gold - gold | hip. metals - <br> elements <br> rs. $\mathrm{Au}-\mathrm{Au}$ <br> rs. $\mathrm{Cu}-\mathrm{Cu}$ <br> rs. $\mathrm{Ag}-\mathrm{Ag}$ <br> rs. $\mathrm{Zn}-\mathrm{Zn}$ <br> rs. $\mathrm{Ni}-\mathrm{Ni}$ | rs. alloy - alloys <br> rs. gold - gold |
| 27 | rs. metals - metal | rs. metals metal | psm. gold - Au | rs. metals - metal psm. Au - gold | rs. gold - gold |
| 28 | psm. copper -Cu | tr. metals - Au | psm. gold - Au | rs. $\mathrm{Au}-\mathrm{Au}$ <br> rs. $\mathrm{Cu}-\mathrm{Cu}$ | psm. gold - Au |
| 29 | tr. metals - gold <br> rs. copper - copper | tr. metals - gold | rs. gold - gold | psm. Au - gold psm. Cu - copper | rs. gold - gold |
| 30 | rs. metals - metal | rs. metals metal | hip. gold - metal | rs. metals - metal | hip. gold - metal |
| 31 | rs. metals - metal | rs. metals metal | rs. gold - gold | rs. metals - metals psm. Au - gold | rs. gold - gold |
| 32 | tr. metals - gold | tr. metals - gold | rs. gold - gold | psm. Au - gold | rs. gold - gold |
| 33 | tr. metals - gold | tr. metals - gold | rs. gold - gold | psm. Au-gold | rs. gold - gold |
| 34 | tr. metals - gold <br> rs. copper - copper <br> rs. nickel - nickel <br> rs. silver - silver | tr. metals - gold | rs. gold - gold | psm. Au - gold <br> psm. Cu - copper <br> psm. Ag - silver <br> psm. Ni - nickel | rs. gold - gold |
| 35 | rs. metals - metals rs. alloy - alloy | rs. alloy - alloy rs. metals metals | rs. gold - gold | rs. metals - metals psm. Au - gold | rs. alloy - alloy hip. necklace jewelry rs. alloy - alloy |
| 36 | rs. alloy - alloys <br> rs. metals - metals <br> rs. copper - copper <br> rs. silver - silver | rs. alloy - alloys rs. metals metals | rs. gold - gold | rs. metals - metals <br> psm. Au - gold <br> psm. Cu - copper <br> psm. Ag - silver | rs. alloy - alloys rs. necklace necklaces rs. gold - gold |


| 11 |  | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 37 |  |  |  |  |  |
| 38 | tr. alloy - jewelry rs. metals - metals rs. nickel - nickel | tr. alloy jewelry rs. metals metals | rs. gold - gold | rs. metals - metals psm. Au - gold psm. Ni - nickel | hip. Ann Landers people <br> hip. necklace jewelry rs. gold - gold |
| 39 | rs. metals - metal | rs. metals metal | rs. gold - gold | rs. metals - metal psm. Au - gold | hip. Ann Landers people rs. gold - gold |
| 40 | tr. metals - gold rs. nickel - nickel | tr. metals - gold | rs. gold - gold | psm. Au - gold psm. Ni - nickel | hip. Ann Landers wearer <br> rs. gold - gold |
| 41 |  |  |  |  |  |
| 42 |  |  |  |  |  |
| 43 | tr. alloy - necklace <br> tr. metals - gold | tr. alloy necklace tr. metals - gold | rs. gold - gold | rs. gold - gold | rs. necklace necklace rs. gold - gold |
| 44 |  |  | tr. colors - green | rs. green - green | rs. green - green hip. Ann Landers person |
| 45 | tr. alloy - necklace tr. metals - gold | tr. alloy necklace tr. metals - gold | rs. gold - gold | rs. gold - gold | tr. alloy - jewelry rs. boyfriend boyfriend rs. necklace necklace rs. gold - gold |
| 46 | rs. metals - metals rs. silver - silver | rs. metals metals | rs. gold - gold | rs. metals - metals psm. Au - gold psm. Ag - silver | rs. gold - gold |

## Anexo



| 16 |  | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 30 | hip. gold - metal |  | rs. metals - metal | rs. metal - metal |  |
| 31 | rs. gold - gold <br> rs. nitric - nitric <br> rs. acid - acid |  | rs. gold - gold <br> rs. aqua - aqua <br> rs. regia - regia <br> rs. nitric - nitric <br> rs. hydrochloric - <br> hydrochloric <br> rs. acid - acid <br> rs. metals - metals <br> rs. dissolved - dissolves | rs. metal - metal |  |
| 32 | rs. gold - gold |  | rs. gold - gold <br> rs. aqua - aqua <br> rs. regia - regia <br> rs. dissolved - dissolved | tr. metal - gold |  |
| 33 | rs. gold - gold |  | rs. gold - gold rs. aqua - aqua rs. regia - regia | tr. metal - gold |  |
| 34 | rs. gold - gold rs. nitric - nitric rs. acid - acid |  | rs. gold - gold <br> rs. nitric - nitric <br> rs. hydrochloric - <br> hydrochloric <br> rs. acid - acid <br> tr. metals - copper | tr. metal - gold |  |
| 35 | rs. gold - gold <br> rs. alloys - alloy | rs. alloy alloy | tr. compositions jewelry <br> rs. gold - gold <br> rs. alloy - alloy <br> rs. base - base <br> rs. metals - metals | rs. jewelry jewelry tr. object - alloy rs. metal - metals | rs. jewelry jewelry |
| 36 | rs. gold - gold <br> rs. alloys - alloys | rs. alloy alloys | tr. compositions necklaces rs. gold - gold rs. alloy - alloys rs. metals - metals | tr. piece of jewelry <br> - necklaces <br> tr. object - alloys <br> rs. metal - metals | tr. jewelry necklaces |
| 37 |  |  |  |  |  |
| 38 | rs. gold - gold <br> tr. alloys - jewelry | tr. alloys jewelry | rs. gold - gold <br> tr. alloy - jewelry <br> rs. base - base <br> rs. metals - metals | rs. jewelry jewelry <br> rs. metal - metals | rs. jewelry jewelry |
| 39 | rs. gold - gold |  | rs. gold - gold <br> rs. metals - metal | tr. jewelry earrings <br> rs. metal - metal | tr. jewelry earrings |

## Anexo

| 16 |  | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 40 | rs. gold - gold |  | rs. gold - gold | tr. metal - gold |  |
| 41 |  |  |  |  |  |
| 42 |  |  |  |  |  |
| 43 | rs. gold - gold <br> tr. alloys - necklace | tr. alloy necklace | tr. compositions necklace rs. gold - gold | tr. piece of jewelry <br> - necklace <br> tr. metal - gold | tr. jewelry necklace |
| 44 |  | psm. carat - K <br> rs. green green | rs. $18-18$ <br> psm. carat $-K$ <br> rs. aqua - aqua <br> rs. regia - regia <br> tr. color - green |  |  |
| 45 | rs. gold - gold <br> tr. alloys - necklace | tr. alloy necklace | tr. compositions necklace <br> rs. gold - gold <br> tr. alloy - jewelry | rs. jewelry jewelry <br> tr. object - necklace <br> tr. metal - gold | rs. jewelry jewelry |
| 46 | rs. gold - gold |  | rs. gold - gold | rs. metal - metals |  |


| 22 |  | 22 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 23 |  | d. oración 22 - this <br> rs. carat - carat <br> rs. weight - weight <br> rc. percentage - percent <br> rs. gold - gold <br> tr. alloy - item | 23 |  |  |
| 24 |  | psm. percentage - <br> fraction <br> tr. gold - metals <br> tr. alloy - mixtures | tr. gold - metals <br> tr. item - mixtures <br> rs. copper - copper <br> rs. mole - mole <br> pc. percent - <br> fraction <br> (percentage) | 24 |  |
| 25 |  | hip. gold - metal | hip. gold - metal | rs. metals - metal | 25 |
| 26 |  | rs. gold - gold <br> rs. alloy - alloys | rs. gold - gold tr. item - alloys psm. copper - Cu | tr. mixtures - alloys tr. metals - gold psm. copper -Cu | tr. metal - gold |
| 27 |  | rs. gold - gold | rs. gold - gold | rs. metals - metal | rs. oxidize - oxidize <br> rs. metal - metal <br> rs. potential potential |
| 28 |  | psm. gold - Au | psm. gold - Au psm. copper -Cu | tr. metals - Au psm. copper - Cu | rc. oxidize oxidizer <br> tr. metal - Au |
| 29 |  | rs. gold - gold | rs. gold - gold <br> rs. copper - copper | tr. metals - gold <br> rs. copper - copper | rs. oxidize - oxidize <br> tr. metal - gold |
| 30 |  | hip. gold - metal | hip. gold - metal | rs. metals - metal | rs. oxidize oxidized rs. metal - metal rs. produce producing rs. potntial potential |
| 31 | rs. acid <br> - acid | rs. gold - gold | rs. gold - gold <br> tr. copper - metals | rs. metals - metal | rc. oxidize oxidant <br> rs. metal - metal |
| 32 |  | rs. gold - gold | rs. gold - gold | tr. metals - gold | tr. metal - gold |
| 33 |  | rs. gold - gold | rs. gold - gold | tr. metals - gold | tr. metal - gold rs. potential potential |
| 34 | rs. acid <br> - acid | rs. gold - gold | rs. gold - gold <br> rs. copper - copper | rs. nickel - nickel <br> tr. metals - gold <br> rs. copper - copper | rc. oxidize oxidation tr. metal - gold |

## Anexo

| 35 | rs. gold - gold rs. alloy - alloy | rs. gold - gold <br> tr. item - alloy <br> tr. copper - metals | tr. mixtures - alloy <br> rs. base - base <br> rs. metals - metals | rc. oxidize - oxidation rs. metal - metals |
| :---: | :---: | :---: | :---: | :---: |
| 36 | rs. gold - gold rs. alloy - alloys | rs. gold - gold <br> tr. item - necklaces <br> rs. copper - copper | rs. mixtures - mixtures <br> rs. metals - metals <br> rs. copper - copper | rs. metal - metals |
| 37 |  |  |  |  |
| 38 | rs. gold - gold tr. alloy - jewelry | rs. gold - gold <br> tr. item - jewelry <br> tr. copper - metals | tr. mixtures - jewelry <br> rs. nickel - nickel <br> rs. base - base <br> rs. metals - metals | rs. metal - metals |
| 39 | rs. gold - gold | rs. gold - gold <br> tr. copper - metal | rs. metals - metal | rs. metal - metal |
| 40 | rs. gold - gold | rs. gold - gold | rs. nickel - nickel tr. metals - gold | tr. metal - gold |
| 41 |  |  |  |  |
| 42 |  |  |  |  |
| 43 | rs. gold - gold tr. alloy - necklace | rs. gold - gold <br> tr. item - necklace | tr. mixtures - necklace <br> tr. metals - gold | tr. metal - gold |
| 44 |  |  |  | tr. colored - green |
| 45 | rs. gold - gold <br> tr. alloy - necklace | rs. gold - gold <br> tr. item - necklace | tr. mixtures - necklace <br> tr. metals - gold | tr. metal - gold |
| 46 | rs. gold - gold | rs. gold - gold hip. copper - metals | rs. metals - metals | rs. metal - metals |



## Anexo

|  | 26 | 27 | 28 | 29 | 30 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 36 | tr. elements metals <br> rs. gold - gold rs. alloys - alloys psm. Ag - silver psm. Cu - copper | rs. gold - gold <br> rs. metal metals <br> rs. solutions solutions | psm. Au - gold psm. Cu - copper | rs. gold - gold <br> rs. copper - copper | rc. prediction predict <br> rs. metal - metals |
| 37 |  |  |  |  |  |
| 38 | tr. elements metals <br> rs. gold - gold tr. alloys - jewelry psm. Ni - nickel | rs. gold - gold rs. metal metals | psm. Au - gold hip. Cu - metals | rs. gold - gold hip. copper - metals | rs. metal - metals |
| 39 | tr. elements metal rs. gold - gold | rs. gold - gold <br> rs. metal - metal | psm. Au - gold <br> tr. $\mathrm{Cu}-$ metal | rs. gold - gold <br> tr. copper - metal | rs. metal - metal |
| 40 | rs. gold - gold psm. Ni - nickel | rs. gold - gold | psm. Au - gold | rs. gold - gold | tr. metal - gold |
| 41 |  |  |  |  | rs. ion - ions |
| 42 |  |  |  |  |  |
| 43 | rs. gold - gold tr. alloys necklace | rs. gold - gold | psm. Au - gold | rs. gold - gold | tr. metal - gold |
| 44 |  |  |  |  |  |
| 45 | rs. gold - gold tr. alloys necklace | rs. gold - gold | psm. Au - gold | rs. gold - gold | tr. metal - gold |
| 46 | rs. gold - gold tr. elements metals | rs. gold - gold rs. metal metals | psm. Au - gold hip. Cu - metals | rs. gold - gold <br> hip. copper - metals | rs. metal - metals |


| 31 |  | 32 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 32 | rs. aqua - aqua <br> rs. regia - regia <br> rs. dissoves - <br> dissolved <br> rs. gold - gold |  |  |  |  |
| 33 | rc. solution dissolution rs. aqua - aqua rs. regia - regia rs. gold - gold | rs. gold - gold rc. dissolved dissolution <br> rs. aqua - aqua <br> rs. regia - regia | 33 |  |  |
| 34 | rs. nitric - nitric <br> rc. oxidant oxidation <br> rs. hydrochloric <br> - hydrochloric <br> rs. acid - acid <br> tr. metals - <br> copper <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | 34 |  |
| 35 | rs. ion - ion rc. oxidant oxidation rs. metals metals rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. oxidation oxidation hip. copper metals rs. gold - gold | 35 |
| 36 | rs. one - one + rs. solution solutions rs. metals metals rs. gold - gold | rs. gold - gold <br> tr. dissolved solutions <br> rc. reaction reactive | rs. gold - gold psm. dissolution solutions | rs. silver - silver <br> rs. copper - copper <br> rs. gold - gold | rs. metals - metals <br> rs. gold - gold <br> tr. jewelry necklaces <br> rs. alloy - alloys |
| 37 |  |  |  |  |  |
| 38 | rs. metals metals rs. gold - gold | rc. reaction reaction rs. gold - gold | rs. gold - gold | hip. copper metals rs. nickel - nickel rs. gold - gold | psp. person people <br> rs. base - base <br> rs. metals - metals <br> rs. gold - gold <br> rs. jewelry jewelry |
| 39 | rs. metal - metal <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | hip. nickel - metal <br> rs. gold - gold | psp. person people <br> rs. metals - metal <br> rs. gold - gold <br> tr. jewelry earrings |
| 40 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. nickel - nickel <br> rs. gold - gold | tr. person - wearer tr. metals - nickel rs. gold - gold |

## Anexo

| 41 | rs. ion - ions | rc. reaction - <br> react |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 42 |  |  |  |  |  |
| 43 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold <br> tr. alloy - necklace |
| 44 | rs. aqua - aqua <br> rs. regia - regia | rs. aqua - aqua <br> rs. regia - regia | rs. aqua - aqua <br> rs. regia - regia |  | rs. person - person |
| 45 | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. gold - gold <br> rs. jewelry - <br> jewelry <br> tr. alloy - necklace |
| 46 | rs. metals - metals <br> rs. gold - gold | rs. gold - gold | rs. gold - gold | rs. silver - silver <br> tr. copper - metals <br> rs. gold - gold | rs. metals - metals <br> rs. gold - gold |



| 42 | rs. sensitized - sensitization |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 42 |  |  |  |
| 43 |  |  |  |  |  |
|  |  |  | 43 |  |  |
| 44 |  |  | s. necklace - it | 44 |  |
| 45 |  |  | rs. necklace necklace <br> rs. gold - gold | rc. unlikely - likely |  |
|  |  |  |  |  | 45 |
| 46 |  |  | rs. gold - gold |  | rs. gold - gold |

2. 3. 2. Matriz con el número de unidades léxicas.


## Anexo




## Anexo




## Anexo




46 | 45 |
| :---: |

## 2. 1. 3. Tabla representativa del número de conexiones entre oraciones.

1. $(-, 8)[8]$
2. $(0,0)[0]$
3. $(0,0)[0]$
4. $(0,0)[0]$
5. $(0,1)[1]$
6. $(1,14)[15]$
7. $(1,9)[10]$
8. $(1,2)$ [3]
9. $(2,4)[6]$
10. $(3,5)[8]$
11. $(1,9)$ [10]
12. $(5,6)[11]$
13. $(0,0)[0]$
14. $(2,8)$ [10]
15. $(9,9)[18]$
16. $(0,6)[6]$
17. $(4,1)[5]$
18. $(9,14)[23]$
19. $(4,4)[8]$
20. $(1,0)[1]$
21. $(1,0)$ [1]
22. $(7,2)[9]$
23. $(9,5)[14]$
24. $(6,5)$ [11]
25. $(0,2)[2]$
26. $(4,6)[10]$
27. $(2,8)$ [10]
28. $(2,4)[6]$
29. $(4,3)[7]$
30. $(2,2)[4]$
31. $(2,1)[3]$
32. $(6,5)$ [11]
33. $(13,5)[18]$
34. $(4,0)[4]$
35. $(10,4)[14]$
36. $(11,2)[13]$
37. $(14,4)[18]$
38. $(0,0)[0]$
39. $(0,0)[0]$
40. $(3,2)[5]$
41. $(3,0)[3]$
42. $(3,0)[3]$
43. $(0,0)[0]$
44. $(0,0)[0]$
45. (3,-) [3]

## 2. 1. 4. Texto resultante tras eliminar las oraciones marginales.

1. On March 12, 1997, Ann Landers advised a writer to believe her boyfriend, who claimed the necklace he had given her for Christmas was "real gold", despite the fact that it kept turning her neck green.
2. To a chemist, "real" gold might imply "pure" gold. 6. The gift necklace was surely not "pure" in a chemical sense, because $100 \%$, or 24 carat gold (also spelled "karat", and always marked as " K ") is too soft to be practical for use in jewelry. 7. Jewelry is usually made of 18 or 14 carat gold, whose weight fraction of gold is 18/24 or $14 / 24$, respectively. 8 . The "carat" system was invented by the British in about the year 1300 to facilitate the use of gold in commerce. 9. In the United States, the lowest allowed carat designation for gold is 10 , but a $1 / 2$-carat error is allowed, so that " 10 K " can be marketed that is only 9.5 K , or $39.6 \%$ by weight gold. 10. In Britain, items that are only 9 K can be sold, but there is no margin for error on the low side; France's lowest carat designation is 18 K . 11. The rest of the material in the alloy can be a variety of other metals; those most often used are copper, nickel, or silver. 12. The composition of the alloy is not disclosed in the "carat" marking, and different alloying metals are used to make different colors.
3. The metals used to make different colors are usually:

Yellow: Au, Cu, Ag, Zn
White: Au, Cu, Ni, Zn
Red: Au, Cu
Green: Au, Ag
15. The alloy called "green gold" (which is only slightly greenish) is rarely used, so the boyfriend of Ann Landers' correspondent was most likely claiming that the gift necklace was one of the recognized alloys whose minimum gold content has been designated in Britain by Hallmarks and there and elsewhere by the carat system.
16. The common phrase "acid text" comes from the practice of testing gold alloys with nitric acid. 17. An alloy of less than about 9 or 10 carat is quickly turned
green. 18. Compositions up to 18 carat gold alloy can be tested with aqua regia (a mixture of nitric and hydrochloric acid, in roughly equal proportions); the small spot subjected to the acid will immediately become pale yellow, as the base metals that provide some of the color are dissolved. 19. Instead of risking damage to the piece of yewelry, tests were often done using a "touchstone", a hard, black, slightly abrasive stone on which the object was rubbed fairly firmly, wiping a small amount of metal onto the stone surface. 20. The tests were done on the stone and the jewelry could easily be repolished to its original condition. 21. It is interesting that so many of the words involved in this testing process have survived to the present time: "Hallmark", "acid test", and "touchstone".
22. A perceptive chemist will recognize that the carat marking specifies the minimum weight percentage of gold (only), but neither the identity nor the concentration of the other parts of the alloy. 23. This means that an 18 carat gold item could have from zero to 25 weight percent copper, which corresponds to zero to 51 mole percent copper. 24. Mixtures involving nickel and zinc result in about the same mole fraction of the base metals because of the similarity of their average atomic masses to that of copper.
25. The question of whether it is possible to oxidize a metal, and therefore to produce the possibility of a colored salt, is largely reflected in the standard potential. 26. For the principal elements of the gold alloys, the pertinent numbers are:

$$
\begin{array}{ll}
\mathrm{Au}^{3+}+3 \dot{e} \rightarrow \mathrm{Au} & \mathrm{E}^{0}=1.42 \mathrm{~V} \\
\mathrm{Ag}^{+}+\mathrm{e} \rightarrow \mathrm{Ag} & \mathrm{E}^{0}=0.80 \mathrm{~V} \\
\mathrm{Cu}^{2+}+2 \dot{e} \rightarrow \mathrm{Cu} & \mathrm{E}^{0}=0.34 \mathrm{~V} \\
\mathrm{Ni}^{2+}+2 \dot{e} \rightarrow \mathrm{Ni} & \mathrm{E}^{0}=0.23 \mathrm{~V} \\
\mathrm{Zn}^{2+} 2 \mathrm{e} \rightarrow \mathrm{Zn} & \mathrm{E}^{0}=0.76 \mathrm{~V}
\end{array}
$$

27. These data suggest why gold is a "noble" metal: the potential required to oxidize it is near the maximum available in aqueous solutions. 28. Consider, for example, combining the half-cells

$$
\mathrm{Au}(\mathrm{~s})=\mathrm{Au}^{3+}+3 \mathrm{e} \quad \mathrm{E}^{0}=-1.42 \mathrm{~V}
$$

or

$$
\mathrm{Cu}(\mathrm{~s})=\mathrm{Cu}^{2+}+2 \dot{e} \quad \mathrm{E}^{0}=-0.34 \mathrm{~V}
$$

with the half-cell for a good oxidizer, such as:

$$
\mathrm{NO}_{3}^{-}+4 \mathrm{H}_{3} \mathrm{O}^{+}+3 \text { é }=\mathrm{NO}+6 \mathrm{H}_{2} \mathrm{O} \quad \mathrm{E}^{0}=0.96 \mathrm{~V}
$$

29. It is obvious that nitric acid will not oxidize gold but will easily oxidize copper. 30. However, the prediction of the conditions under which a metal might be oxidized depends upon more than just the potential for producing the "bare" (or hydrated) metal ion. 31. One must also consider that the metal ion may be stabilized in solution by formation of a complex ion, which is the reason why both the nitric acid oxidant and the hydrochloric acid complexing agent are required when aqua regia (literally, royal water - a phrase coined by alchemists to designate a solvent for "noble" metals) dissolves gold. 32. When gold is dissolved in aqua regia, the reaction is:

$$
\mathrm{Au}(\mathrm{~s})+4 \mathrm{Cl}^{-}+\mathrm{NO}_{3}^{-}+4 \mathrm{H}_{3} \mathrm{O}^{+}=\mathrm{AuCl}_{4}^{-}+\mathrm{NO}(\mathrm{~g})+6 \mathrm{H}_{2} \mathrm{O}
$$

33. Since the potential for

$$
\mathrm{AuCl}_{4}^{-}+3 \mathrm{é}^{-} \mathrm{Au}+4 \mathrm{Cl}^{-}
$$

is 1.00 vol, the dissolution of gold in aqua regia becomes thermodynamically favorable. 34. Oxidation by ordinary air (or air contaminated by sulfides) can tarnish silver, copper, and nickel, but pure gold is impervious to attack, even by concentrated nitric or hydrochloric acid acting independently. 35. The chloride ion in a person's perspiration can facilitate the oxidation of the base metals in a gold jewelry alloy. 36. But another factor impacting on whether these metals are leached out of necklaces, earrings, or
dental work is the fact that mixtures of gold, silver, and copper with other metals are less reactive than one would predict if their alloys were ideal solutions.
38. When people experience an allergic reaction to "real gold" jewelry, it is almost always one of the base metals that is the culprit, and nickel is by far the most notorious in this respect. 39. It seems that some people develop an amazingly acute sensitivity to this metal, and this most often occurs after ears are pierced and goldplated earrings are inserted. 40. Since the gold plating is usually quite thin and it is often applied on top of a layer of nickel plating, it is not too surprising that the wearer is often exposed to significant amounts of nickel as the gold wears, cracks, and is scratched.
44. If eits [the necklace] were "real" 14 K or 18 K , it is unlikely that a person who does not sweat aqua regia would develop a green neck. 45. It is much more likely that the boyfriend had passed off a gold-plated necklace as more expensive jewelry. 46. If some misrepresentation occurred in this case, Georgius Agricola reminds us that it was not the fault of the element: "if by means of gold and silver and gems men can overcome the chastity of women, corrupt the honour of many people, bribe the course of justice and commit innumerable wickednesses, it is not the metals which are to be blamed, but the evil passions of men which become inflamed and ignited"

## 2. 2. Texto 2: Why gold and copper are colored but silver is not.

1. It s well known that $80 \%$ of chemical elements are metals. 2 . When polished, all metals shine owing to reflection of photons by external valence electrons dynamically forming metallic bonds. 3. White light reflects on most metals without color absorption or change to the naked eye; but copper and gold are yellow because they absorb "blue" and "red" photons by electron transitions between spectromeric configurations $n s^{1}(n-1) \mathrm{d}^{10} \quad n s^{2}(n-1) \mathrm{d}^{9}$ of external sublevels.
2. The next question is why silver, with the same external electronic configuration as copper and gold (group 11, IB), is not yellow. 5. The answer is simple, considering atomic radii, ionization potentials and nuclear charge:

|  | Cu | Ag | Au |
| :--- | :--- | :--- | :---: |
| Atomic radius/ pm | 117.3 | 133.9 | 133.6 |
| $1^{\text {st }}$ ionization energy $/ \mathrm{eV}$ | 7.725 | 7.576 | 9.22 |
| $2^{\text {nd }}$ ionization energy $/ \mathrm{eV}$ | 20.29 | 21.48 | 20.52 |
| Nuclear charge | 25 | 35 | 59 |

6. The atomic radius of silver is 16.6 pm larger than that of copper, allowing a bigger difference between sublevels $s$ and d, which is sufficient to restrict the transition $s^{1} d^{10}$
$\leftrightarrow \mathrm{s}^{2} \mathrm{~d}^{9}$ to a lower probability. 7. This is equally supported by the first ionization energry: since it is lower in silver, the fact that one external electron is ejected more easily than in copper atoms is justified.
7. With their higher nuclear charge ( 35 vs 25 ) silver atoms also have larger radii ( $\leftrightarrow=16.6 \mathrm{pm}$ ), and the distance between external sublevels-both spatial and energetic-is too large to freely allow $\mathrm{s} \leftrightarrow \mathrm{d}$ transitions. 9. However, the distance is not large enough to prevent the transitions completely, and after several reflections on two parallel silver mirrors, white light becomes pale yellow.
8. Now we must face an unexpected problem: why is gold yellow? 11. According to the same line of reasoning, gold would be colorless if it had bigger atoms. 12. But gold atoms are not larger than silver; the radii of silver and gold are practically identical owing to lanthanide contraction. 13. Comparing ionization energies, the value 9.22 eV for gold is about $20 \%$ higher than 7.576 eV for silver because gold has a larger nuclear charge ( 59 vs 35 ) while its radius is practically the same. 14. Thus, external s and d sublevels are close enough to allow the necessary transition. 15. As a result, the probability of transition between sublevels is similar to that of copper, and gold is again yellow.
9. We can now perceive the necessary conditions for a metal to be yellow, like copper and gold:
10. Adequate external electronic configuration $s^{1} d^{10} \leftrightarrow s^{2} d^{9}$ (group 11, IB).
11. Sublevels $s$ and d close enough to allow transitions $s^{1} d^{10} \leftrightarrow s^{2} d^{9}$ to occur significantly $(\mathrm{Cu}, \mathrm{Au})$.
12. In contrast, all other metals shine silvery, colorless to the naked eye because they do not possess the necessary electronic external configuration and transition probability to appear colored.
13. Much work has been undertaken in connection with relativistic effects on metal properties (6); however a final question remains: are metals (except for Cu and Au ) really colorless? 19. Various tinges are reported, such as yellow for silver and blue for osmium. 20. How many more will be detected when a complete survey is made? 21. What number of atomic layers must be crossed (twice) in metals to produce a definite color? 22. What about the effect of atomic packing, holes, and impurities? 23. But this
is another story and we would be very happy if research is aroused and enhanced by our questions.

## 2. 2. 1. Matriz de repetición de unidades léxicas.

| 2 | rs. metals metals | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. metals metals | rs. metals - metals <br> rc. reflectionreflects <br> rs. photons photons <br> rs. external external <br> rs. electronselectron | 3 |  |  |
| 4 | tr. metals silver | tr. metals - silver rs. external external rc. electrons electronic | tr. metals - silver <br> rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow <br> rc. electron - <br> electronic <br> rs. configurations configuration rs. external external | 4 |  |
| 5 | tr. metal $\mathrm{Cu}, \mathrm{Ag}, \mathrm{Au}$ | $\begin{aligned} & \text { tr. metal }-\mathrm{Cu}, \mathrm{Ag}, \\ & \mathrm{Au} \end{aligned}$ | psm. copper -Cu psm. gold -Au | e. question -0 <br> psm. silver -Ag <br> psm. copper -Cu <br> psm. gold -Au | 5 |
| 6 | tr. metals silver | tr. metals - silver | tr. metals - silver <br> rs. copper - copper <br> e. electron -0 <br> rs. transitions transition <br> rs. $\mathrm{ns}^{1}(\mathrm{n}-1) \mathrm{d}^{10}$ <br> $n s^{2}(n-1) d^{9}-s^{1} d^{10}$ $\mathrm{s}^{2} \mathrm{~d}^{9}$ <br> e. external - 0 <br> rs. sublevels sublevels | rs. silver - silver <br> rs. copper - copper | rs. atomic - atomic rs. radii - radius psm. Cu - copper psm. Ag - silver |
| 7 | tr. metals silver | tr. metals - silver rs. external external rs. electrons electron | tr. metals - silver <br> rs. copper - copper <br> rs. electron electron <br> rs. external external | rs. silver - silver <br> rs. external external <br> rc. electronic electron <br> rs. copper - copper | rc. atomic - atoms psm. 1st - first rs. ionization ionization <br> rs. energy - energy psm. Cu - copper psm. Ag - silver |
| 8 | tr. metals silver | tr. metals - silver rs. external external | tr. metals - silver <br> e. electron -0 <br> rs. transitions - <br> transitions <br> hip. $\mathrm{ns}^{1}(\mathrm{n}-1) \mathrm{d}^{10}$ <br> $\mathrm{ns}^{2}(\mathrm{n}-1) \mathrm{d}^{9}-\mathrm{s} \quad \mathrm{d}$ <br> rs. external - <br> external <br> rs. sublevels - <br> sublevels | rs. silver - silver rs. external external | rc. atomic - atoms <br> rs. radii - radii <br> pc. potentials - <br> energetic (energy) <br> rs. nuclear - nuclear <br> rs. charge - charge <br> rs. $25-25$ <br> rs. $35-35$ <br> psm. Ag - silver |

## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 9 | tr. metals - silver | tr. metals silver rs. reflection reflections <br> e. photons - 0 | rs. white - white <br> rs. light - light <br> rc. reflects - reflections <br> tr. metals - silver <br> rs. yellow - yellow <br> e. electron -0 <br> rs. transitions - transitions | rs. silver - silver <br> rs. yellow - yellow | psm. Ag - silver |
| 10 | tr. metals <br> - gold | tr. metals gold | rs. gold - gold <br> rs. yellow - yellow | rs. gold - gold <br> rs. yellow - yellow | psm. Au - gold |
| 11 | tr. metals <br> - gold | tr. metals gold | rc. color - colorless <br> rs. gold - gold | rs. gold - gold tr. yellow colorless | rc. atomic - atoms psm. Au - gold |
| 12 | tr. metals - silver | tr. metals silver | tr. metals - silver <br> rs. gold - gold | rs. silver - silver <br> rs. gold - gold | rc. atomic - atoms <br> rs. radii - radii <br> psm. Au - gold <br> psm. Ag - silver |
| 13 | tr. metals <br> - silver | tr. metals silver | tr. metals - silver <br> rs. gold - gold | rs. silver - silver <br> rs. gold - gold | rs. radii - radius <br> rs. ionization ionization psm. potentials energies rs. nuclear - nuclear rs. charge - charge psm. Au - gold psm. Ag - silver rs. $9.22-9.22$ <br> rs. $7.76-7.576$ |
| 14 |  | rs. external external | e. electron -0 <br> rs. transitions - transition hip. $n s^{1}(n-1) d^{10} \quad n s^{2}(n-$ 1) $d^{9}-s$ and $d$ rs. external - external rs. sublevels - sublevels | rs. external external |  |
| 15 | tr. metals <br> - gold | tr. metals gold | rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow <br> e. electron -0 <br> rs. transitions - transition <br> e. external-0 <br> rs. sublevels - sublevels | rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow | psm. Cu - copper <br> psm. Au - gold |
| 16 | rs. metals - metal | rs. metals metal <br> rs. external external rc. electrons -. electronic | rs. metals - metal <br> rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow <br> rc. electron - electronic <br> rs. transitions - transitions <br> rs. configurations - <br> configuration <br> rs. $n s^{1}(n-1) d^{10} \quad n s^{2}(n-1) d^{9}$ <br> $-s^{1} d^{10} \quad s^{2} d^{9}$ <br> rs. external - external <br> rs. sublevels - sublevels | hip. silver - metal <br> rs. external - <br> external <br> rs. electronic electronic <br> rs. configuration configuration <br> rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow | rs. $\mathrm{Cu}-\mathrm{Cu}$ <br> rs. $A u-A u$ |


| 1 |  | 2 | 3 |  | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 17 | rs. metals metals | rs. metals - metals <br> rs. shine - shine <br> rs. external external <br> rc. electrons electronic | rs. metals - metals <br> rc. color - colorless <br> rs. naked - naked <br> rs. eye - eye <br> rc. electron - electronic <br> rs. transitions - transition <br> rs. configurations - <br> configuration <br> rs. external - external | rc. silver - silvery <br> rs. external - external <br> rs. electronic - electronic <br> rs. configuration - <br> configuration <br> tr. yellow - colorless | pc. $\mathrm{Ag}-$ sivery (silver) |
| 18 | rs. metals metals | rs. metals - metals | rs. metals - metals rc. color - colorless psm. copper - Cu psm. gold - Au | rs. question - question <br> hip. silver - metals <br> psm. copper - Cu <br> psm. gold - Au <br> tr. yellow - colorless | rs. $\mathrm{Cu}-\mathrm{Cu}$ <br> rs. $\mathrm{Au}-\mathrm{Au}$ |
| 19 | tr. metals silver | tr. metals - silver | tr. metals - silver <br> rs. yellow - yellow | rs. silver - silver <br> rs. yellow - yellow | psm. Ag- <br> silver |
| 20 |  |  |  |  |  |
| 21 | rs. metals metals | rs. metals - metals | rs. metals - metals rs. color - color | hip. silver - metals <br> hip. yellow - color | rs. atomic atomic <br> hip. Cu , $\mathrm{Ag}, \mathrm{Au}-$ metals |
| 22 |  |  |  |  | rs. atomic atomic |
| 23 |  |  |  | rs. question - questions |  |

## Anexo



|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. sublevels - sublevels <br> rs. $\mathrm{s}-\mathrm{s}$ <br> rs. d-d <br> psp. sufficient - enough <br> a. restrict - allow <br> rs. transition - transition |  | rs. external external rs. sublevels sublevels rs. allow - allow e. $s \quad d-0$ <br> rs. transitions transition | rs. enough enough <br> a. prevent allow <br> rs. transitionstransition |  |
| 15 | tr. silver - gold <br> rs. copper - copper <br> rs. sublevels - sublevels <br> rs. transition - transition <br> rs. probability - probability | tr. silver - gold rs. copper copper | tr. silver - gold <br> e. external - 0 <br> rs. sublevels sublevels <br> e. $s \quad d-0$ <br> rs. transitions transition | rs. transitions transition tr. silver - gold rs. yellow yellow | rs. gold - gold <br> rs. yellow yellow |
| 16 | tr. silver - metal <br> rs. copper - copper <br> rs. sublevels - sublevels <br> rs. s - s <br> rs. $\mathrm{d}-\mathrm{d}$ <br> psp. sufficient - enough <br> a. restrict - allow <br> rs. transition - transitions <br> rs. $s^{1} d^{10} \quad s^{2} d^{9}-s^{1} d^{10} \quad s^{2} d^{9}$ | hip. silver metal <br> rs. external external rc. electron electronic rs. copper copper | hip. silver - metal rs. external external rs. sublevels sublevels rs. allow - allow tr. s d- $\mathrm{s}^{1} \mathrm{~d}^{10}$ $\mathrm{s}^{2} \mathrm{~d}^{9}$ <br> rs. transitions transitions | rs. enough enough <br> a. prevent allow <br> rs. transitions transitions hip. silver metal rs. yellow yellow | rs. we - we+ <br> rs. gold - gold <br> rs. yellow yellow |
| 17 | rc. silver - silvery <br> rs. transition - transition <br> rs. probability - probability | rc. silver silvery rs. external external rc. electron electronic | rc. silver - silvery <br> rs. external external <br> e. $s \mathrm{~d}-0$ <br> rs. transitions transition | rs. transitions transition rc. silver silvery tr. yellow colorless | hip. gold metals tr. yellow colorless |
| 18 | hip. silver - metals psm. copper - Cu | hip. silver metals psm. copper Cu | hip. silver - metals | hip. silver metals tr. yellow colorless | psm. gold Au tr. yellow colorless |
| 19 | rs. silver - silver | rs. silver silver | rs. silver - silver | rs. silver - silver rs. yellow yellow | tr. gold silver rs. yellow yellow |
| 20 |  |  |  |  |  |
| 21 | rs. atomic - atomic hip. silver - metals | hip. silver metals rc. atoms atomic | hip. silver - metals rc. atoms - atomic | hip. silver metals hip. yellow color | hip. gold metals <br> hip. yellow color |
| 22 | rs. atomic - atomic | rc. atoms atomic | rc. atoms - atomic |  |  |
| 23 |  |  |  |  | rs. we - we+ |

## Anexo

| 12 | rs. gold - gold psm. bigger larger <br> rs. atoms atoms | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | rs. gold - gold | rs. gold - gold <br> rs. silver - silver <br> rs. radii - radius <br> rs. practically practically <br> psm. identical same | 13 |  |  |
| 14 |  |  | 14 |  | 15 |
| 15 | rs. gold - gold <br> tr. colorless yellow | rs. gold - gold | rs. gold gold | e. external s and d-0 <br> rs. sublevels - sublevels <br> rs. transition - transition |  |
| 16 | rs. gold - gold tr. colorless yellow | rs. gold - gold hip. silver metal | rs. gold gold | rs. external - external <br> rs. s-s <br> rs. d-d <br> rs. sublevels - sublevels <br> rs. close - close <br> rs. enough - enough <br> rs. allow - allow <br> rs. transition - transitions | rs. transition - transitions <br> rs. sublevels - sublevels <br> rs. copper - copper <br> rs. gold - gold <br> rs. yellow - yellow |
| 17 | hip. gold - <br> metals <br> rs. colorless colorless | rc. silver silvery | hip. silver <br> - metals | rs. external - external rs. necessary - necessary <br> rs. transition - transition | rs. probability probability <br> rs. transition - transition <br> hip. gold - metals <br> tr. yellow - colorless |
| 18 | psm. gold - Au rs. colorless colorless | psm. gold - Au hip. silver metal | $\begin{aligned} & \text { psm. gold } \\ & -\mathrm{Au} \end{aligned}$ |  | psm. copper -Cu psm. gold - Au tr. yellow - colorless |
| 19 | tr. gold - silver tr. colorless yellow | rs. silver - silver | rs. silver silver |  | tr. gold - silver <br> rs. yellow - yellow |
| 20 |  |  |  |  |  |
| 21 | hip. gold metals rc. colorless color rc. atoms atomic | rc. atoms atomic hip. silver metals | hip. gold <br> - metals |  | hip. gold - metals hip. yellow - color |
| 22 | rc. atoms atomic | rc. atoms atomic |  |  |  |
| 23 |  |  |  |  |  |


| 16 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 17 | rs. metal metals <br> tr. yellow colorless psm. adequate necessary rs. external external rs. electronic electronic rs. configuration - configuration rs. transitions transition |  |  |  |  |
| 18 | rs. metal metals <br> tr. yellow colorless <br> psm. copper Cu psm. gold - Au | rs. metals metals <br> rs. colorless colorless | 18 |  |  |
| 19 | rs. yellow yellow | rc. silvery silver tr. colorless yellow | tr. metals silver <br> tr. colorless yellow | 19 |  |
| 20 |  |  | psm. work - <br> survey <br> psm. undertaken <br> - made | e. tinges - 0 |  |
| 21 | rs. metal metals hip. yellow color | rs. metals metals rc. colorless color | rs. metals metals rc. colorless color | hip. yellow - color hip. silver - metals |  |
| 22 |  |  |  |  |  |
| 23 |  |  | psm. work research rs. question questions |  | d. oración 20 - this psm. survey research |


| 22 | rs. atomic - atomic <br> e. in metals to produce a <br> definite color -0 |  |
| :--- | :--- | :--- |
| 23 |  | d. oración 21, 22 - <br> this |

## 2. 2. 2. Matriz con el número de unidades léxicas.

|  | 1 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | 1 | 2 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 3 | 1 | 5 | 3 |  |  |  |  |  |  |  |  |  |  |  |  |
| 4 | 1 | 3 | 7 | 4 |  |  |  |  |  |  |  |  |  |  |  |
| 5 | 1 | 1 | 2 | 4 | 5 |  |  |  |  |  |  |  |  |  |  |
| 6 | 1 | 1 | 7 | 2 | 4 | 6 |  |  |  |  |  |  |  |  |  |
| 7 | 1 | 3 | 4 | 4 | 6 | 4 | 7 |  |  |  |  |  |  |  |  |
| 8 | 1 | 2 | 6 | 2 | 8 | 12 | 3 | 8 |  |  |  |  |  |  |  |
| 9 | 1 | 3 | 7 | 2 | 1 | 8 | 1 | 7 | 9 |  |  |  |  |  |  |
| 10 | 1 | 1 | 2 | 2 | 1 | 1 | 1 | 1 | 2 | 10 |  |  |  |  |  |
| 11 | 1 | 1 | 2 | 2 | 2 | 3 | 2 | 4 | 2 | 2 | 11 |  |  |  |  |
| 12 | 1 | 1 | 2 | 2 | 4 | 4 | 2 | 4 | 1 | 1 | 3 | 12 |  |  |  |
| 13 | 1 | 1 | 2 | 2 | 9 | 2 | 4 | 6 | 1 | 1 | 1 | 5 | 13 |  |  |
| 14 | 0 | 1 | 5 | 1 | 0 | 6 | 0 | 5 | 3 | 0 | 0 | 0 | 0 | 14 |  |
| 15 | 1 | 1 | 7 | 3 | 2 | 5 | 2 | 5 | 3 | 2 | 2 | 1 | 1 | 3 | 15 |
| 16 | 1 | 3 | 10 | 7 | 2 | 9 | 4 | 6 | 5 | 2(3) | 2 | 2 | 1 | 8 | 5 |
| 17 | 1 | 4 | 8 | 5 | 1 | 3 | 3 | 4 | 3 | 2 | 2 | 1 | 1 | 3 | 4 |
| 18 | 1 | 1 | 4 | 5 | 2 | 2 | 2 | 1 | 2 | 2 | 2 | 2 | 1 | 0 | 3 |
| 19 | 1 | 1 | 2 | 2 | 1 | 1 | 1 | 1 | 2 | 2 | 2 | 1 | 1 | 0 | 2 |
| 20 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 21 | 1 | 1 | 2 | 2 | 2 | 2 | 2 | 2 | 2 | 2 | 3 | 2 | 1 | 0 | 2 |
| 22 | 0 | 0 | 0 | 0 | 1 | 1 | 1 | 1 | 0 | 0 | 1 | 1 | 0 | 0 | 0 |
| 23 | 0 | 0 | 0 | 1 | 0 | 0 | 0 | 0 | 0 | 0 (1) | 0 | 0 | 0 | 0 | 0 |



## 2. 2. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 0)[0]$ | 2. $(0,6)[6]$ | 3. $(1,10)[11]$ |
| :--- | :--- | :--- |
| 4. $(2,6)[8]$ | 5. $(1,5)[6]$ | 6. $(2,9)[11]$ |
| 7. $(5,4)[9]$ | 8. $(4,8)[12]$ | 9. $(4,4)[8]$ |
| 10. $(0,0)[0](0,1)[1]$ | 11. $(2,2)[4]$ | 12. $(4,1)[5]$ |
| 13. $(4,0)[4]$ | 14. $(4,3)[7]$ | 15. $(6,3)[9]$ |
| 16. $(9,2)[11](10,2)[12]$ | 17. $(10,0)[10]$ | 18. $(4,0)[4]$ |
| 19. $(0,0)[0]$ | 20. $(0,0)[0]$ | 21. $(1,0)[1]$ |
| 22. $(0,0)[0]$ | 23. $(0,-)[0]$ |  |

## 2. 2. 4. Texto resultante tras eliminar las oraciones marginales.

2. When polished, all metals shine owing to reflection of photons by external valence electrons dynamically forming metallic bonds. 3. White light reflects on most metals without color absorption or change to the naked eye; but copper and gold are yellow because they absorb "blue" and "red" photons by electron transitions between spectromeric configurations $n s^{1}(n-1) \mathrm{d}^{10} \quad n s^{2}(n-1) \mathrm{d}^{9}$ of external sublevels.
3. The next question is why silver, with the same external electronic configuration as copper and gold (group 11, IB), is not yellow. 5. The answer is simple, considering atomic radii, ionization potentials and nuclear charge:

|  | Cu | Ag | Au |
| :--- | :---: | :---: | :---: |
| Atomic radius/ pm | 117.3 | 133.9 | 133.6 |
| $1^{\text {st }}$ ionization energy $/ \mathrm{eV}$ | 7.725 | 7.576 | 9.22 |
| $2^{\text {nd }}$ ionization energy $/ \mathrm{eV}$ | 20.29 | 21.48 | 20.52 |
| Nuclear charge | 25 | 35 | 59 |

6. The atomic radius of silver is 16.6 pm larger than that of copper, allowing a bigger difference between sublevels $s$ and d, which is sufficient to restrict the transition $s^{1} d^{10} \quad s^{2} d^{9}$ to a lower probability. 7. This is equally supported by the first ionization energry: since it is lower in silver, the fact that one external electron is ejected more easily than in copper atoms is justified.
7. With their higher nuclear charge ( 35 vs 25 ) silver atoms also have larger radii ( $=16.6 \mathrm{pm}$ ), and the distance between external sublevels-both spatial and energeticis too large to freely allow $s$ d transitions. 9. However, the distance is not large enough to prevent the transitions completely, and after several reflections on two parallel silver mirrors, white light becomes pale yellow.
${ }^{4} 11$. According to the same line of reasoning, gold would be colorless if it had bigger atoms. 12. But gold atoms are not larger than silver; the radii of silver and gold are practically identical owing to lanthanide contraction. 13. Comparing ionization energies, the value 9.22 eV for gold is about $20 \%$ higher than 7.576 eV for silver because gold has a larger nuclear charge ( 59 vs 35 ) while its radius is practically the same. 14. Thus, external s and d sublevels are close enough to allow the necessary

[^3]transition. 15. As a result, the probability of transition between sublevels is similar to that of copper, and gold is again yellow.
16. We can now perceive the necessary conditions for a metal to be yellow, like copper and gold:

1. Adequate external electronic configuration $s^{1} d^{10} \quad s^{2} d^{9}$ (group 11, IB).
2. Sublevels $s$ and $d$ close enough to allow transitions $s^{1} d^{10} \quad s^{2} d^{9}$ to occur significantly ( $\mathrm{Cu}, \mathrm{Au}$ ).
3. In contrast, all other metals shine silvery, colorless to the naked eye because they do not possess the necessary electronic external configuration and transition probability to appear colored.
4. Much work has been undertaken in connection with relativistic effects on metal properties (6); however a final question remains: are metals (except for Cu and Au ) really colorless? 21. What number of atomic layers must be crossed (twice) in metals to produce a definite color?

## 2. 3. Texto 3: Both nylon and PET fibers burn continuously under atmospheric conditions.

1. We would like to present two series of photographs showing the characteristic burning behaviors of a nylon fiber and a polyethyleneterephthalate (PET) fiber, in order to help people safely handle these fibers in their everyday lives.
2. In many textbooks, especially on textiles, nylon and PET fibers are classified as flammable but self-extinguishing. 3. In other references, we have read that nylon and PET give off combustible gases when they are heated above their decomposition temperatures. 4. According to references, nylon gives propylene ( $8.8 \%$ in volume of total detected gases evolved), cyclopentanone ( $32.2 \%$ ), hexamethylenediamine and other methylene amines (22.5\%), and others (3), and PET gives ethylene $(8.3 \%$ in volume of total detected gases evolved), acetaldehyde ( $10.9 \%$ ), benzoic acid ( $37.5 \%$ ), and other phenyl compounds.
3. On the basis of these pyrolysis data, we were doubtful about the flammable but self-extinguishing classification for nylon and PET. 6. So we very carefully performed experiments to see what would happen when fibers caught fire. 7. We selected typical sewing threads for sewing machine (supplied by Teijin Co., Ltd., and Asahi Chemicals Co., Ltd.) for testing.
4. Thread samples about 50 cm long were hung up just in front of a focused camera and then ignited at the bottom end with a tiny flame from a cigarette lighter. 9. A tiny flame was used because hot air ascended from a big flame and perturbed the thread. 10. Once a part of the terminal end was ignited, it burned continuously, as
shown in the series of photographs in this paper, in contrast to the descriptions in textbooks stating that it "burns slowly but if the sample is removed from the flame it self-extinguishes".
5. In nylon thread, as seen in Figure 1, the flame propagates slowly. 12. In PET thread, shown in Figure 2, the flame propagates more quickly and is accompanied by black smoke. 13. Unlike natural fibers such as cellulose, these materials first melt, then give off combustible gases when the temperature exceeds the decomposition temperature of the polymers in the presence of about $21 \%$ of oxygen (i.e., under atmospheric conditions).14. When the ignition flame was removed, the threads continued to burn. 15. During the course of the burning, pictures were taken of the small spherical fire balls composed of a molten polymer. 16. A shutter speed of one onethousandth of a second and a highly sensitive film (ASA 800) were used. 17. Because the fire ball is changing rapidly, the photographs show scenes that cannot be seen by the naked eye.
6. Caution: We urge you to remember that these small fire balls are composed of viscous molten polymer. 19. They have specific heats that are not only high enough to burn skin but also high enough to cause a big fire. 20. If you want to do this type of experiment, you should wear a glove made of non- flammable fibers so your hand will not be burned.

## 2. 3. 1. Matriz de repetición de unidades léxicas.



| 1 |  | 2 | 3 | 4 | - |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 11 | psp. photographs - figure <br> rs. nylon - nylon | rs. nylon nylon | rs. nylon - nylon | rs. nylon - nylon | rs. nylon - nylon |
| 12 | psp. photographs <br> - figure <br> rs. showing shown <br> rs. PET - PET | rs. PET - PET | rs. PET - PET | rs. PET - PET | rs. PET - PET |
| 13 | hip. nylon/PET polymers <br> rs. fibers - fibers | hip. <br> nylon/PET - <br> polymers <br> rs. fibers - <br> fibers <br> psm. <br> flammable - <br> combustible | hip. nylon/PET polymers <br> rs. give off - give off <br> rs. combustible combustible <br> rs. gases - gases <br> rs. decomposition <br> - decomposition <br> rs. temperatures temperature | hip. nylon/PET polymers psm. gives - give off rs. gases - gases | psm. flammable combustible hip. nylon/PET polymers |
| 14 | rc. burning burn |  |  |  |  |
| 15 | psm. <br> photographs pictures <br> rc. burning burning hip. nylon/PET polymer | hip. <br> nylon/PET - <br> polymer | hip. nylon/PET polymer | hip. nylon/PET polymer | hip. nylon/PET polymer |
| 16 |  |  |  |  |  |
| 17 | rs. photographs <br> - photographs rs. showing show |  |  |  |  |
| 18 | rs. we - we+ hip. nylon/PET polymer | hip. nylon/PET polymer | rs. we - we + hip. nylon/PET polymer | hip. nylon/PET polymer | rs. we - we+ hip. nylon/PET polymer |
| 19 | rc. burning burn |  | rc. heated - heats |  |  |
| 20 | rc. burning burned rc. handle hand rs. fibers - fibers | rs. fibers - <br> fibers <br> rc. flammable <br> - non- <br> flammable | hip. nylon/PET fibers pc. combustible -non-flammable (flammable) | hip. nylon/PET fibers | rc. flammable - nonflammable hip. nylon/PET - fibers |

## Anexo

| 7 | rs. we - we+ |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 7 | 8 | 9 |  |
| 8 | psm. caught fire - ignited | rs. threads thread |  |  |  |
| 9 |  | rs. threads thread | rs. threadthread <br> e. samples - 0 <br> rs. tiny - tiny <br> rs. flame - flame |  |  |
| 10 | psm. caught fire - ignited |  | e. thread - 0 <br> rs. samples sample <br> rs. ignited ignited psm. bottom terminal rs. end - end | rs. flame - flame <br> e. thread - 0 | 10 |
| 11 | tr. fibers - nylon <br> rs. see - seen | rs. threads thread | rs. thread - <br> thread <br> rs. flame - flame | rs. thread - thread <br> rs. flame - flame | psp. photographs figure rs. slowly - slowly rs. flame - flame |
| 12 | tr. fibers - PET | rs. threads thread | rs. thread thread rs. flame - flame | rs. thread - thread <br> rs. flame - flame | rs. shown - shown psp. photographs figure <br> a. slowly - quickly <br> rs. flame - flame |
| 13 | rs. fibers - fibers | hip. threads materials | hip. threads materials | hip. threads materials |  |
| 14 | pc. caught fire ignition (ignited) | rs. threads threads | rs. thread treads <br> e. samples - 0 <br> rc. ignited - <br> ignition <br> rs. flame - flame | rs. flame - flame <br> rs. tread - threads | rc. ignited - ignition <br> psm. burned <br> continuously - <br> continued to burn <br> rs. removed - removed <br> rs. flame - flame |
| 15 | tr. fibers polymer |  |  |  | psp. photographs pictures rc. burns - burning |


| 6 |  |  |  | 7 |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 16 |  |  | rs. used - used |  |  |
| 17 | rs. see - seen |  |  | rs. shown - show <br> rs. photographs - <br> photographs |  |
| 18 | rs. we - we+ <br> tr. fibers - <br> polymer | rs. we - we+ |  |  |  |
| 19 | rs. fire - fire |  |  | rs. burns - burn |  |
| 20 | rs. experiments - <br> experiment <br> rs. fibers - fibers |  | d. oración 8-this |  | rs. burns - burned |

## Anexo

| 12 | tr. nylon - PET rs. thread thread rs. figure - figure rs. flame - flame rs. propagates propagates a. slowly quickly | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | hip. nylon polymers hip. thread materials | hip. PET polymers hip. thread materials | 13 |  |  |
| 14 | rs. thread threads <br> rs. flame - flame | rs. thread threads rs. flame - flame | tr. materials threads | 14 |  |
| 15 | psp. figure pictures hip. nylon polymer | psp. figure pictures hip. PET polymer | pc. melt - molten (melted) rs. polymers polymer | rc. burn - burning | 15 |
| 16 |  |  |  |  |  |
| 17 | psp. figure photographs rs. seen - seen | rs. shown - show psm. figure photographs |  |  | psm. pictures - <br> photographs <br> rs. fire - fire <br> rs. balls - ball |
| 18 | hip. nylon polymer | hip. PET polymer | pc. melt - molten (melted) rs. polymers polymer |  | rs. small - small <br> rs. fire - fire <br> rs. balls - balls <br> rs. composed composed rs. molten - molten rs. polymer polymer |
| 19 |  |  |  | rs. burn - burn | rc. burning - burn s. small ....balls they rs. fire - fire |
| 20 | hip. nylon fibers | hip. PET - fibers | rs. fibers - fibers pc. combustible -non-flammable (flammable) | rs. burn - burned | psm. composed - <br> made* <br> rc. burning - <br> burned <br> tr. polymer - fibers |



## 2. 3. 2. Matriz con el número de unidades léxicas.




## 2. 3. 3. Tabla representativa del número de conexiones entre oraciones.

1. $(-, 8)[8]$
2. $(1,5)[6]$
3. $(2,3)[5]$
4. $(3,2)[5]$
5. $(4,1)[5]$
6. $(0,0)[0]$
7. $(0,0)$ [0]
8. $(0,3)[3]$
9. $(1,0)[1]$
10. $(4,3)[7]$
11. $(1,1)[2]$
12. $(3,0)[3]$
13. $(3,0)[3]$
14. $(2,0)[2]$
15. $(1,3)[4](1,4)[5]$
16. $(0,0)[0]$
17. $(1,0)[1]$
18. $(1,0)[1](1,1)[2]$
19. (1,-) [1] (3,-) [3]

## 2. 3. 4. Texto resultante tras eliminar las oraciones marginales.

1. We would like to present two series of photographs showing the characteristic burning behaviors of a nylon fiber and a polyethyleneterephthalate (PET) fiber, in order to help people safely handle these fibers in their everyday lives.
2. In many textbooks, especially on textiles, nylon and PET fibers are classified as flammable but self-extinguishing. 3. In other references, we have read that nylon and PET give off combustible gases when they are heated above their decomposition temperatures. 4. According to references, nylon gives propylene ( $8.8 \%$ in volume of total detected gases evolved), cyclopentanone (32.2\%), hexamethylenediamine and other methylene amines (22.5\%), and others (3), and PET gives ethylene ( $8.3 \%$ in volume of total detected gases evolved), acetaldehyde (10.9\%), benzoic acid (37.5\%), and other phenyl compounds.
3. On the basis of these pyrolysis data, we were doubtful about the flammable but self-extinguishing classification for nylon and PET.
4. Thread samples about 50 cm long were hung up just in front of a focused camera and then ignited at the bottom end with a tiny flame from a cigarette lighter. 9. A tiny flame was used because hot air ascended from a big flame and perturbed the thread. 10. Once a part of the terminal end was ignited, it burned continuously, as shown in the series of photographs in this paper, in contrast to the descriptions in textbooks stating that it "burns slowly but if the sample is removed from the flame it self-extinguishes".
5. In nylon thread, as seen in Figure 1, the flame propagates slowly. 12. In PET thread, shown in Figure 2, the flame propagates more quickly and is accompanied by black smoke. 13. Unlike natural fibers such as cellulose, these materials first melt, then give off combustible gases when the temperature exceeds the decomposition temperature of the polymers in the presence of about $21 \%$ of oxygen (i.e., under atmospheric conditions).14. When the ignition flame was removed, the threads continued to burn. 15. During the course of the burning, pictures were taken of the small spherical fire balls composed of a molten polymer. 17. Because the fire ball is changing rapidly, the photographs show scenes that cannot be seen by the naked eye.
6. Caution: We urge you to remember that these small fire balls are composed of viscous molten polymer. 19. They have specific heats that are not only high enough to burn skin but also high enough to cause a big fire. 20. If you want to do this type of experiment, you should wear a glove made of non- flammable fibers so your hand will not be burned.

## 2. 4. Texto 4: A chromatographic parable.

1. In thirty years of teaching separations courses, I have often searched for an apt allegory to illustrate the fundamentals of chromatographic processes. 2. The following is one version of such a tale that students seem to find interesting and perhaps even informative.
2. In a small Southern town (it must be a Southern town or the story doesn't work), the people are planning a Fourth of July race from one end of town to the other. 4. The townsfolk have the commonly observed characteristics that most of them are either Saints or Sinners; however, some of the folks are neither Saints nor Sinners (The Agnostic-Teetotalers) and others are both Saints and Sinners (we'11 call this group the Hypocrites). 5. The race will be conducted along the main street of town, and, as in most Southern towns, the street is lined with a suitable collection of churches and bars.
3. During the race the town folks all run at the same speed, but the Saints cannot pass a church without entering to pray for a while, and the Sinners cannot possibly pass by a bar without pausing for a refreshing beer. 7. The immediate question then is who will win the 4th of July race? 8. Most people want the Saints to win the race, but this is not probable because, while they are in church, the Agnostic-Teetotalers are still running. 9. It is fairly obvious, even to college students, that the Agnostic-Teetotalers will win the race, and, quite deservedly, the Hypocrites will come in last. 10. But what about the Saints and Sinners? 11. Who will come in second or third? 12. And finally, what can be done by the City Fathers to alter the outcome of the race next year?
4. So, what will determine the results of the Saints-Sinners race? 14. Let's say there are ten churches, but only three bars, along the main street. 15. Under these conditions, the Sinners will win the race. 16. Right? 17. Watch out! 18. What if it takes longer to drink a beer than it does to say a prayer?
5. The point of the exercise is to illustrate the concept that the results of this particular race are determined by the amount of time the participants spend not racing, that is, drinking or praying as the case may be. 20. The analogy to chromatographic retention times is obvious if somewhat colloquial. 21. Unfortunately, the analogy between the chromatographic stationary phase and a church or bar is perhaps less exemplary.
6. A secondary effect is possible if not all the racers run at exactly the same speed, if some Saints pray longer than others, or if some Sinners have more than one beer. 23. In this case, not all the Sinners will reach the finish line at the same time. 24. It is even possible that some very fast Saints could reach the finish line (elute) before some of the more tipsy Sinners or vice versa. 25. Thus, there would be a distribution of individuals within a group of townsfolk and possible overlap of Saints and Sinners at the finish line. 26. In chromatographic terms, the distribution is known as dispersion (described by the universally dreaded van Deemter equation) and overlap results in poor resolution. 27. Both effects lead to diminished results for a chromatographic separation. 28. In the 4th of July race analogy, it is possible that all the townsfolk (Saints, Sinners, Agnostics, and Hypocrites alike) would finish the race at the same time. 29. In my experience, this is the most probable outcome for most Southern towns, as well as most chromatographic experiments.

## 2. 4. 1. Matriz de repetición de unidades léxicas.



|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 16 |  |  |  |  |  |
| 17 |  |  |  |  |  |
| 18 |  |  |  |  |  |
| 19 | psp. allegory exercise rs. illustrate illustrate | psp. tale exercise | psp. people participants rs. race - race | psp. towsfolk - participants | rs. race - race |
| 20 | rs. chromatographic <br> - chromatographic |  |  |  |  |
| 21 | pc. illustrate exemplary <br> (illustrative) <br> rs. chromatographic <br> - chromatographic |  |  |  | rs. churches church <br> rs. bars - bar |
| 22 |  |  | tr. people - <br> Saints / <br> Sinners <br> rc. race - <br> racers | psp. towsfolk - racers <br> rs. Saints - Saints <br> rs. Sinners - Sinners | rc. race - racers |
| 23 |  |  | tr. people Sinners | rs. Sinners - Sinners |  |
| 24 |  |  | tr. people Saints / Sinners | rs. Saints - Saints <br> rs. Sinners - Sinners |  |
| 25 |  |  | psm. people townsfolk | rs. towsfolk - towsfolk <br> rs. Saints - Saints <br> rs. Sinners - Sinners | pc. race - townsfolk (racers) |
| 26 | rs. chromatographic <br> - chromatographic |  |  |  |  |
| 27 | rs. chromatographic <br> - chromatographic |  |  |  |  |
| 28 |  |  | psm. people townsfolk psm. Fourth $4^{\text {th }}$ <br> rs. July - July <br> rs. race - race | rs. towsfolk - towsfolk <br> rs. Saints - Saints <br> rs. Sinners - Sinners <br> rs. Agnostic - Agnostics <br> rs. Hypocrites - Hypocrites | rs. race - race |
| 29 | rs. I - my+ <br> rs. chromatographic <br> - chromatographic |  | rs. Southern - <br> Southern <br> rs. town towns |  | rs. Southern Southern <br> rs. towns - towns |

## Anexo



| 6 |  | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 20 |  |  |  |  |  |
| 21 | rs. church - church <br> rs. bar - bar |  | rs. church - church |  |  |
| 22 | rc. race - racers <br> rs. run - run <br> rs. same - same <br> rs. speed - speed <br> rs. Saints - Saints <br> rs. pray - pray <br> rs. Sinners - Sinners <br> rs. beer - beer | rc. race racers | rs. Saints - Saints <br> rc. race - racers <br> psm. are in church - pray <br> tr. Agnostic-Teetotalers - <br> Sinners <br> rs. running - run | tr. Agnostic- <br> Teetotalers / <br> Hypocrites - Saints <br> / Sinners <br> rc. race - racers | rs. Saints - <br> Saints <br> rs. Sinners - <br> Sinners |
| 23 | rs. Sinners - Sinners |  | tr. Saints / AgnosticTeetotalers - Sinners | tr. AgnosticTeetotalers /Hypocrites Sinners | rs. Sinners - <br> Sinners |
| 24 | rs. Saints - Saints <br> rs. Sinners - Sinners |  | rs. Saints - Saints <br> tr. Agnostic-Teetotalers - <br> Sinners | tr. Agnostic- <br> Teetotalers / <br> Hypocrites - Saints <br> / Sinners | rs. Saints - <br> Saints <br> rs. Sinners - <br> Sinners |
| 25 | psm. town folks townfolk <br> rs. Saints - Saints <br> rs. Sinners - Sinners | pc. race townsfolk (racers) | rs. Saints - Saints <br> pc. race - townsfolk <br> (racers) <br> tr. Agnostic-Teetotalers Sinners | tr. Agnostic- <br> Teetotalers / <br> Hypocrites - Saints <br> Sinners <br> pc. race - townsfolk <br> (racers) | rs. Saints Saints rs. Sinners Sinners |
| 26 |  |  |  |  |  |
| 27 |  |  |  |  |  |
| 28 | rs. race - race <br> psm. town folks townfolk <br> rs. Saints - Saints <br> rs. Sinners - Sinners | rs. 4th 4th rs. July July rs. race race | rs. Saints - Saints <br> rs. race - race <br> rs. Agnostic - Agnostics | rs. Agnostic Agnostics rs. race - race rs. Hypocrites Hypocrites | rs. Saints - <br> Saints <br> rs. Sinners - <br> Sinners |
| 29 | rs. town - towns |  |  |  |  |

## Anexo




## Anexo



| 26 | rs. distribution - <br> distribution <br> rs. overlap - overlap | 26 |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :---: | :---: |
| 27 |  | rs. chromatographic - <br> chromatographic | 27 |  |  |  |
| 28 | rs. townsfolk - <br> townsfolk <br> rs. Saints - Saints <br> rs. Sinners - Sinners <br> rc. finish - finish | rs. chromatographic - <br> chromatographic | rs. chromatographic - <br> chromatographic | d. oración 28 - this <br> psm. possible - probable |  |  |
| 29 |  |  |  |  |  |  |

2. 4. 2. Matriz con el número de unidades léxicas.


1. 4. 3. Tabla representativa del número de conexiones entre oraciones.
1. $(-, 0)[0]$
2. $(0,0)[0]$
3. $(0,3)[3]$
4. $(0,7)[7]$
5. $(1,2)[3]$
6. $(2,6)[8]$
7. (1,1) [2]
8. $(2,7)$ [9]
9. $(2,2)[4]$
10. $(0,0)[0]$
11. $(0,0)[0]$
12. $(0,0)[0]$
13. $(3,4)[7]$
14. $(1,0)[1]$
15. $(2,0)[2]$
16. $(0,0)[0]$
17. $(0,0)[0]$
18. $(0,1)[1]$
19. $(3,1)[4]$
20. $(0,0)[0]$
21. $(0,0)[0]$
22. $(6,2)[8]$
23. $(0,3)[3]$
24. $(1,2)[3]$
25. $(7,1)[8]$
26. $(0,0)[0]$
27. $(0,0)$ [0]
28. $(11,0)[11]$
29. (0,-) [0]

## 2. 4. 4. Texto resultante tras eliminar las oraciones marginales.

3. In a small Southern town (it must be a Southern town or the story doesn't work), the people are planning a Fourth of July race from one end of town to the other.4. The townsfolk have the commonly observed characteristics that most of them are either Saints or Sinners; however, some of the folks are neither Saints nor Sinners (The Agnostic-Teetotalers) and others are both Saints and Sinners (we'11 call this group the Hypocrites). 5. The race will be conducted along the main street of town,
and, as in most Southern towns, the street is lined with a suitable collection of churches and bars.
4. During the race the town folks all run at the same speed, but the Saints cannot pass a church without entering to pray for a while, and the Sinners cannot possibly pass by a bar without pausing for a refreshing beer. 7. The immediate question then is who will win the 4th of July race? 8. Most people want the Saints to win the race, but this is not probable because, while they are in church, the AgnosticTeetotalers are still running. 9. It is fairly obvious, even to college students, that the Agnostic-Teetotalers will win the race, and, quite deservedly, the Hypocrites will come in last.
5. So, what will determine the results of the Saints-Sinners race? 14. Let's say there are ten churches, but only three bars, along the main street. 15. Under these conditions, the Sinners will win the race. 18. What if it takes longer to drink a beer than it does to say a prayer?
6. The point of the exercise is to illustrate the concept that the results of this particular race are determined by the amount of time the participants spend not racing, that is, drinking or praying as the case may be.
7. A secondary effect is possible if not all the racers run at exactly the same speed, if some Saints pray longer than others, or if some Sinners have more than one beer. 23. In this case, not all the Sinners will reach the finish line at the same time. 24. It is even possible that some very fast Saints could reach the finish line (elute) before some of the more tipsy Sinners or vice versa. 25. Thus, there would be a distribution of individuals within a group of townsfolk and possible overlap of Saints and Sinners at the finish line. 28. In the 4th of July race analogy, it is possible that all the townsfolk (Saints, Sinners, Agnostics, and Hypocrites alike) would finish the race at the same time.

## 2. 5. Texto 5: High flying polymer.

1. A new type of fire-resistant polymer could improve your chances of survival in a plane crash, according to Phillip Westmoreland, professor of chemical engineering at the University of Massachusetts Amherst in the US.
2. Much of today's aircraft interiors are made of polymers because they are lightweight and versatile - they can be dyed different colours and formed into many shapes. 3. They are used in seats, windows, wall panels, floor carpets, wiring, insulation, 'just about everything except the metal chair supports', says Westmoreland.
3. When a plane crashes and catches fire, polymers decompose from the heat, releasing combustible gases, which in turn also catch fire. 5. According to Westmoreland's co-researcher Richard Lyon, Federal Aviation Authority (FAA), programme manager for fire research and fire safety, 40 per cent of the fatalities that occur in impact survivable air accidents are a result of fire. 6. Fire-resistant polymers are therefore an important target.
4. Westmoreland and his team focused on polyhydroxyamide (PHA) as a potential candidate for a fire - resistant polymer. 8. The backbone structure of PHA meant that it could be a useful thermoplastic (softens on heating) for forming into films and fibres. 9. Also at temperatures of ca $180-200^{\circ} \mathrm{C}$, PHA converts with very little mass loss to water and a different polymer, ie the rigid high - strength polybenzoxazole (PBO, 2), which decomposes only at very high temperatures ( $\mathrm{ca} 600^{\circ} \mathrm{C}$ ). 10. 'PBO has the best non-flammability of any material we know of, but you just can't use the stuff', commented Westmoreland. 11. PBO is too hard to form into useful products, such as fabrics or panels.
5. Researchers at the University of Massachusetts synthesised several structural variants of PHA, from the simplest form $(\mathrm{R}=\mathrm{H})$, to phosphate-containing R.-groups, to see which had the lowest flammability. 13. At the same time, a team at the FAA developed a new microcalorimeter that could evaluate the polymers' ability to burn in milligram quantities, a method with advantages over conventional tests which involve much larger samples - eg 'taking an aircraft seat and setting fire to it'. 14. The results revealed that all forms of PHA had low flammability, but the best polymer was the simplest - ie when $\mathrm{R}=\mathrm{H}$. 15. In tests, this form of PHA gave passengers ca 10 times longer to get out of an aircraft than the best existing polymer.

## 2. 5. 1. Matriz de repetición de unidades léxicas.

| 2 | rs. polymer polymers psm. plane aircraft | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rs. Westmoreland <br> - Westmoreland | s. polymers they | 3 |  |  |
| 4 | rs. fire - fire <br> rs. polymer polymers <br> rs. plane - plane <br> rc. crash - crashes | psm. aircraft plane rs. polymers polymers |  | 4 |  |
| 5 | rs. fire - fire rc. survival survivable <br> psm. crash impact rs. Westmoreland - Westmoreland |  | rs. <br> Westmoreland - <br> Westmoreland | rs. fire - fire | 5 |
| 6 | rs. fire - fire rs. resistant resistant rs. polymer polymers | rs. polymers polymers |  | rs. fire - fire <br> rs. polymers - polymers | rs. fire - fire |
| 7 | rs. fire - fire <br> rs. resistant resistant <br> rs. polymer polymer rs. Westmoreland - Westmoreland | rs. polymers polymer | rs. <br> Westmoreland - <br> Westmoreland | rs. fire - fire <br> rs. polymers - polymer | rs. Westmoreland Westmoreland rs. fire - fire |
| 8 | tr. polymer PHA | tr. polymer PHA <br> rs. formed forming | rc. used - useful | tr. polymers PHA <br> rc. heat - heating |  |
| 9 | rs. polymer polymer | rs. polymers polymer |  | rs. polymers - polymer <br> rs. decompose decomposes |  |
| 10 | tr. polymer PBO rs. Westmoreland - Westmoreland | tr. polymers PBO | $\begin{aligned} & \text { rs. used - use } \\ & \text { psm. says - } \\ & \text { commented } \\ & \text { rs. } \\ & \text { Westmoreland - } \\ & \text { Westmoreland } \end{aligned}$ | tr. polymers PBO pc. combustible flammability <br> (flammable) | rs. Westmoreland Westmoreland |

## Anexo

| 1 |  |  |  | 2 |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 11 | tr. polymer - PBO | tr. polymers- PBO <br> rs. formed - form | rc. used - useful <br> rs. panels - panels | tr. polymers - PBO |  |
| 12 | tr. polymer - PHA <br> rs. University - <br> University <br> rs. Massachusetts <br> - Massachusetts | tr. polymers - PHA |  | tr. polymers - PHA <br> pc. combustible - <br> flammability <br> (flammable) |  |
| 13 | rs. fire - fire <br> rs. polymer - <br> polymers <br> psm. plane - <br> aircraft | rs. aircraft - aircraft <br> rs. polymers - <br> polymers | rs. seats - seat | psm. plane - <br> aircraft <br> rs. fire - fire <br> rs. polymers- <br> polymers |  |
| 14 | rs. polymer - <br> polymer | rs. polymers - <br> polymer | rs. FAA - FAA <br> rs. fire - fire |  |  |
| 15 | psm. type - form <br> rs. polymer - <br> polymer <br> psm. plane - <br> aircraft | rs. aircraft - aircraft <br> rs. polymers - <br> polymer | rs. polymers - <br> polymer <br> pc. combustible - <br> flammability <br> (flammable) |  |  |



## Anexo

| 12 | tr. PBO - PHA |  |  |  |
| :--- | :--- | :--- | :--- | :--- |
| 13 | hip. PBO - polymers | hip. PHA - polymers |  |  |
| 14 | hip. PBO - polymer | rs. PHA - PHA <br> rs. simplest - simplest <br> rs. form - forms <br> rs. R=H - R=H <br> rs. had - had <br> rc. lowest - low <br> rs. flammability - flammability | rs. polymers - polymer |  |
| 15 | hip. PBO - polymer | rs. PHA - PHA <br> rs. form - form <br> pc. see - tests (test) | rs. polymers - polymer <br> rs. tests - tests <br> rs. aircraft - aircraft | rs. forms - form <br> rs. PHA - PHA <br> rs. polymer - polymer |

## 2. 5. 2. Matriz con el número de unidades léxicas.


2. 5. 3. Tabla representativa del número de conexiones entre oraciones.

1. $(-, 7)[7]$
2. $(0,0)[0]$
3. $(0,1)[1]$
4. $(1,1)[2]$
5. $(1,0)[1]$
6. $(1,1)[2]$
7. $(2,1)[3]$
8. $(0,0)[0]$
9. $(0,0)[0]$
10. $(1,2)[3]$
11. $(0,0)[0]$
12. $(2,2)[4]$
13. $(3,1)[4]$
14. $(2,1)[3]$
15. $(4,-)[4]$

## 2. 5. 4. Texto resultante tras eliminar las oraciones marginales.

1. A new type of fire-resistant polymer could improve your chances of survival in a plane crash, according to Phillip Westmoreland, professor of chemical engineering at the University of Massachusetts Amherst in the US.
2. <They> [polymers] are used in seats, windows, wall panels, floor carpets, wiring, insulation, 'just about everything except the metal chair supports', says Westmoreland.
3. When a plane crashes and catches fire, polymers decompose from the heat, releasing combustible gases, which in turn also catch fire. 5. According to Westmoreland's co-researcher Richard Lyon, Federal Aviation Authority (FAA), programme manager for fire research and fire safety, 40 per cent of the fatalities that
occur in impact survivable air accidents are a result of fire. 6. Fire-resistant polymers are therefore an important target.
4. Westmoreland and his team focused on polyhydroxyamide (PHA) as a potential candidate for a fire - resistant polymer. 10. 'PBO has the best nonflammability of any material we know of, but you just can't use the stuff', commented Westmoreland.
5. Researchers at the University of Massachusetts synthesised several structural variants of PHA, from the simplest form ( $\mathrm{R}=\mathrm{H}$ ), to phosphate-containing R.groups, to see which had the lowest flammability. 13. At the same time, a team at the FAA developed a new microcalorimeter that could evaluate the polymers' ability to burn in milligram quantities, a method with advantages over conventional tests which involve much larger samples - eg 'taking an aircraft seat and setting fire to it'. 14. The results revealed that all forms of PHA had low flammability, but the best polymer was the simplest - ie when $\mathrm{R}=\mathrm{H}$. 15. In tests, this form of PHA gave passengers ca 10 times longer to get out of an aircraft than the best existing polymer.

## 2. 6. Texto 6: Flash of inspiration wins Nobel prize for chemistry.

1. Chemist Ahmed Zewail of the California Institute of Technology (Caltech) (pictured top right) was the recipient of the 1999 Nobel prize for chemistry for a flash of inspiration that is revolutionising our understanding of chemical reactions. 2. Using brief bursts of light from lasers, he developed a way to take 'snapshots of individual molecules as they change during a chemical reaction.
2. Modern lasers can produce a very short burst of light, lasting a few femtoseconds ie a million-billionth of a second. 4. Like a fast camera that freezes a dancer in motion, the laser beam can illuminate a molecule as it is transformed from one shape and structure to another during a chemical reaction - its transition state. 5. This transition state, which exists between the reactant and the product, lasts for only femtoseconds, so observing it before it disappears was, until Zewail's experiments, almost impossible. 6. Being able to observe this state is helping chemists to find out exactly how particular reactions work and allowing them to predict the outcome of other related reactions as well as the complex interactions of, for example, a drug molecule with a biological receptor.
3. The earliest attempt to look at reactions as they happen was by H. Hartridge and EJ. Roughton in the 1920s. 8. They used a spectrophotometer to observe what happens when two compounds are mixed and saw chemical reactions taking place in a thousandth of a second. 9. In the 1960s, Ronald Norrish and George Porter came up with the idea of using a flash-lamp to freeze the reactions - the shorter the flash, the
more transient the reactions they could see. 10. They observed chemistry on the millisecond and microsecond timescales - a thousand times shorter than that possible in the 1920s. 11. Poter and Norrish shared the 1967 Nobel prize with the German chemist Manfred Eigen, who used heat and pressure shock methods to trigger a reaction and observe 'almost' the instant at which it was happening (Eigen was also working at the milli-to micro-second timescale).
4. During the early 1980s, Dudley Herschbach, Yuan Lee and John Polanyi had improved the ability to observe chemical reactions down to the picosecond scale using a vacuum collision experiments - for this work they received the 1986 Nobel prize for chemistry. 13. With shorter and shorter timescales, chemists began to reveal the intermediate chemical species in reactions - not, the transition states, they were still too fleeting, but the structures either side that lasted just long enough for them to record. 14. Once chemists had reached the picoscale, they only needed to take one step further to reach the femto timescale. 15. The femtosecond $-10^{-15} \mathrm{~s}$ - represents the frequency at which molecules vibrate, without which there would be no interaction and no chemical change. 16. If chemists could watch molecular vibrations they would have reached the limit of observation.
5. Zewail realised that to observe molecules at this level his flashlamp would have to be very fast, a pulsing laser that flashes once every femtosecond, he reasoned, would do the job. 18. For their simplest experiment, Zewail and his colleagues chose a unimolecular reaction, ie where a single substance changes into another without the involvement of a second chemical, and formed a molecular beam in a vacuum chamber. 19. By blasting this beam with a 'pump pulse' of laser light they excited the molecules and triggered a change. 20. Then, by applying a weaker, 'probe pulse' from a laser lasting a few femtoseconds - at a frequency to coincide with the absorption frequency of the suspected transition state of the substance - Zewail and his team obtained a characteristic spectrum from the light emitted by the transition state. $\mathbf{2 1}$. They had frozen the reaction.
6. The chemists compared the characteristic spectrum with the theoretical pattern obtained by using the methods of last year's Nobel chemists John Pople and Walter Kohn (Educ. Chem., 1999, 36(1), 7) who provided them with the means to predict molecular structure and so their characteristic spectra. 23. Zewail's first
unimolecular reaction - the one that started the whole femtochemistry field - was the dissociation of iodine cyanide (ICN), which takes just 200 femtoseconds. 24. His results were published in 1987 in the journal of physical chemistry and showed the transition state just as the carbon-iodine bond in the molecule is about to break to form the cyano radical and an iodine atom.
7. Zewail and his colleagues then moved on to bimolecular reactions, which involve two interacting chemical species. 26. They studied the reaction of hydrogen with carbon dioxide, which produces carbon monoxide and hydroxy radicals. 27. Zewail's flash revealed that the reaction passes through a transitionary HOCO molecule, which exists fleetingly for a mere picosecond (1000fs). 28. His team also began to look at a puzzle that had occupied chemical minds for some time - ie would two seemingly identical bonds in a molecule break simultaneously in, for instance, a dissociation reaction. 29. For the dissociation of tetrafluorodiiodoethane it turns out that the 'equivalent' $\mathrm{C}-1$ bonds do not break at the same time - there is a delay of 200 fs following the splitting of the first.
8. Since Zewail's pioneering studies in the 1980s and 1990s, many other research teams have begun to use femtochemistry to look at diverse reactions watching them happen in real-time.

## 2. 6. 1. Matriz de repetición de unidades léxicas.

1

| 2 | s. Ahmed Zewail he rs. chemical chemical <br> rs. reactions reaction | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 |  | psm. brief - short rs. bursts - burst rs. light - light rs. lasers - lasers | 3 |  |  |
| 4 | rs. chemical chemical rs. reactions reaction | ```psp. bursts - flash rs. lasers - laser rs. molecules - molecule psm. change - transformed rs. chemical - chemical rs. reaction - reaction``` | rs. lasers - laser psp. burst - flash | 4 |  |
| 5 | rs. Zewail - Zewail rc. reactions reactant | psm. take 'snapshots' observing <br> rc. reaction - reactant | rs. lasting lasts* rs. femtoseconds - femtoseconds | rc. reaction reactant rs. transition transition rs. state - state | 5 |
| 6 | rs. chemist chemists rs. reactions reactions | psm. take 'snapshots' observe <br> rs. molecules - molecule <br> rc. chemical - chemists <br> rs. reaction - reactions |  | rs. molecule molecule rc. chemical chemists rs. reaction reactions rs. state - state | rs. state - state <br> rc. reactant - reactions <br> rs. observing observe |
| 7 | rs. reactions reactions | psm. take 'snapshots' look at rs. reaction - reactions |  | rs. reaction reactions | rc. reactant - reactions psm. observing - look at |
| 8 | rs. chemical chemical rs. reactions reactions | rs. using - used <br> psm. take 'snapshots' observe <br> rs. chemical - chemical <br> rs. reaction - reactions | rs. second second* | rs. chemical chemical rs. reaction reactions | rc. reactant - reactions <br> rc. femtoseconds second rs. observing observe |
| 9 | psp. inspiration idea <br> rs. reactions reactions | ```rs. using - using pc. brief - shorter (short) psm. bursts - flah psm. take 'snapshots' - see rs. reaction - reactions``` | rc. short - <br> shorter <br> psm. burst - <br> flash | rs. flash - flash rs. freezes freeze rs. reaction reactions rc. transition transient | rc. transition transient rc. reactant - reactions psm. observing - see |
| 10 | rs. chemistry chemistry | psm. take 'snaoshots' observed rc. chemical - chemistry | rc. second millisecond* | rc. chemical chemistry | tr. femtoseconds millisecond rs. observing observed |


|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 11 | rs. chemist - chemist <br> rs. Nobel - Nobel <br> rs. prize - prize <br> rs. reactions - reaction | rs. using - used psm. way - methods psm. take 'snaoshots' observe <br> rc. chemical - chemist <br> rs. reaction - reaction | rc. second microsecond* | rc. chemical chemist rs. reaction reaction | rc. reactant reaction tr. femtoseconds microsecond rs. observing observe |
| 12 | pc. recipient received (receiver) <br> rs. Nobel - Nobel <br> rs. prize - prize <br> rs. chemistry chemistry rs. chemical chemical <br> rs. reactions reactions | rs. using - using <br> psm. take 'snapshots - <br> observe <br> rs. chemical - <br> chemical <br> rs reaction - reactions | rc. second picosecond* | rs. chemical chemical rs. reaction reactions | rc. reactant reactions tr. femtoseconds picosecond rs. observing observe rs. experiments experiments |
| 13 | rs. chemist - chemists <br> rs. chemical chemical <br> rs. reactions reactions | rs. chemical chemical rs. reaction - reactions | rs. lasting lasted* | rs. structure structures rs. chemical chemical <br> rs. reaction reactions <br> rs. transition transition rs. state states | rs. transition transition <br> rs. state - states <br> rc. reactant reactions <br> rs. lasts - lasted |
| 14 | rs. chemist - chemists | rc. chemical - chemists |  | rc. chemical chemists |  |
| 15 | rs. chemical chemical psm. reactions change | rs. molecules molecules rs. chemical chemical rc. change - change | rs. femtoseconds <br> - femtosecond | rs. molecule molecules <br> rs. chemical chemical psm. reaction - change | rs. femtoseconds femtosecond pc. reactant change (reaction) |
| 16 | rs. chemist - chemists | psm. take 'snapshots' watch rc. molecules molecular rc. chemical - chemists |  | rc. molecule molecular rc. chemical chemists | rc. observing observation |
| 17 | rs. Zewail - Zewail | pc. bursts - flashes <br> (flash) <br> rs. lasers - laser <br> rs. he - he <br> psm. take 'snapshots' - <br> observe <br> rs. molecules molecules | rs. lasers - laser pc. burst flashes (flash) rs. light - light | rs. fast - fast rc. flash flashes rs. laser laser rs. molecule molecules | rs. femtoseconds femtosecond rs. observing observe <br> rs. Zewail - Zewail |


|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 18 | rs. Zewail - Zewail <br> rc. chemical chemical <br> rs. reactions reaction | rs. he - his <br> rc. molecules molecular <br> rs. change - changes <br> rc. chemical - <br> chemical <br> rs. reaction - reaction |  | rs. beam - beam <br> rc. molecule - <br> molecular <br> psm. transformed - <br> changes <br> rc. chemical - <br> chemical <br> rs. reaction reaction | rc. reactant reaction <br> rs. Zewail - <br> Zewail <br> rs. experiments experiment |
| 19 | psm. reactions change | pc. bursts - blasting <br> (blast) <br> rs. light - light <br> rs. lasers - laser <br> rs. molecules - <br> molecules <br> rc. change - change | rs. lasers - laser pc. burst blasting (blast) rs. light - light | tr. flash - blasting <br> rs. laser - laser <br> rs. beam - beam <br> rs. molecule molecules <br> psm. reaction change | pc. reactant change (reaction) |
| 20 | rs. Zewail - Zewail | rs. light - light* rs. lasers - laser rs. he - his | rs. lasers - laser <br> rs. light - light* <br> rs. lasting lasting <br> rs. few - few <br> rs. femtoseconds <br> - femtoseconds | rs. laser - laser rs. transition transition rs. state - state | rs. transition transition <br> rs. state - state <br> rs. lasts lasting* <br> rs. femtoseconds - femtoseconds <br> rs. Zewail Zewail |
| 21 | rs. reactions reaction | rs. reaction - reaction |  | rs. freezes - frozen <br> rs. reaction reaction | rc. reactant reaction |
| 22 | rs. chemist chemists rs. Nobel - Nobel psm. chemistry chem.* | rs. using - using psm. way - methods rc. molecules molecular rc. chemical chemists |  | rc. molecule molecular rs. structure structure rc. chemical chemists |  |
| 23 | rs. Zewail - Zewail <br> rc. chemistry femtochemistry rs. reactions reaction | rc. molecules unimolecular rc. chemical femtochemistry rs. reaction - reaction | rs. femtoseconds <br> - femtoseconds | rc. molecule unimolecular rs. reaction reaction | rc. reactant reaction psm. lasts takes rs. femtoseconds - femtoseconds rs. Zewail Zewail |
| 24 | s. Zewail - his rs. chemistry chemistry* | rs. he - his rs. molecules molecule rc. chemical chemistry* |  | rs. molecule molecule rc. chemical chemistry* rs. transition transition rs. state - state | rs. transition transition rs. state - state s. Zewail - his |

## Anexo

| 1 |  | 2 |  | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 25 | rs. Zewail - Zewail <br> rs. chemical chemical rs. reactions reactions | rs. he - his <br> rc. molecules bimolecular <br> rs. chemical chemical rs. reaction - reactions |  | rc. molecule bimolecular rs. chemical chemical rs. reaction reactions | rc. reactant reactions rs. Zewail - Zewail |
| 26 | rs. reactions - reaction | psp. take 'snapshots' studied rs. reaction - reaction |  | rs. reaction reaction | rc. reactant reaction psm. observing studied |
| 27 | rs. Zewail - Zewail <br> rs. reactions - reaction | psp. bursts - flash <br> rs. molecules molecule <br> rs. reaction - reaction | psp. bursts - <br> flash <br> psm. <br> femtoseconds - fs | rs. flash - flash rs. molecule molecule rs. reaction reaction rc. transition transitionary | rc. transition transitionary rs. exists - exists rc. reactant reaction tr. femtoseconds picosecond rs. Zewail - Zewail |
| 28 | rc. chemist - chemical <br> s. Zewail - his <br> rs. reactions - reaction | rs. he - his <br> psm. take 'snapshots' look at rs. molecules molecule rs. chemical chemical rs. reaction - reaction |  | rs. molecule molecule rs. chemical chemical rs. reaction reaction | rc. reactant reaction psm. observing look at <br> s. Zewail - his |
| 29 |  |  | psm. <br> femtoseconds - fs |  | psm. femtoseconds $-\mathrm{fs}$ |
| 30 | rs. Zewail - Zewail rc. chemistry femtochemistry rs. reactions reactions | rs. using - use <br> psm. take 'snapshots' - <br> look at <br> rc. chemical - <br> femtochemistry <br> rs. reaction - reactions |  | rc. chemical femtochemistry rs. reaction reactions | rc. reactant reaction psm. observing watching <br> rs. Zewail - Zewail psm. experiments studies |



## Anexo

|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 15 | rc. chemists chemical psm. reactions change rs. interactions interaction rs. molecule molecules | psm. reactions change | rs. chemical chemical psm. reactions change rc. second femtosecond | psm. reactions change | rc. chemistry chemical tr. microsecond femtosecond |
| 16 | psp. being able could rc. observe observation rs. chemists - chemists rc. molecules molecular | psm. look at watch | rc. observe observation rc. chemical chemists | psm. see - watch | rc. observed observation rc. chemistry chemists |
| 17 | rs. observe - observe rs. molecule molecules | psm. look at observe | rs. observe observe rc. second femtosecond | psm. see observe rs. flashlamp flashlamp rc. flash flashes | rs. observed - observe <br> tr. microsecond femtosecond |
| 18 | rc. chemists chemical rs. reactions - reaction rc. molecule molecular | rs. reactions reaction | rc. chemical chemical rs. reactions reaction | rs. reactions reaction | rc. chemistry chemical |
| 19 | psm. reactions change rs. molecule molecules | psm. reactions change | psm. reactions change | psm. reactions - <br> change <br> tr. flash - <br> blasting |  |
| 20 | rs. state - state |  | rc. second femtoseconds | rc. trasient transition | tr. microsecond femtoseconds |
| 21 | rs. reactions - reaction | rs. reactions reaction | rs. reactions reaction | rs. freeze frozen rs. reactions reaction |  |
| 22 | rs. chemists - chemists rs. predict - predict rc. molecule molecular |  | rs. used - using rc. chemical chemists | rs. using - using | rc. chemistry chemists |
| 23 | rc. chemists femtochemistry rs. reactions - reaction rc. molecule unimolecular | rs. reactions reaction | rs. reactions reaction rc. second femtoseconds | rs. reactions reaction | tr. microsecond femtoseconds |


| 6 |  | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 24 | rs. state - state rc. chemists chemistry* rs. molecule molecule |  | rc. chemical chemistry* | rc. transient transition | rs. chemistrychemistry* |
| 25 | rc. chemists chemical rs. reactions reactions rc. molecule bimolecular | rs. reactions reactions | rs. chemical chemical rs. reactions reactions | rs. reactions reactions | rc. chemistry chemical |
| 26 | psp. observe - studied rs. reactions - reaction | psp. look at studied rs. reactions reaction | psp. observe studied rs. reactions reaction | rs. reactions reaction psp. see studied | psp. observed studied |
| 27 | rs. reactions - reaction rs. molecule molecule | rs. reactions reaction | rs. reactions reaction rc. second picosecond | rs. reactions reaction rs. flash - flash rc. transient trasitionary | tr. microsecond picosecond |
| 28 | psm. observe - look at rc. chemists chemical <br> rs. reactions - reaction rs. molecule molecule | rs. look at - look at <br> rs. reactions reaction | psm. observe look at tr. chemical chemical rs. reactions reaction | rs. reactions reaction psm. see - look at | psm. observed - look at rc. chemistry chemical |
| 29 |  |  | pc. second -fs (femtosecond) |  | pc. second -fs (femtosecond) |
| 30 | psm. observe watching rc. chemists femtochemistry rs. reactions reactions | rs. look at - look at <br> rs. reactions reactions <br> rs. happen happen | psm. observe look at rs. happen happen rc. chemical femtochemistry rs. reactions reactions | rs. using - use rs. reactions reactions psm. see - look at | psm. observed - look at rc. chemistry femtochemistry |

## Anexo

| 12 | rs. Nobel - Nobel <br> rs. prize - prize <br> rc. chemist chemistry <br> rs. used - using <br> rs. reaction - reactions <br> rs. observe - observe <br> rc. working - work <br> tr. micro-second - <br> picosecond <br> rc. timescale - scale | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | rs. chemist - chemists <br> rs. reaction - reactions <br> pc. instant - fleeting <br> (instantaneous) <br> rs. timescale timescales | rs. chemical chemical rs. reactions reactions rc. scale timescales rc. chemistry chemists | 13 |  |  |
| 14 | rs. chemist - chemists rs. timescale timescale | rc. scale picoscale rc. chemistry chemists | rs. timescales timescale rs. chemists chemists | 14 |  |
| 15 | rc. chemist - chemical psm. reaction - change tr. microsecond femtosecond | rs. chemical chemical psm. reactions change tr. picosecond femtosecond | rs. chemical chemical psm. reactions change | rc. chemists chemical | 15 |
| 16 | rs. chemist - chemists <br> rc. observe - <br> observation | rc. observe observation rc. chemistry chemists | rs. chemists chemists | rs. chemists chemists rs. reach reached | rc. molecules molecular <br> rc. vibrate - vibrations <br> rc. chemical - <br> chemists |
| 17 | rs. observe - observe tr. microsecond femtosecond | rs. observe observe tr. picosecond femtosecond |  |  | rs. femtosecond femtosecond rs. molecules molecules |
| 18 | rc. chemist - chemical <br> rs. reaction - reaction | rc. chemical chemical rs. reactions reaction <br> rs. vacuum vacuum rs. experiments experiment | rc. chemical chemical rs. reactions reaction | rc. chemists chemical | rc. molecules unimolecular rc. chemical chemical psm. change - reaction |
| 19 | rs. trigger - triggered psm. reaction - change | psm. reaction change | psm. reactions change |  | rs. molecules molecules rs. change - change |


| 11 |  | 12 | 13 |
| :--- | :--- | :--- | :--- | :--- | :--- |


| 11 |  | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 28 | rc. chemist - chemical rs. reaction - reaction psm. observe - look at | psm. observe look at rs. reactions reaction rc. chemistry chemical | rc. chemists chemical rs. began began rs. reactions reaction | rc chemists chemical | rs. molecules molecule tr. chemical chemical psm. change - reaction |
| 29 | tr. microsecond - fs (femtosecond) | tr. picosecond fs (femtosecond) |  |  | psm. femtosecond - fs |
| 30 | rc. chemist femtochemistry rs. used - use rs. reaction - reactions psm. observe - look at rs. happening happen | psm. observe look at rs. reactions reactions rs. using - use rc. chemistry femtochemistry | rc. chemists femtochemistry rs. began begun rs. reactions reactions | rc. chemists femtochemistry | rc. chemical femtochemistry psm. change reactions |


| 16 |  | 17 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 17 | rc. molecular molecules <br> d. watch molecular vibrations - this rc. observation observe |  |  |  |  |
|  |  |  | 18 |  |  |
| 18 | rc. chemists chemical rc. molecular unimolecular | rs. Zewail - Zewail rc. molecules unimolecular |  |  |  |
| 19 | rc. molecular molecules | rs. molecules molecules <br> rc. pulsing - pulse <br> rs. laser - laser <br> tr. flashes - blasting | s. Zewail and his colleages - they rc. changes change rc. molecular molecules rs. beam - beam | 19 |  |
| 20 |  | rs. Zewail - Zewail <br> rc. pulsing - pulse <br> rs. laser - laser <br> rs. femtosecond femtoseconds | rs. Zewail - Zewail rs. his - his psm. colleagues team rs. substance substance | rs. pulse - pulse rs. laser - laser rs. light - light* | 20 |
| 21 |  |  | s. Zewail and his colleagues - they rs. reaction reaction | rs. they - they psm. change reaction |  |
| 22 | rs. chemists chemists rs. molecular molecular | rc. molecules molecular | co-ref. Zewail and his colleagues chemists rc. chemical chemists rs. molecular molecular | rs. they - them rc. molecules molecular | co-ref. Zewail and his colleagues - chemists rs. obtained - obtained rs. characteristic characteristic rs. spectrum spectrum |
| 23 | rc. chemists femtochemistry rc. molecular unimolecular | rs. Zewail - Zewail <br> rc. molecules unimolecular <br> rs. femtosecond femtoseconds | rs. Zewail - Zewail rs. unimolecular unimolecular rs. reaction reaction rc. chemical femtochemistry | rc. molecules unimolecular psm. change reaction | psm. lasting - takes* rs. femtoseconds femtoseconds rs. Zewail - Zewail |
| 24 | rc. chemists chemistry* rc. molecular molecule | s. Zewail - his rs. molecules molecule | s. Zewail - his rc. chemical chemistry* rc. molecular molecule | rs. molecules molecule | s. Zewaqil - his rs. transition transition rs. state - state |


|  | 16 | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 25 | rc. chemists chemical rc. molecular bimolecular | rs. Zewail Zewail rc. molecules bimolecular | rs. Zewail - Zewail <br> rs. his - his <br> rs. colleagues colleagues <br> rc. unimolecular bimolecular <br> rs. reaction - reactions <br> rc. involvement - involve <br> rc. chemical - chemical | rc. molecules bimolecular psm. change reactions | rs. Zewail - Zewail rs. his - his psm. team colleagues |
| 26 | psp. watch studied | psp. observe studied | s. Zewail and his colleagues - they rs. reaction - reaction | rs. they - they psm. change reaction | s. Zewail and his team - they |
| 27 | rc. molecular molecule | rs. Zewail Zewail rs. molecules molecule rc. flashes flash | rs. Zewail - Zewail rs. reaction - reaction rc. unimolecular molecule | tr. blasting flash rs. molecules molecule psm. change reaction | psm. lasting - <br> exits* <br> psm. femtoseconds - fs <br> rs. Zewail - Zewail <br> rc. transition - <br> transitionary |
| 28 | rc. chemists chemical psm. watch look at rc. molecular molecule | s. Zewail - his psm. observe look at rs. molecules molecule | s. Zewail - his psm. colleagues - team rc. unimolecular molecule <br> rs. reaction - reaction <br> tr. chemical - chemical | rs. molecules molecule psm. change reaction | tr. substance molecule rs. his - his rs. team - team |
| 29 |  | psm. <br> femtosecond - fs |  |  | psm. femtoseconds $-\mathrm{fs}$ |
| 30 | rc. chemists femtochemistry rs. watch watching pc. observation look at (observe) | rs. Zewail - <br> Zewail <br> psm. observe - <br> look at | rs. Zewail - Zewail psm. colleagues - teams rs. reaction - reactions rc. chemical femtochemistry | psm. change reactions | rs. Zewail - Zewail <br> rs. team - teams |



## Anexo

| 27 | rs. reaction - reaction |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  | 27 | 28 |  |
| 28 | psp. studied - look at <br> rs. reaction - reaction | s. Zewail's - his <br> rs. reaction - reaction <br> rs. molecule - molecule |  |  |
| 29 |  | rs. fs - fs | rs. bonds - bonds <br> rs. break - break <br> psm. simultaneously - at the same time rs. dissociation - dissociation | 29 |
| 30 | rc. studied - studies <br> rs. reaction reactions | rs. Zewial - Zewail <br> rs. reaction - reactions | rs. team - teams <br> rs. began - begun <br> rs. look at - look at <br> rc. chemical - femtochemistry <br> rs. reaction - reactions |  |

2. 6. 2. Matriz con el número de unidades léxicas.

|  | 1 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | 3 | 2 |  |  |  |  |  |  |  |  |  |  |  |  |
| 3 | 0 | 4 | 3 |  |  |  |  |  |  |  |  |  |  |  |
| 4 | 2 | 6 | 2 | 4 |  |  |  |  |  |  |  |  |  |  |
| 5 | 2 | 2 | 1(2) | 3 | 5 |  |  |  |  |  |  |  |  |  |
| 6 | 2 | 4 | 0 | 4 | 3 | 6 |  |  |  |  |  |  |  |  |
| 7 | 1 | 2 | 0 | 1 | 2 | 2 | 7 |  |  |  |  |  |  |  |
| 8 | 2 | 4 | 0 (1) | 2 | 3 | 3 | 4 | 8 |  |  |  |  |  |  |
| 9 | 2 | 5 | 2 | 4 | 3 | 3 | 2 | 3 |  |  |  |  |  |  |
| 10 | 1 | 2 | 0 (1) | 1 | 2 | 2 | 2 | 3 | 2 | 10 |  |  |  |  |
| 11 | 4 | 5 | $0(1)$ | 2 | 3 | 3 | 3 | 6 | 5 | 5 | 11 |  |  |  |
| 12 | 6 | 4 | 0 (1) | 2 | 4 | 4 | 2 | 5 | 3 | 4 | 9 | 12 |  |  |
| 13 | 3 | 2 | 0 (1) | 5 | 4 | 3 | 1 | 2 | 2 | 3 | 4 | 4 | 13 |  |
| 14 | 1 | 1 | 0 | 1 | 0 | 1 | 0 | 1 | 0 | 2 | 2 | 2 | 2 | 14 |
| 15 | 2 | 3 | 1 | 3 | 2 | 4 | 1 | 3 | 1 | 2 | 3 | 3 | 2 | 1 |
| 16 | 1 | 3 | 0 | 2 | 1 | 4 | 1 | 2 | 1 | 2 | 2 | 2 | 1 | 2 |
|  | 1 | 5 | 3 | 4 | 3 | 2 | 1 | 2 | 3 | 2 | 2 | 2 | 0 | 0 |
| 18 | 3 | 5 | 0 | 5 | 3 | 3 | 1 | 2 | 1 | 1 | 2 | 4 | 2 | 1 |
| 19 | 1 | 5 | 3 | 5 | 1 | 2 | 1 | 1 | 2 | 0 | 2 | 1 | 1 | 0 |
| 20 | 1 | $2(3)$ | 4(5) | 3 | 4(5) | 1 | 0 | 1 | 1 | 1 | 1 | 1 | 2 | 0 |
| 21 | 1 | 1 | 0 | 2 | 1 | 1 | 1 | 1 | 2 | 0 | 1 | 1 | 1 | 0 |
| 22 | $2(3)$ | 4 | 0 | 3 | 0 | 3 | 0 | 2 | 1 | 1 | 4 | 3 | 1 | 1 |
|  | 3 | 3 | 1 | 2 | 4 | 3 | 1 | 2 | 1 | 1 | 3 | 3 | 3 | 1 |
| 24 | 1(2) | $2(3)$ | 0 | 3(4) | 3 | $2(3)$ | 0 | 0 (1) | 1 | 0 (1) | 0 (1) | 0 (1) | 3(4) | $0(1)$ |
| 25 | 3 | 4 | 0 | 3 | 2 | 3 | 1 | 2 | 1 | 1 | 2 | 2 | 3 | 1 |
| 26 | 1 | 2 | 0 | 1 | 2 | 2 | 2 | 2 | 2 | 1 | 2 | 2 | 1 | 0 |
| 27 | 2 | 3 | 2 | 4 | 5 | 2 | 1 | 2 | 3 | 1 | 3 | 2 | 5 | 0 |
| 28 | 3 | 5 | 0 | 3 | 3 | 4 | 2 | 3 | 2 | 2 | 3 | 3 | 3 | 1 |
| 29 | 0 | 0 | 1 | 0 | 1 | 0 | 0 | 1 | 0 | 1 | 1 | 1 | 0 | 0 |


$c$
30
2. 6. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 9)(-, 10)$ | $[9][10]$ | 2. $(1,18)(1,20)[19][21]$ |
| :--- | :--- | :--- |
| 4. $(1,14)[15]$ | S. $(1,14)[15]$ | 3. $(3,3)[4]$ |
| 7. $(0,3)[3]$ | 8. $(4,7)[11]$ | 9. $(5,5)[10]$ |
| 10. $(1,3)[4]$ | 11. $(8,8)[16]$ | 12. $(8,7)[15]$ |
| 13. $(7,6)[13]$ | 14. $(0,0)[0]$ | 15. $(6,6)[12]$ |
| 16. $(3,3)[6]$ | 17. $(6,5)[11]$ | 18. $(7,8)(7,9)[15][16]$ |
| 19. $(5,1)(5,2)[6][7]$ | 20. $(5,5)(7,6)[10][13]$ | 21. $(0,0)[0]$ |
| 22. $(7,0)(8,0)[7][8]$ | 23. $(10,6)(11,6)[16][17]$ | 24. $(5,2)(8,2)[7][10]$ |
| 25. $(9,3)[12]$ | 26. $(0,0)[0]$ | 27. $(14,1)[15]$ |
| 28. $(18,2)[20]$ | 29. $(2,0)[2]$ | 30. $(15,-)[15]$ |

## 2. 6. 4. Texto resultante tras eliminar las oraciones marginales.

1. Chemist Ahmed Zewail of the California Institute of Technology (Caltech) (pictured top right) was the recipient of the 1999 Nobel prize for chemistry for a flash of inspiration that is revolutionising our understanding of chemical reactions. 2. Using brief bursts of light from lasers, he developed a way to take 'snapshots of individual molecules as they change during a chemical reaction.
2. Modern lasers can produce a very short burst of light, lasting a few femtoseconds ie a million-billionth of a second. 4. Like a fast camera that freezes a dancer in motion, the laser beam can illuminate a molecule as it is transformed from one shape and structure to another during a chemical reaction - its transition state. 5. This transition state, which exists between the reactant and the product, lasts for only femtoseconds, so observing it before it disappears was, until Zewail's experiments, almost impossible. 6. Being able to observe this state is helping chemists to find out exactly how particular reactions work and allowing them to predict the outcome of other related reactions as well as the complex interactions of, for example, a drug molecule with a biological receptor.
3. The earliest attempt to look at reactions as they happen was by H. Hartridge and EJ. Roughton in the 1920s. 8. They used a spectrophotometer to observe what happens when two compounds are mixed and saw chemical reactions taking place in a thousandth of a second. 9. In the 1960s, Ronald Norrish and George Porter came up with the idea of using a flash-lamp to freeze the reactions - the shorter the flash, the more transient the reactions they could see. 10. They observed chemistry on the millisecond and microsecond timescales - a thousand times shorter than that possible in the 1920s. 11. Poter and Norrish shared the 1967 Nobel prize with the German chemist Manfred Eigen, who used heat and pressure shock methods to trigger a reaction and observe 'almost' the instant at which it was happening (Eigen was also working at the milli-to micro-second timescale).
4. During the early 1980s, Dudley Herschbach, Yuan Lee and John Polanyi had improved the ability to observe chemical reactions down to the picosecond scale using a vacuum collision experiments - for this work they received the 1986 Nobel prize for chemistry. 13. With shorter and shorter timescales, chemists began to reveal the intermediate chemical species in reactions - not, the transition states, they were still too fleeting, but the structures either side that lasted just long enough for them to record. 15. The femtosecond $-10^{-15} \mathrm{~s}$ - represents the frequency at which molecules vibrate, without which there would be no interaction and no chemical change. 16. If
chemists could watch molecular vibrations they would have reached the limit of observation.
5. Zewail realised that to observe molecules at this level his flashlamp would have to be very fast, a pulsing laser that flashes once every femtosecond, he reasoned, would do the job. 18. For their simplest experiment, Zewail and his colleagues chose a unimolecular reaction, ie where a single substance changes into another without the involvement of a second chemical, and formed a molecular beam in a vacuum chamber. 19. By blasting this beam with a 'pump pulse' of laser light they excited the molecules and triggered a change. 20. Then, by applying a weaker, 'probe pulse' from a laser lasting a few femtoseconds - at a frequency to coincide with the absorption frequency of the suspected transition state of the substance - Zewail and his team obtained a characteristic spectrum from the light emitted by the transition state.
6. The chemists compared the characteristic spectrum with the theoretical pattern obtained by using the methods of last year's Nobel chemists John Pople and Walter Kohn (Educ. Chem., 1999, 36(1), 7) who provided them with the means to predict molecular structure and so their characteristic spectra. 23. Zewail's first unimolecular reaction - the one that started the whole femtochemistry field - was the dissociation of iodine cyanide (ICN), which takes just 200 femtoseconds. 24. His results were published in 1987 in the journal of physical chemistry and showed the transition state just as the carbon-iodine bond in the molecule is about to break to form the cyano radical and an iodine atom.
7. Zewail and his colleagues then moved on to bimolecular reactions, which involve two interacting chemical species. 27. Zewail's flash revealed that the reaction [of hydrogen with carbon dioxide, which produces carbon monoxide and hydroxy radicals] passes through a transitionary HOCO molecule, which exists fleetingly for a mere picosecond (1000fs). 28. His team also began to look at a puzzle that had occupied chemical minds for some time - ie would two seemingly identical bonds in a molecule break simultaneously in, for instance, a dissociation reaction. 29. For the dissociation of tetrafluorodiiodoethane it turns out that the 'equivalent' $\mathrm{C}-1$ bonds do not break at the same time - there is a delay of 200 fs following the splitting of the first.
8. Since Zewail's pioneering studies in the 1980s and 1990s, many other research teams have begun to use femtochemistry to look at diverse reactions watching them happen in real-time.

## 2. 7. Texto 7: Pressure to change solvents.

1. Decent decaffeinated coffee has been around since 1960s, when chemist Kurt Zosel found an alternative to using the toxic and unpleasant tasting benzene to extract the caffeine. 2. He discovered that a 19th century chemical curiosity, known as a supercritical fluid (SCF), could dissolve out the caffeine but leave no solvent residue. $\mathbf{3}$. Supercritical fluids while still curious are now being used to destroy toxic waste, make industrial chemicals without toxic and highly flammable volatile organic compounds (VOCs) and are even making it easier to take your medicine. 4. So what are these strange materials and why are they so supercritical?
2. If you apply enough pressure to some gases while heating them they liquefy but keep their gaseous energy. 6. Conversely, heating some liquids while you apply pressure gives them gaseous energy but without losing their density. 7. These fluids are caught between the liquid and gas phase above a certain critical temperature and pressure - they are supercritical fluids, see Fig 1. 8. Many common chemicals can become supercritical, from carbon dioxide and water to the noble gas xenon.
3. Water, for instance, becomes a supercritical fluid when it is heated above $374^{\circ} \mathrm{C}$ and put under a pressure of 218 atmos. 10. The fluid looks like a liquid but strangely, on the one hand can be mixed with oil but on the other will no longer dissolve ordinary table salt. 11. These effects can be explained by the changes in the bonds between water molecules which, in the supercritical state, become weaker than normal. 12. So, oily molecules can squeeze in between them but they are too weak to hold the sodium and chloride ions from salt. 13. Amazingly, oxygen dissolved in supercritical water supports 'flameless' combustion. 14. Scientists at Sandia National Laboratories in New Mexico are using this property to destroy industrial and domestic waste without the need for conventional incineration. 15. Dissolved salts and metals come out of the solution and can be recycled or disposed of safely, while the organic content is broken down into carbon dioxide and water by the oxidation process. 16. The process works at lower temperatures than incineration, so there are no nitrogen oxide pollutants produced.
4. Organic chemists from the University of Leeds have also been quick to latch on to Zosel's early discovery and have been using SCFs to extract natural products from plants and other organic materials for years. 18. Natural flavour molecules, such as vanilla, for instance, can be cleanly extracted from the pod using an SCF. 19. More recently, though, chemists have turned to SCFs to dissolve reactants that usually need a toxic and flammable VOC or do not dissolve at all.
5. Synthetic chemists are using SCFs in the manufacture of new types of polymer and other molecules that could function as industrial catalysts, thus avoiding the use of harmful solvents. 21. Joseph DeSimone's group at the University of North Carolina in Chapel Hill, for example, is using supercritical carbon dioxide to make new types of fluorine-containing polymer. 22. Adding fluorine atoms to a polymer chain is used to make some tough, smooth and chemically inert materials. 23.

Polytetrafluoroethene (PTFE or Teflon) was one of the earliest fluoropolymers, and is still used to coat non-stick frying pans! 24. Modern fluoropolymers have more high-tech applications, such as acting as 'dry' lubricating layers for the moving parts in computers, eg hard drives, where a drop of oil would wreck the electronics. 25 . The problem with making these new fluoropolymers, however, is that fluorine atoms have a residual negative charge, which makes them polar so they dissolve best in water. 26. This makes it difficult to process them further because any other chemicals added will usually be soluble only in organic solvents.
27. DeSimone's team has got around this problem by using supercritical carbon dioxide instead. 28. The chemists can now control the length of the polymer chains and their precise chemical structure. 29. This leads to consistent materials for high-tech. aerospace and electronic applications.
30. Martyn Poliakoff and his team at the University of Nottingham, meanwhile, are exploring how SCFs can help them make new industrial catalysts. 31. They have discovered that they can make organometallic compounds such as metal carbonyls, many of which are too unstable to prepare by conventional methods. 32. Metal carbonyls are used in various industrial reactions as catalysts for speeding up the production of simple materials such as formic acid and formaldehyde and more complex compounds, like pharmaceuticals and polymers. 33. Carbonyl compounds in which nitrogen or hydrogen molecules have been substituted for a carbonyl group can catalyse more complex reactions still. 34. For example, novel piano-stool shaped manganese carbonyls with an attached dihydrogen might be a useful polymerisation catalyst. 35. The problem in making them is that hydrogen and nitrogen gases do not dissolve well in conventional organic solvents at room temperature so it is hard to add the atoms to the starting molecule. 36. The Nottingham team, however, has found that hydrogen mixes very well with supercritical carbon dioxide at $80-100$ atmos, allowing the reaction to add hydrogen or nitrogen atoms as needed to the carbonyl compound.
37. Once the reaction is over, the SCF can be quickly recycled by releasing the pressure and trapping the carbon dioxide gas that escapes. 38. This is one of the major advantages of SCFs over other solvents. 39. VOCs, for instance, become contaminated during a reaction and it is expensive and wasteful to purify them. 40. SCFs avoid this problem because once they become a gas again they leave behind any impurities,
41. SCFs are also much less viscous than liquid solvents, so they flow more easily through a reaction system. 42. They can also get into the smallest of crevices and pits inside the reactor system. 43. By flushing the system with an SCF once a reaction is complete any impurities can be washed out, leaving the system pristine and ready to be used again.
44. But, what about SCFs making it easier to take medicines? 45. Scientists are now using SCFs to help them make drugs that normally have to be injected work when taken by mouth instead. 46. A collaborative team from the US, Canada and Norway has found they can make sub-microscopic particles of the immunosuppressant drug cyclosporin, which is used to prevent transplanted organ rejection, by preparing it in supercritical carbon dioxide and then blasting it into normal water by releasing the pressure. 47. The blast makes billions upon billions of tiny drug particles just fractions of a micrometre in size. 48. These particles are so small that the researchers hope they will be absorbable by the gut so that patients avoid getting the needle.
49. Amazing what a little warmth and a squeeze will do.

## 2. 7. 1. Matriz de repetición de unidades léxicas.

| 2 | rc. chemist chemical <br> s. Zosel - he hip. benzene solvent rs. caffeine caffeine | 2 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | rc. chemist chemicals rs. using used rs. toxic toxic hip. benzene VOCs | rc. chemical chemicals rc. curiosity - curious rs- supercritical supercritical rs. fluid - fluids tr. solvent - VOCs | 3 |  |  |
| 4 |  | pc. curiosity - strange (curious) <br> rs- supercritical supercritical hip. fluid - materials | rs- supercritical - supercritical hip. fluid materials psm. curious strange | 4 |  |
| 5 |  |  |  | 5 |  |
| 6 |  |  |  |  | rs. you - you + <br> rs. apply - apply <br> rs. pressure - pressure <br> rs. heating - heating <br> rc. liquefy - liquids <br> pc. keep - losing <br> rs. gaseous - gaseous <br> rs. energy - energy |
| 7 |  | rs- supercritical supercritical rs. fluid - fluids | rs- supercritical - supercritical rs. fluids - fluids | tr. materials - fluids rs- supercritical supercritical | rs. pressure - pressure <br> rs. gases - gas <br> rc. liquefy - liquid |
| 8 | rc. chemist chemicals | rc. chemical chemicals rs- supercritical supercritical | rs- supercritical <br> - supercritical rs. chemicals chemicals | rs- supercritical supercritical | rs. gases - gas |
| 9 |  | rs- supercritical supercritical rs. fluid - fluid | rs- supercritical - supercritical rs. fluids - fluid | tr. materials - fluid rs- supercritical supercritical | psm. apply - put under rs. pressure - pressure rs. heating - heated |
| 10 |  | pc. curiosity strangely (curiously) <br> rs. fluid - fluid <br> rs. dissolve dissolve* | rs. fluids - fluid pc. curious strangely (strange) | rc. strange strangely tr. materials - fluid | rc. liquefy - liquid |
| 11 |  | rs- supercritical supercritical | rs- supercritical <br> - supercritical | rs- supercritical supercritical |  |


|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 12 |  |  |  |  |  |
| 13 |  | rs. supercritical supercritical rs. dissolve dissolved | rs. supercritical supercritical | rs. supercritical supercritical |  |
| 14 | hip. chemist scientists rs. using - using* | tr. chemicalscientists | rs. used - using* <br> rs. destroy -destroy <br> rs. waste - waste <br> tr. chemicals - scientists |  |  |
| 15 |  | rs. dissolve dissolved* | psm. destroy - disposed of |  |  |
| 16 |  |  |  |  |  |
| 17 | rs. chemist chemists <br> rs. Zosel - Zosel <br> pc. found discovery <br> (discover) <br> rs. using - using <br> rs. extract - extract <br> hip. caffeine - <br> products | rc. discovered discovery rc. chemical chemists <br> rs. SCF - SCFs hip. caffeine products | psm. supercritical fluids - <br> SCFs <br> rs. used - using <br> rc. chemicals - chemists |  |  |
| 18 | rs. using - using <br> rs. extract extracted hip. caffeine molecules | rs. SCF - SCF hip. caffeine molecules | psm. supercritical fluids SCF <br> rs. used - using |  |  |
| 19 | rs. chemist chemists rs. toxic - toxic hip. benzene VOC | rc. chemical chemists <br> rs. SCF - SCFs <br> rs. dissolve dissolve <br> tr. solvent VOC | psm. supercritical fluids SCFs <br> rc. chemicals - chemists <br> rs. toxic - toxic <br> rs. flammable - flammable <br> rs. VOCs - VOC |  |  |
| 20 | rs. chemist chemists rs. using - using hip. benzene solvents | rc. chemical chemists rs. SCF - SCFs rs. solvent solvents | psm. supercritical fluids SCFs <br> rs. used - using <br> pc. make - manufacture (making) <br> rs. industrial - industrial <br> rc. chemicals - chemists <br> hip. VOC - solvents |  |  |

## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 21 | rs. using - using | rs. supercritical supercritical | rs. supercritical supercritical rs. used - using rs. make - make | rs. supercritical supercritical |  |
| 22 | rc. chemist chemically rs. using - used* | rc. chemical - chemically | rs. used - used* <br> rs. make - make <br> rc. chemicals chemically |  |  |
| 23 | rs. using - used* |  | rs. used - used* |  |  |
| 24 | pc. using applications* (use) |  | $\begin{aligned} & \text { pc. used - applications* } \\ & \text { (use) } \end{aligned}$ |  |  |
| 25 |  | rs. dissolve - dissolve <br> rc. residue - residual | rs. make - making |  |  |
| 26 | rc. chemist chemicals hip. benzene solvents | rc. chemical - chemicals <br> rs. solvent - solvents | rs. chemicals - chemicals <br> rs. organic - organic <br> hip. VOCs - solvents |  |  |
| 27 | rs. using - using | rs. supercritical supercritical | rs. supercritical supercritical rs. used - using | rs. supercritical supercritical |  |
| 28 | rs. chemist chemists | rc. chemical - chemists | rc. chemicals - chemists |  |  |
| 29 | pc. using applications* (use) |  | $\begin{aligned} & \text { pc. used - applications* } \\ & \text { (use) } \end{aligned}$ |  |  |
| 30 |  | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | psm. supercritical fluids <br> - SCFs <br> rs. make - make <br> rs. industrial - industrial |  |  |
| 31 | psm. found discovered | rs. discovered discovered | rs. make - make |  |  |
| 32 | rs. using - used* |  | rs. used - used* <br> pc. make - production* <br> (making) <br> rs. industrial - industrial |  |  |


|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 33 |  |  |  |  |  |
| 34 | rc. using - useful* |  | rc. used - useful* |  |  |
| 35 | hip. benzene solvents | rs. dissolve - dissolve <br> rs. solvent - solvents | rs. make - making rs. organic - organic hip. VOCs - solvents |  |  |
| 36 | rs. found - found | psm. discovered - found rs. supercritical supercritical | rs. supercritical supercritical | rs. supercritical supercritical |  |
| 37 |  | rs. $\mathrm{SCF}-\mathrm{SCF}$ | psm. supercritical - <br> fluids - SCF |  | rs. pressure <br> - pressure <br> rs. gases - <br> gas |
| 38 | hip. benzene solvents | rs. SCF - SCFs <br> rs. solvent - solvents | psm. supercritical fluids - SCFs hip. VOCs - solvents |  |  |
| 39 | hip. benzene VOCs | tr. solvent - VOCs | rs. VOCs - VOCs |  |  |
| 40 |  | rs. $\mathrm{SCF}-\mathrm{SCFs}$ <br> rs. leave - leave | psm. supercritical fluids - SCFs |  | rs. gases gas |
| 41 | hip. benzene solvents | rs. SCF - SCFs <br> rs. solvent - solvents | psm. supercritical fluids - SCFs hip. VOCs - solvents |  |  |
| 42 |  | s. SCF - they | s. supercritical fluids they |  |  |
| 43 |  | rs. SCF - SCF <br> rs. leave - leaving* | psm. supercritical - <br> fluids - SCF |  |  |
| 44 |  | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | psm. supercritical - <br> fluids - SCFs <br> rs. making - making <br> rs. easier - easier <br> rs. take - take <br> rs. medicine - medicines |  |  |

## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 45 | hip. chemist scientists rs. using - using | tr. chemical - scientists <br> rs. SCF - SCFs | psm. supercritical - <br> fluids - SCFs <br> rs. used - using <br> rs. make - make <br> tr. chemicals - scientists <br> psm. medicine - drugs |  |  |
| 46 | rs. found - found <br> rs. using - used* | psm. discovered - found rs. supercritical supercritical | rs. supercritical supercritical rs. used - used* rs. make - make psm. medicine - drug | rs. supercritical supercritical | rs. pressure <br> - pressure |
| 47 |  |  | psm. medicine - drug |  |  |
| 48 | hip. chemist researchers | tr. chemical - researchers | tr. chemicals researchers |  |  |
| 49 |  |  |  |  |  |


| 7 | rs. liquids - liquid <br> rs. pressure pressure <br> rc gaseous - gas |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 7 | 8 |  |  |
| 8 | rc gaseous - gas | rs. supercritical supercritical |  |  |  |
| 9 | rs. heating - heated psm. apply - put under <br> rs. pressure pressure | rs. pressure pressure rs. supercritical supercritical rs. fluids - fluid | rs. become becomes rs. supercritical supercritical rs. water - water | 9 |  |
| 10 | rs. liquids - liquid | rs. liquid - liquid <br> rs. fluids - fluid |  | rs. fluid - fluid | 10 |
|  |  |  |  |  |  |
| 11 |  | rs. supercritical supercritical | rs. become become rs. supercritical supercritical rs. water - water | rs. water - water <br> rs. becomes become* <br> rs. supercritical supercritical | d. oración 10 these |
| 12 |  |  |  |  | rc. oil - oily rs. salt - salt |
| 13 |  | rs. supercritical supercritical | rs. supercritical supercritical rs. water - water | rs. water - water rs. supercritical supercritical |  |
| 14 |  |  | tr. chemicals scientists |  |  |
| 15 |  |  |  |  |  |
| 16 |  |  |  |  |  |
| 17 |  | psm. <br> supercritical <br> fluids - SCFs | rc. chemicals chemists hip. carbon dioxide .... gas xenon SCFs | psm. supercritical fluids - SCFs |  |
| 18 |  | psm. <br> supercritical <br> fluids - SCF | hip. carbon dioxide .... gas xenon SCF | psm. supercritical fluids - SCF |  |

## Anexo



|  | 6 | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 34 |  |  |  |  |  |
| 35 | rc. gaseous gases | rs. temperature temperature |  |  |  |
| 36 |  | rs. supercritical supercritical | rs. supercritical supercritical rs. carbon - carbon rs. dioxide - dioxide | rs. supercritical supercritical |  |
| 37 | rs. pressure pressure rc. gaseous gas | rs. pressure - pressure psm. supercritical fluids - SCF | hip. carbon dioxide .... gas xenon - SCF | psm. supercritical <br> fluid - SCF <br> rs. pressure pressure |  |
| 38 |  | psm. supercritical fluids - SCFs | hip. carbon dioxide .... gas xenon-SCFs | psm. supercritical fluid - SCFs |  |
| 39 |  |  |  |  |  |
| 40 | rc. gaseous gas | rs. gas - gas psm. supercritical fluids - SCFs | hip. carbon dioxide .... gas xenon-SCFs | psm. supercritical fluid - SCFs |  |
| 41 |  | psm. supercritical fluids - SCFs | hip. carbon dioxide .... gas xenon - SCFs | psm. supercritical fluid - SCFs |  |
| 42 |  |  |  |  |  |
| 43 |  | psm. supercritical fluids - SCF | hip. carbon dioxide .... gas xenon - SCF | psm. supercritical fluid - SCF |  |
| 44 |  | psm. supercritical fluids - SCFs | hip. carbon dioxide .... gas xenon-SCFs | psm. supercritical fluid - SCFs |  |
| 45 |  | psm. supercritical fluids - SCFs | tr. chemicals - scientists hip. carbon dioxide .... gas xenon-SCFs | psm. supercritical fluid - SCFs |  |
| 46 | rs. pressure pressure | rs. pressure - pressure <br> rs. supercritical supercritical | rs. supercritical supercritical <br> rs. carbon - carbon <br> rs. dioxide - dioxide | rs. supercritical supercritical rs. pressure pressure |  |


| 6 |  | 7 |  | 8 | 9 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 47 |  |  |  |  |  |
| 48 |  |  |  |  |  |
| 49 |  |  | tr. chemicals - researchers |  |  |


| 12 | s. bonds - them rs. molecules molecules rc. weaker - weak |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 12 | 13 | 14 |  |
| 13 | rs. water - water rs. supercritical supercritical |  |  |  |  |
| 14 |  |  | d. oración 13 - this |  |  |
|  |  |  |  |  | 15 |
| 15 |  |  | rs. dissolved dissolved* | $\begin{aligned} & \text { psm. destroy - disposed } \\ & \text { of } \end{aligned}$ |  |
| 16 |  |  |  | rs. incineration incineration | rc. dioxide - oxide <br> rs. process - process |
| 17 |  |  | hip. supercritical water - SCFs | tr. scientists - chemists rs. using - using* |  |
| 18 |  |  | hip. supercritical water - SCF | rs. using - using* |  |
| 19 |  |  | rs. dissolved dissolve hip. supercritical water - SCFs rc. flameless flammable | tr. scientists - chemists | rs. dissolved dissolve* |
| 20 |  |  | hip. supercritical water - SCFs | tr. scientists - chemists rs. using - using* |  |
| 21 | rs. supercritical supercritical |  | tr. supercritical water - supercritical carbon dioxide | rs. using - using* |  |
| 22 |  |  |  |  |  |
| 23 |  |  |  |  |  |
| 24 |  |  |  |  |  |
| 25 | rs. water - water |  | rs. dissolved dissolve <br> rs. water - water |  | rs. dissolved dissolve* |

## Anexo

|  | 11 | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 26 |  |  |  |  |  |
| 27 | rs. supercritical supercritical |  | tr. supercritical water supercritical carbon dioxide | rs. using - using* |  |
| 28 |  |  |  | tr. scientists - chemists |  |
| 29 |  |  |  |  |  |
| 30 |  |  | hip. supercritical water SCFs |  |  |
| 31 |  |  |  |  |  |
| 32 |  |  |  |  |  |
| 33 |  |  |  |  |  |
| 34 |  |  |  |  |  |
| 35 |  |  | rs. dissolved - dissolve |  |  |
| 36 | rs. supercritical supercritical |  | tr. supercritical water supercritical carbon dioxide |  |  |
| 37 |  |  | hip. supercritical water SCF |  |  |
| 38 |  |  | hip. supercritical water SCFs |  |  |
| 39 |  |  |  |  |  |
| 40 |  |  | hip. supercritical water SCFs |  |  |
| 41 |  |  | hip. supercritical water SCFs |  |  |


| 11 |  |  |  |  | 13 |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :---: | :---: | :---: |
| 43 |  |  | hip. supercritical water - <br> SCF |  |  |  |  |  |
| 44 |  |  | hip. supercritical water - <br> SCFs |  |  |  |  |  |
| 45 |  |  | hip. supercritical water - <br> SCFs | rs. scientists - scientists <br> rs. using - using* |  |  |  |  |
| 46 |  |  |  |  |  |  |  |  |
| 47 |  |  |  | tr. supercritical water - <br> supercritical carbon dioxide |  |  |  |  |

## Anexo



|  | 16 | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 27 |  | rs. using - using <br> tr. SCFs - <br> supercritical carbon dioxide | rs. using - using <br> tr. SCF - supercritical carbon dioxide | tr. SCFs supercritical carbon dioxide | rs. using - using <br> tr. SCFs - supercritical carbon dioxide |
| 28 |  | rs. chemists - chemists |  | rs. chemists chemists | rs. chemists - chemists <br> rs. polymer - polymer |
| 29 |  | pc. using applications* (use) | pc. using applications* (use) |  | $\begin{aligned} & \text { pc. using - applications* } \\ & \text { (use) } \end{aligned}$ |
| 30 |  | rs. SCFs - SCFs | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | rs. SCFs - SCFs | rs. SCFs - SCFs <br> pc. manufacture - make (making) <br> rs. new - new <br> rs. industrial - industrial <br> rs. catalysts - catalysts |
| 31 |  | rc. discovery discovered |  |  | pc. manufacture - make (making) |
| 32 |  | rs. using - used* | rs. using - used* | rc. reactants reactions | rs. using - used* <br> psm. manufacture production <br> rs. polymer - polymers <br> rs. industrial - industrial <br> rs. catalysts - catalysts |
| 33 |  |  |  | rc. reactants reactions | rc. catalysts - catalyse |
| 34 |  | rc. using - useful* | rc. using - useful* |  | rc. using - useful* <br> rc. polymer polymerisation <br> rs. catalysts - catalyst |
| 35 |  |  |  | rs. dissolve dissolve hip. VOC solvents | psm. manufacture making <br> rs. solvents - solvents |
| 36 |  | pc. discovery - found (discover) <br> tr. SCFs - <br> supercritical carbon dioxide | tr. SCF - supercritical carbon dioxide | tr. SCF supercritical carbon dioxide rc. reactants reactions | tr. SCFs - supercritical carbon dioxide |
| 37 |  | rs. $\mathrm{SCFs}-\mathrm{SCF}$ | rs. SCF - SCF | rs. SCFs - SCF rc. reactants reaction | rs. $\mathrm{SCFs}-\mathrm{SCF}$ |

## Anexo

|  | 16 | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 38 |  | rs. SCFs - SCFs | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | rs. SCFs - SCFs hip. VOC solvents | rs. SCFs - SCFs <br> rs. solvents - solvents |
| 39 |  |  |  | rc. reactants reaction <br> rs. VOC - VOCs | tr. solvents - VOCs |
| 40 |  | rs. SCFs - SCFs | rs. SCF - SCFs | rs. SCFs - SCFs | rs. SCFs - SCFs |
| 41 |  | rs. SCFs - SCFs | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | rs. SCFs - SCFs <br> rc. reactants reaction <br> hip. VOC - <br> solvents | rs. SCFs - SCFs <br> rs. solvents - solvents |
| 42 |  | s. SCFs - they |  | rc. reactants reactor | s. SCFs - they |
| 43 |  | rs. $\mathrm{SCFs}-\mathrm{SCF}$ | rs. SCF - SCF | rs. SCFs - SCF rc. reactants reaction | rs. $\mathrm{SCFs}-\mathrm{SCF}$ |
| 44 |  | rs. SCFs - SCFs | rs. $\mathrm{SCF}-\mathrm{SCFs}$ | rs. SCFs - SCFs | rs. SCFs - SCFs |
| 45 |  | hip. chemists scientists <br> rs. using - using <br> rs. SCFs - SCFs | rs. using - using <br> rs. $\mathrm{SCF}-\mathrm{SCFs}$ | hip. chemists scientists <br> rs. SCFs - SCFs | hip. chemists - scientists rs. using - using <br> rs. SCFs - SCFs <br> pc. manufacture - make (making) |
| 46 |  | pc. discovery - found (discover) <br> rs. using - used <br> tr. SCFs - <br> supercritical carbon dioxide | rs. using - used* <br> tr. SCF - supercritical carbon dioxide | tr. SCFs supercritical carbon dioxide | rs. using - used* <br> tr. SCFs - supercritical carbon dioxide pc. manufacture - make (making) |
| 47 |  |  |  |  |  |
| 48 |  | hip. chemists researchers |  | hip. chemists researchers | hip. chemists researchers |
| 49 |  |  |  |  |  |



## Anexo

|  | 21 | 22 | 23 | 24 | 25 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 31 | rs. make - make | rs. make - make |  |  |  |
| 32 | rs. using - used* <br> pc. make - production <br> (produce) <br> rs. polymer polymers | rs. polymer polymers <br> rs. used - used <br> pc. make - production <br> (produce) <br> rs. materials materials | rs. used - used rc. <br> fluoropolymers polymers | rc. <br> fluoropolymers polymers | rc. <br> fluoropolymers polymers |
| 33 |  |  |  |  |  |
| 34 | rc. using - useful* <br> rc. polymer polymerisation | rc. used - useful rc. polymer polymerisation | rc. used - useful tr. <br> fluoropolymers polymerisation | tr. <br> fluoropolymers polymerisation | tr. <br> fluoropolymers polymerisation |
| 35 | rs. make - making | rs. adding - add <br> rs. atoms - atoms <br> rs. make - making |  |  | rs. problem problem rs. making making rs. atoms atoms rs. dissolve dissolve |
| 36 | psm. group - team <br> rs. supercritical supercritical rs. carbon - carbon rs. dioxide - dioxide | rs. adding - add <br> rs. atoms - atoms |  |  | rs. atoms atoms |
| 37 | hip. supercritical carbon dioxide - SCF |  |  |  |  |
| 38 | hip. supercritical carbon dioxide - SCFs |  |  |  |  |
| 39 |  |  |  |  |  |
| 40 | hip. supercritical carbon dioxide - SCFs |  |  |  |  |
| 41 | hip. supercritical carbon dioxide - SCFs |  |  |  |  |
| 42 |  |  |  |  |  |


| 43 | hip. supercritical <br> carbon dioxide - SCF |  |  | 24 | 25 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 44 | hip. supercritical <br> carbon dioxide - SCFs |  |  |  |  |
| 45 | rs. using - using <br> hip. supercritical <br> carbon dioxide - SCFs <br> rs. make - make | rs. used - using* | rs. used - using* | pc. applications - <br> using* (use) |  |
| 46 | psm. group - team <br> rs. using - used* <br> rs. <br> supercrititical - <br> rs. carbon - carbon <br> s. dioxide - ioxide <br> s. make - make |  |  |  |  |
| 47 |  |  |  |  |  |
| 48 |  |  |  |  |  |
| 49 |  |  |  |  |  |

## Anexo

| 27 | rs. this - this |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 27 | 28 | 29 |  |
| 28 | rc. chemicals chemists |  |  |  |  |
|  |  |  |  |  |  |
| 29 |  |  | d. oración 28 - this |  |  |
| 30 |  | rs. team - team hip. supercritical carbon dioxide SCFs |  | 30 |  |
|  |  |  |  |  |  |
| 31 |  |  |  |  | rs. can - can <br> rs. them - they <br> rs. make - make |
| 32 |  | rs. using - used* | rs. polymer polymers |  | pc. make - production (produce) <br> rs. industrial industrial <br> rs. catalysts - catalysts |
| 33 |  |  |  |  | rc. catalysts - catalyse |
| 34 |  | rc. using - useful* | rc. polymer polymerisation |  | rs. catalysts - catalyst |
| 35 | rs. added - add <br> rs. organic - organic <br> rs. solvents - solvents |  |  |  | rs. make - making |
| 36 |  | rs. team - team <br> rs. supercritical - <br> supercritical <br> rs. carbon - carbon <br> rs. dioxide - dioxide |  |  | rs. team - team <br> rs. Nottingham - <br> Nottingham <br> tr. SCFs - supercritical carbon dioxide |
| 37 |  | hip. supercritical carbon dioxide - SCF |  |  | rs. $\mathrm{SCFs}-\mathrm{SCF}$ |
| 38 | rs. solvents - solvents | hip. supercritical carbon dioxide SCFs |  |  | rs. SCFs - SCFs |


|  | 26 | 27 | 28 | 29 | 30 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 39 | tr. solvents - VOCs |  |  |  |  |
| 40 |  | rs. problem - problem hip. supercritical carbon dioxide - SCFs |  |  | rs. SCFs - SCFs |
| 41 | rs. solvents solvents | hip. supercritical carbon dioxide - SCFs |  |  | rs. SCFs - SCFs |
| 42 |  |  |  |  | s. SCFs - they |
| 43 |  | hip. supercritical carbon dioxide - SCF |  |  | rs. $\mathrm{SCFs}-\mathrm{SCF}$ |
| 44 |  | hip. supercritical carbon dioxide - SCFs |  |  | rs. SCFs - SCFs |
| 45 |  | rs. using - using hip. supercritical carbon dioxide-SCFs | hip. chemists scientists |  | rs. SCFs - SCFs rs. help - help rs. make - make |
| 46 |  | rs. team - team <br> rs. using - used* <br> rs. supercritical supercritical <br> rs. carbon - carbon <br> rs. dioxide - dioxide |  |  | rs. team - team <br> tr. SCFs - supercritical <br> carbon dioxide <br> rs. make - make |
| 47 |  |  |  |  |  |
| 48 |  |  | hip. chemists researchers |  |  |
| 49 |  |  |  |  |  |

## Anexo



| 31 |  |  | 32 |  | 33 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 42 |  | rc. reactions - reactor | rc. reactions - reactor |  |  |
| 43 |  |  |  |  |  |
| 44 |  |  |  |  |  |
| 45 | rs. make - make |  |  |  |  |
| 46 | psm. discovered - found <br> rs. make - make <br> rs. prepare - preparing |  |  |  |  |
| 47 |  |  |  |  |  |
| 48 |  |  |  |  |  |
| 49 |  |  |  |  |  |

## Anexo

| 37 | hip. supercritical carbon dioxide - SCF <br> rs. reaction - reaction | 37 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | 38 |  |  |
| 38 | hip. supercritical carbon dioxide - SCFs | d. oración 37 - this <br> rs. SCF - SCFs |  |  |  |
|  |  |  |  | 39 |  |
| 39 | rs. reaction - reaction | rs. reaction - reaction | tr. solvents VOCs |  |  |
| 40 | hip. supercritical carbon dioxide - SCFs | rs. SCF - SCFs | rs. SCFs SCFs | d. oración 39 <br> - this <br> rc. purify impurities | 40 |
| 41 | hip. supercritical carbon dioxide - SCFs <br> rs. reaction - reaction | rs. reaction - reaction <br> rs. SCF - SCFs | rs. SCFs SCFs | hip. VOCs - <br> solvents <br> rs. reaction reaction | rs. SCFs - SCFs |
| 42 | rc. reaction - reactor | rc. reaction - reactor | s. SCFs - they | rc. reaction reactor | s. SCFs - they |
| 43 | hip. supercritical carbon dioxide - SCF <br> rs. reaction - reaction | rs. reaction - reaction psm. is over - is complete rs. SCF - SCF | rs. SCFs SCF | rs. reaction reaction rc. purify impurities | rs. SCFs - SCF <br> rs. leave - leaving rs. impurities impurities |
| 44 | hip. supercritical carbon dioxide - SCFs | rs. SCF - SCFs | rs. SCFs SCFs |  | rs. SCFs - SCFs |
| 45 | hip. supercritical carbon dioxide - SCFs | rs. SCF - SCFs | rs. SCFs SCFs |  | rs. SCFs - SCFs |
| 46 | rs. team - team <br> rs. found - found <br> rs. supercritical supercritical <br> rs. carbon - carbon <br> rs. dioxide - dioxide | tr. SCF - supercritical carbon dioxide rs. releasing releasing <br> rs. pressure - pressure | tr. SCFs supercritical carbon dioxide |  | tr. SCFs supercritical carbon dioxide |
| 47 |  |  |  |  |  |
| 48 |  |  |  |  |  |
| 49 |  |  |  |  |  |


| 42 | s. SCFs they rc. reaction - reactor rs. system - system | 42 |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 43 | rs. $\mathrm{SCFs}_{-}$ SCF <br> rs. reaction - reaction rs. system - system | rc. reactor <br> - reaction <br> rs. system <br> - system | 43 |  |  |  |  |  |
| 44 | rs. $\mathrm{SCFs}_{-}$ <br> SCFs |  | rs. SCF - <br> SCFs | 44 |  |  |  |  |
| 45 | rs. SCFs - <br> SCFs |  | rs. SCF - <br> SCFs | rs. SCFs SCFs rs. take taken psm. medicines drugs | 45 |  |  |  |
| 46 | tr. $\mathrm{SCFs}_{-}$ <br> supercrit. <br> carbon <br> dioxide |  | tr. SCF - <br> supercrit. <br> carbon <br> dioxide | tr. SCFs - <br> supercrit. <br> carbon <br> dioxide <br> psm. <br> medicines - <br> drug | rs. using used* <br> tr. SCFs - <br> supercrit. <br> carbon <br> dioxide <br> rs. make - <br> make <br> rs. drugs drug | 46 |  |  |
| 47 |  |  |  | psm. <br> medicines - <br> drug | rs. drugs drug | rs. drugs drug rc. blasting - blast | 47 |  |
| 48 |  |  |  |  | psm. <br> scientists researchers |  | rs. particles particles |  |
|  |  |  |  |  |  |  |  | 48 |
| 49 |  |  |  |  |  |  |  |  |

2. 7. 2. Matriz con el número de unidades léxicas.


|  | 1 | 2 | 3 |  |  | 6 | 7 | 8 |  | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 23 | 0 (1) | 0 | $0(1)$ | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
|  | 0 (1) | 0 | $0(1)$ | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 25 | 0 | 2 | 1 | 0 | 0 | 0 | 0 | 1 | 1 | 0 | 1 |
|  | 2 | 2 | 3 | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 0 |
| 26 |  |  |  |  |  |  |  |  |  |  |  |
| 27 | 1 | 1 | 2 | 1 | 0 | 0 | 1 | 3 | 1 | 0 | 1 |
| 28 | 1 | 1 | 1 | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 0 |
|  | 0 (1) | 0 | 0 (1) | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 29 |  |  |  |  |  |  |  |  |  |  |  |
| 30 | 0 | 1 | 3 | 0 | 0 | 0 | 1 | 1 | 1 | 0 | 0 |
| 31 | 1 | 1 | 1 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 32 | 0 (1) | 0 | $1(3)$ | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 33 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 34 | 0 (1) | 0 | 0 (1) | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
|  | 1 | 2 | 3 | 0 | 0 | 1 | 1 | 0 | 0 | 0 | 0 |
| 35 |  |  |  |  |  |  |  |  |  |  |  |
| 36 | 1 | 2 | 1 | 1 | 0 | 0 | 1 | 3 | 1 | 0 | 1 |
| 37 | 0 | 1 | 1 | 0 | 2 | 2 | 2 | 1 | 2 | 0 | 0 |
|  | 1 | 2 | 2 | 0 | 0 | 0 | 1 | 1 | 1 | 0 | 0 |
| 39 | 1 | 1 | 1 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 40 | 0 | 2 | 1 | 0 | 1 | 1 | 2 | 1 | 1 | 0 | 0 |
| 41 | 1 | 2 | 2 | 0 | 0 | 0 | 1 | 1 | 1 | 0 | 0 |
| 42 | 0 | 1 | 1 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 43 | 0 | 1(2) | 1 | 0 | 0 | 0 | 1 | 1 | 1 | 0 | 0 |
| 44 | 0 | 1 | 5 | 0 | 0 | 0 | 1 | 1 | 1 | 0 | 0 |


|  | 1 | 2 | 3 |  |  |  |  |  |  | $10 \quad 11$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 2 | 2 | 5 | 0 | 0 | 0 | 1 | 2 | 1 | 0 | 0 |
|  | 1(2) | 2 | 3(4) | 1 | 1 | 1 | 2 | 3 | 2 | 0 | 1 |
| 47 | 0 | 0 | 1 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
|  | 1 | 1 | 1 | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 0 |
| 49 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |



## Anexo




|  |  |  |  |  | 27 |  |  | 3 |  |  |  | 33 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 0 | 0 | 0 | 0 | 4(5) | 0 | 0 | 3 | 3 | 0 | 0 |  |
|  | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |  |
| 48 | 0 | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 0 | 0 | 0 |  |
|  | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |  |
| 49 |  |  |  |  |  |  |  |  |  |  |  |  |



## 2. 7. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 6)[6]$ | 2. $(1,5)[6](1,6)[7]$ | 3. $(2,12)[14](2,14)[16]$ |
| :--- | :--- | :--- |
| 4. $(2,0)[2]$ | 5. $(0,3)[3]$ | 6. $(1,2)[3]$ |
| 7. $(2,1)[3]$ | 8. $(0,6)[6]$ | 9. $(4,0)[4](4,1)[5]$ |
| 10. $(0,0)[0](0,1)[1]$ | 11. $(1,1)[2](2,1)[3]$ | 12. $(1,0)[1]$ |
| 13. $(0,1)[1]$ | 14. $(1,0)[1]$ | 15. $(0,0)[0]$ |
| 16. $(0,0)[0]$ | 17. $(3,5)[8]$ | 18. $(2,0)[2]$ |
| 19. $(4,2)[6]$ | 20. $(5,6)[11](5,8)[13]$ | 21. $(4,7)[11](4,8)[12]$ |
| 22. $(2,4)[6](3,4)[7]$ | 23. $(0,0)[0]$ | 24. $(0,1)[1]$ |
| 25. $(3,2)[5]$ | 26. $(2,1)[3]$ | 27. $(2,2)[4]$ |
| 28. $(1,0)[1]$ | 29. $(1,0)[1]$ | 30. $(3,5)[8]$ |
| 31. $(1,4)[5]$ | 32. $(4,2)[6](6,3)[9]$ | 33. $(1,3)[4]$ |
| 34. $(2,2)[4](3,2)[5]$ | 35. $(7,1)[8]$ | 36. $(8,1)[9](9,1)[10]$ |
| 37. $(0,2)[2]$ | 38. $(0,0)[0]$ | 39. $(0,0)[0]$ |
| 4. $(0,1)[1]$ | 41. $(1,2)[3]$ | 42. $(1,0)[1]$ |
| 43. $(3,0)[3]$ | 44. $(1,1)[2]$ | 45. $(6,1)[7]$ |
| 46. $(10,0)[10](11,0)[11]$ | 47. $(0,0)[0]$ | 48. $(0,0)[0]$ |
| 49. $(0,-)[0]$ |  |  |

## 2. 7. 4. Texto resultante tras eliminar las oraciones marginales.

1. Decent decaffeinated coffee has been around since 1960s, when chemist Kurt Zosel found an alternative to using the toxic and unpleasant tasting benzene to extract the caffeine. 2. He discovered that a 19th century chemical curiosity, known as a supercritical fluid (SCF), could dissolve out the caffeine but leave no solvent residue. 3. Supercritical fluids while still curious are now being used to destroy toxic waste, make industrial chemicals without toxic and highly flammable volatile organic compounds (VOCs) and are even making it easier to take your medicine. 4. So what are these strange materials and why are they so supercritical?
2. If you apply enough pressure to some gases while heating them they liquefy but keep their gaseous energy. 6. Conversely, heating some liquids while you apply pressure gives them gaseous energy but without losing their density. 7. These fluids are caught between the liquid and gas phase above a certain critical temperature and pressure - they are supercritical fluids, see Fig 1. 8. Many common chemicals can become supercritical, from carbon dioxide and water to the noble gas xenon.
3. Water, for instance, becomes a supercritical fluid when it is heated above $374^{\circ} \mathrm{C}$ and put under a pressure of 218 atmos. ${ }^{5} 11$. [That the fluid looks like a liquid but strangely, on the one hand can be mixed with oil but on the other will no longer dissolve ordinary table salt] <The effects> can be explained by the changes in the bonds between water molecules which, in the supercritical state, become weaker than normal. 12. So, oily molecules can squeeze in between them but they are too weak to hold the sodium and chloride ions from salt. 13. Amazingly, oxygen dissolved in supercritical water supports 'flameless' combustion. 14. Scientists at Sandia National Laboratories in New Mexico are using this property to destroy industrial and domestic waste without the need for conventional incineration.

[^4]17. Organic chemists from the University of Leeds have also been quick to latch on to Zosel's early discovery and have been using SCFs to extract natural products from plants and other organic materials for years. 18. Natural flavour molecules, such as vanilla, for instance, can be cleanly extracted from the pod using an SCF. 19. More recently, though, chemists have turned to SCFs to dissolve reactants that usually need a toxic and flammable VOC or do not dissolve at all.
20. Synthetic chemists are using SCFs in the manufacture of new types of polymer and other molecules that could function as industrial catalysts, thus avoiding the use of harmful solvents. 21. Joseph DeSimone's group at the University of North Carolina in Chapel Hill, for example, is using supercritical carbon dioxide to make new types of fluorine-containing polymer. 22. Adding fluorine atoms to a polymer chain is used to make some tough, smooth and chemically inert materials. 24. Modern fluoropolymers have more high-tech applications, such as acting as 'dry' lubricating layers for the moving parts in computers, eg hard drives, where a drop of oil would wreck the electronics. 25. The problem with making these new fluoropolymers, however, is that fluorine atoms have a residual negative charge, which makes them polar so they dissolve best in water. 26. This makes it difficult to process them further because any other chemicals added will usually be soluble only in organic solvents.
27. DeSimone's team has got around this problem by using supercritical carbon dioxide instead. 28. The chemists can now control the length of the polymer chains and their precise chemical structure. 29. This leads to consistent materials for high-tech. aerospace and electronic applications.
30. Martyn Poliakoff and his team at the University of Nottingham, meanwhile, are exploring how SCFs can help them make new industrial catalysts. 31. They have discovered that they can make organometallic compounds such as metal carbonyls, many of which are too unstable to prepare by conventional methods. 32. Metal carbonyls are used in various industrial reactions as catalysts for speeding up the production of simple materials such as formic acid and formaldehyde and more complex compounds, like pharmaceuticals and polymers. 33. Carbonyl compounds in which nitrogen or hydrogen molecules have been substituted for a carbonyl group can catalyse more complex reactions still. 34. For example, novel piano-stool shaped manganese carbonyls with an attached dihydrogen might be a useful polymerisation catalyst. 35. The problem in making them is that hydrogen and nitrogen gases do not dissolve well in conventional organic solvents at room temperature so it is hard to add the atoms to the starting molecule. 36. The Nottingham team, however, has found that hydrogen mixes very well with supercritical carbon dioxide at 80-100atmos, allowing the reaction to add hydrogen or nitrogen atoms as needed to the carbonyl compound.
37. Once the reaction is over, the SCF can be quickly recycled by releasing the pressure and trapping the carbon dioxide gas that escapes. 40. SCFs avoid this problem [to become contaminated] because once they become a gas again they leave behind any impurities,
41. SCFs are also much less viscous than liquid solvents, so they flow more easily through a reaction system. 42. They can also get into the smallest of crevices and pits inside the reactor system. 43. By flushing the system with an SCF once a reaction is complete any impurities can be washed out, leaving the system pristine and ready to be used again.
44. But, what about SCFs making it easier to take medicines? 45. Scientists are now using SCFs to help them make drugs that normally have to be injected work when taken by mouth instead. 46. A collaborative team from the US, Canada and Norway has found they can make sub-microscopic particles of the immunosuppressant drug cyclosporin, which is used to prevent transplanted organ rejection, by preparing it in
supercritical carbon dioxide and then blasting it into normal water by releasing the pressure.

## 2. 8. Texto 8: A healthy spread.

1. Cholesterol, an essential constituent of all cell membranes, forms part of the casing that protects nerve fibres and is a precursor in the production of vitamin D , steroid hormones and bile salts. 2. However, too much cholesterol in the blood is associated with heart disease. 3. While reducing elevated cholesterol levels cannot guarantee a healthy heart, scientists and doctors agree that it can reduce the risk of problems. 4. Here we consider how this can be done through dietary considerations, by reducing the use of food components that raise cholesterol and by adding cholesterollowering ingredients - ie functional foods or 'nutraceuticals'.
2. Most of the cholesterol we need is manufactured in our liver, ca 600 mg day. 6. Research suggests that if a healthy adult absorbs ca 80 mg day of cholesterol from foods such as animal products and eggs, the liver synthesises ca nine times as much (ca 720 mg day). 7. Reducing cholesterol in our diet therefore has only a modest effect on lowering blood cholesterol levels. 8. Scientists therefore considered which other components in food have a significant effect on cholesterol levels.
3. Cholesterol is insoluble in water and has to be carried around the blood stream as lipoproteins (ie all the insoluble lipid molecules in the body, attached to proteins). $\mathbf{1 0}$. Different combinations of lipids and proteins produce complexes of different densities. 11. Low density lipoproteins (LDLs), for example, supply cholesterol to cells, increased levels of which are associated with atherosclerosis - ie an accumulation of lipids in plaques on artery walls, which narrows the arteries and restricts the blood flow to the heart (ischaemia) and brain (stroke). 12. In contrast, high density lipoproteins (HDLs) transport cholesterol away from artery walls and therefore act as cardio-protectors. 13. To reduce the risk of heart disease, people therefore need to lower both their total cholesterol levels and their LDL-cholesterol levels in the plasma.
4. Dietary fats, both animal and vegetable, are made up of a mixture of triglycerides. 15. They are the major food constituents known to have a significant effect on cholesterol levels. 16. Animal fats, in butter for example, consist of a relatively high proportion of saturated fatty acids, some of which according to Judy Donnelly, nutritional biochemist at Trinity and All Saints University College Leeds,
'increase the proportion of LDL-cholesterol in the blood, compared with HDLcholesterol. 17. Cutting down on the amount of saturated fatty acids we eat could therefore lower our risk of heart disease. 18. In contrast, vegetable oils, such as those found in margarines, consist of long - chain polyunsaturated and monounsaturated fatty acids, which are associated with lowering LDL-cholesterol levels.
5. As people become more conscious of the benefits of cutting down excess intake of fats, especially saturated fats, spreads that contain $<80$ per cent fat are gaining in popularity. 20. It is the saturated fatty acid content that makes butters and margarines solid so we can spread them. 21. In lower fat spreads, fat substitutes are sometimes added to achieve the desired consistency and attributes. 22. Sometimes the substitute is water (in butter-milk and skimmed milk with added salts and preservatives), but it may be that starch molecules or whey proteins, which have been processed to give the particles a uniform size and thus a smooth feel in the mouth, are added. 23. Many of the resulting spreads, however, are not as popular with consumers because, for example, they lack the saturated fatty acids that give butter its distinctive flavour. 24. To improve the acceptability of low fat spreads, researchers are investigating synthetic replacements to animal fats, or 'structural fatitutes'. 25. Such compounds provide many similar properties, such as taste and texture, but they are not digested or absorbed from the gut into the blood and therefore cannot raise LDL-cholesterol levels. 26. They are used in the US in crisps and savoury products, but have not yet been added to fat spreads.
6. In the past few years the focus of research has shifted to adding ingredients (nutraceuticals) to food to reduce LDL-cholesterol levels. 28. Since the early 1950s scientists have known that plant sterols, and their hydrogenated counterparts, stanols, have cholesterol-lowering properties. 29. Unfortunately, these compounds are not naturally abundant in the food we eat. 30. Over the years scientists have come to realise that these compounds are very effective at lowering LDL-cholesterol levels when sufficient is eaten, for example in rich fat spreads. 31. Such products have recently been developed by esterifying the compounds with fatty acids to increase their fat solubility.
7. Two fat spreads - Benecol and Flora Proactive - are currently on the market for reducing LDL-cholesterol levels. 33. Benecol contains plant stanol esters (sitostanol esters), and Flora Proactive contains sterol esters. 34. Clinical trials, on people with
elevated cholesterol levels, have shown that these products reduce total plama cholesterol levels and LDL-cholesterol levels by 8-13 per cent, without effecting HDL levels. 35. Both products appear to have no adverse health effects and are non-toxic even in high doses, though a few people with the rare condition, phytosterolaemia cannot metabolise sterols and should avoid them.
8. According to Donnelly, there are two mechanisms by which these compounds are thought to lower cholesterol levels. 37. 'Choresterol is not very soluble in the gut and its absorption is slow. 38. Since you have other fats also being absorbed from the gut, cholesterol is one of the last to go through' she explained. 39. 'Plant sterols and stanols have similar structures to cholesterol so they also get left behind. $\mathbf{4 0}$. As the concentration of sterols/stanols increases, a threshold level is reached when the cholesterol molecules and the sterols/stanols coprecipitate into a solid crystalline form which cannot be absorbed by the gut' 41. According to Donnelly, another possibility focuses on micelles, which are clusters of molecules that transport fats across the gut membrane. 42. There is limited capacity for carrying cholesterol, and the plant sterols and stanols compete with cholesterol to get into the micelles, which limits the amount of cholesterol that can be absorbed. 43. 'These mechanisms do not just reduce the absorption of dietary cholesterol', said Donnelly, 'but they also hinder reabsorption of some of the cholesterol produced by the body, which has been used in producing bile salts’ 44. Bile salts are used in the intestine to breakdown the fatty acids that we eat. 45. Normally, the cholesterol in the bile salts would be recycled by re-absorption in the gut, but in this case they are excreted. 46. Essentially more of the cholesterol produced has to go in to producing more bile salts, reducing the amounts in the blood plasma.
9. Cholesterol-lowering spreads are some of the first functional foods on the market, but scientists are continually identifying ingredients that have potential health benefits. 48. As new advances in food technology allow their incorporation into products, we will see a lot more on the supermarket shelves. 49. Although these products can be beneficial, Donnelly says that she hopes 'people do not begin to rely on them because they are not miracle cures and there are many other factors involved in heart disease, which these products do not address.

## 2. 8. 1. Matriz de repetición de unidades léxicas.



| 1 |  | 2 |  | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 12 | rs. cholesterol cholesterol rc. protects protectors | rs. cholesterol cholesterol psm. heart cardio | rs. cholesterol - cholesterol psm. heart - cardio | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 13 | rs. cholesterol cholesterol | rs. cholesterol cholesterol rs. heart - heart rs. disease disease | psm. reducing - lower <br> rs. cholesterol - cholesterol <br> rs. levels - levels <br> pc. healthy - disease <br> (health) <br> rs. heart - heart <br> rs. reduce - reduce <br> rs. risk - risk | rs. reducing reducing* <br> a. raise - lower* <br> rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 14 |  |  |  | rs. dietary dietary |  |
| 15 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol <br> rs. levels - levels | rs. food - food psm. <br> components constituents rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 16 | rs. cholesterol cholesterol | rs. cholesterol cholesterol <br> rs. blood - blood | a. reducing - increase <br> rs. cholesterol - cholesterol | psm. raise increase rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 17 | rs. cholesterol cholesterol | rs. heart - heart rs. disease disease | ```psm. reducing - cutting down* pc. healthy - disease (health) rs. heart - heart psm. reduce - lower rs. risk - risk``` | rs. we - we+ psm. reducing cutting down rs. lowering lower* | rs. we - we + |
| 18 | rs. cholesterol cholesterol | rs. cholesterol cholesterol rs. associated associated* | $\begin{aligned} & \text { psm. reducing - lowering } \\ & \text { rs. cholesterol - cholesterol } \\ & \text { rs. levels - levels } \end{aligned}$ | rs. cholesterol cholesterol rs. lowering lowering | rs. cholesterol <br> - cholesterol |
| 19 |  |  | psm. reducing - cutting down* <br> a. risk - benefits | psm. reducing cutting down |  |
| 20 |  |  |  | rs. we - we + psm. components content | rs. we - we+ |
| 21 |  |  |  | rs. adding added |  |


| 1 |  | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 22 |  |  |  | rs. adding added |  |
| 23 |  |  |  |  |  |
| 24 |  |  | psm. scientists researchers |  |  |
| 25 |  | rs. cholesterol cholesterol <br> rs. blood - blood | a. reducing - raise rs. cholesterol cholesterol rs. levels - levels | rs. raise - raise rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 26 |  |  |  | rs. adding added |  |
| 27 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. reducing - reduce <br> rs. cholesterol cholesterol <br> rs. levels - levels pc. scientists - research (researcher) | rs. food - food <br> rs. cholesterol cholesterol <br> rs. adding adding <br> psm. lowering reduce <br> rs. ingredients ingredients rs. nutraceuticals - nutraceuticals | rs. cholesterol <br> - cholesterol |
| 28 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | ```psm. reducing - lowering rs. cholesterol - cholesterol rs. scientists - scientists``` | rs. cholesterol cholesterol rs. lowering lowering | rs. cholesterol <br> - cholesterol |
| 29 |  |  |  | rs. we - we+ rs. food - food | rs. we - we+ |
| 30 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | psm. reducing - lowering <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels <br> rs. scientists - scientists | rs. cholesterol cholesterol rs. lowering lowering | rs. cholesterol <br> - cholesterol |
| 31 |  |  |  |  |  |
| 32 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. reducing - reducing <br> rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol psm. lowering reducing | rs. cholesterol <br> - cholesterol |
| 33 |  |  |  |  |  |

## Anexo

| 1 |  | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 34 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. reducing - reduce <br> rs. elevated - elevated <br> rs. cholesterol - cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol psm. lowering reduce | rs. cholesterol <br> - cholesterol |
| 35 |  | a. disease health | rc. healthy - health <br> tr. risk - adverse |  |  |
| 36 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | psm. reducing - lower rs. cholesterol - cholesterol rs. levels - levels | rs. cholesterol cholesterol rs. lowering lower | rs. cholesterol <br> - cholesterol |
| 37 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 38 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 39 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 40 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 41 |  |  |  |  |  |
| 42 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 43 | rs. cholesterol cholesterol rc. production producing rs. bile - bile rs. salts - salts | rs. cholesterol cholesterol | rs. reducing - reduce* <br> rs. cholesterol - cholesterol | rs. dietary dietary rs. cholesterol cholesterol psm. lowering reduce | rs. cholesterol <br> - cholesterol |
| 44 | rs. bile - bile <br> rs. salts - salts |  |  |  | rs. we - we+ |
| 45 | rs. cholesterol cholesterol rs. bile - bile rs. salts - salts | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol |
| 46 | rs. cholesterol cholesterol rc. production producing rs. bile - bile rs. salts - salts | rs. cholesterol cholesterol <br> rs. blood - blood | rs. reducing - reducing <br> rs. cholesterol - cholesterol | rs. reducing reducing* rs. cholesterol cholesterol | rs. cholesterol - cholesterol psm. manufactured - produced |


| 1 |  | 2 |  | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 47 | rs. cholesterol <br> - cholesterol | rs. cholesterol cholesterol <br> a. disease - health | psm. reducing - lowering rs. cholesterol - cholesterol rc. healthy - health <br> rs. scientists - scientists <br> a. risk - benefits | rs. cholesterol cholesterol rs. lowering lowering <br> rs. ingredients ingredients rs. functional functional rs. foods - foods | rs. cholesterol <br> - cholesterol |
| 48 |  |  |  | rs. we - we + pc. adding incorporation (addition) rs. foods - food | rs. we - we+ |
| 49 |  | psm. associated - <br> involved <br> rs. heart - heart <br> rs. disease - disease | pc. healthy - disease <br> (health) <br> rs. heart - heart <br> pc. risk - beneficial <br> (benefit) |  |  |

## Anexo

| 7 | rs. cholesterol cholesterol | 7 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | 8 |  |  |
| 8 | pc. research scientists (researcher) rs. cholesterol cholesterol rs. foods - food | a. modest - <br> significant <br> rs. effect - effect <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels |  |  |  |
| 9 | rs. cholesterol cholesterol | rs. blood - blood rs. cholesterol cholesterol | rs. cholesterol - cholesterol | 9 |  |
| 10 |  |  |  | rs. lipid - lipids rs. proteins proteins | 10 |
| 11 | tr. healthy ischaemia rs. cholesterol cholesterol | a. lowering increased <br> rs. blood - blood rs. cholesterol cholesterol <br> rs. levels - levels | rs. cholesterol - cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol <br> rs. blood - blood <br> rs. lipoproteins lipoproteins <br> rs. lipid - lipids | rs. lipids lipids rc. proteins lipoproteins rs. densities density |
| 12 | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol psm. carried transport rs. lipoproteins lipoproteins | rc. proteins lipoproteins rs. densities density |
| 13 | pc. healthy disease (health) hip. adult people rs. cholesterol cholesterol | rs. reducing reduce* <br> rs. lowering lower <br> rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol - cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol |  |
| 14 | rs. animal animal | rc. diet - dietary |  |  |  |
| 15 | rs. cholesterol cholesterol rs. foods - food | a. modest - <br> significant <br> rs. effect - effect <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | psm. components - <br> constituents <br> rs. food - food <br> rs. have - have <br> rs. significant - significant <br> rs. effect - effect <br> rs. cholesterol - cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol |  |
| 16 | rs. cholesterol cholesterol rs. animal animal | a. lowering increase <br> rs. blood - blood rs. cholesterol cholesterol | rs. cholesterol - cholesterol | rs. cholesterol cholesterol <br> rs. blood - blood |  |



## Anexo

| 6 |  | 7 | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 30 | pc. research scientists (researcher) rs. cholesterol - cholesterol rs. cholesterol - cholesterol | rc. effect - effective <br> rs. lowering - lowering <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | rs. scientists - scientists <br> rc. effect - effective <br> rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol |  |
| 31 |  |  |  |  |  |
| 32 | rs. cholesterol <br> - cholesterol | psm. lowering - reducing <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol |  |
| 33 |  |  |  |  |  |
| 34 | hip. adult - <br> people <br> rs. cholesterol <br> - cholesterol | psm. lowering - reduce <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol |  |
| 35 | hip. adult people | rs. effect - effects* | rs. effect - effects* |  |  |
| 36 | rs. cholesterol <br> - cholesterol | rs. lowering - lower <br> rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol rs. levels - levels | rs. cholesterol cholesterol |  |
| 37 | rc. absorbs absorption rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol rc. insoluble soluble |  |
| 38 | rs. absorbs absorbed rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |
| 39 | rs. cholesterol <br> - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |
| 40 | rs. absorbs absorbed rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |
| 41 |  |  |  |  |  |
| 42 | rs. absorbs absorbed rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol <br> rs. carried carrying |  |
| 43 | rc. absorbs absorption rs. cholesterol - cholesterol | rs. reducing - reduce <br> rc. diet - dietary <br> rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |


| 6 |  | 7 |  | 9 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 44 |  | rs. our - we+ |  |  |  |
| 45 | rc. absorbs - re-absorption <br> rs. cholesterol - cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |
| 46 | rs. cholesterol - cholesterol | rs. cholesterol cholesterol psm. lowering reducing rs. blood - blood | rs. cholesterol cholesterol | rs. cholesterol cholesterol <br> rs. blood - blood |  |
| 47 | pc. research - scientists (researcher) <br> rc. healthy - health <br> rs. cholesterol - cholesterol <br> rs. foods - foods | rs. lowering lowering rs. cholesterol cholesterol | rs. scientists scientists rs. food - foods rs. cholesterol cholesterol | rs. cholesterol cholesterol |  |
| 48 | rs. foods - food <br> rs. products - products | rs. our - we+ | rs. food - food |  |  |
| 49 | pc. healthy - disease <br> (health) <br> hip. adult - people <br> rs. products - products |  |  |  |  |

## Anexo

| 12 | a. low - high <br> rs. density - density <br> rs. lipoproteins lipoproteins rs. cholesterol cholesterol rs. artery - artery rs. walls - walls psm. heart - cardio | 12 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | rs. LDLs - LDL <br> rs. cholesterol cholesterol <br> a. increased - lower <br> rs. levels - levels <br> rs. heart - heart hip. ischaemia - disease | rs. cholesterol - cholesterol psm. cardio heart | 13 |  |  |
| 14 |  |  |  | 14 |  |
| 15 | rs. cholesterol cholesterol <br> rs. levels - levels | rs. cholesterol <br> - cholesterol | rs. cholesterol cholesterol <br> rs. levels - levels | s. triglycerides they | 15 |
| 16 | rs. LDLs - LDL <br> rs. cholesterol cholesterol rs. increased - increase rs. blood - blood | rs. HDLs HDL <br> rs. cholesterol <br> - cholesterol | a. lower - increase <br> rs. LDL - LDL <br> rs. cholesterol cholesterol | rs. fats - fats <br> rs. animal - animal psm. made up of consist of | rs. cholesterol <br> - cholesterol |
| 17 | a. increased - lower* <br> rs. heart - heart <br> hip. ischaemia - disease | psm. cardio heart | psm. reduce - lower rs. risk - risk <br> rs. heart - heart <br> rs. disease - disease psm. lower - cutting down* | rc. fats - fatty |  |
| 18 | rs. LDLs - LDL <br> rs. cholesterol cholesterol <br> a. increased - lowering <br> rs. levels - levels <br> rs. associated associated* | rs. cholesterol <br> - cholesterol | rs. lower - lowering <br> rs. LDL - LDL <br> rs. cholesterol cholesterol <br> rs. levels - levels | rc. fats - fatty <br> rs. vegetable vegetable <br> psm. made up of consist of | rs. cholesterol - cholesterol rs. levels levels |
| 19 | $\begin{aligned} & \text { a. increased - cutting } \\ & \text { down* } \end{aligned}$ |  | a. risk - benefits rs. people - people psm. lower - cutting down* | rs. fats - fats |  |
| 20 |  |  |  | rc. fats - fatty | psm. <br> constituents content |
| 21 |  |  |  | rs. fats - fat |  |
| 22 |  |  |  |  |  |


| 11 |  | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 23 |  |  | tr. people - consumers | rc. fats - fatty |  |
| 24 |  |  |  | rs. fats - fats rs.- animal animal |  |
| 25 | rs. LDLs - LDL <br> rs. cholesterol cholesterol psm. increased - raise rs. levels - levels rs. blood - blood | rs. cholesterol cholesterol | a. lower - raise <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels |  | rs. cholesterol <br> - cholesterol <br> rs. levels levels |
| 26 |  |  |  | rs. fats - fat |  |
| 27 | rs. LDLs - LDL <br> rs. cholesterol cholesterol <br> a. increased - reduce <br> rs. levels - levels | rs. cholesterol cholesterol | psm. lower - reduce <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels |  | rs. cholesterol <br> - cholesterol <br> rs. levels levels |
| 28 | rs. cholesterol cholesterol <br> a. increased - lowering | rs. cholesterol cholesterol | rs. lower - lowering <br> rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 29 |  |  |  |  | rs. food - food |
| 30 | rs. LDLs - LDL <br> rs. cholesterol cholesterol <br> a. increased - lowering <br> rs. levels - levels | rs. cholesterol cholesterol | rs. lower - lowering <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels | rs. fats - fat | rc. effect effective rs. cholesterol - cholesterol rs. levels levels |
| 31 |  |  |  | rs. fats - fat |  |
| 32 | rs. LDLs - LDL <br> rs. cholesterol cholesterol <br> a. increased - reducing <br> rs. levels - levels | rs. cholesterol cholesterol | psm. lower - reducing <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels | rs. fats - fat | rs. cholesterol <br> - cholesterol <br> rs. levels levels |
| 33 |  |  |  |  |  |
| 34 | rs. LDLs - LDL <br> rs. cholesterol cholesterol psm. increased elevated rs. levels - levels | rs. HDLs HDL <br> rs. cholesterol cholesterol | rs. people - people <br> psm. lower - reduce <br> rs. total - total <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels <br> rs. plasma - plasma |  | rs. cholesterol <br> - cholesterol <br> rs. levels levels |
| 35 | tr. ischaemia - health |  | tr. risk - adverse <br> a. disease - health <br> rs. people - people |  | rs. effect effects* |

## Anexo

| 11 |  | 12 | 13 | 14 | 15 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 36 | rs. cholesterol cholesterol <br> a. increased - lower <br> rs. levels - levels | rs. cholesterol <br> - cholesterol | rs. lower - lower <br> rs. cholesterol - cholesterol <br> rs. levels - levels |  | rs. cholesterol <br> - cholesterol <br> rs. levels levels |
| 37 | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 38 | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol - cholesterol | rs. fats - fats | rs. cholesterol <br> - cholesterol |
| 39 | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 40 | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 41 |  |  |  | rs. fats - fats |  |
| 42 | rs. cholesterol cholesterol | psm. transport <br> - carrying <br> rs. cholesterol <br> -cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 43 | rs. cholesterol cholesterol <br> a. increased - reduce* | rs. cholesterol <br> - cholesterol | psm. lower - reduce <br> rs. cholesterol - cholesterol | rs. dietary dietary | rs. cholesterol <br> - cholesterol |
| 44 |  |  |  | rc. fats fatty |  |
| 45 | rs. cholesterol cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol <br> - cholesterol |
| 46 | rs. cholesterol cholesterol <br> a. increased - reducing* <br> rs. blood - blood | rs. cholesterol <br> - cholesterol | psm. lower - reducing rs. cholesterol - cholesterol rs. plasma - plasma |  | rs. cholesterol <br> - cholesterol |
| 47 | rs. cholesterol cholesterol <br> a. increased lowering* | rs. cholesterol <br> - cholesterol | a. risk - benefits <br> a. disease - health <br> rs. lower - lowering <br> rs. cholesterol - cholesterol |  | rs. food - <br> foods <br> rs. cholesterol <br> - cholesterol |
| 48 |  |  |  |  | rs. food - food |
| 49 | rs. heart - heart <br> hip. ischaemia - disease | psm. cardio heart | pc. risk - beneficial <br> (benefit) <br> rs. heart - heart <br> rs. disease - disease <br> rs. people - people |  |  |


| 16 |  | 17 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 17 | rs. saturated saturated rs. fatty - fatty rs. acids - acids a. increase lower* |  |  |  |  |
| 18 | rs. consist consist rc. saturated polyunsaturated rs. fatty - fatty rs. acids - acids a. increase lowering rs. LDL - LDL rs. cholesterol cholesterol | rc. saturated polyunsaturated rs. fatty - fatty rs. acids - acids rs. lower lowering* | 18 |  |  |
| 19 | rs. saturated saturated rs. fats - fats a. increase cutting down* | rs. cutting down - cutting down rs. saturated saturated rc. fatty - fats <br> a. risk - benefits | rc. polyunsaturated <br> - saturated <br> rc. fatty - fats <br> psm. lowering - <br> cutting down* | 19 |  |
| 20 | rs. butter butters rs. saturated saturated rs. fatty - fatty rs. acids - acid | rs. saturated saturated rs. fatty - fatty rs. acids - acid rs. we - we+ | rs. margarines margarines rc. polyunsaturated - saturated rs. fatty - fatty rs. acids - acid | rs. saturated - <br> saturated <br> rc. fats - fatty <br> rc. spreads - spread | 20 |
| 21 | rs. fats - fat | rc. fatty - fat | rc. fatty - fat | rs. fats - fat <br> rs. spreads - spreads | rc. fatty - fat |
| 22 | rs. butter - butter |  |  |  | rs. butters - butter |
| 23 | rs. butter - butter <br> rs. saturated saturated rs. fatty - fatty <br> rs. acids - acids | rs. saturated saturated rs. fatty - fatty rs. acids - acids | rc. polyunsaturated <br> - saturated <br> rs. fatty - fatty <br> rs. acids - acids | tr. people - consumers <br> rs. saturated saturated rc. fats - fatty rs. spreads - spreads rc. popularity popular | rs. saturated saturated rs. fatty - fatty rs. acid - acids rs. butters - butter |
| 24 | rs. animal animal <br> rs. fats - fats | rc. fatty - fat | rc. fatty - fat | rs. fats - fat rs. spreads - spreads psm. popularity acceptability | rc. fatty - fat |
| 25 | $\begin{aligned} & \text { psm. increase - } \\ & \text { raise } \\ & \text { rs. LDL - LDL } \\ & \text { rs. cholesterol - } \\ & \text { cholesterol } \\ & \text { rs. blood - blood } \end{aligned}$ | a. lower - raise* | a. lowering - raise <br> rs. LDL - LDL <br> rs. cholesterol cholesterol rs. levels - levels | a. cutting down raise* |  |

## Anexo

|  | 16 | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 26 | rs. fats - fat | rc. fatty - fat | rc. fatty - fat | rs. fats - fat rs. spreads spreads | rc. fatty - fat |
| 27 | a. increase reduce <br> rs. LDL - LDL <br> rs. cholesterol cholesterol | psm. lower reduce* | psm. lowering - reduce <br> rs. LDL - LDL <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | psm. cutting down - reduce* |  |
| 28 | a. increase lowering rs. cholesterol cholesterol | rs. lower lowering* | rs. lowering - lowering rs. cholesterol cholesterol | psm. cutting down - lowering* |  |
| 29 |  | rs. we - we + <br> rs. eat - eat |  |  | rs. we - we+ |
| 30 | rs. fats - fat <br> a. increase lowering rs. LDL - LDL rs. cholesterol cholesterol | rc. fatty - fat rs. eat - eaten rs. lower lowering* | rc. fatty - fat <br> rs. lowering - lowering <br> rs. LDL - LDL <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | psm. cutting down - lowering* rs. fats - fat rs. spreads spreads | rc. fatty - fat |
| 31 | rs. fatty - fatty <br> rs. acids - acids | rs. fatty - fatty <br> rs. acids - acids | rs. fatty - fatty <br> rs. acids - acids | rs. fats - fat hip. spreads products | rs. fatty - fatty <br> rs. acid - acids |
| 32 | rs. fats - fat <br> a. increase reducing rs. LDL - LDL rs. cholesterol cholesterol | rc. fatty - fat psm. lower reducing* | rc. fatty - fat psm. lowering - reducing rs. LDL - LDL <br> rs. cholesterol cholesterol rs. levels - levels | psm. cutting down - reducing* rs. fats - fat rs. spreads spreads | rc. fatty - fat |
| 33 |  |  |  | rs. contain contains |  |
| 34 | a. increase reduce rs. LDL - LDL rs. cholesterol cholesterol | psm. lower reduce* | psm. lowering - reduce <br> rs. LDL - LDL <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels | rs. people - people psm. cutting down - reduce* |  |
| 35 |  | tr. risk - adverse <br> a. disease health |  | rs. people - people |  |
| 36 | rs. Donnelly Donnelly <br> a. increase lower rs. cholesterol cholesterol | rs. lower lower* | rs. lowering - lower rs. cholesterol cholesterol rs. levels - levels | psm. cutting down - lower* |  |


|  | 16 | 17 | 18 | 19 | 20 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 37 | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |  |  |
| 38 | rs. fats - fats <br> s. Donnelly - she <br> rs. cholesterol - cholesterol | rc. fatty - fats | rc. fatty - fats rs. cholesterol cholesterol | rs. fats - fats | rc. fatty - fats |
| 39 | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |  |  |
| 40 | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |  |  |
| 41 | rs. fats - fats <br> rs. Donnelly - Donnelly | rc. fatty - fats | rc. fatty - fats | rs. fats - fats | rc. fatty - fats |
| 42 | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |  |  |
| 43 | rs. Donnelly - Donnelly <br> a. increase - reduce <br> rs. cholesterol - cholesterol | psm. lower - reduce* | psm. loweringreduce rs. cholesterol cholesterol | psm. cutting down reduce* |  |
| 44 | rs. fatty - fatty rs. acids - acids | rs. fatty - fatty <br> rs. acids - acids <br> rs. we - we+ <br> rs. eat - eat | rs. fatty - fatty <br> rs. acids - acids | rc. fats - fatty | rs. fatty fatty rs. acid acids <br> rs. we - we+ |
| 45 | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |  |  |
| 46 | a. increase - reducing <br> rs. cholesterol - cholesterol <br> rs. blood - blood | psm. lower reducing* | psm. lowering reducing rs. cholesterol cholesterol | psm. cutting down reducing* |  |
| 47 | hip. butter - foods <br> a. increase - lowering <br> rs. cholesterol - cholesterol | rs. lower - lowering* <br> a. risk - benefits <br> a. disease - health | hip. margarines foods rs. lowering lowering rs. cholesterol cholesterol | psm. cutting downlowering* rs. spreads spreads |  |
| 48 |  | rs. we - we+ |  |  | rs. we - we+ |
| 49 | rs. Donnelly - Donnelly | pc. risk - beneficial (benefit) rs. heart - heart rs. disease - disease |  | rs. people people |  |

## Anexo



| 21 |  | 22 | 23 | 24 | 25 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 34 |  |  | hip. consumers people |  | a. raise - reduce <br> rs. LDL - LDL <br> rs. cholesterol - cholesterol <br> rs. levels - levels |
| 35 |  |  | hip. consumers people |  |  |
| 36 |  |  |  |  | a. raise - lower <br> rs. cholesterol - cholesterol <br> rs. levels - levels |
| 37 |  |  |  |  | rc. absorbed - absorption <br> rs. gut - gut <br> rs. cholesterol - cholesterol |
| 38 | rs. fat - fats |  | rc. fatty - fats | rs. fat - fats | rs. absorbed - absorbed <br> rs. gut - gut <br> rs. cholesterol - cholesterol |
| 39 |  |  |  |  | rs. cholesterol - cholesterol |
| 40 |  |  |  |  | rs. absorbed - absorbed <br> rs. gut - gut <br> rs. cholesterol - cholesterol |
| 41 | rs. fat - fats |  | rc. fatty - fats | rs. fat - fats | rs. gut - gut |
| 42 |  |  |  |  | rs. absorbed - absorbed <br> rs. cholesterol - cholesterol |
| 43 |  |  |  |  | rc. absorbed - absorption <br> a. raise - reduce <br> rs. cholesterol - cholesterol |
| 44 | rc. fat - fatty |  | rs. fatty - fatty <br> rs. acids - acids | rc. fat - fatty | psm. gut - intestine |
| 45 |  |  |  |  | rc. absorbed - re-absorption <br> rs. gut - gut <br> rs. cholesterol - cholesterol |
| 46 |  |  |  |  | a. raise - reducing <br> rs. cholesterol - cholesterol |
| 47 | rs. spreads spreads psm. <br> substitutes ingredients | psm. substitutesingredients | rs. spreads spreads | rs. spreads - <br> spreads <br> psm. researchers - <br> scientists <br> psm. replacements <br> - ingredients | hip. compounds ingredients <br> a. raise - lowering <br> rs. cholesterol - cholesterol |
| 48 | pc. added incorporation (addition) | pc. added incorporation (addition) |  |  |  |
| 49 |  |  | hip. consumers people |  |  |

## Anexo

| 26 |  | 27 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 27 | rs. added adding |  |  |  |  |
| 28 |  | pc. research - scientists (researcher) <br> psm. reduce - lowering rs. cholesterol cholesterol | 28 |  |  |
| 29 |  | rs. food - food | hip. sterols/stanols - compounds | 29 |  |
| 30 | rs. fat - fat <br> rs. spreads <br> - spreads | pc. research - scientists (researcher) psm. reduce - lowering rs. LDL - LDL <br> rs. cholesterol cholesterol rs. level-levels | rs. scientists scientists hip. sterols/stanols - compounds rs. cholesterol cholesterol rs. lowering lowering | rs. compounds - compounds rs. eat - eaten | 30 |
| 31 | rs. fat - fat hip. <br> spreads products |  | hip. sterols/stanols <br> - compounds | rs. compounds <br> - compounds | rs. compounds compounds rs. fat - fat hip. spreads - products |
| 32 | rs. fat - fat <br> rs. spreads <br> - spreads | rs. reduce - reducing <br> rs. LDL - LDL <br> rs. cholesterol cholesterol rs. level-levels | rs. cholesterol cholesterol psm. lowering reducing |  | psm. lowering - reducing <br> rs. LDL - LDL <br> rs. cholesterol cholesterol <br> rs. levels - levels <br> rs. fat - fat <br> rs. spreads - spreads |
| 33 |  |  | rs. plant - plant <br> rs. sterols - sterol <br> rs. stanols - stanol | hip. <br> compounds - <br> stanol/sterol | hip. compounds stanol/sterol |
| 34 |  | rs. reduce - reduce <br> rs. LDL - LDL <br> rs. cholesterol cholesterol rs. level-levels | rs. cholesterol cholesterol psm. lowering reduce |  | psm. lowering - reduce <br> rs. LDL - LDL <br> rs. cholesterol - <br> cholesterol <br> rs. levels - levels |
| 35 |  |  | rs. sterols - sterols | tr. compounds <br> - sterols | tr. compounds - sterols |
| 36 |  | psm. reduce - lower rs. cholesterol cholesterol rs. level-levels | hip. sterols/stanols <br> - compounds <br> rs. cholesterol - <br> cholesterol <br> rs. lowering - lower | rs. compounds <br> - compounds | rs. compounds compounds <br> rs. lowering - lower <br> rs. cholesterol cholesterol rs. levels - levels |
| 37 |  | rs. cholesterol cholesterol | rs. cholesterol cholesterol |  | rs. cholesterol cholesterol |

26

| 38 | rs. fat - fats | rs. cholesterol cholesterol | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol rs. fat - fats |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 39 |  | rs. cholesterol cholesterol | rs. plant - plant <br> rs. sterols - sterols <br> rs. stanols - stanols <br> rs. cholesterol - cholesterol | hip. compounds - sterols/stanols | hip. compounds - sterols/stanols rs. cholesterol cholesterol |
| 40 |  | rs. cholesterol cholesterol | rs. sterols - sterols <br> rs. stanols - stanols <br> rs. cholesterol - cholesterol | hip. compounds - sterols/stanols | hip. compounds - sterols/stanols rs. cholesterol cholesterol |
| 41 | rs. fat - fats |  |  |  | rs. fat - fats |
| 42 |  | rs. cholesterol cholesterol | rs. plant - plant <br> rs. sterols - sterols <br> rs. stanols - stanols <br> rs. cholesterol - cholesterol | hip. compounds <br> - sterols/stanols | hip. compounds <br> - sterols/stanols rs. cholesterol cholesterol |
| 43 |  | rs. reduce reduce rs. cholesterol cholesterol | rs. cholesterol - cholesterol <br> rs. lowering - reduce |  | psm. lowering reduce rs. cholesterol cholesterol |
| 44 | rc. fat - fatty |  |  | rs. we - we + <br> rs. eat - eat | rs. eaten - eat rc. fat - fatty |
| 45 |  |  | rs. cholesterol - cholesterol |  | rs. cholesterol cholesterol |
| 46 |  | rs. reduce reducing rs. cholesterol cholesterol | rs. cholesterol - cholesterol psm. lowering - reducing |  | psm. lowering reducing rs. cholesterol cholesterol |
| 47 |  | pc. research scientists (researcher) rs. ingredients ingredients rs. food - foods psm. reduce lowering rs. cholesterol cholesterol | rs. scientists - scientists <br> rs. cholesterol - cholesterol <br> rs. lowering - lowering | rs. food - foods | rs. scientists scientists rs. lowering lowering rs. cholesterol cholesterol rs. spreads spreads |
| 48 | rs. products products pc. added incorporation (addition) | pc. adding incorporation (addition) <br> s. ingredients their rs. food - food |  | rs. food - food <br> rs. we - we+ |  |
| 49 | rs. products products |  |  |  |  |

## Anexo



|  | 31 | 32 | 33 | 34 | 35 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 44 | rs. fatty - fatty rs. acids - acids | rc. fat - fatty |  |  |  |
| 45 |  | rs. cholesterol cholesterol |  | rs. cholesterol cholesterol |  |
| 46 |  | rs. reducing - reducing <br> rs. cholesterol cholesterol |  | rs. cholesterol - <br> cholesterol <br> rs. reduce - reducing <br> rs. plasma - plasma |  |
| 47 | tr. products spreads | rs. spreads - spreads <br> rs. market - market <br> psm. reducing - lowering <br> rs. cholesterol - <br> cholesterol |  | rs. cholesterol cholesterol <br> tr. products - spreads psm. reduce lowering | tr. products spreads rs. have - have pc. adverse benefits (adversity) rs. health - health |
| 48 |  | rc. market - supermarket |  |  |  |
| 49 |  |  |  | rs. people - people | tr. adverse beneficial rs. people - people |

## Anexo



| 36 | 37 | 38 | 40 |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 47 | rs. lower - lowering <br> rs. cholesterol - <br> cholesterol | rs. cholesterol - <br> cholesterol | rs. cholesterol - <br> cholesterol | rs. cholesterol <br> - cholesterol | rs. cholesterol- <br> cholesterol |
| 48 |  |  | rs. she - she <br> psm. explained - <br> says |  |  |
| 49 | rs. Donnelly - <br> Donnelly |  |  |  |  |

## Anexo



2. 8. 2. Matriz con el número de unidades léxicas.


## Anexo

|  |  | 2 | 3 | 4 | 5 | 6 | 7 | 8 |  | 1 | 11 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 23 | 0 | 0 | 0 | 0 | 0 | 1 | 0 | 0 | 0 | 0 | 0 |
|  | 0 | 0 | 1 | 0 | 0 | 2 | 0 | 1 | 0 | 0 | 0 |
| 25 | 0 | 2 | 3 | 2 | 1 | 2 | 4 | 2 | 2 | 0 | 5 |
|  | 0 | 0 | 0 | 1 | 0 | 1 | 0 | 0 | 0 | 0 | 0 |
| 26 |  |  |  |  |  |  |  |  |  |  |  |
| 27 | 1 | 1 | 4 | 6 | 1 | 3 | 3 | 4 | 1 | 0 | 4 |
| 28 | 1 | 1 | 3 | 2 | 1 | 2 | 2 | 2 | 1 | 0 | 2 |
|  | 0 | 0 | 0 | 1(2) | 0 (1) | 1 | $0(1)$ | 1 | 0 | 0 | 0 |
| 29 |  |  |  |  |  |  |  |  |  |  |  |
| 30 | 1 | 1 | 4 | 2 | 1 | 3 | 4 | 4 | 1 | 0 | 4 |
| 31 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
|  | 1 | 1 | 3 | 2 | 1 | 1 | 3 | 2 | 1 | 0 | 4 |
| 32 |  |  |  |  |  |  |  |  |  |  |  |
| 33 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 34 | 1 | 1 | 4 | 2 | 1 | 2 | 3 | 2 | 1 | 0 | 4 |
|  | 0 | 1 | 2 | 0 | 0 | 1 | $0(1)$ | 0 (1) | 0 | 0 | 1 |
| 35 |  |  |  |  |  |  |  |  |  |  |  |
| 36 | 1 | 1 | 3 | 2 | 1 | 1 | 3 | 2 | 1 | 0 | 3 |
| 37 | 1 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 2 | 0 | 1 |
|  | 1 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 1 | 0 | 1 |
| 39 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 0 | 1 |
| 40 | 1 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 1 | 0 | 1 |
| 41 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 42 | 1 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 2 | 0 | 1 |
| 43 | 4 | 1 | 1(2) | 3 | 1 | 2 | 3 | 1 | 1 | 0 | 1(2) |
| 44 | 2 | 0 | 0 | 0 | $0(1)$ | 0 | $0(1)$ | 0 | 0 | 0 | 0 |


|  | 1 |  |  | 4 | 5 |  | 7 |  |  |  | 11 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 3 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 1 | 0 | 1 |
|  | 4 | 2 | 2 | 1(2) | 2 | 1 | 3 | 1 | 2 | 0 | 2 (3) |
| 47 | 1 | 2 | 5 | 5 | 1 | 4 | 2 | 3 | 1 | 0 | 1(2) |
|  | 0 | 0 | 0 | 2(3) | 0 (1) | 2 | 0 (1) | 1 | 0 | 0 | 0 |
|  | 0 | 3 | 3 | 0 | 0 | 3 | 0 | 0 | 0 | 0 | 2 |

## Anexo



|  | 12 |  |  | 15 |  | 17 |  | 19 | 20 |  | 22 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 35 | 0 | 3 | 0 | 0 (1) | 0 | 2 | 0 | 1 | 0 | 0 | 0 |
|  | 1 | 3 | 0 | 2 | 3 | 0 (1) | 3 | 0 (1) | 0 | 0 | 0 |
| 37 | 1 | 1 | 0 | 1 | 1 | 0 | 1 | 0 | 0 | 0 | 0 |
|  | 1 | 1 | 1 | 1 | 3 | 1 | 2 | 1 | 1 | 1 | 0 |
| 38 |  |  |  |  |  |  |  |  |  |  |  |
| 39 | 1 | 1 | 0 | 1 | 1 | 0 | 1 | 0 | 0 | 0 | 0 |
| 40 | 1 | 1 | 0 | 1 | 1 | 0 | 1 | 0 | 0 | 0 | 0 |
|  | 0 | 0 | 1 | 0 | 2 | 1 | 1 | 1 | 1 | 1 | 0 |
| 42 | 2 | 1 | 0 | 1 | 1 | 0 | 1 | 0 | 0 | 0 | 0 |
| 43 | 1 | 2 | 1 | 1 | 3 | 0 (1) | 2 | 0 (1) | 0 | 0 | 0 |
| 44 | 0 | 0 | 1 | 0 | 2 | 3(4) | 2 | 1 | $2(3)$ | 1 | 0 |
| 45 | 1 | 1 | 0 | 1 | 1 | 0 | 1 | 0 | 0 | 0 | 0 |
| 46 | 1 | 3 | 0 | 1 | 3 | O(1) | 2 | 0 (1) | 0 | 0 | 0 |
| 47 | 1 | 4 | 0 | 2 | 3 |  | 3 | 1(2) | 0 | 2 | 1 |
| 48 | 0 | 0 | 0 | 1 | 0 | O(1) | 0 | 0 | 0 (1) | 1 | 1 |
| 49 | 1 | 4 | 0 | 0 | 1 | 3 | 0 | 1 | 0 | 0 | 0 |

## Anexo




## Anexo




## 2. 8. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 3)[3]$ | 2. $(0,4)[4]$ | 3. $(1,17)[18]$ |
| :--- | :--- | :--- |
| 4. $(1,6)[7](1,9)[10]$ | 5. $(0,1)[1]$ | 6. $(2,6)[8]$ |
| 7. $(2,14)[16](2,15)[17]$ | 8. $(4,4)[8]$ | 9. $(0,2)[2]$ |
| 10. $(0,1)[1]$ | 11. $(5,10)[15](5,12)[17]$ | 12. $(2,0)[2]$ |
| 13. $(5,13)[18](6,14)[20]$ | 14. $(0,2)[2]$ | 15. $(3,1)[4]$ |
| 16. $(4,14)[18](4,15)[19]$ | 17. $(3,6)[9](6,8)[14]$ | 18. $(7,9)[16](7,10)[17]$ |
| 19. $(1,3)[4](4,5)[9]$ | 20. $(4,1)[5](4,2)[6]$ | 21. $(0,2)[2]$ |
| 22. $(0,0)[0]$ | 23. $(5,3)[8]$ | 24. $(3,3)[6]$ |
| 25. $(6,11)[17]$ | 26. $(3,0)[3]$ | 27. $(10,7)[17]$ |
| 28. $(2,7)[9]$ | 29. $(0,0)[0]$ | 30. $(13,5)[18](15,5)[20]$ |
| 31. $(2,0)[2]$ | 32. $(9,4)[13](10,4)[14]$ | 33. $(2,2)[4]$ |
| 34. $(10,3)[13]$ | 35. $(1,1)[2]$ | 36. $(12,1)[13]$ |
| 37. $(1,3)[4]$ | 38. $(3,3)[6]$ | 39. $(2,2)[4]$ |
| 40. $(5,2)[7]$ | 41. $(0,0)[0]$ | 42. $(4,1)[5]$ |
| 43. $(5,2)[7](8,2)[10]$ | 44. $(1,1)[2](2,1)[3]$ | 45. $(7,1)[8]$ |
| 46. $(7,0)[7](8,0)[8]$ | 47. $(15,1)[16](16,1)[17]$ | 48. $(2,0)[2](3,0)[3]$ |
| 49. $(5,-)[5]$ |  |  |

## 2. 8. 4. Texto resultante tras eliminar las oraciones marginales.

1. Cholesterol, an essential constituent of all cell membranes, forms part of the casing that protects nerve fibres and is a precursor in the production of vitamin $D$, steroid hormones and bile salts. 2. However, too much cholesterol in the blood is associated with heart disease. 3. While reducing elevated cholesterol levels cannot guarantee a healthy heart, scientists and doctors agree that it can reduce the risk of problems. 4. Here we consider how this can be done through dietary considerations, by reducing the use of food components that raise cholesterol and by adding cholesterollowering ingredients - ie functional foods or 'nutraceuticals'.
2. Most of the cholesterol we need is manufactured in our liver, ca 600 mg day. 6. Research suggests that if a healthy adult absorbs ca 80 mg day of cholesterol from foods such as animal products and eggs, the liver synthesises ca nine times as much (ca 720 mg day). 7. Reducing cholesterol in our diet therefore has only a modest effect on lowering blood cholesterol levels. 8. Scientists therefore considered which other components in food have a significant effect on cholesterol levels.
3. Cholesterol is insoluble in water and has to be carried around the blood stream as lipoproteins (ie all the insoluble lipid molecules in the body, attached to proteins). 10. Different combinations of lipids and proteins produce complexes of different densities. 11. Low density lipoproteins (LDLs), for example, supply cholesterol to cells, increased levels of which are associated with atherosclerosis - ie an accumulation of lipids in plaques on artery walls, which narrows the arteries and restricts the blood flow to the heart (ischaemia) and brain (stroke). 12. In contrast, high density lipoproteins (HDLs) transport cholesterol away from artery walls and therefore act as cardio-protectors. 13. To reduce the risk of heart disease, people therefore need to lower both their total cholesterol levels and their LDL-cholesterol levels in the plasma.
4. Dietary fats, both animal and vegetable, are made up of a mixture of triglycerides. 15. They are the major food constituents known to have a significant
effect on cholesterol levels. 16. Animal fats, in butter for example, consist of a relatively high proportion of saturated fatty acids, some of which according to Judy Donnelly, nutritional biochemist at Trinity and All Saints University College Leeds, 'increase the proportion of LDL-cholesterol in the blood, compared with HDL-cholesterol. 17. Cutting down on the amount of saturated fatty acids we eat could therefore lower our risk of heart disease. 18. In contrast, vegetable oils, such as those found in margarines, consist of long - chain polyunsaturated and monounsaturated fatty acids, which are associated with lowering LDL-cholesterol levels.
5. As people become more conscious of the benefits of cutting down excess intake of fats, especially saturated fats, spreads that contain $<80$ per cent fat are gaining in popularity. $\mathbf{2 0}$. It is the saturated fatty acid content that makes butters and margarines solid so we can spread them. 21. In lower fat spreads, fat substitutes are sometimes added to achieve the desired consistency and attributes. 23. Many of the resulting spreads, however, are not as popular with consumers because, for example, they lack the saturated fatty acids that give butter its distinctive flavour. 24. To improve the acceptability of low fat spreads, researchers are investigating synthetic replacements to animal fats, or 'structural fatitutes'. 25. Such compounds provide many similar properties, such as taste and texture, but they are not digested or absorbed from the gut into the blood and therefore cannot raise LDL-cholesterol levels. 26. They are used in the US in crisps and savoury products, but have not yet been added to fat spreads.
6. In the past few years the focus of research has shifted to adding ingredients (nutraceuticals) to food to reduce LDL-cholesterol levels. 28. Since the early 1950s scientists have known that plant sterols, and their hydrogenated counterparts, stanols, have cholesterol-lowering properties. 30. Over the years scientists have come to realise that these compounds are very effective at lowering LDL-cholesterol levels when sufficient is eaten, for example in rich fat spreads. 31. Such products have recently been developed by esterifying the compounds with fatty acids to increase their fat solubility.
7. Two fat spreads - Benecol and Flora Proactive - are currently on the market for reducing LDL-cholesterol levels. 33. Benecol contains plant stanol esters (sitostanol esters), and Flora Proactive contains sterol esters. 34. Clinical trials, on people with elevated cholesterol levels, have shown that these products reduce total plama cholesterol levels and LDL-cholesterol levels by 8-13 per cent, without effecting HDL levels. 35. Both products appear to have no adverse health effects and are nontoxic even in high doses, though a few people with the rare condition, phytosterolaemia cannot metabolise sterols and should avoid them.
8. According to Donnelly, there are two mechanisms by which these compounds are thought to lower cholesterol levels. 37. 'Choresterol is not very soluble in the gut and its absorption is slow. 38. Since you have other fats also being absorbed from the gut, cholesterol is one of the last to go through' she explained. 39. 'Plant sterols and stanols have similar structures to cholesterol so they also get left behind. 40. As the concentration of sterols/stanols increases, a threshold level is reached when the cholesterol molecules and the sterols/stanols coprecipitate into a solid crystalline form which cannot be absorbed by the gut' 42. There is limited capacity for carrying cholesterol, and the plant sterols and stanols compete with cholesterol to get into the micelles, which limits the amount of cholesterol that can be absorbed. 43. 'These mechanisms do not just reduce the absorption of dietary cholesterol', said Donnelly, 'but they also hinder reabsorption of some of the cholesterol produced by the body, which has been used in producing bile salts' 44 . Bile salts are used in the intestine to breakdown the fatty acids that we eat. 45. Normally, the cholesterol in the bile salts would be recycled by re-absorption in the gut, but in this case they are excreted. 46.

Essentially more of the cholesterol produced has to go in to producing more bile salts, reducing the amounts in the blood plasma.
47. Cholesterol-lowering spreads are some of the first functional foods on the market, but scientists are continually identifying ingredients that have potential health benefits. 48. As new advances in food technology allow their incorporation into products, we will see a lot more on the supermarket shelves. 49. Although these products can be beneficial, Donnelly says that she hopes 'people do not begin to rely on them because they are not miracle cures and there are many other factors involved in heart disease, which these products do not address.

## 2. 9. Texto 9: Apatite for destruction.

1. The industrial revolution of the 18th and 19th centuries brought great prosperity to the UK, but not without a price. 2. The environment Agency estimates that 300000 hectares of the UK is contaminated as a result of industrial pollution, for example cadmium and lead contamination associated with the iron, steel and paint industries. 3. Now with the increasing demand for housing, which places pressure on the countryside, the Government requires that 60 per cent of all new housing should be built on reclaimed sites. 4. Using current techniques of remediation - 'dig and dump’ and 'soil washing' - the cost of reclaiming this land is estimated at $£ 20$ billion. 5. However, scientists at the Natural History Museum believe they have found a costeffective solution to treating heavy metal pollution by using bone-meal. 6. Their method, presented by Dr Eugenia Valsami-Jones, at the BA festival of science, in London in September, involves 'immobilising' polluting metals as insoluble phosphates. 7.The work is sponsored by the BOC Foundation and the Environment Agency.
2. Bone-meal, widely used as a garden fertiliser, is sterilised, crushed animal bone comprising two main components. 9. There is an organic component, ie a fibrous protein (collagen) and an inorganic component, ie the crystalline mineral hydroxyapatite $\left(\mathrm{Ca}_{10}\left(\mathrm{PO}_{4}\right)_{6} \mathrm{OH}_{2}\right) . \mathbf{1 0}$. It is the hydroxyapatite, with phosphate ions locked in its crystal structure, that allows bone-meal to trap heavy metals. 11. The treatment of contaminated land with bone-meal is based on two reactions. 12. First, on mixing with soil, bone-meal dissolves in the pore/rain water, releasing phosphate ions from the crystal structure, along with calcium ions and some hydroxide ions. 13. Secondly, free phosphate ions react with the metal pollutant, forming insoluble metal phosphates. 14.

This reaction locks the polluting metal into a rigid mineral structure, thus acting as a 'micro barrier' between the pollutant and the environment. 15. Lab trials of bone-meal as a treatment for heavy metal pollution at the Natural History Museum using Scanning Electron Microscopy (SEM) confirm the formation of metal phosphate minerals with aluminium, copper, zinc, cadmium, nickel, lead and uranium.
16. In the short-term future the team hopes to demonstrate that the method will work at a variety of contaminated sites, thus establishing the long-term stability of the remediated metals. 17. 'In the future, we hope to see the method being used and contributing to the improvement of the lives of people affected by heavy metal pollution', said Dr Valsami-Jones

## 2. 9. 1. Matriz de repetición de unidades léxicas.



## Anexo

|  | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 15 |  | rs. pollution - pollution <br> rs. cadmium - cadmium <br> rs. lead - lead |  | rs. using - using | rs. Natural - Natural <br> rs. History - History <br> rs. Museum - Museum <br> rc. treating - treatment <br> rs. heavy - heavy <br> rs. metal - metal <br> rs. pollution - pollution <br> rs. using - using <br> rs. bone - bone <br> rs. meal - meal |
| 16 | hip. UK <br> - sites | hip. UK - sites <br> rs. contaminated - contaminated hip. cadmium and lead - metals | rs. sites sites | rc. remediation remediated psm. land - sites | hip. scientists - team <br> rs. metal - metals <br> pc. pollution contaminated (pollute) |
| 17 |  | rs. pollution - pollution hip. cadmium and lead - metal |  | rs. using - used | rs. heavy - heavy <br> rs. metal - metal <br> rs. pollution - pollution <br> rs. using - used |

6


## Anexo

6
7
8
9
10
11

| 16 | tr. their - team <br> rs. method - method <br> psm. polluting - <br> contaminated* <br> rs. metals - metals |  |  | rs. metals - metals | rs. contaminated - <br> contaminated <br> psm. land - sites |
| :---: | :--- | :--- | :--- | :--- | :--- |
| 17 | rs. method - method <br> rs. Valsami-Jones - <br> Valsami-Jones <br> rc. polluting - pollution <br> rs. metals - metal |  |  | rs. heavy - heavy <br> rs. metals - metal | pc. contaminated - <br> pollution* (pollute) |


| 13 | pc. releasing - free (freeing) rs. phosphate - phosphate rs. ions ions | 13 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14 | rs. structure structure | rs. react reaction rs. metal - metal rs. pollutant pollutant | 14 |  |  |
| 15 | rs. bone bone <br> rs. meal meal <br> rs. phosphate <br> - phosphate | rs. metal - metal rc. pollutant pollution rc. forming formation rs. metal - metal rs. phosphates phosphate | rc. polluting pollution rs. metal - metal rs. mineral minerals | 15 |  |
| 16 |  | rs. metal metals | psm. polluting contaminated* rs. metal metals | rs. metal - metals pc. pollution contaminated* (pollute) hip. SEM - method | 16 |
| 17 |  | rs. metal - metal rc. pollutant pollution | rc. polluting pollution <br> rs. metal - metal | rs. heavy - heavy <br> rs. metal - metal <br> rs. pollution - pollution <br> rs. using - used <br> hip. SEM - method | rs. future - future <br> rs. hopes - hope <br> rs. method - method <br> pc. contaminated - <br> pollution* (pollute) <br> rs. metals - metal |

2. 9. 2. Matriz con el número de unidades léxicas.

1. 9. 3. Tabla representativa del número de conexiones entre oraciones.
1. $(-, 0)[0]$
2. $(0,3)[3]$
3. $(0,0)[0]$
4. $(0,0)[0]$
5. $(0,6)[6]$
6. $(1,4)[5]$
7. $(0,0)$ [0]
8. $(0,0)[0]$
9. $(0,0)$ [0]
10. $(1,4)[5]$
11. $(1,1)[2]$
12. $(1,2)[3]$
13. $(3,2)[5]$
14. $(3,1)[4]$
15. $(8,2)$ [10]
16. $(3,1)[4](4,1)[5]$
17. (4,-) [4]

## 2. 9. 4. Texto resultante tras eliminar las oraciones marginales.

2. The environment Agency estimates that 300000 hectares of the UK is contaminated as a result of industrial pollution, for example cadmium and lead contamination associated with the iron, steel and paint industries. 5. However, scientists at the Natural History Museum believe they have found a cost-effective solution to treating heavy metal pollution by using bone-meal. 6. Their method, presented by Dr Eugenia Valsami-Jones, at the BA festival of science, in London in September, involves 'immobilising' polluting metals as insoluble phosphates.
3. It is the hydroxyapatite [( $\left.\mathrm{Ca}_{10}\left(\mathrm{PO}_{4}\right)_{6} \mathrm{OH}_{2}\right)$ ], with phosphate ions locked in its crystal structure, that allows bone-meal to trap heavy metals. 11. The treatment of contaminated land with bone-meal is based on two reactions. 12. First, on mixing with soil, bone-meal dissolves in the pore/rain water, releasing phosphate ions from the crystal structure, along with calcium ions and some hydroxide ions. 13. Secondly, free phosphate ions react with the metal pollutant, forming insoluble metal phosphates. 14. This reaction locks the polluting metal into a rigid mineral structure, thus acting as a 'micro barrier' between the pollutant and the environment. 15. Lab trials of bone-meal as a treatment for heavy metal pollution at the Natural History Museum using Scanning Electron Microscopy (SEM) confirm the formation of metal phosphate minerals with aluminium, copper, zinc, cadmium, nickel, lead and uranium.
4. In the short-term future the team hopes to demonstrate that the method will work at a variety of contaminated sites, thus establishing the long-term stability of the remediated metals. 17. 'In the future, we hope to see the method being used and contributing to the improvement of the lives of people affected by heavy metal pollution', said Dr Valsami-Jones

## 2. 10. Texto 10: Hair-raising ideas.

1. Hair could tell other people a lot more about you than you might want them to know. 2. Two new methods of hair analyses presented at the American Chemical Society meeting in Washington in August both use supercritical fluid technologies to identify the perpetrators of crime. 3. Typically, hair samples collected at crime scenes are inspected under microscope to determine colour, thickness and morphology (straightness). 4. But, without resorting to DNA analysis, this frequently gives a profile that is far from unique.
2. At the US National Institute for Standards and Technology, Bruce Benner has come up with an analytical technique based on supercritical fluid (SF) extraction combined with GC-MS that can provide a more reliable chemical hair profile. 6. By exploiting the powerful solubilising ability of $\mathrm{SFCO}_{2}$, Benner is able to strip away from the hair a much greater proportion of the surrounding lipids and other ingredients, including several hormones and other proteins. 7. Recent analyses of a variety of hair samples using the approach have revealed that the technique is highly reproducible, so
criminals won't simply be able to disguise themselves by changing the shampoo or conditioner they use.
3. The external composition of hair also depends on a variety of other factors, Benner says, including what you eat, your gender and ethnic type, as well as your general health and well-being. 9. In fact, looking at the general lipid composition of hair may even be a good way of detecting different illness, he adds.
4. Getting deeper inside the hair shaft can be even more revealing, according to Janet Morrison and Alison Rada at Trinity College, Conneticut. 11. Here, researchers are interested in looking for signs of drug abuse by sample provider - in particular to detect illicit use of amphetamines, which includes increasingly common drugs such as MDMA (Ecstacy). 12. Conventional procedures for detecting these drugs in blood and urine samples are notoriously time-consuming and involve a two step process that involves liquid-liquid or solid-phase extraction followed by lengthy derivatisation of the drugs to make analogues suitable for GC-MS analysis.13. Although even the $\mathrm{SFCO}_{2}$ used for this new extraction process is not powerful enough to dissolve the amphetamines directly, the researchers are able to speed up this process enormously by incorporating the derivatising reagents in this extraction solvent.
5. By performing both extraction and derivatisation in one step, the researchers are able to reduce the time needed to carry out this detective work from several days to just over an hour. 15. Morrison has already applied a similar technique for cocaine analyses in hair, but both methods will need to be validated by the courts before they can become routinely adopted by toxicologists. 16. Not only do they promise to catch culprits more quickly, but hair greatly expands the time window for drug detection compared with urine and blood. 17. Knowing that hair grows by 1 cm per month, it is possible to obtain an accurate date for when the abuse took place.

## Anexo

## 2. 10. 1. Matriz de repetición de unidades léxicas.

| 1 |  | 2 |  | 4 |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | rs. hair - hair |  |  |  |  |
| 3 | rs. hair - hair | rs. hair - hair psm. identify determine rs. crime - crime | 3 |  |  |
| 4 |  | rs. analyses - analysis | d. oración 3 this |  |  |
| 5 | rs. hair - hair | psm. methods - <br> technique <br> rs. hair - hair <br> rc. analyses - analytical <br> rs. supercritical - <br> supercritical <br> rs. fluid - fluid | rs. hair - hair | rc. analysis - <br> analytical <br> psm. gives - <br> provide <br> rs. profile - profile | 5 |
| 6 | rs. hair - hair | rs. hair - hair | rs. hair - hair |  | rs. Benner - Benner pc. extraction - strip away (extract) rs. hair - hair |
| 7 | rs. hair - hair | psm. methods - <br> technique <br> rs. hair - hair <br> rs. analyses - analyses <br> rs. use - using <br> rc. crime - criminals | rs. hair - hair <br> rs. samples - <br> samples <br> rc. crime - <br> criminals | rs. analysis analyses | rc. analytical - analyses <br> rs. technique - <br> technique <br> rs. hair - hair |
| 8 | rs. hair - hair <br> rs. you - you + | rs. hair - hair | rs. hair - hair |  | rs. Benner - Benner rs. hair - hair |
| 9 | rs. hair - hair | rs. hair - hair | rs. hair - hair psm. inspected - looking at |  | s. Benner - he rs. hair - hair |
| 10 | rs. hair - hair pc. tell revealing (telling) | rs. hair - hair | rs. hair - hair |  | rs. hair - hair |
| 11 |  |  | rs. samples sample |  |  |
| 12 |  | a. new - conventional psm. methods procedures rs. analyses - analysis | rs. samples samples* | rs. analysis analysis | rc. analytical - analysis <br> rs. extraction extraction <br> rs. GC-MS - GC-MS |
| 13 |  |  |  |  | rs. extraction extraction |
| 14 |  |  |  |  | rs. extraction extraction |


| 1 |  | 2 | 3 | 4 | 5 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 15 | rs. hair - hair | rs. methods - methods <br> rs. hair - hair <br> rs. analyses - analyses <br> rc. technologies - <br> technique | rs. hair - hair | rs. analysis - <br> analyses | rc. analytical - analyses <br> rs. technique - technique <br> rs. hair - hair |
| 16 | rs. hair - hair | s. methods - they <br> rs. hair - hair <br> tr. crime - culprits | rs. hair - hair |  | rs. hair - hair |
| 17 | rs. hair - hair | rs. hair - hair | rs. hair - hair |  | rs. hair - hair |

## Anexo

| 6 |  | 7 | 8 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 7 | hip. $\mathrm{SFCO}_{2}-$ technique rs. hair - hair |  |  |  |  |  |
| 8 | rs. Benner Benner rs. hair - hair | rs. hair - hair |  |  |  |  |
| 9 | s. Benner - he rs. hair - hair rs. lipids lipid | rs. hair - hair | rs. <br> composition composition rs. hair - hair s. Benner - he psm. says adds | 9 |  |  |
| 10 | rs. hair - hair | rs. hair - hair rc. revealed revealing | rs. hair - hair | rs. hair hair | 10 |  |
| 11 |  | rs. samples sample |  | rs. detecting <br> - detect | co-ref. Morrison and Rada - researchers | 11 |
| 12 | pc strip away - extraction (extract) |  |  | rs. detecting <br> - detecting |  | rs. sample sample <br> rs. detect detecting <br> rs. drugs - drugs |
| 13 | rs. powerful powerful <br> rs. $\mathrm{SFCO}_{2}-$ $\mathrm{SFCO}_{2}$ psm. strip away dissolve |  |  |  | co-ref. Morrison and Rada - researchers | rs. researchers researchers rs. amphetamines amphetamines |
| 14 | pc strip away - extraction (extract) |  |  | rc. detecting <br> - detective | co-ref. Morrison and Rada - researchers | rs. researchers researchers rc. detect detective |
| 15 | hip. $\mathrm{SFCO}_{2}-$ technique rs. hair - hair | rs. analyses analyses rs. hair - hair rs. technique technique | rs. hair - hair | rs. hair hair | rs. Morrison Morrison rs. hair - hair | tr. drugs cocaine |
| 16 | rs. hair - hair | rs. hair - hair hip. criminals - culprits | rs. hair - hair | rs. hair hair <br> rc. detecting <br> - detection | rs. hair - hair | rs. drug - drug rs. detect detection |
| 17 | rs. hair - hair | rs. hair - hair | rs. hair - hair | rs. hair hair | rs. hair - hair | e. drug -0 <br> rs. abuse - abuse |

12

2. 10. 2. Matriz con el número de unidades léxicas.

| 2 | 1 | 2 |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3 | 1 | 3 | 3 |  |  |  |  |  |  |  |  |
| 4 | 0 | 1 | 1 | 4 |  |  |  |  |  |  |  |
| 5 | 1 | 5 | 1 | 3 | 5 |  |  |  |  |  |  |
| 6 | 1 | 1 | 1 | 0 | 3 |  |  |  |  |  |  |
| 7 | 1 | 5 | 3 | 1 | 3 | 2 | 7 |  |  |  |  |
| 8 | 1(2) | 1 | 1 | 0 | 2 | 2 | 1 | 8 |  |  |  |
| 9 | 1 | 1 | 2 | 0 | 2 | 3 | 1 | 4 |  |  |  |
| 10 | 2 | 1 | 1 | 0 | 1 | 1 | 2 | 1 | 1 | 10 |  |
| 11 | 0 | 0 | 1 | 0 | 0 | 0 | 1 | 0 | 1 | 1 | 11 |
| 12 | 0 | 3 | $0(1)$ | 1 | 3 | 1 | 0 | 0 | 1 | 0 | 3 |
| 13 | 0 | 0 | 0 | 0 | 1 | 3 | 0 | 0 | 0 | 1 | 2 |
| 14 | 0 | 0 | 0 | 0 | 1 | 1 | 0 | 0 | 1 | 1 | 2 |
| 15 | 1 | 4 | 1 | 1 | 3 | 2 | 3 | 1 | 1 | 2 | 1 |
| 16 | 1 | 3 | 1 | 0 | 1 | 1 | 2 | 1 | 2 | 1 | 2 |
| 17 | 1 | 1 | 1 | 0 | 1 | 1 | 1 | 1 | 1 | 1 | 2 |


2. 10. 3. Tabla representativa del número de conexiones entre oraciones.

| 1. $(-, 0)[0]$ | 2. $(0,6)[6]$ | 3. $(1,1)[2]$ |
| :--- | :--- | :--- |
| 4. $(0,1)[1]$ | 5. $(2,4)[6]$ | 6. $(1,2)[3]$ |
| 7. $(3,1)[4]$ | 8. $(0,1)[1]$ | 9. $(2,0)[2]$ |
| 10. $(0,0)[0]$ | 11. $(0,1)[1]$ | 12. $(3,3)[6]$ |
| 13. $(2,1)[3]$ | 14. $(2,0)[2]$ | 15. $(3,1)[4]$ |
| 16. $(3,0)[3]$ | 17. $(0,-)[0]$ |  |

## 2. 10. 4. Texto resultante tras eliminar las oraciones marginales.

2. Two new methods of hair analyses presented at the American Chemical Society meeting in Washington in August both use supercritical fluid technologies to identify the perpetrators of crime. 3. Typically, hair samples collected at crime scenes are inspected under microscope to determine colour, thickness and morphology (straightness). 4. But, without resorting to DNA analysis, this frequently gives a profile that is far from unique.
3. At the US National Institute for Standards and Technology, Bruce Benner has come up with an analytical technique based on supercritical fluid (SF) extraction combined with GC-MS that can provide a more reliable chemical hair profile. 6. By exploiting the powerful solubilising ability of $\mathrm{SFCO}_{2}$, Benner is able to strip away from the hair a much greater proportion of the surrounding lipids and other ingredients, including several hormones and other proteins. 7. Recent analyses of a variety of hair samples using the approach have revealed that the technique is highly reproducible, so criminals won't simply be able to disguise themselves by changing the shampoo or conditioner they use.
4. The external composition of hair also depends on a variety of other factors, Benner says, including what you eat, your gender and ethnic type, as well as your general health and well-being. 9. In fact, looking at the general lipid composition of hair may even be a good way of detecting different illness, he adds.
5. Here, researchers are interested in looking for signs of drug abuse by sample provider - in particular to detect illicit use of amphetamines, which includes increasingly common drugs such as MDMA (Ecstacy). 12. Conventional procedures for detecting these drugs in blood and urine samples are notoriously time-consuming and involve a two step process that involves liquid-liquid or solid-phase extraction followed by lengthy derivatisation of the drugs to make analogues suitable for GC-MS analysis.13. Although even the $\mathrm{SFCO}_{2}$ used for this new extraction process is not powerful enough to dissolve the amphetamines directly, the researchers are able to speed up this process enormously by incorporating the derivatising reagents in this extraction solvent.
6. By performing both extraction and derivatisation in one step, the researchers are able to reduce the time needed to carry out this detective work from several days to just over an hour. 15. Morrison has already applied a similar technique for cocaine analyses in hair, but both methods will need to be validated by the courts before they can become routinely adopted by toxicologists. 16. Not only do they promise to catch culprits more quickly, but hair greatly expands the time window for drug detection compared with urine and blood.
7. LISTADO DE UNIDADES LÉXICAS QUE HAN ESTABLECIDO REPETICIÓN.

| UNIDAD LÉXICA | IAI | AAI | UNIDAD LÉXICA | IAI | AAI |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $4^{\text {TH }}$ |  | $\bullet$ | ANALYZE | $\bullet$ |  |
| ABILITY |  | $\bullet$ | ANALYZERS | $\bullet$ |  |
| ABLE |  | $\bullet$ | ANALYZING | $\bullet$ |  |
| ABSORBED |  | $\bullet$ | ANIMAL |  | $\bullet$ |
| ABSORBS |  | $\bullet$ | APPLICATIONS |  | $\bullet$ |
| ABSORPTION |  | $\bullet$ | APPLIED | $\bullet$ |  |
| ABUSE |  | $\bullet$ | APPLY |  | $\bullet$ |
| ACCEPTABILITY |  | $\bullet$ | AQUA |  | $\bullet$ |
| ACID |  | $\bullet$ | AQUEOUS | $\bullet$ |  |
| ACIDS |  | $\bullet$ | ARTERY |  | $\bullet$ |
| ADD |  | $\bullet$ | ASSOCIATED |  | $\bullet$ |
| ADDED | $\bullet$ | $\bullet$ | AT THE SAME TIME |  | $\bullet$ |
| ADDING |  | $\bullet$ | ATMOSPHERE | $\bullet$ |  |
| ADDS |  | $\bullet$ | ATMOSPHERIC | $\bullet$ |  |
| ADEQUATE | $\bullet$ | $\bullet$ | ATOMIC |  | $\bullet$ |
| ADOPTED | $\bullet$ |  | ATOMS |  | $\bullet$ |
| ADOPTION | $\bullet$ |  | ATTACHED |  | - |
| ADSORBED | $\bullet$ |  | ATTACK | $\bullet$ |  |
| ADULT |  | $\bullet$ | ATTACKS | $\bullet$ |  |
| ADVERSE | $\bullet$ | $\bullet$ | ATTRIBUTES |  | $\bullet$ |
| AEROSOLS | $\bullet$ |  | AU |  | $\bullet$ |
| AFFECT |  | $\bullet$ | AUTOMATED | $\bullet$ |  |
| AG |  | $\bullet$ | BAKERY | $\bullet$ |  |
| AGENCY |  | $\bullet$ | BALL |  | $\bullet$ |
| AGNOSTIC |  | $\bullet$ | BALLOON | - |  |
| AGNOSTICS |  | $\bullet$ | BALLS |  | $\bullet$ |
| AIR | $\bullet$ | $\bullet$ | BAR |  | $\bullet$ |
| AIRCRAFT |  | $\bullet$ | BARS |  | $\bullet$ |
| ALCOHOL | $\bullet$ |  | BASE |  | $\bullet$ |
| ALCOHOLIC | $\bullet$ |  | BASED | $\bullet$ |  |
| ALL | $\bullet$ |  | BATCH | $\bullet$ |  |
| ALLEGORY |  | $\bullet$ | BEAM |  | $\bullet$ |
| ALLOW |  | $\bullet$ | BECOME |  | $\bullet$ |
| ALLOWING |  | $\bullet$ | BECOMES |  | $\bullet$ |
| ALLOY |  | $\bullet$ | BEER |  | $\bullet$ |
| ALLOYS |  | $\bullet$ | BEGAN |  | $\bullet$ |
| ALTERNATIVE | $\bullet$ |  | BEGUN |  | $\bullet$ |
| ALTERNATIVES | $\bullet$ |  | BEHAVIOR | $\bullet$ |  |
| AMAZING |  | $\bullet$ | BENECOL |  | $\bullet$ |
| AMAZINGLY |  | $\bullet$ | BENEFICIAL |  | $\bullet$ |
| AMOUNT |  | $\bullet$ | BENEFITS |  | $\bullet$ |
| AMOUNTS | - | $\bullet$ | BENZENE |  | $\bullet$ |
| AMPHETAMINES |  | $\bullet$ | BIGGER |  | $\bullet$ |
| ANALOGY |  | $\bullet$ | BILE |  | $\bullet$ |
| ANALYSERS | - |  | BIMOLECULAR |  | $\bullet$ |
| ANALYSES |  | $\bullet$ | BINDING | $\bullet$ |  |
| ANALYSING | $\bullet$ |  | BIOSPHERE | $\bullet$ |  |
| ANALYSIS | $\bullet$ | - | BIRTH | $\bullet$ |  |
| ANALYSTS | $\bullet$ |  | BLAST |  | $\bullet$ |
| ANALYTICAL | $\bullet$ | $\bullet$ | BLASTING |  | $\bullet$ |


| BLOOD | $\bullet$ | - | CHEMICALLY |  | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| BODY |  | $\bullet$ | CHEMICALS |  | $\bullet$ |
| BOND |  | $\bullet$ | CHEMIST |  | $\bullet$ |
| BONDS |  | $\bullet$ | CHEMISTRY |  | $\bullet$ |
| BONE |  | $\bullet$ | CHEMISTS |  | $\bullet$ |
| BORE | $\bullet$ |  | CHOLESTEROL |  | $\bullet$ |
| BOROHYDRIDE | $\bullet$ |  | CHROMATOGRAPHIC | $\bullet$ | $\bullet$ |
| BOTTOM |  | - | CHROMATOGRAPHY | $\bullet$ |  |
| BOYFRIEND |  | $\bullet$ | CHURCH |  | $\bullet$ |
| BREAK |  | $\bullet$ | CHURCHES |  | $\bullet$ |
| BREAST | $\bullet$ |  | CIGARETTE | $\bullet$ |  |
| BRIEF |  | $\bullet$ | CITY |  | $\bullet$ |
| BRITAIN |  | $\bullet$ | CLAIMED |  | $\bullet$ |
| BRITISH |  | $\bullet$ | CLAIMING |  | $\bullet$ |
| BURN |  | $\bullet$ | CLASSICAL | $\bullet$ |  |
| BURNED |  | $\bullet$ | CLASSIFICATION |  | $\bullet$ |
| BURNING |  | $\bullet$ | CLASSIFIED |  | $\bullet$ |
| BURNS |  | $\bullet$ | CLEAN | $\bullet$ |  |
| BURST |  | $\bullet$ | CLOSE |  | $\bullet$ |
| BURSTS |  | $\bullet$ | $\mathrm{CO}_{2}$ | $\bullet$ |  |
| BUTTER |  | $\bullet$ | COCAINE |  | $\bullet$ |
| BUTTERS |  | - | COFERMENTATION | $\bullet$ |  |
| C | $\bullet$ |  | COLLEAGUES |  | $\bullet$ |
| CA |  | - | COLLECTING | $\bullet$ |  |
| CADMIUM |  | $\bullet$ | COLLECTION | $\bullet$ |  |
| CAFFEINE |  | $\bullet$ | COLOR |  | $\bullet$ |
| CAN |  | $\bullet$ | COLORED |  | $\bullet$ |
| CANNOT |  | - | COLORLESS |  | $\bullet$ |
| CAPILLARY | $\bullet$ |  | COLORS |  | $\bullet$ |
| CARAT |  | $\bullet$ | COLOURS | $\bullet$ |  |
| CARBON | $\bullet$ | $\bullet$ | COLUMN | $\bullet$ |  |
| CARBONACEOUS | $\bullet$ |  | COLUMNS | $\bullet$ |  |
| CARBONYL |  | $\bullet$ | COMBUSTIBLE |  | $\bullet$ |
| CARBONYLS |  | $\bullet$ | COME |  | $\bullet$ |
| CARCINOGEN | $\bullet$ |  | COMMENTED |  | $\bullet$ |
| CARCINOGENS | $\bullet$ |  | COMPARISON | $\bullet$ |  |
| CARDIO |  | $\bullet$ | COMPLETE |  | $\bullet$ |
| CARRIED |  | $\bullet$ | COMPONENT |  | $\bullet$ |
| CARRYING |  | $\bullet$ | COMPONENTS |  | $\bullet$ |
| CATALYSE |  | $\bullet$ | COMPOSED |  | $\bullet$ |
| CATALYST |  | $\bullet$ | COMPOSITION | $\bullet$ | $\bullet$ |
| CAUGHT |  | - | COMPOSITIONS |  | $\bullet$ |
| CELL | $\bullet$ | $\bullet$ | COMPOUND | $\bullet$ | $\bullet$ |
| CELLS | $\bullet$ | $\bullet$ | COMPOUNDS | $\bullet$ | $\bullet$ |
| CHAIN |  | $\bullet$ | CONCENTRATION | $\bullet$ |  |
| CHANGE | $\bullet$ | $\bullet$ | CONCENTRATIONS | $\bullet$ |  |
| CHANGES | $\bullet$ | $\bullet$ | CONCERN | $\bullet$ |  |
| CHARACTERISTIC |  | $\bullet$ | CONDITIONS | $\bullet$ |  |
| CHARACTERIZATION | $\bullet$ |  | CONDUCTED | $\bullet$ |  |
| CHARACTERIZE | $\bullet$ |  | CONDUCTIVITY | $\bullet$ |  |
| CHARGE |  | - | CONDUCTOMETRIC | $\bullet$ |  |
| CHEESE | $\bullet$ |  | CONFIGURATION |  | $\bullet$ |
| CHEM. |  | $\bullet$ | CONFIGURATIONS |  | $\bullet$ |
| CHEMICAL | $\bullet$ | $\bullet$ | CONFORMATION | $\bullet$ |  |


| CONSIDERABLY | - |  | DESCRIBE | - |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| CONSIDERED |  | $\bullet$ | DESCRIBED | $\bullet$ |  |
| CONSIST |  | - | DESIGNATED |  | - |
| CONSTITUENTS |  | - | DESIGNATION |  | - |
| CONSTRAINS | - |  | DESIGNED | $\bullet$ |  |
| CONSUMERS |  | $\bullet$ | DESIGNS | - |  |
| CONTACT | - |  | DESTROY |  | - |
| CONTAIN | - | - | DETAILED | - |  |
| CONTAINING | $\bullet$ |  | DETAILS | - |  |
| CONTAMINATED |  | $\bullet$ | DETECT |  | - |
| CONTENT | - | $\bullet$ | DETECTING |  | $\bullet$ |
| CONTINUED |  | - | DETECTION |  | - |
| CONTINUOUS | - |  | DETECTIVE |  | - |
| CONTINUOUSLY |  | - | DETERMINAND | - |  |
| CONVENTIONAL | - | - | DETERMINATION | - |  |
| COOLING | - |  | DETERMINE | - | - |
| COPPER | $\bullet$ | - | DETERMINED | - | - |
| CORN | - |  | DETERMINING | - |  |
| COST |  | $\bullet$ | DEVELOP | $\bullet$ | - |
| COULD |  | $\bullet$ | DEVELOPED | $\bullet$ |  |
| COULOMETRIC | $\bullet$ |  | DEVELOPMENT | - |  |
| COW | $\bullet$ |  | DEVICE | $\bullet$ |  |
| CRASH |  | - | DIALYSATE | - |  |
| CRASHES |  | $\bullet$ | DIALYSATES | - |  |
| CRIME |  | $\bullet$ | DIALYSIS | - |  |
| CRIMINALS |  | $\bullet$ | DIESEL | $\bullet$ |  |
| CRITERIA | - |  | DIET |  | $\bullet$ |
| CRYSTAL |  | - | DIETARY | - | - |
| CRYSTALLINE |  | $\bullet$ | DIFFERENCE |  | $\bullet$ |
| CU |  | $\bullet$ | DIFFERENT | - | $\bullet$ |
| CULPRITS |  | - | DIHYDROGEN |  | - |
| CULTURE | - |  | DIOXIDE |  | - |
| CURIOSITY |  | $\bullet$ | DISADVANTAGES | - |  |
| CURIOUS |  | - | DISCOVERED |  | $\bullet$ |
| CUTTING DOWN |  | $\bullet$ | DISCOVERY |  | $\bullet$ |
| CYANIDE | - | $\bullet$ | DISCUSSION | - |  |
| CYANO |  | $\bullet$ | DISEASE | - | - |
| CYT C/BGH | $\bullet$ |  | DISPOSED OF |  | $\bullet$ |
| D |  | - | DISSOCIATION |  | $\bullet$ |
| DATA | - |  | DISSOLUTION |  | $\bullet$ |
| DAY |  | - | DISSOLVE |  | $\bullet$ |
| DEATH | - |  | DISSOLVED |  | - |
| DECOMPOSE |  | $\bullet$ | DISSOLVES |  | $\bullet$ |
| DECOMPOSES |  | $\bullet$ | DISTANCE |  | - |
| DECOMPOSITION |  | $\bullet$ | DISTRIBUTION |  | - |
| DECREASE |  | - | DMS | - |  |
| DEFINED | $\bullet$ |  | DONE |  | - |
| DEFORESTAT | $\bullet$ |  | DRINK |  | $\bullet$ |
| DELETERIOUS | $\bullet$ |  | DRINKING |  | $\bullet$ |
| DENSITIES |  | $\bullet$ | DRUG |  | - |
| DENSITY |  | - | DRUGS |  | - |
| DERIVATISING |  | $\bullet$ | DUAL | - |  |
| DERIVATIZATION | - | $\bullet$ | DYE | - |  |
| DERIVATIZATIONS | - |  | DYES | - |  |


| EARRINGS |  | $\bullet$ | FED | $\bullet$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| EASIER |  | $\bullet$ | FEMTOCHEMISTRY |  | $\bullet$ |
| EAT |  | $\bullet$ | FEMTOSECOND |  | $\bullet$ |
| EATEN |  | $\bullet$ | FEMTOSECONDS |  | $\bullet$ |
| EC | $\bullet$ |  | FERMENT | $\bullet$ |  |
| EFFECT |  | - | FERMENTATING | $\bullet$ |  |
| EFFECTIVE |  | $\bullet$ | FERMENTATION | $\bullet$ |  |
| EFFECTS | $\bullet$ | - | FERMENTED | - |  |
| EFFICIENCY | $\bullet$ |  | FERMENTING | $\bullet$ |  |
| EFFICIENT | $\bullet$ |  | FEW |  | $\bullet$ |
| ELECTRON |  | $\bullet$ | FIBERS |  | $\bullet$ |
| ELECTRONIC |  | $\bullet$ | FIELD | $\bullet$ |  |
| ELECTRONS |  | $\bullet$ | FIGURE |  | $\bullet$ |
| ELEMENTAL | $\bullet$ |  | FINISH |  | $\bullet$ |
| ELEMENTS | - | $\bullet$ | FIRE |  | $\bullet$ |
| ELEVATED |  | $\bullet$ | FIRST | $\bullet$ | $\bullet$ |
| EMPLOYING | $\bullet$ |  | FLAME |  | $\bullet$ |
| END |  | - | FLAMELESS |  | $\bullet$ |
| ENERGETIC |  | $\bullet$ | FLAMMABILITY |  | $\bullet$ |
| ENERGIES |  | $\bullet$ | FLAMMABLE |  | $\bullet$ |
| ENERGY |  | $\bullet$ | FLASH |  | $\bullet$ |
| ENGINE | $\bullet$ |  | FLASHES |  | $\bullet$ |
| ENOUGH |  | - | FLASHLAMP |  | $\bullet$ |
| ENRICHMENT | $\bullet$ |  | FLASK | $\bullet$ |  |
| ENVIRONMENT |  | - | FLASKS | $\bullet$ |  |
| ENVIRONMENTAL | $\bullet$ |  | FLAVOUR |  | $\bullet$ |
| ERROR |  | - | FLEETING |  | $\bullet$ |
| ESTABLISHED | $\bullet$ |  | FLEETINGLY |  | $\bullet$ |
| ESTIMATED |  | $\bullet$ | FLORA |  | $\bullet$ |
| ESTIMATES |  | $\bullet$ | FLUID |  | $\bullet$ |
| ETHANOL | $\bullet$ |  | FLUIDS | $\bullet$ | $\bullet$ |
| EXAMINE | $\bullet$ |  | FLUORESCENCE | $\bullet$ |  |
| EXAMINED | $\bullet$ |  | FLUORINE |  | $\bullet$ |
| EXCESSIVE | $\bullet$ |  | FLUOROPOLYMER |  | $\bullet$ |
| EXEMPLARY |  | - | FLUX | $\bullet$ |  |
| EXERCISE |  | $\bullet$ | FLUXES | $\bullet$ |  |
| EXHAUST | $\bullet$ |  | FOLKS |  | $\bullet$ |
| EXISTS |  | $\bullet$ | FOOD | $\bullet$ | $\bullet$ |
| EXPERIMENT |  | $\bullet$ | FOODS | $\bullet$ | $\bullet$ |
| EXPERIMENTS |  | $\bullet$ | FORM |  | $\bullet$ |
| EXPLAINED |  | $\bullet$ | FORMATION |  | $\bullet$ |
| EXPOSED |  | - | FORMED |  | $\bullet$ |
| EXPOSURE | $\bullet$ | $\bullet$ | FORMING |  | $\bullet$ |
| EXTERNAL |  | $\bullet$ | FORMS | $\bullet$ | $\bullet$ |
| EXTRACT |  | $\bullet$ | FORMULA | $\bullet$ |  |
| EXTRACTED |  | $\bullet$ | FOSSIL | $\bullet$ |  |
| EXTRACTION | $\bullet$ | $\bullet$ | FOUND | $\bullet$ | $\bullet$ |
| EYE |  | $\bullet$ | FOURTH |  | $\bullet$ |
| FAA |  | $\bullet$ | FRACTION |  | $\bullet$ |
| FAST |  | $\bullet$ | FRECUENCY |  | $\bullet$ |
| FAT |  | $\bullet$ | FREE |  | $\bullet$ |
| FATITUTES |  | $\bullet$ | FREEZE |  | $\bullet$ |
| FATS |  | $\bullet$ | FREEZES |  | $\bullet$ |
| FATTY |  | $\bullet$ | FROZEN |  | $\bullet$ |


| FS |  | $\bullet$ | HUMAN | $\bullet$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| FUEL | $\bullet$ |  | HUMANS | $\bullet$ |  |
| FUELS | $\bullet$ |  | HYDROCHLORIC |  | $\bullet$ |
| FULL-TERM | $\bullet$ |  | HYDROGEN |  | $\bullet$ |
| FUNCTIONAL |  | $\bullet$ | HYDROXYAPATITE |  | $\bullet$ |
| FUTURE |  | $\bullet$ | IDEA |  | $\bullet$ |
| GAS |  | $\bullet$ | IDENTICAL |  | $\bullet$ |
| GASEOUS |  | $\bullet$ | IDENTIFY | $\bullet$ | $\bullet$ |
| GASES |  | $\bullet$ | IDENTITIES | $\bullet$ |  |
| GC-GC | $\bullet$ |  | IDENTITY | $\bullet$ |  |
| GC-MS |  | $\bullet$ | IGNITED |  | $\bullet$ |
| GIFT |  | $\bullet$ | ILLUSTRATE |  | $\bullet$ |
| GIVE | - | $\bullet$ | IMPACT |  | $\bullet$ |
| GIVE OFF |  | $\bullet$ | IMPART | $\bullet$ |  |
| GIVES |  | $\bullet$ | IMPEDED | $\bullet$ |  |
| GIVING | $\bullet$ |  | IMPURITIES |  | $\bullet$ |
| GLUCOSE | $\bullet$ |  | IN SITU | $\bullet$ |  |
| GOLD |  | $\bullet$ | IN VITRO | $\bullet$ |  |
| GREEN |  | $\bullet$ | INCINERATION |  | $\bullet$ |
| GREENHOUSE GASES | - |  | INCLUDE | $\bullet$ |  |
| GROUP |  | $\bullet$ | INCLUDING | $\bullet$ |  |
| GROUPS | $\bullet$ |  | INCORPORATION |  | $\bullet$ |
| GROWTH | $\bullet$ |  | INCREASE | $\bullet$ | $\bullet$ |
| GUT |  | $\bullet$ | INCREASED |  | $\bullet$ |
| HAD |  | $\bullet$ | INDUSTRIAL |  | $\bullet$ |
| HAIR |  | $\bullet$ | INDUSTRIES | $\bullet$ |  |
| HALLMARKS |  | $\bullet$ | INDUSTRY | $\bullet$ |  |
| HAND |  | $\bullet$ | INETERCOMPARISON | $\bullet$ |  |
| HANDLE |  | $\bullet$ | INFANCY | $\bullet$ |  |
| HAPPEN |  | $\bullet$ | INFANT | $\bullet$ |  |
| HAPPENING |  | $\bullet$ | INGREDIENTS |  | $\bullet$ |
| HAPPENS |  | $\bullet$ | INJECTED | $\bullet$ |  |
| HAS |  | $\bullet$ | INOCULUM | $\bullet$ |  |
| HAVE |  | $\bullet$ | INSOLUBLE |  | $\bullet$ |
| HDL |  | $\bullet$ | INSPECTED |  | $\bullet$ |
| HDLS |  | $\bullet$ | INSPIRATION |  | $\bullet$ |
| HEALTH | $\bullet$ | $\bullet$ | INSTANT |  | $\bullet$ |
| HEALTHY |  | $\bullet$ | INSTRUMENT | $\bullet$ |  |
| HEART |  | $\bullet$ | INSTRUMENTATION | $\bullet$ |  |
| HEAT |  | $\bullet$ | INSTRUMENTS | $\bullet$ |  |
| HEATED |  | $\bullet$ | INTERACT | $\bullet$ |  |
| HEATING |  | $\bullet$ | INTERACTING |  | $\bullet$ |
| HEATS |  | $\bullet$ | INTERACTION | $\bullet$ | $\bullet$ |
| HEAVY |  | $\bullet$ | INTERACTIONS | $\bullet$ | $\bullet$ |
| HELP |  | $\bullet$ | INTERCOMPARISON | $\bullet$ |  |
| HIGH | $\bullet$ | $\bullet$ | INTERLABORATORY | $\bullet$ |  |
| HIGHER | $\bullet$ | $\bullet$ | INTESTINE |  | $\bullet$ |
| HINDER |  | $\bullet$ | INTRODUCED | $\bullet$ |  |
| HIPOCRITES |  | $\bullet$ | INVESTIGATE | $\bullet$ |  |
| HISTORY |  | $\bullet$ | INVESTIGATED | $\bullet$ |  |
| HOPE |  | $\bullet$ | INVESTIGATING | $\bullet$ |  |
| HOPES |  | $\bullet$ | INVESTIGATION | $\bullet$ |  |
| HPLC | $\bullet$ |  | INVOLVE |  | $\bullet$ |
| HS | $\bullet$ |  | INVOLVED |  | $\bullet$ |


| INVOLVEMENT |  | $\bullet$ | LOCKS |  | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| IODINE |  | $\bullet$ | LOOK AT |  | $\bullet$ |
| ION |  | $\bullet$ | LOOKING AT |  | $\bullet$ |
| IONIZATION |  | $\bullet$ | LOSING |  | $\bullet$ |
| IONS |  | $\bullet$ | LOW | $\bullet$ | $\bullet$ |
| IRON | $\bullet$ |  | LOWER | $\bullet$ | $\bullet$ |
| IS OVER |  | $\bullet$ | LOWERING |  | $\bullet$ |
| ISCHAEMIA |  | $\bullet$ | LOWEST |  | $\bullet$ |
| ISOLATION | $\bullet$ |  | MADE |  | $\bullet$ |
| ITEM |  | $\bullet$ | MADE UP OF |  | $\bullet$ |
| ITEMS |  | $\bullet$ | MAIN |  | $\bullet$ |
| JEWELRY |  | $\bullet$ | MAKE |  | $\bullet$ |
| JULY |  | $\bullet$ | MAKING |  | $\bullet$ |
| K |  | $\bullet$ | MANAGEMENT | $\bullet$ |  |
| KEEP |  | $\bullet$ | MANGANESE | $\bullet$ |  |
| KITE | $\bullet$ |  | MANUFACTURE |  | $\bullet$ |
| KNOWLEDGE | $\bullet$ |  | MANUFACTURED |  | $\bullet$ |
| KNOWN | $\bullet$ |  | MARGARINES |  | $\bullet$ |
| LABORATORIES | $\bullet$ |  | MARKED |  | $\bullet$ |
| LACTOSE | $\bullet$ |  | MARKET |  | $\bullet$ |
| LAND |  | $\bullet$ | MARKETED |  | $\bullet$ |
| LARGE | $\bullet$ | $\bullet$ | MARKING |  | $\bullet$ |
| LARGER | $\bullet$ | $\bullet$ | MASH | $\bullet$ |  |
| LASER |  | $\bullet$ | MASS | $\bullet$ |  |
| LASERS |  | $\bullet$ | MASSACHUSETTS |  | $\bullet$ |
| LASTED |  | $\bullet$ | MATERIAL | $\bullet$ |  |
| LASTING |  | $\bullet$ | MATERIALS |  | $\bullet$ |
| LASTS |  | $\bullet$ | MATERNAL | - |  |
| LDLS |  | $\bullet$ | MEAL |  | $\bullet$ |
| LEAD |  | $\bullet$ | MEAN | $\bullet$ | $\bullet$ |
| LEAVE |  | $\bullet$ | MEASURE | $\bullet$ |  |
| LEAVING |  | $\bullet$ | MEASURED | $\bullet$ |  |
| LENGTHY |  | $\bullet$ | MEASUREMENT | $\bullet$ |  |
| LEVEL | $\bullet$ | $\bullet$ | MEASUREMENTS | $\bullet$ |  |
| LEVELS | $\bullet$ | $\bullet$ | MEASURING | $\bullet$ |  |
| LIFE | $\bullet$ |  | MECHANISMS |  | $\bullet$ |
| LIGHT |  | $\bullet$ | MEDICINE |  | $\bullet$ |
| LIGHTWEIGHT | $\bullet$ |  | MEDICINES |  | $\bullet$ |
| LIKELY |  | $\bullet$ | MEDIUM | $\bullet$ |  |
| LIMITATION | $\bullet$ |  | MELT |  | $\bullet$ |
| LIMITATIONS | $\bullet$ |  | MEMBRANES | $\bullet$ |  |
| LIMITED | $\bullet$ |  | METABOLISM | $\bullet$ |  |
| LIMITS |  | $\bullet$ | METAL |  | $\bullet$ |
| LINE | $\bullet$ | $\bullet$ | METALLOPROTEINS | $\bullet$ |  |
| LINES | $\bullet$ |  | METALS |  | $\bullet$ |
| LIPID |  | $\bullet$ | METHOD | $\bullet$ | $\bullet$ |
| LIPIDS |  | $\bullet$ | METHODS | $\bullet$ | $\bullet$ |
| LIPOPROTEINS |  | $\bullet$ | MG |  | $\bullet$ |
| LIQUEFY |  | $\bullet$ | MICELLES |  | $\bullet$ |
| LIQUID |  | $\bullet$ | MICROCONSTITUENTS | $\bullet$ |  |
| LIQUIDS |  | $\bullet$ | MICRONUTRIENTS | $\bullet$ |  |
| LIVER |  | $\bullet$ | MICROSECOND |  | $\bullet$ |
| LOCATION | $\bullet$ |  | MICROSENSOR | $\bullet$ |  |
| LOCKED |  | $\bullet$ | MICROSENSORS | $\bullet$ |  |


| MILK | $\bullet$ |  | OBSERVATION |  | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| MILLI |  | $\bullet$ | OBSERVE |  | $\bullet$ |
| MILLISECOND |  | $\bullet$ | OBSERVED | $\bullet$ | $\bullet$ |
| MINERAL |  | $\bullet$ | OBSERVING |  | $\bullet$ |
| MINERALS |  | $\bullet$ | OBTAINED | $\bullet$ | $\bullet$ |
| MINIMUM |  | $\bullet$ | OC | $\bullet$ |  |
| MIXING | $\bullet$ |  | OCEANS | $\bullet$ |  |
| MIXTURES |  | - | OFF-LINE | $\bullet$ |  |
| ML | $\bullet$ |  | OFFSITE | $\bullet$ |  |
| MODELS | $\bullet$ |  | OIL |  | $\bullet$ |
| MODEST |  | - | OILY |  | $\bullet$ |
| MODIFIED | $\bullet$ |  | OLIGOSACCHARIDES | $\bullet$ |  |
| MOLE |  | - | ON-LINE | $\bullet$ |  |
| MOLECULAR | $\bullet$ | $\bullet$ | OPERATED | $\bullet$ |  |
| MOLECULE |  | $\bullet$ | OPERATIONAL | $\bullet$ |  |
| MOLECULES | $\bullet$ | $\bullet$ | OPTICAL | $\bullet$ |  |
| MOLTEN |  | $\bullet$ | ORGANIC | $\bullet$ | $\bullet$ |
| MONITORING | $\bullet$ |  | ORIENTATION | $\bullet$ |  |
| MONTHS | $\bullet$ |  | OUTCOME |  | $\bullet$ |
| MOTHER | $\bullet$ |  | OVEN | $\bullet$ |  |
| MUSEUM |  | - | OVERLAP |  | $\bullet$ |
| NAKED |  | - | OXIDANT |  | $\bullet$ |
| NATURAL | $\bullet$ | - | OXIDATION |  | $\bullet$ |
| NATURE | $\bullet$ |  | OXIDE |  | $\bullet$ |
| NDIR | $\bullet$ |  | OXIDIZE |  | $\bullet$ |
| NECESSARY | $\bullet$ | $\bullet$ | OXIDIZED |  | $\bullet$ |
| NECK |  | $\bullet$ | OXIDIZER |  | $\bullet$ |
| NECKLACE |  | $\bullet$ | PANELS |  | $\bullet$ |
| NECKLACES |  | $\bullet$ | PAPER | $\bullet$ |  |
| NEEDED | $\bullet$ |  | PARTICIPANTS |  | $\bullet$ |
| NERVE | $\bullet$ |  | PARTICIPATED | $\bullet$ |  |
| NERVOUS | $\bullet$ |  | PARTICLES |  | $\bullet$ |
| NEUROREGULATING | $\bullet$ |  | PARTICULATE | $\bullet$ |  |
| NEW |  | - | PAST | $\bullet$ |  |
| NEWBORN | $\bullet$ |  | PATTERN | $\bullet$ |  |
| NI |  | $\bullet$ | PBO |  | $\bullet$ |
| NICKEL |  | - | PEOPLE |  | $\bullet$ |
| NIOSH | $\bullet$ |  | PEPTIDE | $\bullet$ |  |
| NITRIC |  | $\bullet$ | PEPTIDES | $\bullet$ |  |
| NITROGEN |  | $\bullet$ | PERCENT |  | $\bullet$ |
| NOBEL |  | $\bullet$ | PERCENTAGE |  | $\bullet$ |
| NOBLE |  | $\bullet$ | PERFORM | $\bullet$ |  |
| NONCANCER | $\bullet$ |  | PERFORMED | $\bullet$ |  |
| NON-FLAMMABILITY |  | $\bullet$ | PERIOD | $\bullet$ |  |
| NON-FLAMMABLE |  | $\bullet$ | PERMITTED | $\bullet$ |  |
| NOTTINGHAM |  | $\bullet$ | PERSON |  | $\bullet$ |
| NPD | $\bullet$ |  | PET |  | $\bullet$ |
| NUCLEAR |  | - | PHA |  | $\bullet$ |
| NUMBER | $\bullet$ |  | PHASE | $\bullet$ |  |
| NUMBERS | $\bullet$ |  | PHOSPHATE |  | $\bullet$ |
| NUTRACEUTICALS |  | - | PHOSPHATES |  | $\bullet$ |
| NUTRITIVE | $\bullet$ |  | PHOTOGRAPHS |  | $\bullet$ |
| NYLON |  | $\bullet$ | PHOTONS |  | $\bullet$ |
| OBJECT |  | $\bullet$ | PHYSICAL | $\bullet$ |  |


| PHYSICOCHEMICAL | - |  | PRODUCED | - | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PICOSCALE |  | $\bullet$ | PRODUCING |  | $\bullet$ |
| PICOSECOND |  | - | PRODUCTION | $\bullet$ | $\bullet$ |
| PICTURES |  | $\bullet$ | PRODUCTS | $\bullet$ | $\bullet$ |
| PIECE |  | $\bullet$ | PROFILE | $\bullet$ | $\bullet$ |
| PLANE |  | $\bullet$ | PROGRESSION | $\bullet$ |  |
| PLANT |  | $\bullet$ | PROPAGATES |  | $\bullet$ |
| PLANTED |  | $\bullet$ | PROPERTIES |  | $\bullet$ |
| PLANTING |  | $\bullet$ | PROPOSED | $\bullet$ |  |
| PLASMA | - | $\bullet$ | PROTECTORS |  | $\bullet$ |
| PLATELET | $\bullet$ |  | PROTECTS |  | $\bullet$ |
| PLATFORMS | $\bullet$ |  | PROTEIN | $\bullet$ |  |
| POLLUTANT |  | - | PROTEINS | $\bullet$ | $\bullet$ |
| POLLUTANTS | $\bullet$ |  | PROTEOLYTIC | $\bullet$ |  |
| POLLUTED | $\bullet$ |  | PROTEOME | $\bullet$ |  |
| POLLUTING |  | $\bullet$ | PROVIDE | $\bullet$ |  |
| POLLUTION | $\bullet$ | $\bullet$ | PROVIDE |  | $\bullet$ |
| POLYMER |  | $\bullet$ | PUBLISHED | - |  |
| POLYMERISATION |  | $\bullet$ | PULSE |  | $\bullet$ |
| POLYMERS |  | $\bullet$ | PULSING |  | $\bullet$ |
| POLYPEPTIDES | - |  | PURE |  | $\bullet$ |
| POLYUNSATURATED |  | $\bullet$ | PURIFY |  | $\bullet$ |
| POPULAR |  | $\bullet$ | PUT |  | $\bullet$ |
| POPULARITY |  | $\bullet$ | QUANTITATIVE | $\bullet$ |  |
| POSSIBLE |  | $\bullet$ | QUANTITIES | $\bullet$ |  |
| POTENTIAL |  | $\bullet$ | QUESTION |  | $\bullet$ |
| POTENTIALS |  | $\bullet$ | QUESTIONS |  | $\bullet$ |
| POWERFUL |  | $\bullet$ | QUICKLY |  | $\bullet$ |
| PRACTICALLY |  | $\bullet$ | RACE |  | $\bullet$ |
| PRAY |  | $\bullet$ | RACERS |  | $\bullet$ |
| PRAYER |  | $\bullet$ | RACING |  | $\bullet$ |
| PRAYING |  | $\bullet$ | RADICAL |  | $\bullet$ |
| PRECISION | $\bullet$ |  | RADICALS |  | $\bullet$ |
| PREDICT |  | $\bullet$ | RADII |  | $\bullet$ |
| PREDICTION |  | $\bullet$ | RADIUS |  | $\bullet$ |
| PREMATURE | $\bullet$ |  | RAISE |  | $\bullet$ |
| PRENATAL | $\bullet$ |  | RATIO | $\bullet$ |  |
| PREPARATION | $\bullet$ |  | RATIOS | $\bullet$ |  |
| PREPARE |  | $\bullet$ | RE-ABSORPTION |  | $\bullet$ |
| PREPARING |  | $\bullet$ | REACH |  | $\bullet$ |
| PRESSURE | $\bullet$ | $\bullet$ | REACHED |  | $\bullet$ |
| PRE-TERM | - |  | REACT |  | $\bullet$ |
| PREVENT |  | $\bullet$ | REACTANT |  | $\bullet$ |
| PRIZE |  | $\bullet$ | REACTANTS |  | $\bullet$ |
| PROACTIVE |  | $\bullet$ | REACTION |  | $\bullet$ |
| PROBABILITY |  | $\bullet$ | REACTIONS |  | $\bullet$ |
| PROBABLE |  | $\bullet$ | REACTIVE |  | $\bullet$ |
| PROBLEM |  | $\bullet$ | REACTOR |  | $\bullet$ |
| PROCEDURES | $\bullet$ |  | REAL |  | $\bullet$ |
| PROCESS |  | $\bullet$ | RECEIVED |  | $\bullet$ |
| PROCESSED | $\bullet$ |  | RECENTLY | $\bullet$ |  |
| PROCESSING | $\bullet$ |  | RECIPIENT |  | $\bullet$ |
| PROCESSORS | $\bullet$ |  | RECLAIMED |  | $\bullet$ |
| PRODUCE | - | $\bullet$ | RECLAIMING |  | $\bullet$ |


| REDUCE |  | $\bullet$ | SAVOURY |  | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| REDUCING |  | $\bullet$ | SAW |  | $\bullet$ |
| REDUCTANT | $\bullet$ |  | SAYS |  | $\bullet$ |
| REDUCTION | $\bullet$ |  | SCALE |  | $\bullet$ |
| REFERENCES |  | $\bullet$ | SCF |  | $\bullet$ |
| REFLECTION |  | $\bullet$ | SCFS |  | $\bullet$ |
| REFLECTIONS |  | $\bullet$ | SCIENTISTS |  | $\bullet$ |
| REFLECTS |  | $\bullet$ | SEAT |  | $\bullet$ |
| REGIA |  | $\bullet$ | SEATS |  | $\bullet$ |
| REGION | $\bullet$ |  | SECOND |  | $\bullet$ |
| REGIONS | $\bullet$ |  | SEE |  | $\bullet$ |
| REGULATION | $\bullet$ |  | SEEN |  | - |
| RELATED | $\bullet$ | $\bullet$ | SELECTED | $\bullet$ |  |
| RELEASED | $\bullet$ |  | SELENIUM | $\bullet$ |  |
| RELEASING |  | $\bullet$ | SELF- EXTINGUISHES |  | $\bullet$ |
| REMEDIATED |  | $\bullet$ | SELF-EXTINGUISHING |  | $\bullet$ |
| REMEDIATION |  | $\bullet$ | SEM |  | $\bullet$ |
| REMOVED |  | $\bullet$ | SENSITIVITY | - | $\bullet$ |
| REPLACEMENTS |  | $\bullet$ | SENSITIZATION |  | $\bullet$ |
| REPORT | $\bullet$ |  | SENSITIZED |  | $\bullet$ |
| REPORTED | $\bullet$ |  | SEPARATION | $\bullet$ |  |
| REPORTS | - |  | SERIES |  | $\bullet$ |
| REQIRED |  | $\bullet$ | SEROTONIN | $\bullet$ |  |
| REQUIREMENT | $\bullet$ |  | SERUM | $\bullet$ |  |
| RESEARCH | $\bullet$ | $\bullet$ | SETTINGS | $\bullet$ |  |
| RESEARCHERS |  | $\bullet$ | SEVEN | $\bullet$ |  |
| RESERVOIR |  |  | $\mathrm{SFCO}_{2}$ |  | $\bullet$ |
| RESIDUAL |  | $\bullet$ | SHINE |  | $\bullet$ |
| RESIDUE |  | $\bullet$ | SHORT |  | $\bullet$ |
| RESISTANT |  | $\bullet$ | SHORTER |  | $\bullet$ |
| RESTRICT |  | $\bullet$ | SHOW |  | $\bullet$ |
| RESULTS | $\bullet$ | $\bullet$ | SHOWED |  | $\bullet$ |
| RETENTION | $\bullet$ |  | SHOWING |  | $\bullet$ |
| REVEAL |  | $\bullet$ | SHOWN |  | $\bullet$ |
| REVEALED |  | $\bullet$ | SIGNIFICANT |  | $\bullet$ |
| REVEALING |  | $\bullet$ | SIGNIFICANTLY | $\bullet$ |  |
| REVERSED | $\bullet$ |  | SILVER |  | $\bullet$ |
| RFC | $\bullet$ |  | SILVERY |  | $\bullet$ |
| RICH |  | $\bullet$ | SIMPLEST |  | $\bullet$ |
| RISE | - |  | SIMULTANEOUSLY |  | $\bullet$ |
| RISK |  | $\bullet$ | SINGLE | $\bullet$ |  |
| RP-HPLC | $\bullet$ |  | SINKS | $\bullet$ |  |
| RSD | $\bullet$ |  | SINNERS |  | $\bullet$ |
| RUN |  | $\bullet$ | SITES |  | $\bullet$ |
| RUNNING |  | $\bullet$ | SLOWLY |  | $\bullet$ |
| SAID |  | $\bullet$ | SMALL | $\bullet$ | $\bullet$ |
| SAINTS |  | $\bullet$ | SMOKE | $\bullet$ |  |
| SALT |  | $\bullet$ | SMOKES | $\bullet$ |  |
| SALTS |  | $\bullet$ | SNAPSHOTS |  | $\bullet$ |
| SAME | $\bullet$ | $\bullet$ | SOLD |  | $\bullet$ |
| SAMPLE | $\bullet$ | $\bullet$ | SOLUBLE |  | $\bullet$ |
| SAMPLES | $\bullet$ | $\bullet$ | SOLUTE | $\bullet$ |  |
| SAMPLING | $\bullet$ |  | SOLUTION |  | $\bullet$ |
| SATURATED |  | $\bullet$ | SOLUTIONS |  | $\bullet$ |


| SOLVENT |  | $\bullet$ | SUPERCRITICAL |  | $\bullet$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| SOLVENTS |  | $\bullet$ | SUPERMARKET |  | $\bullet$ |
| SORBENT | $\bullet$ |  | SUPPLIED | $\bullet$ |  |
| SORBENTS | $\bullet$ |  | SUPPLY | $\bullet$ |  |
| SOURCES | $\bullet$ |  | SURFACE | $\bullet$ |  |
| SOUTHERN |  | $\bullet$ | SURPRISING |  | $\bullet$ |
| SOY | $\bullet$ |  | SURVEY |  | $\bullet$ |
| SPE | $\bullet$ |  | SURVIVABLE |  | $\bullet$ |
| SPECIAL | $\bullet$ |  | SURVIVAL |  | $\bullet$ |
| SPECIATION | $\bullet$ |  | SYNTHETIC | $\bullet$ |  |
| SPECIES |  | $\bullet$ | SYSTEM | $\bullet$ | $\bullet$ |
| SPECIFIC | $\bullet$ |  | TAKE |  | $\bullet$ |
| SPECIFIES |  | $\bullet$ | TAKEN |  | $\bullet$ |
| SPECTROMETRY | $\bullet$ |  | TAKES |  | $\bullet$ |
| SPECTRUM |  | $\bullet$ | TALE |  | $\bullet$ |
| SPEED |  | $\bullet$ | TASTE |  | $\bullet$ |
| SPEED UP |  | $\bullet$ | TEAM |  | $\bullet$ |
| SPREAD |  | $\bullet$ | TEAMS |  | $\bullet$ |
| SPREADS |  | $\bullet$ | TECH |  | $\bullet$ |
| STABILITY | - |  | TECHNICAL | $\bullet$ |  |
| STAGE | $\bullet$ |  | TECHNIQUE | $\bullet$ | $\bullet$ |
| STANOL |  | $\bullet$ | TECHNIQUES | $\bullet$ |  |
| STANOLS |  | $\bullet$ | TECHNOLOGIES |  | $\bullet$ |
| STARCH | $\bullet$ |  | TEETOTALERS |  | $\bullet$ |
| STARCHY | $\bullet$ |  | TELL |  | $\bullet$ |
| STATE |  | $\bullet$ | TEMPERATURE | $\bullet$ | $\bullet$ |
| STATES |  | $\bullet$ | TEMPERATURES | $\bullet$ | $\bullet$ |
| STEP |  | $\bullet$ | TENDENCY | $\bullet$ |  |
| STEROL |  | $\bullet$ | TERMINAL |  | $\bullet$ |
| STEROLS |  | $\bullet$ | TEST |  | $\bullet$ |
| STONE |  | $\bullet$ | TESTED |  | $\bullet$ |
| STORES | $\bullet$ |  | TESTING |  | $\bullet$ |
| STORY |  | $\bullet$ | TESTS |  | $\bullet$ |
| STRANGE |  | $\bullet$ | TEXTBOOKS |  | $\bullet$ |
| STRANGELY |  | $\bullet$ | THERMAL | $\bullet$ |  |
| STREET |  | $\bullet$ | THREAD |  | $\bullet$ |
| STRIP AWAY |  | $\bullet$ | THREADS |  | $\bullet$ |
| STRUCTURAL | $\bullet$ | $\bullet$ | THROMBOCYTES | $\bullet$ |  |
| STRUCTURE | $\bullet$ | $\bullet$ | TIME |  | $\bullet$ |
| STRUCTURES |  | $\bullet$ | TIMES | $\bullet$ |  |
| STUDENTS |  | $\bullet$ | TIMESCALE |  | $\bullet$ |
| STUDIED |  | $\bullet$ | TIMESCALES |  | $\bullet$ |
| STUDIES | $\bullet$ | $\bullet$ | TISSUE | $\bullet$ |  |
| STUDY | $\bullet$ |  | TISSUES | $\bullet$ |  |
| STUDYING | $\bullet$ |  | TOTAL |  | $\bullet$ |
| SUBLEVELS |  | $\bullet$ | TOUCHSTONE |  | $\bullet$ |
| SUBSTANCE | $\bullet$ | $\bullet$ | TOWN |  | $\bullet$ |
| SUBSTITUTE |  | $\bullet$ | TOWNS |  | $\bullet$ |
| SUBSTITUTES |  | $\bullet$ | TOWSFOLK |  | $\bullet$ |
| SUCCESSFULLY | $\bullet$ |  | TOXIC |  | $\bullet$ |
| SUFFICIENT |  | $\bullet$ | TRACE | $\bullet$ |  |
| SUGAR | $\bullet$ |  | TRANSFER | $\bullet$ |  |
| SUGARS | $\bullet$ |  | TRANSFERRED | $\bullet$ |  |
| SULFUR | $\bullet$ |  | TRANSFORMED |  | $\bullet$ |


| TRANSIENT |  | $\bullet$ | VARY | $\bullet$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| TRANSITION |  | $\bullet$ | VEGETABLE |  | $\bullet$ |
| TRANSITIONS |  | $\bullet$ | VERTICAL | $\bullet$ |  |
| TRANSPORT |  | $\bullet$ | VIA | $\bullet$ |  |
| TRAP |  | $\bullet$ | VIBRATE |  | $\bullet$ |
| TRASITIONARY |  | $\bullet$ | VIBRATIONS |  | $\bullet$ |
| TREATING |  | $\bullet$ | VOC |  | $\bullet$ |
| TREATMENT |  | $\bullet$ | VOCS |  | $\bullet$ |
| TREND | $\bullet$ |  | VOLUMES | $\bullet$ |  |
| TRIGGER |  | $\bullet$ | WALLS |  | $\bullet$ |
| TRIGGERED |  | $\bullet$ | WASTE | $\bullet$ | $\bullet$ |
| TUMOR | $\bullet$ |  | WASTES | $\bullet$ |  |
| TUMORS | $\bullet$ |  | WATCH |  | $\bullet$ |
| TURNING |  | $\bullet$ | WATCHING |  | $\bullet$ |
| TYPE |  | $\bullet$ | WATER | $\bullet$ | $\bullet$ |
| TYPES |  | $\bullet$ | WATERS | $\bullet$ |  |
| UK |  | $\bullet$ | WAY |  | $\bullet$ |
| UNDERTAKEN |  | $\bullet$ | WEAK |  | $\bullet$ |
| UNIMOLECULAR |  | $\bullet$ | WEAKER |  | $\bullet$ |
| UNIVERSITY |  | $\bullet$ | WEARER |  | $\bullet$ |
| UNKNOWN | $\bullet$ |  | WEIGHT | $\bullet$ | $\bullet$ |
| UNLIKELY |  | $\bullet$ | WELL | $\bullet$ |  |
| UNPOLLUTED | $\bullet$ |  | WHEY | $\bullet$ |  |
| URINE |  | $\bullet$ | WHITE |  | $\bullet$ |
| USE | $\bullet$ | $\bullet$ | WHOLE | $\bullet$ |  |
| USED | $\bullet$ | $\bullet$ | WIN |  | $\bullet$ |
| USEFUL | $\bullet$ | $\bullet$ | WORK | $\bullet$ | $\bullet$ |
| USES | $\bullet$ |  | WORKERS | $\bullet$ |  |
| USING | $\bullet$ | $\bullet$ | WORKING |  |  |
| UTILIZATION | $\bullet$ |  | WORKPLACE | $\bullet$ |  |
| VACUUM |  | $\bullet$ | WORKPLACES | $\bullet$ |  |
| VALUE | $\bullet$ |  | YEARS | $\bullet$ |  |
| VALUES | $\bullet$ |  | YEAST | $\bullet$ |  |
| VAPOR | $\bullet$ |  | YELLOW |  | $\bullet$ |
| VARIABILITY | $\bullet$ |  | YIELD | $\bullet$ |  |
| VARIATION | $\bullet$ |  | ZINC | $\bullet$ | $\bullet$ |
| VARIATIONS | $\bullet$ |  | ZN |  | $\bullet$ |
| VARIES | $\bullet$ |  |  |  |  |


[^0]:    1 La oración 7 establece un una conexión mediante un enlace dudoso. Puede eliminarse y el texto presenta la misma coherencia que el texto original.

[^1]:    ${ }^{2}$ La oración 3, que establece una conexión mediante un enlace dudoso, puede eliminarse, ya que su información está contenida en la 4, mediante la utilización de this. Presenta la misma coherencia que el texto original.

[^2]:    3 La oración 13 establece una conexión mediante un enlace dodoso. Como puede observarse puede eliminarse sin que la coherencia del texto original se vea afectada.

[^3]:    4 La oración 10 establece una conexión mediante un enlace dudoso. Puede eliminarse sin afectar a la coherencia del texto original.

[^4]:    5 La oración 10, que establece conexiones mediante enlaces dudosos, se puede eliminar sin que la coherencia del texto original se vea afectada.

